

# **Aerosol composition and sources during high and low pollution periods in Ningbo, China**

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1 **Abstract**

2 Due to the rapid industrialization of the Yangtze River Delta (YRD) region in China, heavy  
3 air pollution episodes have occurred frequently over the past five years which are of great  
4 concern due to their environmental and health impacts. To investigate the chemical  
5 characteristics of the highly polluted aerosols in this region, a sampling campaign had been  
6 conducted in Ningbo from 3 December 2012 to 27 June 2013, during which a month long  
7 high pollution episode had been captured. Daily average PM<sub>2.5</sub> concentrations during high  
8 and low pollution periods were 111 µg m<sup>-3</sup> and 53 µg m<sup>-3</sup>, respectively. The most polluted  
9 day was 8 January 2013 with a PM<sub>2.5</sub> concentration up to 175 µg m<sup>-3</sup>. To understand the  
10 origin of the highly polluted aerosols, meteorological conditions, air mass backward  
11 trajectories, distribution of fire spots in surrounding areas and various categories of aerosol  
12 pollutants were analysed, including trace metals, inorganic species, PAHs and anhydrosugars.  
13 Total metal concentrations were 3.8 and 1.6 µg m<sup>-3</sup> for the high and low pollution episodes,  
14 respectively, accounting for 3.4% and 3.1% of the total PM<sub>2.5</sub> mass. Total concentrations of  
15 ionic species accounted for more than 50.0% of the PM<sub>2.5</sub> by mass, with dominant ions  
16 (nitrate, sulphate, ammonium) accounting for over 42.0% of the PM<sub>2.5</sub> mass concentrations in  
17 both periods. During the high pollution episode, enhanced Cd – Pb and biomarker  
18 (levoglucosan, mannosan) levels indicated the contributions from coal combustion, traffic  
19 and biomass burning to fine aerosol PM<sub>2.5</sub>. The average diagnostic ratio of Fla/(Fla+Pyr) was  
20 0.54 in high pollution episode, which was intermediate between that for wood (>0.50) and  
21 coal combustion (0.58). BaP/Bpe was 0.49 and 0.30 for the highly and lightly polluted  
22 aerosols respectively, associated with the significant non-traffic emissions (<0.60). In  
23 addition, stagnant weather conditions during the high pollution period and long-range  
24 transport of air masses from heavy industries and biomass burning from northern China to  
25 Ningbo could be considered as the main factors for the formation of the aerosols during high  
26 pollution period.

27 **Keywords:** aerosol, PM<sub>2.5</sub>, PAHs, levoglucosan, mannosan, Ningbo

## 28 **1. Introduction**

29 Atmospheric aerosols can greatly affect the Earth's radiation budget and climate change  
30 along with greenhouse gases (Gkikas et al., 2016; Xin et al., 2016), and they have been  
31 verified to have significant impacts not only on visibility/haze, but also on air quality and  
32 public health, particularly those fine particles with aerodynamic diameters less than or equal  
33 to 2.5  $\mu\text{m}$  (PM<sub>2.5</sub>) (Wang et al., 2014). Highly polluted aerosols can potentially lead to lung  
34 cancer, respiratory diseases and cardiopulmonary mortality for long term exposure (Pope Iii  
35 et al., 2002; Tie et al., 2009). Generally, atmospheric aerosols can be divided into primary  
36 ones directly emitted from various sources and secondary ones formed through gas-to-  
37 particle transformation processes. In recent decades, many regions have encountered heavy  
38 aerosol pollution, including Indonesia (Field et al., 2004; Forsyth, 2014; Langmann, 2007),  
39 United States (Odman et al., 2009; Park et al., 2006; Schichtel et al., 2001), Northern Europe  
40 (Toledano et al., 2012), and China (Tao et al., 2014; Wang et al., 2015c; Zhang et al., 2015a).

41 In China, a number of cities have experienced severe aerosol pollution with an Air Pollution  
42 Index (API) higher than 500 that is categorized as the unhealthiest level by China's Ministry  
43 of Environmental Protection (MEP). Less than 1% of the top 500 cities in China can meet the  
44 World Health Organization air quality standards (Li and Zhang, 2014). Power plants, heavy  
45 industry and vehicles were reported to be mainly responsible for the occurrence of severe  
46 aerosol pollution episode especially in winter when domestic coal consumption increased  
47 significantly (Li and Zhang, 2014). From satellite observations, northern and eastern China  
48 was reported to be affected by hazardous dense aerosol pollution the most frequently (Tie et  
49 al., 2006). The Yangtze River Delta (YRD), located at the eastern coast of China bordering  
50 the East China Sea, has experienced many aerosol pollution events due to its remarkable  
51 economic growth and accelerated urbanization over the past 30 years (Liao et al., 2014),  
52 especially the rapid development of heavy industries, such as iron and steel, automobile  
53 manufacturing, oil and gas (Cheng et al., 2014). Additionally, another significant contributor  
54 of severe aerosol pollution in this area could be the open burning of biomass including  
55 agricultural waste which tends to be a common practice for land clearance by local farmers  
56 (Cheng et al., 2014). Previous studies have revealed high levels of aerosol pollution and  
57 extremely low visibilities in the YRD (Fu et al., 2008; Gao et al., 2011). Meteorological  
58 stations observed the average visibility of this region has shown a trend of 2.4 km decrease  
59 per decade from 25 to < 20 km in the period of 1981 ~ 2005 (Gao et al., 2011). A few studies  
60 have been conducted to investigate the aerosol pollution episodes occurring in megacities of

61 the YRD including Nanjing, Shanghai and Hangzhou (Cheng et al., 2013b; Wang et al.,  
62 2014), but non of these studies have reported detailed information on particulate  
63 compositions.

64 Located at the south of YRD region, Ningbo is adjacent to Hangzhou and Shaoxing, about  
65 15km to the west coast of the East China Sea. As the second largest city of Zhejiang Province,  
66 it has a population of approximately 8 million and covers an area of around 10,000 km<sup>2</sup>.  
67 Before 2000, the number of days affected by severe aerosol pollution in Ningbo was reported  
68 to be less than 15 per year. The visibility of Ningbo ranged from 8.6 to 14.9 km in 1980.  
69 However, the number of heavy aerosol pollution days was rapidly increased to 50 per year  
70 after 2001 and the visibility observed in 2003 ranged from 3.8 to 11.7 km, which was an  
71 obvious decrease compared to that of 1980 (Cheng et al., 2013b). In January 2013, a long  
72 lasting aerosol pollution episode occurred in central and eastern China and it was considered  
73 as the most severe aerosol pollution since 2000. However, only a few studies have reported  
74 this particular event (Andersson et al., 2015; Cheng et al., 2013b; Ji et al., 2014; Wang et al.,  
75 2014). The previous studies mostly focused on the study of aerosol number concentrations,  
76 visibility, OC and EC in YRD. Only one report discussed the source apportionment of  
77 combustion-derived black carbon aerosols by using carbon isotopes (Andersson et al., 2015).  
78 Their results preliminarily show that biomass combustion contributed around 30% to the  
79 severe aerosol pollution in North China Plain (NCP, Beijing) and Yangtze River Delta (YRD,  
80 Shanghai). For black carbon, it was found that the petroleum usage and coal combustion  
81 could account for 46% and 66% of BC in YRD and NCP, respectively. In this study, the  
82 chemical characteristics including trace metals, ionic species, polycyclic aromatic  
83 hydrocarbons (PAHs) and biomarkers for high and low aerosol pollution periods in Ningbo  
84 have been investigated, and also diagnostic ratios, organic tracers and air mass backward  
85 trajectories have been adopted for a qualitative source analysis for this particular aerosol  
86 pollution event.

## 87 **2. Experimental**

### 88 **2.1 Sampling site and aerosol collection**

89 The sampling site (29.80N, 121.56E) is located at the southern city of YRD- Ningbo, shown  
90 in Fig. 1(a) and (b). It is less than 10km away from the central business district (CBD). A 24-  
91 hour sampling was conducted at the air monitoring station on the rooftop of Science and  
92 Engineering Building (SEB) in the University of Nottingham Ningbo Campus (UNNC) from

93 December 3<sup>rd</sup> 2012 to June 27<sup>th</sup> 2013, using a high volume sampler (Model: TH-1000H,  
94 Tianhong Instrument CO., Ltd. Wuhan, China) with the flow rate of 1.05 m<sup>3</sup> min<sup>-1</sup>.  
95 20cm×25cm glass fibre filter (Huitong Instrument CO., Tianjin, China) was loaded to capture  
96 PM<sub>2.5</sub>. In total, 32 PM<sub>2.5</sub> samples were collected and blank filters were obtained every two  
97 weeks from the sampler without their pump on.

## 98 **2.2 Air mass backward trajectory and fire-spot analysis**

99 To investigate the effects of medium and long-range transport of aerosols on local air quality,  
100 air mass transport pathways were studied through backward trajectory analysis which was  
101 carried out using the Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPPLIT) 4.9  
102 model (Draxler, 2013; Rolph, 2013). A 4-day (96h) backward trajectory was started at the  
103 sampling site every 6 hours during each sampling day at 500 m above ground level (agl) and  
104 then these computed trajectories were clustered by applying TrajStat 1.2.1.0  
105 (<http://www.arl.noaa.gov/HYSPPLIT.php>) (Wang et al., 2009). TrajStat is a geographic  
106 information system (GIS) based software which can identify aerosol potential sources from  
107 long-term measurement data by using various trajectory statistical analysis methods. In this  
108 software, there are two clustering models- Euclidean distance and angle distance (Turpin and  
109 Huntzicker, 1995). The Euclidean distance model has been applied in this study because it  
110 concerns both the directions and distances of the trajectories while the angle distance only  
111 concerns the directions of the trajectories. In addition, Moderate Resolution Imaging  
112 Spectroradiometer (MODIS) fire-spots in this study were obtained from Fire Information for  
113 Resource Management System (FIRMS) Web Fire Mapper. Each fire-spot that was detected  
114 by satellites represents the centre of an approximately 1km pixel marked as containing one or  
115 more fires, or other thermal activities.

## 116 **2.3 Chemical analysis of aerosol samples**

### 117 **2.3.1 Quality assurance and control**

118 Before sampling, fresh blank filters were prebaked for 4 hours at 550°C in a muffle furnace  
119 in order to remove any absorbed carbonaceous compounds. Equilibration of filters was  
120 carried out at constant temperature of 22°C ± 1°C and relative humidity (RH) at 30% ± 5%  
121 for 24 hours before and after sampling prior to gravimetric measurement by an electronic  
122 balance (Model: AL 104, Mettler Toledo, precision 0.1mg) and stored in refrigerator at -20°C  
123 to avoid any possible volatilization before analysis. Four portions were cut from each filter  
124 for the analyses of trace metals, PAHs, ionic species and anhydrosugar compounds. All

125 extracts were filtrated with 0.45  $\mu\text{m}$  pore size membranes. Calibrations were carried out by  
126 using external standard obtained from Sigma-Aldrich (St. Louis, MO, USA) and all analytical  
127 results were corrected by subtracting the values obtained from blank filters.

### 128 **2.3.2 Trace metals**

129 Water-extracted and acid-digested trace metals were analysed to investigate the water-soluble  
130 and total metal concentrations, respectively. Water-soluble metals were extracted  
131 ultrasonically with 15mL Milli-Q water for 2.5 hours at room temperature. Total metals were  
132 extracted by 11mL concentrated nitric acid (65%, Sinopharm Chemical Reagent Co.,Ltd.)  
133 and hydrochloric acid (37%, Sinopharm Chemical Reagent Co.,Ltd.) mixture (volume ratio  
134 3:1) in a microwave digester (MARS 5, CEM CO., U.S.). The digestion temperature program  
135 was set as below: ramp to 185°C in 15 minutes and held for 25 minutes. After cooling, all  
136 extracts were filtered and further diluted with Milli-Q water to 100mL and stored at 4°C until  
137 analysis. Water extracts were acidified by HNO<sub>3</sub> to 2% prior to analysis. In total, 13 trace  
138 metals, including Mn, Zn, Co, Cd, Cu, Al, Cr, Ni, Pb, V, Fe, Ti and As, were determined by  
139 Inductively Coupled Plasma – Mass Spectrometry (ICP-MS, NexION™ 300X).

### 140 **2.3.3 Polycyclic aromatic hydrocarbons (PAHs)**

141 PAHs were extracted from filters with 20 mL hexane and acetone mixture (volume ratio 3:1)  
142 by a microwave digester, the conditions of which were set at: temperature (50°C), microwave  
143 energy (150W) and held for 20 minutes. Extracts were then filtrated and evaporated to 1mL  
144 and analysed by gas chromatograph (Agilent 7890A) - mass selective detector (Agilent  
145 5975C) (GC-MSD). GC was equipped with a capillary column (HP-5MS, 0.25 $\mu\text{m}$  film  
146 thickness, 30m $\times$ 0.25mm i.d., Agilent J&W.), using helium as carrier gas. In this study, 17  
147 PAHs were investigated, including retene and 16 EPA priority PAHs: Naphthalene (Nap),  
148 acenaphthene (Ace), acenaphthylene (Acy), fluorene (Flu), phenanthrene (Phe), anthracene  
149 (Ant), fluoranthene (Fla), pyrene (Pyr), benzo[a]anthracene (BaA), chrysene (Chr),  
150 benzo[b]fluoranthene (BbF), benzo[k]fluoranthene (BkF), benzo[a]pyrene (BaP),  
151 indeno[1,2,3-cd]pyrene (InP), dibenzo[a,h]anthracene (DBA) and benzo-[ghi]perylene (Bpe).  
152 During analysis, 1 $\mu\text{L}$  sample was injected in splitless mode with 5-minute solvent delay and  
153 the temperature program was set as follows (Karthikeyan et al., 2006): held at 50°C for 2  
154 minutes, ramp to 200°C at a rate of 10°C min<sup>-1</sup> and held for 8 minutes, then ramp to 300°C at  
155 a rate of 5°C min<sup>-1</sup> and held for 8 minutes. PAHs were identified based on their specific m/z  
156 at different retention time and quantified by calibration with authentic standards.

#### 157 **2.3.4 Ionic species**

158 Filters were extracted ultrasonically by 20 mL of Milli-Q water for 45 minutes and then the  
159 aqueous extracts were filtrated and stored at 4°C before analysis. In total, 11 inorganic ions,  
160 namely Fluoride (F<sup>-</sup>), Chloride (Cl<sup>-</sup>), Bromide (Br<sup>-</sup>), Nitrate (NO<sub>3</sub><sup>-</sup>) and Sulfate (SO<sub>4</sub><sup>2-</sup>),  
161 Lithium (Li<sup>+</sup>), Sodium (Na<sup>+</sup>), Ammonium (NH<sub>4</sub><sup>+</sup>), Potassium (K<sup>+</sup>), Magnesium (Mg<sup>2+</sup>) and  
162 Calcium (Ca<sup>2+</sup>), were determined by Ion Chromatograph (ICS-1600). The system  
163 comprises an autosampler (Dionex AS-DV), an analytical column (Dionex, IonPac™ AS23  
164 for anions, IonPac™ CS12A for cations), a guard column (Dionex, IonPac™ AG23 for  
165 anions, IonPac™ CG12A for cations) and a self-regenerating suppressor (Dionex, ASRS™  
166 300 for anions, IonPac™ CS12A for cations). A constant eluent (4.5mM Na<sub>2</sub>CO<sub>3</sub> + 0.8mM  
167 NaHCO<sub>3</sub>, 1 mL min<sup>-1</sup>) with a suppressor current at 25mA were used for anion detection.  
168 Cations were investigated applying 1mL min<sup>-1</sup> 20mM Methanesulfonic acid (MSA) as eluent  
169 with a current at 59mA.

#### 170 **2.3.5 Anhydrosugar compounds**

171 Filters were extracted ultrasonically by 4 mL Milli-Q water for 30 minutes under room  
172 temperature. The extracts then filtered and analyzed by High Performance Liquid  
173 Chromatograph (Shimadzu 30A) - Electrospray Ionisation - tandem Mass Spectrometry  
174 (ABSciex 3200 Q trap) (HPLC-ESI-MS/MS) with an anion-exchange analytical column  
175 (Dionex, Carbopac PA1, 250mm×4mm) and guard column (Dionex, Carbopac PA1,  
176 50mm×4mm). Similar detection conditions can be found in the work of Piot et al. (2012) with  
177 different mobile phase. Due to the crystallization effect of sodium hydroxide solution in ion  
178 source, instead of applying 0.5mM sodium hydroxide solution, the mobile phase used in this  
179 study was approximately 0.5mM ammonium hydroxide (NH<sub>3</sub>·H<sub>2</sub>O, HPLC level, Sigma-  
180 Aldrich) in isocratic mode with a flow of 0.5 mL min<sup>-1</sup>. Columns were flushed and  
181 equilibrated between two samples for 3 minutes, applying the same flow rate. Parameters of  
182 the acquisition method were optimized to achieve the best Collision Induced Dissociation  
183 efficiency with selective current of daughter ions which are m/z 101+113, and m/z 101+129  
184 for levoglucosan (1, 6-anhydro-β-D-glucopyranose, Sigma-Aldrich) and mannosan (1, 6-  
185 anhydro-β-D-mannopyranose, Sigma-Aldrich), respectively.

## 186 **3. Results and discussion**

### 187 **3.1 Mass concentration of PM<sub>2.5</sub>**

188 During the entire sampling period, 32 samples were collected. The temporal distribution of  
189 the daily-averaged PM<sub>2.5</sub> concentration is plotted in Fig.2. Based on the variation of PM<sub>2.5</sub>  
190 concentrations, the sampling period was classified as two types: high and low pollution  
191 periods. For the 18 samples collected during the winter sampling period (2012/12/03-  
192 2013/01/25), high ambient PM<sub>2.5</sub> concentrations were obtained. The average PM<sub>2.5</sub>  
193 concentration of this period is  $110.9 \pm 30.8 \mu\text{g m}^{-3}$ , much higher than the latest GB 3095-  
194 2012 Chinese “Ambient Air Quality Standard (AAQS)” Grade II standard ( $75.0 \mu\text{g m}^{-3}$  for  
195 24h-averaged PM<sub>2.5</sub>, applicable to residential, commercial, cultural, industrial and rural areas)  
196 (MEP, 2012). Therefore 2012/12/03-2013/01/25 was categorized as a high pollution period.  
197 For this period, the PM<sub>2.5</sub> concentration ranged from 60.8 to  $175.3 \mu\text{g m}^{-3}$ , and the most  
198 polluted day occurred on 8 January 2013 with the highest PM<sub>2.5</sub> concentration of  $175.3 \mu\text{g m}^{-3}$ .  
199 This is in very good agreement with the study conducted by Wang et al. (2014) who  
200 reported the mean PM<sub>2.5</sub> concentration in eight cities of YRD ranging from 110.8-175.6  $\mu\text{g m}^{-3}$   
201  $\text{m}^{-3}$  in January 2013. As shown in Table 6, the average PM<sub>2.5</sub> concentration of this study  
202 ( $110.9 \mu\text{g m}^{-3}$ ) is also comparable with that of Tianjin ( $> 94 \mu\text{g m}^{-3}$ ) (Han et al., 2014) in  
203 northern China, but it is much lower than the results obtained in Handan ( $160.1 \pm 77.9 \mu\text{g m}^{-3}$ )  
204 (Wei et al., 2014) and Beijing ( $258 \pm 100 \mu\text{g m}^{-3}$ ) (Ho et al., 2016) of northern China and  
205 Xi’an ( $233 \pm 52 \mu\text{g m}^{-3}$ ) (Ho et al., 2016) in central China. For the 14 samples that were  
206 collected in the remaining sampling period (2013/02/25-2013/06/27), the PM<sub>2.5</sub>  
207 concentrations varied between 34.9 and  $67.7 \mu\text{g m}^{-3}$  with an average of  $52.6 \pm 11.9 \mu\text{g m}^{-3}$ ,  
208 which was only half of that for the high pollution period and lower than the above mentioned  
209 threshold, hence it is defined as a low pollution period accordingly.

### 210 **3.2 Meteorological conditions**

211 Stagnant weather conditions are favorable for the accumulation of atmospheric contaminants  
212 while flowing air is beneficial for their dispersion. In this study, meteorological data were  
213 collected from the UNNC meteorological station (WatchDog 2900ET weather station,  
214 Spectrum® Technologies, Inc.) located at the same rooftop as the samplers. Original data  
215 (temperature, rainfall, wind speed, relative humidity) were recorded at 10-minute intervals,  
216 and were then converted to daily-averaged data and summarized in Table 1.



217 Ningbo experiences a marine monsoon subtropical climate, featuring distinct seasons with  
218 hot, humid summers and cool dry winters (Haas and Ban, 2014). This high pollution period  
219 occurred in the winter in slightly windy conditions (2.1km/h), but the ambient temperature  
220 (6.5°C), rainfall (0.015mm) and relative humidity (67.9%) are relatively low. Under such  
221 conditions, a lower planetary boundary layer and a stronger surface inversion could be  
222 expected (Seidel et al., 2010), which would induce a more stagnant atmosphere. The low  
223 pollution period possessed had a higher average temperature (18.8°C) and rainfall (0.040mm),  
224 which favoured the dispersion and scavenging of aerosols.

### 225 **3.3 Air mass backward trajectory and fire-spot analysis for both high and low pollution** 226 **periods**

227 As mentioned before, all the computed trajectories for this sampling campaign have been  
228 clustered by TrajStat model. According to the change or turning point in total spatial variance  
229 (TSV) during the Hysplit cluster analysis, 3 has been chosen as the number of clusters (Wang  
230 et al., 2015), which deems most suitable and indicative in this study by the software. After  
231 comparing the clustered results between “display means” and “display clusters”, it is decided  
232 to use the statistically mean trajectory to represent the pathways of each cluster for a more  
233 concise presentation by merging through the Euclidean calculation. In Figure 3, both  
234 trajectory clustering results and fire spot data have been included; different air mass  
235 backward trajectory clusters have been presented as lines in various colours with their  
236 relative percentage shown at the left bottom of the map. Due to numerous fire-spots emerged  
237 during the sampling period (2012/12/03-2013/06/27), representative fire-spots for both  
238 periods were selected.

239 Fig. 3 (a) shows that the air arriving in Ningbo during the high pollution period originated  
240 from northwest direction, 60.3% of the 96-h backward trajectories originated from northern  
241 China, and air masses transported through Hebei, Shandong and Jiangsu provinces to Ningbo.  
242 While the rest 39.7% trajectories came from Mongolia and crossed Inner Mongolia, Shanxi,  
243 Hebei, Shandong and Jiangsu provinces to the receptor site through long-range transport.  
244 During the high pollution period, trajectories were observed to pass through the north China  
245 plain with heavy industries and large farming area with plenty of fire-spots, indicating long-  
246 range contributions from industrial emission and biomass burning to the high pollution  
247 aerosols in Ningbo.

248 Different from the trajectories in the high pollution period, only 17.3% of the trajectories  
249 came from Mongolia, passing through Inner Mongolia and crossing the Bo-Hai Sea and  
250 Yellow Sea to reach Ningbo, as shown in Fig. 3 (b). 42.3% of these trajectories derived from  
251 northern China, and also travelled through the Bo-Hai Sea, Yellow Sea and East China Sea to  
252 Ningbo. While the rest 40.4% 96-h backward trajectories originated from Taiwan and crossed  
253 the strait of Taiwan to Ningbo with a short distance.

254 Even though the low pollution period seemed to have slightly denser fire spot distribution  
255 than the high pollution period, as shown in Fig. 3, trajectories reaching Ningbo during low  
256 pollution period did not pass the dense fire-spots area in the northern region but were more of  
257 oceanic origins, indicating less influence of long-range transport of air pollutants including  
258 biomass burning emission to the slightly polluted aerosols in Ningbo.

### 259 **3.4 Characteristics of both high and low pollution aerosols**

#### 260 **3.4.1 Trace metals**

##### 261 (1) Metal concentrations

262 A comparison of the metal concentrations between the high and low pollution periods is  
263 listed in Table 2 which includes water soluble ( $C_{ws}$ ) and total ( $C_T$ ) metal concentrations, their  
264 corresponding water solubility and enrichment factors. The sum of total metal concentrations  
265 during the high pollution episode ( $3.8 \mu\text{g m}^{-3}$ ) was approximately 2.4 times that for the low  
266 pollution period ( $1.6 \mu\text{g m}^{-3}$ ), accounting for 3.4% and 3.1% of total  $\text{PM}_{2.5}$  mass  
267 concentrations, respectively. These results in high pollution period are in good agreement  
268 with a study conducted in Beijing, which shows the sum of total 13 metal concentrations is  
269  $3.74 \mu\text{g m}^{-3}$ , accounting for 2.6% of its total  $\text{PM}_{2.5}$  mass concentration (Tan et al., 2016)  
270 (Table 6). The sum of water soluble metal concentrations during high and low pollution  
271 periods were only 0.8 and  $0.5 \mu\text{g m}^{-3}$ , accounting for 0.7% and 1.0% of total  $\text{PM}_{2.5}$  mass  
272 concentrations, respectively. The averaged total metal abundance in the aerosol samples were  
273 in following order:  $\text{Al} > \text{Fe} > \text{Zn} > \text{Mn} > \text{Pb} > \text{Cu} > \text{As} > \text{Ni} > \text{Cr} > \text{V} > \text{Cd} > \text{Ti} > \text{Co}$  (High  
274 Pollution) and  $\text{Al} > \text{Fe} > \text{Zn} > \text{Mn} > \text{Pb} > \text{Cu} > \text{Ni} > \text{As} > \text{Cr} > \text{V} > \text{Ti} > \text{Cd} > \text{Co}$  (Low Pollution).  
275 The highest Al concentrations were 1.1 and  $0.6 \mu\text{g m}^{-3}$  in the high and low pollution episodes,  
276 respectively. Compared with low pollution period, Mn, Co, Cu, As, V and Fe doubled in high  
277 pollution period. Zn and Pb were tripled, and Cd in high pollution episode even increased  
278 about 5 folds. The concentration of Cd in winter of Ningbo is  $47.0 \text{ ng m}^{-3}$ , which is  
279 comparable with that in winter of a southern Chinese city- Foshan ( $42.6 \text{ ng m}^{-3}$ ) (Tan et al.,

280 2014). Nevertheless, the differences of Ni and Ti between the high and low pollution periods  
281 were not as significant as those of the above mentioned others. Besides the stronger  
282 accumulation of locally emitted aerosols due to the more stagnant atmospheric condition,  
283 higher metals concentrations of PM<sub>2.5</sub> during high pollution episode may be contributed by  
284 the long-range transport of air pollutants from heavy industries in Northern China, especially  
285 when large amount of particulates were released from coal combustion for centralized and  
286 decentralized house-heating in urban areas and countryside, respectively (Li and Zhang,  
287 2014).

## 288 (2) Water solubility

289 In addition to the comparison of metal concentrations between the high and low pollution  
290 samples, their water solubility was also compared. Water solubility is defined as the ratio of  
291 the water-extracted metals to the acid-extracted metals concentrations in this study. Water  
292 solubility is essential to evaluate the health effects caused by metals, because only those  
293 water-soluble fractions of metals are more likely to cause health problems due to their  
294 bioavailability (Birmili et al., 2006). In general, the water solubility varied for different  
295 metals, ranging from 8.4% (Cd) to 46.4% (Cu) in the highly polluted aerosols and 12.0% (Cd)  
296 to 60.3% (Cu) in the slightly polluted aerosols. Cu was found with highest solubility in both  
297 high and low pollution episodes, which could be explained that it existed dominantly as  
298 soluble salts such as sulfate in aerosol (Manousakas et al., 2014). Other metals including Ti,  
299 Fe, As, Cr and Pb also had moderately good water solubility (20%-60%), whereas, Zn, Co,  
300 Cd and Mn were observed to have lower solubility (<20%) in high pollution aerosols.  
301 Desboeufs et al (2005) summarized that the solubility of individual metals is related to the  
302 origins of the samples and they have found the metals in more alumino-silicated particles  
303 would be less water soluble, which may explain the lower water solubility in high pollution  
304 samples as mentioned above. It has been observed that more contributions were from long-  
305 range transported air masses to the local high pollution particles and during the transport  
306 process more aluminum containing dusts were incorporated into the aerosols in this study.  
307 Toxic metal like Pb, with high concentrations (high pollution period 291.5 ng m<sup>-3</sup>, low  
308 pollution period 92.4 ng m<sup>-3</sup>) and moderately good water solubility (high pollution period  
309 20.0%, low pollution period 29.6%) was expected to greatly influence human health.  
310 Generally, most metals were observed to have lower water solubility in the high pollution  
311 samples. Compared to the results from other studies conducted in East China (Hsu et al.,  
312 2010; Jiang et al., 2014), the water solubility of Cu, Pb, V and As were in good agreement;

313 however, the water solubility of Cd and Zn in highly polluted samples in this study were  
314 lower than the reported data, while Fe and Ti of both periods showed higher water solubility  
315 than those collected in Hong Kong (Jiang et al., 2014). In addition to the various extraction  
316 methods applied, the particular aerosol matrix affected by diverse yet distinct sources in  
317 different studied areas may result in the discrepancy of recovery efficiency of individual  
318 metals.

### 319 (3) Enrichment factor

320 The enrichment factors (EFs) can be applied to investigate whether the metals were  
321 originated from anthropogenic or natural sources. EF of each metal ( $EF_i$ ) in this study was  
322 calculated by dividing the relative abundance of each metal in a  $PM_{2.5}$  sample by its  
323 corresponding average abundance in the upper continental crust and it was normalized by a  
324 commonly used reference metal- Al, which was chosen due to its stability in chemical  
325 analysis (Birmili et al., 2006; Zhou et al., 2014). The calculation is carried out by the  
326 following equation (1) (Zhou et al., 2014):

$$327 \quad EF_i = \left(\frac{C_i}{C_{ref}}\right)_{atmosphere} / \left(\frac{C_i}{C_{ref}}\right)_{crust} \quad (1)$$

328 Where,  $(C_i/C_{ref})_{atmosphere}$  is the ratio of specified metal ( $C_i$ ) and reference metal ( $C_{ref}$ )  
329 concentrations in aerosol samples, while  $(C_i/C_{ref})_{crust}$  is the ratio of specified metal ( $C_i$ ) and  
330 reference metal ( $C_{ref}$ ) concentration in the upper continental crust. Metal concentrations in  
331 upper continental crust were 6.62% for Al, 2.94% for Fe, 0.38% for Ti and 583, 74.2, 12.7,  
332 0.097, 22.6, 61, 26.9, 26, 11.2 and 82.4  $mg\ kg^{-1}$  for Mn, Zn, Co, Cu, Cr, Ni, Pb, As and V,  
333 respectively, reported by China's National Environmental Monitoring Centre (CNEMC)  
334 (CNEMC, 1990).

335 In this study, most metals have EF values greater than 10, suggesting their primarily  
336 anthropogenic sources, such as vehicles and industrial emissions (Zhou et al., 2014).  
337 However, the EF value of Fe is only 1.7 and 1.2 in high and low pollution samples,  
338 respectively. This might be because Fe has the second highest crustal background  
339 concentration  $(C_{Fe})_{crust}$  which have led to relatively lower EF value, even though the  
340 concentration of Fe ranked the second in all metals. It is noteworthy that EF value of Ti is  
341 less than 1, which indicates Ti is depleted in the environment and crustal sources are  
342 dominant. Mn, Co, Cr and V were found moderately enriched ( $10 < EF < 100$ ), implying  
343 majority of them were emitted from human activities. Toxic As was found to have  
344 consistently high EF ( $>100$ ) values in both high and low pollution samples, as well as Zn, Cu,

345 Pb, Ni and Cd, suggesting they were all significantly related with anthropogenic sources. The  
346 mean EF value of them are ranked as: Cd > Pb > Zn > As > Cu > Ni, which showed very  
347 good agreement with the atmospheric metal study carried in Foshan that the top five EF  
348 values were in the same order and Cd even had the highest daily maximum EF value of  
349 18,357.0 (Tan et al., 2014). The EF value of Cd in this study is 29, 509.0 (haze), which is  
350 nearly 3.2 times to the low pollution period (9,269.6). The EF of Pb in high pollution period  
351 (648.8) shows approximately 1.6 times higher than that of low pollution period (397.7). The  
352 EF value of Zn in high pollution period (604.1) increased by a factor of 1.9 compared to low  
353 pollution period (318.1). While the EF value of Ni and Ti were reversely higher in low  
354 pollution period. Higher EF values of metals during high pollution period imply that cool and  
355 dry winter with less rainfall favored the accumulation of these metals onto particles in  
356 addition to those greatly contributed by long-range transport from north China.

#### 357 (4) Correlations among trace metals

358 Correlations among trace metals can be used to determine whether these metals have similar  
359 sources, therefore, the correlation coefficients of each trace metal for both high and low  
360 pollution periods are summarized in Table 3. In low pollution period, there are no significant  
361 correlations between metals. Nevertheless, few metals were observed to have good  
362 correlations in high pollution period. V, Fe, Ti, Cr, Mn and As have high correlation  
363 coefficients ( $0.72 < r < 0.95$ ). As is characterized as one of the representatives of coal burning  
364 in China (Kang et al., 2011). V is released to the ambient environment from oil burning and  
365 the fossil fuel combustion is reported to account for 85% of total V emissions in China (Duan  
366 and Tan, 2013). The correlated V, Fe, Ti, Cr, Mn and As may suggest that oil burning and  
367 coal combustion could be the major sources of these metals. The ratio of Cu/As was reported  
368 to be 1.2 (279/224, ug/g in PM) in honeycomb coal burning (Ge et al., 2004), while in this  
369 study it is approximately 1.8 during the high pollution period, indicating coal combustion  
370 may not be the unique source of Cu, and worn tires and automobile brake pads abrasion  
371 particles could be another contributors of Cu (Okuda et al., 2008). Cd and Pb, which were  
372 reported to correlate in coal-fired power plants in China (Deng et al., 2014), were also found  
373 with a significant correlation coefficient of 0.9581 in the high pollution period. As reported  
374 by Ge et al. (2004) and Karanasiou et al. (2007), Cd and Pb existed predominantly in fine  
375 particles other than in coarse particles of coal burning, part of aerosols containing Cd and Pb  
376 could be possibly transported from Northern China (as shown in Fig. 3 (a)), where heavy  
377 industries and centralized / decentralized house-heating are responsible for coal emissions.

### 378 3.4.2 Polycyclic aromatic hydrocarbons (PAHs)

379 PAHs are usually predominantly derived from incomplete combustion or high-temperature  
380 pyrolysis processes of fossil fuels and biomass (Ravindra et al., 2008). The average  
381 concentrations of each PAH during the high and low pollution periods are plotted in Fig.4.  
382 The total PAH concentration during the high pollution episode ranged from 15.2 to 331.9 ng  
383 m<sup>-3</sup> with an average of 90.6 ng m<sup>-3</sup>, accounting for 0.8‰ of total PM<sub>2.5</sub> mass concentration.  
384 These results are comparable with those in other cities (Nantong, Wuxi and Suzhou) of YRD  
385 (range: 13.9-229.0 ng m<sup>-3</sup>, average: 88.2 ng m<sup>-3</sup>) (Zhang et al., 2013). As seen in Table 6, the  
386 average total PAH concentration in this study is also consistent with the result in PRD (91.5 ±  
387 36.1 ng m<sup>-3</sup>) (Huang et al., 2014), but lower than the result in Zhengzhou of northern China,  
388 which is 211 ng m<sup>-3</sup> and accounts for 1.2‰ of total PM<sub>2.5</sub> mass concentration (Wang et al.,  
389 2015a). The total PAHs level in the low pollution episode ranged from 14.3 to 59.6 with an  
390 average of 34.8 ng m<sup>-3</sup>. Results in this work were higher than the PAHs concentrations  
391 obtained in Guangzhou (low pollution period: 13.3 ng m<sup>-3</sup>, high pollution period: 59.8 ng m<sup>-3</sup>)  
392 (Tan et al., 2011). Compared to the low pollution episode, BkF, Flu, Chr, Bpe, BbF, BaA,  
393 BaP and InP were about tripled, while Phe, Ant, Ace, Fla, Pyr and Bpe were about doubled  
394 and the rest pollutants increased less than 50%. In addition to the strong accumulation of  
395 local aerosols and long-range transport of those from outside of this region, another reason  
396 for higher occurrence levels of PAHs during the high pollution period (winter time) could be  
397 due to the relatively lower temperature which would promote more PAHs to distribute and  
398 condense onto particulate matter via the gas-particle partitioning process (He and  
399 Balasubramanian, 2009).

400 During the high pollution episode BkF, BbF, InP, Bpe and Chr dominated. BkF exhibited the  
401 highest concentration at 13.3 ng m<sup>-3</sup> and Acy was the lowest at 0.017 ng m<sup>-3</sup> in high pollution  
402 aerosols. The concentration of highly carcinogenic BaP was 3.8 ng m<sup>-3</sup> during the high  
403 pollution episode, similar to the result in northeast China (Jin et al., 2012). Retene has been  
404 identified as a tracer for soft wood burning especially conifers (Azevedo et al., 2002) and its  
405 concentration was doubled during the high pollution period, indicating the increased  
406 contribution from soft wood burning.

407 The low molecular weight 2 and 3 ring PAHs - Nap, Ace, Acy, Flu and Ant, were observed in  
408 low abundance in particle phases (< 1 ng m<sup>-3</sup>), while the high molecular weight PAHs (4-6  
409 rings) accounted for 87.5% of total PAHs. This result is in good agreement with the study  
410 conducted in Nanjing where 4-6 ring PAHs accounted for more than 80% of the total PAHs

411 (Meng et al., 2015). Basically, PAHs with 2 or 3 rings are present mainly in gaseous phase  
412 due to their higher volatility and they would more actively participate in photochemical  
413 reactions and, thus, their atmospheric lifetimes are reported to be a few hours or less, much  
414 shorter than those with more than three rings (Oliveira et al., 2014).

415 The diagnostic ratios of PAHs in particulates have been commonly used as indicators for  
416 source apportionment of PAHs (Bourotte et al., 2005; Esen et al., 2008; Harrison et al., 1996).  
417 Previous studies have shown that PAH concentrations varied largely based on their  
418 composition and different emission sources (Alves et al., 2014), therefore, some PAH ratios  
419 are adopted to study the source origins. The mean value of Fla/(Fla+Pyr) during the high  
420 pollution episode ranged from 0.32 to 0.70, with an average value of 0.54, which was in the  
421 range of wood combustion ( $>0.50$ ), approaching 0.58 for coal combustion (Bravo-Linares et  
422 al., 2012; Xu et al., 2012). The diagnostic ratio of BaP/Bpe was calculated at 0.49 and 0.30  
423 for the high and low pollution aerosols respectively, associated with the result of non-traffic  
424 emissions ( $<0.6$ ) (Bravo-Linares et al., 2012), indicating less influence of traffic on PAH  
425 concentrations and PM<sub>2.5</sub> concentrations. This is reasonable since the sampling site is around  
426 half kilometre away from the main roads where the air pollution attributed to traffic  
427 emissions could be reduced to low levels (Zhu et al., 2002). To summarize, aerosols were not  
428 significantly influenced by traffic emissions, mainly arising from pyrogenic sources, such as  
429 biomass burning and coal combustion.

### 430 **3.4.3 Inorganic ions**

431 The ion concentrations, their relative abundance in PM<sub>2.5</sub> and their ratios for the high and low  
432 pollution episodes are summarized in Table 4. Total ionic mass concentrations in the high  
433 and low pollution periods were 59.5 and 32.6  $\mu\text{g m}^{-3}$ , accounting for 55.3% and 62.0% of  
434 PM<sub>2.5</sub> mass concentrations, respectively. As compared in Table 6, the total ionic mass  
435 concentration during high pollution period in this study (59.5  $\mu\text{g m}^{-3}$ , 55.3% in PM<sub>2.5</sub>) is  
436 higher than that of Hangzhou in YRD (41.7  $\mu\text{g m}^{-3}$ , 38.5% in PM<sub>2.5</sub>) (Liu et al., 2015), but  
437 lower than that of Handan in northern China (77.3  $\mu\text{g m}^{-3}$ , 48.3% in PM<sub>2.5</sub>) (Wei et al., 2014).  
438 Sulfate, nitrate and ammonium (namely SNA) ranked as top three ions in the high pollution  
439 period and their total concentration reached up to 45.2  $\mu\text{g m}^{-3}$ , accounting for 78.0% of total  
440 ions masses and 42.0% of PM<sub>2.5</sub> mass, respectively. This result was in good agreement with  
441 the studies conducted in Nanjing, Hangzhou and Shanghai of YRD, which have shown that  
442 SNA accounting for 41.0-61.0% of PM<sub>2.5</sub> (Fu et al., 2008). Furthermore, the result is also  
443 comparable with two studies conducted in Beijing of northern China with SNA accounting

444 for 35.8% and 45.2% of PM<sub>2.5</sub> (Tan et al., 2016, Ho et al., 2016). Nevertheless, the result in  
 445 this study is slightly lower compared with that of Xi'an in central China, which has shown  
 446 that SNA contributed to 53.9% of total PM<sub>2.5</sub> concentration (Ho et al., 2016). In low pollution  
 447 aerosols, the abundance of sulfate, nitrate and ammonium accounted for even higher  
 448 percentage (44.0%) of aerosols with total average concentration of 23.2 µg m<sup>-3</sup>, accounting  
 449 for 71.3% of total ions masses. Similar result was also found in Chengdu where sulfate,  
 450 nitrate and ammonium accounted for 72.9% of total ion concentration (Tao et al., 2013).  
 451 When comparing them individually, SO<sub>4</sub><sup>2-</sup> and NO<sub>3</sub><sup>-</sup> were doubled in high pollution period,  
 452 while NH<sub>4</sub><sup>+</sup> was about 2.7 times higher. In low pollution episode, SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup>, NH<sub>4</sub><sup>+</sup> and K<sup>+</sup>  
 453 were obviously decreased, especially NH<sub>4</sub><sup>+</sup> which decreased from 7.2 µg m<sup>-3</sup> (High Pollution)  
 454 to 2.7 µg m<sup>-3</sup> (Low Pollution), while the rest ions such as Na<sup>+</sup> was not significantly changed.

455 Since the sampling site is located in an international port city- Ningbo, marine contribution to  
 456 ionic species needs to be eliminated to evaluate the inputs from other sources in this study.  
 457 Measured Na<sup>+</sup> was assumed to be derived from sea salts. Non-sea salt (nss) components were  
 458 calculated using the following equation (2) (Kong et al., 2014):

$$459 \text{ nss-X} = X_i - \text{Na}^+_i \times (\text{X/Na}^+)_{\text{sea}} \quad (2)$$

460 where, X<sub>i</sub> represents the ion concentration in samples, Na<sup>+</sup><sub>i</sub> is the concentration of Na<sup>+</sup> in  
 461 sample and (X/Na<sup>+</sup>)<sub>sea</sub> is seawater ratio which is 0.0385 for Ca<sup>2+</sup>, 0.037 for K<sup>+</sup> and 0.2516 for  
 462 SO<sub>4</sub><sup>2-</sup> respectively based on seawater composition (Balasubramanian et al., 2003). For the  
 463 high pollution episode, nss-SO<sub>4</sub><sup>2-</sup> (25.98 µg m<sup>-3</sup>), nss-Ca<sup>2+</sup> (2.44 µg m<sup>-3</sup>) and nss-K<sup>+</sup> (1.40 µg  
 464 m<sup>-3</sup>) accounted for 95.7%, 93.3% and 89.2% of SO<sub>4</sub><sup>2-</sup>, Ca<sup>2+</sup> and K<sup>+</sup> mass respectively. As for  
 465 low pollution episode, nss-SO<sub>4</sub><sup>2-</sup>/SO<sub>4</sub><sup>2-</sup>, nss-Ca<sup>2+</sup>/Ca<sup>2+</sup> and nss-K<sup>+</sup>/K<sup>+</sup> were 94.0%, 93.6% and  
 466 83.4% respectively, similar to high pollution aerosols. These results indicated that marine  
 467 sources are not the dominant origin of these ions. The mass ratio of Na<sup>+</sup>/SO<sub>4</sub><sup>2-</sup> is 0.17 and  
 468 0.24 in the high and low pollution samples, respectively, much lower than that of seawater  
 469 (3.98), reconfirming that marine source was not predominant (Yuan et al., 2015). The high  
 470 value of nss-Ca<sup>2+</sup>/Ca<sup>2+</sup> indicate a large amount of continental contribution to atmospheric  
 471 Ca<sup>2+</sup>. A few studies have revealed K<sup>+</sup> is a tracer of biomass burning (Chow et al., 2004). Two  
 472 times higher nss-K<sup>+</sup> was observed in the high pollution episode compared with low pollution  
 473 episode. This is not surprising since the air in high pollution episode passed through the  
 474 region with more dense fire-spots than low pollution episode, which could have carried more  
 475 biomass burning pollution to affect the sampling site. Further investigation on biomass  
 476 burning impact on local air quality would be discussed in section 3.4.4.



477 Other than marine-derived sulfate and nitrate, these ions may also be formed through gas-  
478 particle transition. Therefore, to investigate the transformation of SO<sub>2</sub> to SO<sub>4</sub><sup>2-</sup> and NO<sub>2</sub> to  
479 NO<sub>3</sub><sup>-</sup>, sulfur oxidation ratio (SOR) and nitrogen oxidation ratio (NOR) were applied and  
480 caculated by following equations (Lin, 2002):

$$481 \text{ SOR} = [\text{nss-SO}_4^{2-}] / ([\text{nss-SO}_4^{2-}] + [\text{SO}_2]) \quad (3)$$

$$482 \text{ NOR} = [\text{NO}_3^-] / ([\text{NO}_3^-] + [\text{NO}_2]) \quad (4)$$

483 Where [X] refers to molar concentration, higher value of SOR and NOR would imply the  
484 greater oxidation of gaseous species like SO<sub>2</sub> and NO<sub>2</sub> and the increase of secondary aerosol  
485 formation. In this work, NOR in high and low pollution episodes were 0.14 and 0.08  
486 respectively, indicating the formation of nitrate during the high pollution episode, which is  
487 consistent with the result obtained in Shanghai where the NOR value was 0.18 and 0.084  
488 during high and low pollution days respectively (Hua et al., 2015). It is reported that the  
489 photochemical oxidation of SO<sub>2</sub> occurred when SOR>0.1 (Ohta and Okita, 1990). The SOR  
490 value for the low pollution episode was 0.44, suggesting a considerable conversion from SO<sub>2</sub>  
491 to SO<sub>4</sub><sup>2-</sup>. While the SOR value for the high pollution samples (0.27) was surprisingly lower  
492 than that of low pollution samples, but comparable with the SOR values of high pollution  
493 aerosols in Taiwan (0.30) and Guangzhou (0.29) (Lin, 2002; Tan et al., 2009). Lower SOR in  
494 high pollution aerosols may probably due to the rapid increase of SO<sub>2</sub> and relatively less  
495 formation of sulphate during high pollution episode: (1) During the high pollution episode,  
496 with more stagnant weather during the winter at Ningbo, more SO<sub>2</sub> would preferably  
497 accumulate under poorer air circulation/ dispersion and weaker solar radiation. (2)  
498 Additionally, it has also reported that high NO<sub>2</sub> concentrations and weak photochemical  
499 process cannot generate sufficient oxidants for the formation of sulphate (Hua et al., 2008).  
500 The ratio of (NO<sub>3</sub><sup>-</sup>/ nss-SO<sub>4</sub><sup>2-</sup>) to (NO<sub>2</sub>/ SO<sub>2</sub>) in high pollution aerosols was more than 4 times  
501 higher than low pollution aerosols, suggesting nitrate formation is more significant than  
502 sulphate formation in high pollution episode. The higher formation of nitrate may have  
503 hindered the generation of conversion oxidants for the formation of sulphate and  
504 consequently led to lower SOR value in high pollution period.

505 In this study, the correlation between NH<sub>4</sub><sup>+</sup> and SO<sub>4</sub><sup>2-</sup> was only 0.12 during the low pollution  
506 episode, while NH<sub>4</sub><sup>+</sup> was highly correlated with NO<sub>3</sub><sup>-</sup> with coefficient of 0.70, indicating  
507 nitrate dominated in secondary inorganic aerosols formation in low pollution episode. In high  
508 pollution aerosols, NH<sub>4</sub><sup>+</sup> was found both highly correlated with SO<sub>4</sub><sup>2-</sup> and NO<sub>3</sub><sup>-</sup> with

509 coefficients of 0.72 and 0.73 respectively, indicating the formation of  $(\text{NH}_4)_2\text{SO}_4$ ,  $\text{NH}_4\text{HSO}_4$   
510 and  $\text{NH}_4\text{NO}_3$ , which is consistent with the results in Shanghai during high pollution period  
511 (Du et al., 2011). The correlation coefficients between SOR and NOR-temperature were 0.31  
512 and 0.08, indicating very little influence of temperature. No obvious relation was observed  
513 between SOR, NOR and relative humidity. The correlation coefficients between SOR, NOR  
514 and  $\text{PM}_{2.5}$  concentrations were 0.16 and 0.42, respectively, indicating higher aerosol  
515 concentration (higher surface area) may have potentially favored the secondary aerosol  
516 formation though the number concentrations have not been considered here.

#### 517 **3.4.4 Anhydrosugar compounds**

518 Anhydrosugars such as levoglucosan (L) and mannosan (M) are reported as unique and stable  
519 components of aerosols emitted from biomass burning (Saarnio et al., 2010), thus, they are  
520 used as tracers to investigate contributions from biomass burning to high pollution aerosols in  
521 this study. The average concentrations of levoglucosan and mannosan during high and low  
522 pollution episodes are listed in Table 5, as well as the ratio of L and M. The average  
523 concentrations of levoglucosan during dry high pollution episode and wet low pollution  
524 period were  $76.1 \pm 27.7 \text{ ng m}^{-3}$  and  $14.7 \pm 6.7 \text{ ng m}^{-3}$ , accounting for 0.7‰ and 0.3‰ of total  
525  $\text{PM}_{2.5}$  mass concentration, respectively. Compared with the results above, a suburb area  
526 located in Pearl River Delta (PRD) region of south China was observed with higher  
527 concentration of levoglucosan during dry season ( $181.0 \pm 124.0 \text{ ng m}^{-3}$ ) but with a lower  
528 concentration during the wet season ( $7.5 \pm 8.7 \text{ ng m}^{-3}$ ) (Zhang et al., 2015b). As shown in  
529 Table 6, significantly higher levoglucosan concentrations were observed in Beijing of  
530 Northern China and Xi'an of Central China than in this study. In Beijing the average  
531 levoglucosan concentrations are reported to be  $307 \text{ ng m}^{-3}$  (3.1‰ in  $\text{PM}_{2.5}$ ) (Zhang et al.,  
532 2008) and  $359.3 \pm 130.2 \text{ ng m}^{-3}$  (1.4‰ in  $\text{PM}_{2.5}$ ) (Ho et al., 2016), while in Xi'an even higher  
533 level of levoglucosan is found at  $653.3 \pm 191.4 \text{ ng m}^{-3}$  and accounting for 2.8‰ of  $\text{PM}_{2.5}$   
534 mass concentration (Ho et al., 2016). Significantly higher concentrations of levoglucosan in  
535 northern and central China could be possibly due to large amount of softwood/grass burning  
536 for decentralized house-heating in the country area especially during the winter season  
537 (Cheng et al., 2013a). The average concentration of mannosan in this study during the dry  
538 high pollution episode ( $14.9 \pm 6.3 \text{ ng m}^{-3}$ ) was about 5 fold higher compared to the wet low  
539 pollution episode ( $2.6 \pm 1.4 \text{ ng m}^{-3}$ ). The concentration of mannosan during the dry high  
540 pollution episode was comparable with that of dry season in PRD region ( $10.0 \pm 6.2 \text{ ng m}^{-3}$ )  
541 (Zhang et al., 2015b). Higher levoglucosan and mannosan concentrations in the high

542 pollution aerosols indicated the strong contribution from biomass burning to the organic  
543 matter present. In this study, levoglucosan was correlated very well with mannosan, with  
544 coefficient of 0.99 and both of them were positively correlated to PM<sub>2.5</sub>, with linear  
545 coefficients of 0.67 and 0.60, respectively, further confirming that biomass burning  
546 contributed to the high pollution aerosols.

547 The L/M ratios can also be used to distinguish emissions from specific types of biomass.  
548 Based on combustion chamber studies, Engling et al. (2013) summarized the following L/M  
549 ratios for the differentiation of various biomass sources: softwood (3-5), hard wood (14-15),  
550 peat (11) and grass (5.5). In addition, Sang et al. (2013) reported average L/M ratios were  
551  $32.6 \pm 19.1$  for crop residues. In this study, the high pollution episode has displayed a  
552 consistent L/M ratio with average of 5.2, while L/M of low pollution episode is in between of  
553 4.8 and 7.2, with average of 5.9, which falls into the L/M range of softwood and grass  
554 burning. Given that the high pollution episode occurred from December 2012 to January  
555 2013 when rice harvest season had passed in YRD, it might be safe to infer that rice straw  
556 burning was not the dominant form of biomass burning. Based on the air mass backward  
557 trajectory analysis, long-range transport of these organic tracers from north China might be  
558 the main origin where the softwood/grass in addition to coal would be used in large amounts  
559 for decentralized heating in the country area (Cheng et al., 2013a).

#### 560 **4. Conclusions**

561 1. For the 32 samples analysed, the high pollution samples clearly exhibited a higher average  
562 concentration of PM<sub>2.5</sub> ( $111 \mu\text{g m}^{-3}$ ) than the low pollution samples ( $53 \mu\text{g m}^{-3}$ ).

563 2. High pollution aerosols had higher concentrations of metals, PAHs, inorganic ions and  
564 organic tracers. Total ionic mass concentrations in the high and low pollution periods were  
565  $59.5$  and  $32.6 \mu\text{g m}^{-3}$ , respectively, accounting for 55.3% and 62.0% of PM<sub>2.5</sub> mass  
566 concentrations, respectively, with nitrate, sulphate and ammonium accounting for over 42.0%  
567 of PM<sub>2.5</sub> mass concentrations in both periods. Total metals concentrations were  $3.8 \mu\text{g m}^{-3}$   
568 and  $1.6 \mu\text{g m}^{-3}$  in the high and low pollution episodes respectively. Al was the highest in both  
569 episodes. PAHs in Ningbo were comparable with those in other cities of YRD. The high  
570 molecular weight PAHs (4-6 rings) accounted for 87.5% of total PAHs.

571 3. Heavy metals including Cd and Pb indicated that there is a major contribution from  
572 anthropogenic sources especially from coal combustion. Diagnostic ratios Fla/(Fla+Pyr) and  
573 BaP/Bpe indicated the predominant sources of PAHs were both coal and wood combustion,

574 rather than traffic emissions. The elevated nss-K<sup>+</sup>, retene, levoglucosan and mannosan  
575 concentrations, and the ratio of L/M around 5 during the high pollution episode implied  
576 contributions from softwood and grass burning to the aerosols.

577 4. Through the comparisons of above mentioned chemical components in highly polluted  
578 aerosols between this study and other studies in China, those medium-sized and megacities in  
579 Northern and Central China seemed to be experiencing generally more severe aerosol  
580 pollution than in YRD region in terms of the occurrence levels of PM<sub>2.5</sub>, PAHs, inorganic  
581 ions and biomass burning tracer levoglucosan for the past decade.

582 5. Long-range trajectories showed that the air in Ningbo during the high pollution episode  
583 had been transported from heavily polluted northern areas and passed through regions with  
584 more dense fire spots than the low pollution episode. To summarize, the high pollution  
585 aerosols in Ningbo arise from a combination of local emissions, long-range transport of air  
586 pollution and formation of secondary aerosols with the assistance of stagnant atmospheric  
587 conditions in this region.

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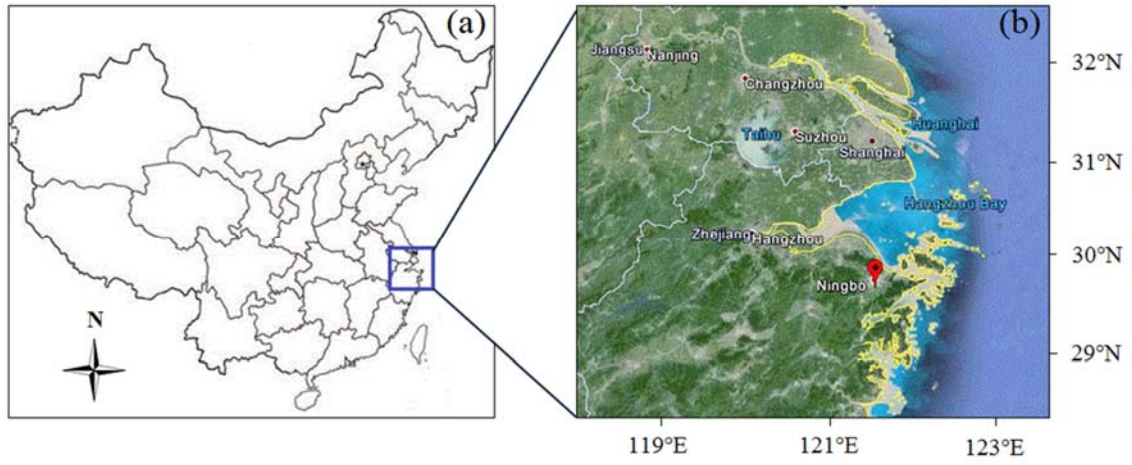
822 **Figure Captions**

823 Fig. 1 (a) Location of YRD region in China and (b) Location of Ningbo in YRD

824 Fig. 2 Daily-averaged PM<sub>2.5</sub> concentrations during the high and low pollution periods of  
825 Ningbo from 2012/12/03 to 2013/06/27

826 Fig. 3 Air mass backward trajectories and representative fire spots during the (a) high and (b)  
827 low pollution periods in Ningbo

828 Fig. 4 Comparison of PAH concentrations in the high and low pollution periods

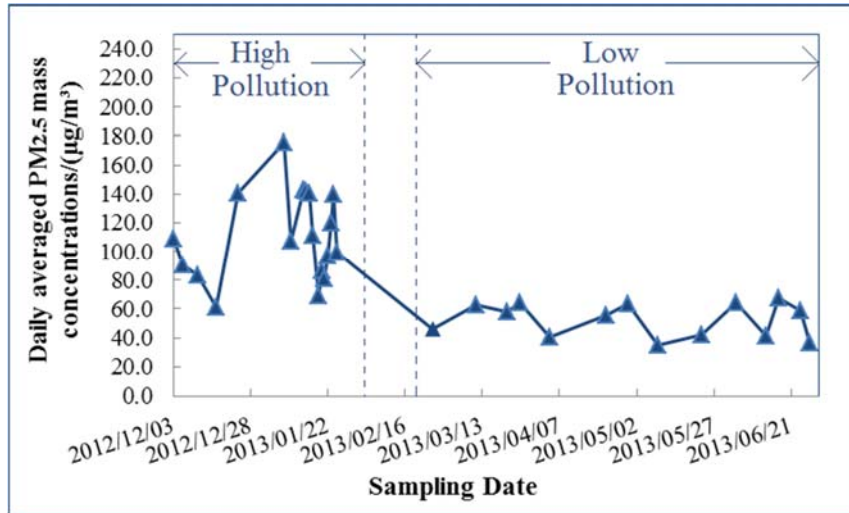


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Fig. 1 (a) Location of YRD region in China and (b) Location of Ningbo in YRD

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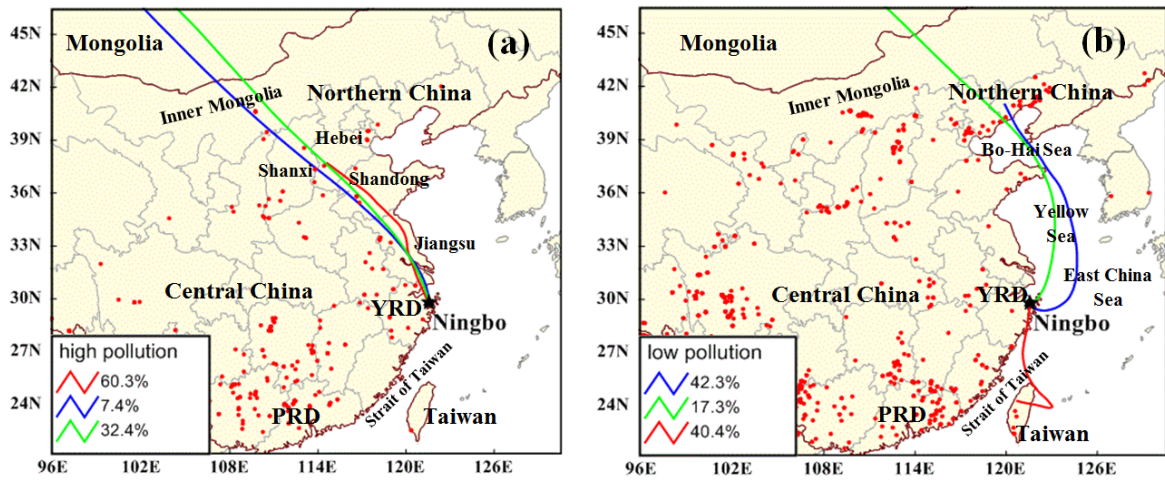
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Fig. 2 Daily-averaged PM<sub>2.5</sub> concentrations during the high and low pollution periods of Ningbo from

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2012/12/03 to 2013/06/27

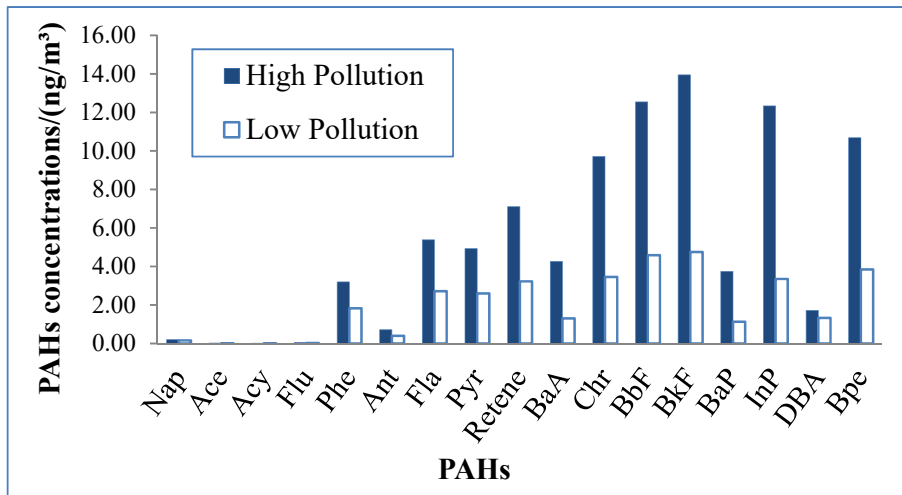


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836 Fig. 3 Air mass backward trajectories and representative fire spots during the (a) high and (b) low pollution  
 837 periods in Ningbo

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Fig. 4 Comparison of PAH concentrations in the high and low pollution periods

842 **Table Captions**

843 Table 1 Average daily values for the meteorological data during the high and low pollution  
844 periods

845 Table 2 Water soluble ( $C_{WS}$ ) and total ( $C_T$ ) metal concentrations, corresponding water  
846 solubility and enrichment factors (EF) of the high and low pollution samples

847 Table 3 Correlation matrix for the trace metal concentrations

848 Table 4 Average concentrations of ionic species and their abundance in  $PM_{2.5}$

849 Table 5 Average concentrations of levoglucosan and mannosan during the high and low  
850 pollution periods

851 Table 6 Comparisons of chemical compositions and their relative abundance in highly  
852 polluted aerosols from different regions in China

853 Table 1 Average daily values for the meteorological data during the high and low pollution periods

	Sampling Days	PM <sub>2.5</sub> / ( μg m <sup>-3</sup> )	Temperature (°C)	Rainfall (mm)	Wind Speed (km/h)	Relative Humidity (%)
High Pollution	18	110.9 ± 30.8	6.5	0.015	2.1	67.9
Low Pollution	14	52.6 ± 11.9	18.8	0.040	1.9	76.8

854

855 Table 2 Water soluble ( $C_{ws}$ ) and total ( $C_T$ ) metal concentrations, corresponding water solubility and enrichment  
 856 factors (EF) of the high and low pollution samples

	High Pollution					Low Pollution				
	$C_{ws}$ ( $\mu\text{g m}^{-3}$ )	$C_T$ ( $\mu\text{g m}^{-3}$ )	$(C_T/PM_{2.5})$ /%	Water solubility/%	EF**a	$C_{ws}$ ( $\mu\text{g m}^{-3}$ )	$C_T$ ( $\mu\text{g m}^{-3}$ )	$(C_T/PM_{2.5})$ /%	Water solubility/%	EF
Al	0.200	1.087	0.98	18.4	Ref**b	0.147	0.589	1.12	24.9	Ref
Mn	0.036	0.366	0.33	9.8	38.3	0.022	0.173	0.33	12.8	33.4
Zn	0.136	0.736	0.66	18.5	604.2	0.075	0.210	0.40	35.7	318.0
Co	0.002	0.009	0.0081	17.6	41.7	0.001	0.004	0.0076	21.7	37.7
Cd	0.005	0.047	0.042	9.7	29342.0	0.002	0.008	0.015	26.5	9495.7
Cu	0.069	0.155	0.14	44.3	419.1	0.040	0.068	0.13	58.1	339.8
Cr	0.011	0.053	0.048	21.4	53.0	0.008	0.029	0.055	26.5	53.2
Ni	0.011	0.072	0.065	15.0	162.4	0.005	0.053	0.10	9.3	219.9
Pb	0.055	0.277	0.25	20.0	649.1	0.031	0.092	0.17	33.1	399.7
As	0.022	0.087	0.079	24.8	472.6	0.010	0.037	0.070	26.6	372.6
V	0.011	0.052	0.047	20.6	38.5	0.004	0.023	0.044	17.0	31.0
Fe	0.261	0.816	0.74	32.0	1.7	0.160	0.321	0.61	49.8	1.2
Ti	0.007	0.019	0.017	36.4	0.3	0.005	0.012	0.023	41.3	0.4
Sum	0.8	3.8				0.5	1.6			
PM <sub>2.5</sub>	110.9	110.9				52.6	52.6			
( $\Sigma$ metals)/PM <sub>2.5</sub>	0.7%	3.4%				1.0%	3.1%			

857 \*<sup>a</sup> EF - Enrichment factor of metal, is defined as dividing the relative abundance of each metal in sample by its  
 858 corresponding average abundance in the upper continental crust.

859 
$$EF_i = \left(\frac{C_i}{C_{ref}}\right)_{atmosphere} / \left(\frac{C_i}{C_{ref}}\right)_{crust}$$

860 \*<sup>b</sup> Al was chosen as a reference metal for the calculation of enrichment factors



861 Table 3 Correlation matrix for the trace metal concentrations

	Mn	Zn	Co	Cd	Cu	Al	Cr	Ni	Pb	As	V	Fe	Ti
High Pollution													
Mn	1												
Zn	0.4069	1											
Co	0.2748	0.6224	1										
Cd	0.6139	0.4107	0.3286	1									
Cu	0.5289	0.1341	0.4003	0.5093	1								
Al	0.0061	0.1871	0.3222	0.1075	0.0218	1							
Cr	<b>0.8803</b>	0.4436	0.2122	0.4824	0.3332	0.0236	1						
Ni	0.4433	0.1323	0.2142	0.0872	0.5352	0.0059	0.3524	1					
Pb	0.6618	0.3397	0.2772	<b>0.9581</b>	0.5257	0.0327	0.4959	0.1012	1				
As	<b>0.8351</b>	0.5981	0.3306	0.7216	0.4179	0.0110	<b>0.7222</b>	0.3050	0.6976	1			
V	<b>0.8470</b>	0.3951	0.2354	0.5305	0.4756	0.0236	0.6940	0.4341	0.5678	<b>0.9029</b>	1		
Fe	<b>0.8082</b>	0.6936	0.363	0.7463	0.3254	0.0145	0.7202	0.186	0.7337	<b>0.9477</b>	<b>0.8126</b>	1	
Ti	0.7885	0.4903	0.334	0.6958	0.5649	0.0209	0.6418	0.440	0.6583	<b>0.948</b>	<b>0.8828</b>	<b>0.8173</b>	1
Low Pollution													
Mn	1												
Zn	0.0078	1											
Co	0.2630	0.0530	1										
Cd	0.0194	0.0005	0.5251	1									
Cu	0.2443	0.0008	0.5907	0.4194	1								
Al	0.2297	0.2543	0.4380	0.4679	0.5528	1							
Cr	0.6450	0.2871	0.3748	0.1072	0.4597	0.0124	1						
Ni	0.5568	0.0179	0.8301	0.2303	0.5075	0.4323	0.1926	1					
Pb	0.0024	0.0023	0.5275	0.6891	0.1763	0.2760	0.1086	0.2663	1				
As	0.0365	0.4745	0.0576	0.2961	0.2480	0.7264	0.0250	0.0398	0.0504	1			
V	0.0130	0.0434	0.1481	0.0064	0.0011	0.1111	0.1882	0.1163	0.1847	0.5334	1		
Fe	0.0374	0.5416	0.1324	0.1731	0.1456	0.6172	0.0638	0.1527	0.2764	0.4387	0.0463	1	
Ti	0.3469	0.2167	0.1670	0.0040	0.1763	0.0019	0.6092	0.1636	0.0063	0.0687	0.1229	0.2030	1

862

863 Table 4 Average concentrations of ionic species and their abundance in PM<sub>2.5</sub>

	High Pollution		Low Pollution		C <sub>ion</sub> (High Pollution)/ C <sub>ion</sub> (Low Pollution)
	C <sub>ion</sub> * / ug m <sup>-3</sup>	(C <sub>ion</sub> /PM <sub>2.5</sub> ) /%	C <sub>ion</sub> * / ug m <sup>-3</sup>	(C <sub>ion</sub> /PM <sub>2.5</sub> ) /%	
F <sup>-</sup>	0.06	0.06	0.05	0.09	1.37
Cl <sup>-</sup>	3.32	3.09	2.44	4.65	1.36
Br <sup>-</sup>	0.013	0.01	0.007	0.01	1.78
NO <sub>3</sub> <sup>-</sup>	12.39	11.51	5.93	11.27	<b>2.09</b>
SO <sub>4</sub> <sup>2-</sup>	27.08	25.16	14.65	27.86	<b>1.85</b>
Li <sup>+</sup>	0.06	0.06	0.05	0.10	1.20
Na <sup>+</sup>	4.37	4.06	3.42	6.50	1.28
NH <sub>4</sub> <sup>+</sup>	7.15	6.64	2.67	5.07	<b>2.68</b>
K <sup>+</sup>	1.56	1.45	0.77	1.46	<b>2.04</b>
Mg <sup>2+</sup>	0.84	0.78	0.57	1.08	1.48
Ca <sup>2+</sup>	2.61	2.43	2.07	3.93	1.27
sum	59.5	55.3	32.6	62.0	-

864 \* C<sub>ion</sub> was defined as the average value of daily mean ionic concentration during high/ low pollution episode.

865 Table 5 Average concentrations of levoglucosan and mannosan during the high and low pollution periods

	Levoglucosan/ ( ng m <sup>-3</sup> )	Mannosan/ ( ng m <sup>-3</sup> )	L/M
High Pollution	76.1 ± 27.7	14.9 ± 6.3	5.2
Low Pollution	14.7 ± 6.7	2.6 ± 1.4	5.9

866 Table 6 Comparisons of chemical compositions and their relative abundance in highly polluted aerosols from different regions in China

Location	Sampling period (yyyy/mm-yyyy/mm)	PM <sub>2.5</sub> (µg m <sup>-3</sup> )	Metals (µg m <sup>-3</sup> )	PAHs (ng m <sup>-3</sup> )	Inorganic ions (µg m <sup>-3</sup> )	Levoglucosan (ng m <sup>-3</sup> )	Reference
Ningbo, YRD region	2012/12-2013/01	110.9 ± 30.8	3.78 (3.4% <sup>*1</sup> )	90.6 (0.8‰)	59.5 (55.3%)	76 ± 28 (0.7‰)	This study
Suzhou etc, YRD region	2009-07/2010/04			88.2			Zhang et al., 2013
Hangzhou, YRD region	2004/04-2005/03	108.2 ± 43.2			41.7 (38.5%)		Liu et al., 2015
PRD region	2010/11-2010/12			91.5 ± 36.1			Huang et al., 2014
PRD region	2010/05-2010/06 2010/11-2010/12					181.0 ± 124.0	Zhang et al., 2015b
Beijing, Northern China	2002/07-2003/07	99.2				307 (3.1‰)	Zhang et al., 2008
Tianjin, Northern China	2013/01-2013/01	> 94					Han et al., 2014
Handan, Northern China	2012/12-2013/01	160.1 ± 77.9			77.3 (48.3%)		Wei et al., 2014
Zhengzhou, Northern China	2011/03-2014/01	194 ± 109		211 (1.1‰)			Wang et al., 2015a
Beijing, Northern China	2006/12-2006/12	142.3 ± 46.0	3.74 (2.6%)		50.9 (SNA <sup>*2</sup> )		Tan et al., 2016
Beijing, Northern China	2013/01-2013/02	258 ± 100			116.5 (SNA)	359.3 ± 130.2	Ho et al., 2016
Xi'an, Central China	2013/01-2013/02	233 ± 52			125.6 (SNA)	653.3 ± 191.4	Ho et al., 2016

867 <sup>\*1</sup> Percentages in the table are the relative abundance of each chemical component in PM<sub>2.5</sub>, calculated as its mass concentration divided by its corresponding PM<sub>2.5</sub>  
868 concentration.

869 <sup>\*2</sup> SNA – SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup> and NH<sub>4</sub><sup>+</sup> (namely SNA)