



## Research Article

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# Effects of $\text{Fe}_2\text{O}_3$ addition and annealing on the mechanical and dissolution properties of MgO- and CaO-containing phosphate glass fibres for bio-applications

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**Abstract:** This paper investigated the preparation of phosphate glass fibres (PGFs) in the following systems: i)  $45\text{P}_2\text{O}_5\text{-}5\text{B}_2\text{O}_3\text{-}5\text{Na}_2\text{O}\text{-}(29\text{-}x)\text{CaO}\text{-}16\text{MgO}\text{-}(x)\text{Fe}_2\text{O}_3$  and ii)  $45\text{P}_2\text{O}_5\text{-}5\text{B}_2\text{O}_3\text{-}5\text{Na}_2\text{O}\text{-}24\text{CaO}\text{-}(21\text{-}x)\text{MgO}\text{-}(x)\text{Fe}_2\text{O}_3$  (where  $x = 5, 8$  and  $11$  mol%) for biomedical applications. Continuous fibres of  $23 \pm 1$   $\mu\text{m}$  diameter were prepared via a melt-draw spinning process. Compositions with higher  $\text{Fe}_2\text{O}_3$  content and higher MgO/CaO ratio required higher melting temperature and longer heating time to achieve glass melts for fibre pulling. The effects of  $\text{Fe}_2\text{O}_3$  addition and annealing treatment on mechanical properties and degradation behaviours were also investigated. Adding  $\text{Fe}_2\text{O}_3$  was found to increase the tensile strength from  $523 \pm 63$  (Ca-Fe5) to  $680 \pm 75$  MPa (Ca-Fe11), improve the tensile modulus from  $72 \pm 4$  (Ca-Fe5) to  $78 \pm 3$  GPa (Ca-Fe11) and decrease the degradation rate from  $4.0$  (Mg-Fe5) to  $1.9 \times 10^{-6}$   $\text{kg m}^{-2} \text{s}^{-1}$  (Mg-Fe11). The annealing process reduced the fibre tensile strength by 46% (Ca-Fe5), increased the modulus by 19.6% (Ca-Fe8) and decreased the degradation rate by 89.5% (Mg-Fe11) in comparison to the corresponding as-drawn fibres. Additionally, the annealing process also impeded the formation of precipitate shells and revealed co-existence of the precipitation and the pitting corrosion as fibre degradation behaviours.

**Keywords:** Phosphate glass fibres, iron, mechanical properties, annealing, degradation

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## 1 Introduction

Phosphate glasses can be made with formulations similar to the inorganic component of bone and have been widely investigated as novel biomaterials for bone repair applications [1–4]. They have been produced and processed into different forms such as powders [5], chopped fibres [6, 7] or continuous fibres to reinforce resorbable polymeric matrices [8–11]. It is known that high  $\text{P}_2\text{O}_5$  content phosphate glasses have low chemical durability due to their susceptibility to hydration and hydrolysis [10]. Conversely, phosphate glasses with higher phosphate content are easier to draw into continuous fibres due to their longer chain structure [12]. In order to improve the fibre drawing performance of more chemically durable and lower content phosphate glasses,  $\text{B}_2\text{O}_3$  was added to optimise the glass structure as it has been shown that addition of  $\text{B}_2\text{O}_3$  in-

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creased the effective phosphate chain lengths ( $Q^2$  species) [13–16] without compromising on chemical durability.

In our previous work [17], phosphate glasses with compositions of  $45P_2O_5-5B_2O_3-5Na_2O-(29-x)CaO-16MgO-(x)Fe_2O_3$  and  $45P_2O_5-5B_2O_3-5Na_2O-24CaO-(21-x)MgO-(x)Fe_2O_3$  (where  $x = 5, 8$  and  $11$ ) were prepared and investigated. It was found that the local phosphate network was mainly based on pyro- and metaphosphate units ( $Q^1$  and  $Q^2$ ) even though the O/P ratio was higher than 3.5. This was due to the addition of  $B_2O_3$ , which introduced  $BO_3/BO_4$  units into the phosphate chain structure to form a mixed borophosphate glass. It was anticipated that this extended borophosphate structure would improve fibre drawing performance without suffering the poor chemical durability that would be observed in an extended pure phosphate network.

The degradation rate of resorbable materials plays a very important role for their intended application [18]. In this respect, phosphate glasses offer a distinct advantage over other materials like Si-based bioglass, as their degradation rates can easily be tailored via simply modifying the addition of modifier oxides to suit the end application [7, 13, 19–22]. In addition, it has also been reported that the degradation rate of phosphate glasses has a direct influence on their biological performance, as their biocompatibility is enhanced with higher chemical durability [4, 23–28].  $Fe_2O_3$  addition to phosphate glasses has successfully been shown to efficiently improve their chemical durability, which in turn showed improvement in the biocompatibility of the glasses [17, 29–33]. However, the study [34] reported that  $Fe_2O_3$  addition could lead to more stringent fibre drawing conditions for phosphate glasses, since the glass drawing temperature was close to the maximum temperature limit for the in-house melt-drawn fibre preparation system that was used.

A number of studies have investigated the effect of iron oxide on phosphate glass systems, such as  $P_2O_5-Fe_2O_3-BaO$  [35],  $P_2O_5-Fe_2O_3-PbO$  [36] and  $P_2O_5-Fe_2O_3-TiO_2-CaF_2$  [37]. When mono- or divalent metal oxide modifiers were replaced by  $Fe_2O_3$ , these studies showed that the phosphate chain structure increased in cross-linking density [17, 38]. Both  $Fe^{2+}$  and  $Fe^{3+}$  can exist in the iron phosphate glasses, forming strong chemically resistant P-O-Fe(II) and P-O-Fe(III) bonds instead of P-O-P bonds [29, 39]. Furthermore, both  $Fe^{2+}$  and  $Fe^{3+}$  play different roles in the glass structure, with  $Fe^{2+}$  suggested to be involved in cross-linking, whereas  $Fe^{3+}$  could go in the phosphate network [32]. In the case of  $P_2O_5-B_2O_3-Na_2O-CaO-MgO-Fe_2O_3$ , cross-linking between the phosphate chains occurred due to the higher field strength of Fe ions compared to sodium, calcium and magnesium [23, 40, 41], with the

ionic radius decreasing for  $Na^+(1.13 \text{ \AA}) > Ca^{2+}(0.99 \text{ \AA}) > Mg^{2+}(0.72 \text{ \AA}) > Fe^{3+}(0.64 \text{ \AA})$  (six-fold coordination for  $Mg^{2+}$  and  $Ca^{2+}$ ) [13, 42]. In addition, Albon *et al.* [29] investigated the  $P_2O_5-Fe_2O_3-Na_2O$  glasses by using Mössbauer spectrometry and found an increase in  $Fe^{3+}/Fe^{2+}$  ratio with increasing  $Fe_2O_3$  content in the glass. They also observed that the  $Fe^{2+}$  ions were oxidized to  $Fe^{3+}$  in the glass during the heat treatment. Therefore, it was expected that both  $Fe_2O_3$  content and heat treatment would affect the glass structure and consequently the physical properties.

Phosphate glasses have been prepared into fibres with sufficient mechanical strength for investigations into bone repair applications [1]. Furthermore, phosphate glass fibre (PGF) reinforced polymeric composites have also been manufactured with suitable mechanical strength for use as bone repair devices. For example, the flexural properties of PGF/PLA composites have been reported to be 80–500 MPa for flexural strength and 5–25 GPa for modulus, with varying fibre formulations, fibre volume fractions and fibre alignment [8, 43–45]; while the flexural properties of cortical bone are stated to be 135–193 MPa for strength and 10–20 GPa for modulus [46–48].

The aim of the present work was to utilise the extended borophosphate network structure to improve the fibre drawing characteristics of phosphate glasses containing less than 50% phosphate, whilst investigating the effects of varying the ratio of divalent and trivalent additives (Ca, Mg and Fe) on the mechanical and chemical behaviour. Most studies to date substitute between monovalent and other species, whereas here the exchange is between divalent and trivalent species. The effects of heat treatment on the fibres were also considered. This work will provide glass formulations that provide fibres with superior mechanical and chemical properties, but that are also suited to continuous production as reinforcement materials. The bulk glass of these formulations has been studied previously [17] and the present work analysed the glass fibres from these formulations which has not been investigated before.

## 2 Materials and methods

### 2.1 Glass preparation and compositional analysis

Five different glass compositions in two series of  $45P_2O_5-5B_2O_3-5Na_2O-(29-x)CaO-16MgO-(x)Fe_2O_3$  and  $45P_2O_5-5B_2O_3-5Na_2O-24CaO-(21-x)MgO-(x)Fe_2O_3$  (where  $x = 5, 8$  and  $11$ ) were prepared via a melt quenching method and

**Table 1:** The real glass formulations (in mol%) of the batch used for fibre preparation characterized by using ICP-MS, the corresponding O/P ratio and glass transition temperature ( $T_g$ ) as analysed in previous work [17].

Glass series	Sample codes	P <sub>2</sub> O <sub>5</sub>	B <sub>2</sub> O <sub>3</sub>	Na <sub>2</sub> O	CaO	MgO	Fe <sub>2</sub> O <sub>3</sub>	O/P	$T_g$ (°C)
45P <sub>2</sub> O <sub>5</sub> -5B <sub>2</sub> O <sub>3</sub>	Ca-Fe5	44.0 ± 0.1	4.8 ± 0.1	6.2 ± 0.1	23.0 ± 0.1	16.8 ± 0.1	5.3 ± 0.1	3.37	513
-5Na <sub>2</sub> O-(29-x)CaO	Ca-Fe8	43.6 ± 0.1	4.8 ± 0.1	6.7 ± 0.1	20.3 ± 0.1	16.2 ± 0.1	8.4 ± 0.1	3.45	517
-16MgO-(x)Fe <sub>2</sub> O <sub>3</sub>	Ca-Fe11	43.4 ± 0.1	4.7 ± 0.1	7.3 ± 0.1	17.4 ± 0.1	16.0 ± 0.1	11.2 ± 0.1	3.52	521
45P <sub>2</sub> O <sub>5</sub> -5B <sub>2</sub> O <sub>3</sub>	Mg-Fe5	44.0 ± 0.1	4.8 ± 0.1	6.2 ± 0.1	23.0 ± 0.1	16.8 ± 0.1	5.3 ± 0.1	3.37	513
-5Na <sub>2</sub> O-24CaO	Mg-Fe8	43.8 ± 0.1	4.9 ± 0.1	6.8 ± 0.1	23.1 ± 0.1	13.2 ± 0.1	8.2 ± 0.1	3.44	517
-(21-x)MgO-(x)Fe <sub>2</sub> O <sub>3</sub>	Mg-Fe11	43.7 ± 0.1	4.8 ± 0.1	7.3 ± 0.1	22.7 ± 0.1	10.1 ± 0.1	11.3 ± 0.1	3.51	523

compositionally confirmed by inductively coupled plasma mass spectrometry (ICP-MS) as discussed in the previous work [17]. Table 1 shows the ICP results, O/P ratio and glass transition temperature of the investigated glass batches. In brief, the glass compositions were prepared using reagent grade P<sub>2</sub>O<sub>5</sub>, B<sub>2</sub>O<sub>3</sub>, NaH<sub>2</sub>PO<sub>4</sub>, CaHPO<sub>4</sub>, MgHPO<sub>4</sub>·3H<sub>2</sub>O and FePO<sub>4</sub>·2H<sub>2</sub>O (Sigma Aldrich, UK) powder precursors which were mixed into a Pt/5% Au crucible (Birmingham Metal Co., UK). After preheating at 350°C for 30 min, the crucible was transferred to a furnace and the salt mixtures were subsequently melted at 1200°C for 90 min. Finally, the molten glass was poured onto a steel plate and left to cool.

## 2.2 Phosphate glass fibres (PGFs) preparation

Continuous fibres of ~25 µm diameter were produced via a melt-draw spinning process using a dedicated in-house facility [49]. The pulling temperature was adjusted to around 1200°C. The molten glass was pushed through the bushing by hydrostatic pressure and collected on a rotating drum at a speed of 900 rpm.

The resulting fibres were annealed for 90 min at 10°C above the  $T_g$  (see Table 1). The heating cycle involved a 5°C min<sup>-1</sup> ramp to  $T_g - 200^\circ\text{C}$ , followed by 1°C min<sup>-1</sup> to  $T_g + 10^\circ\text{C}$ , 90-min dwell and cooling to  $T_g - 200^\circ\text{C}$  by 0.25°C min<sup>-1</sup> before another 5°C min<sup>-1</sup> ramp to room temperature.

## 2.3 Fibre Degradation Analysis

A degradation study was conducted on both the annealed and non-annealed fibres. Around 300 mg fibres, 50 mm length were put into a glass vial containing 30 mL phosphate buffer saline solution (PBS; pH = 7.4 ± 0.1) and the vial was placed into an oven at 37.0 ± 0.5°C. At specified time points, the solution pH was measured. Then the

PBS solution was removed carefully and the fibres were dried in a drying oven at 50°C overnight before weighing. The mass loss per surface area of the fibres was plotted against the immersion time to explore the degradation rate (kg m<sup>-2</sup> s<sup>-1</sup>). This analysis was carried out for 28 days (time points of 0, 1, 7, 14, 21 and 28) and the PBS solution was changed at each time point.

## 2.4 Single fibre tensile test (SFTT)

SFTT was conducted in accordance with ISO 11566 [49]. Thirty fibres were mounted individually onto plastic tabs for each sample with a 25-mm-gauge length testing setup. The ends of each fibre were bonded to the plastic tab with an acrylic adhesive (Dymax 3099, Dymax, Europe) and the adhesive was cured using UV light. The diameter of the fibres was measured using an LSM 6200 laser scan micrometre (Mitutoyo, Japan), which was previously calibrated with the glass fibre of known diameter (determined by SEM) and the error on diameter measurements was considered to be ±0.3 µm [34]. The tensile test was performed using a LEX 810 tensile tester (Diastron, UK) at room temperature with a load cell capacity of 1 N and a crosshead speed of 0.017 mm s<sup>-1</sup>.

The Weibull distribution is an accepted statistical tool used to characterise the failure mode of brittle fibres [50]. The Weibull modulus and normalising stress are described as the shape and scale parameters, respectively, where the normalising stress (Weibull scale) can be regarded as the most probable stress at which a fibre of gauge length will fail [34]. In this study, Weibull parameters were obtained from the tensile strength data calculated using Minitab 15 (version 3.2.1).

## 2.5 Scanning electron microscopy (SEM) analysis

SEM images were taken to examine the change in surface morphology of the annealed and non-annealed fibres. The chopped fibres were fixed on a sample stage with conductive adhesive tape and coated with gold using a Leica EM SCD 500 high vacuum sputter coater prior to use. The micrographs were taken at an accelerating voltage of 3 kV using SE2 mode on a Zeiss Sigma/VP SEM.

## 2.6 Statistical analysis

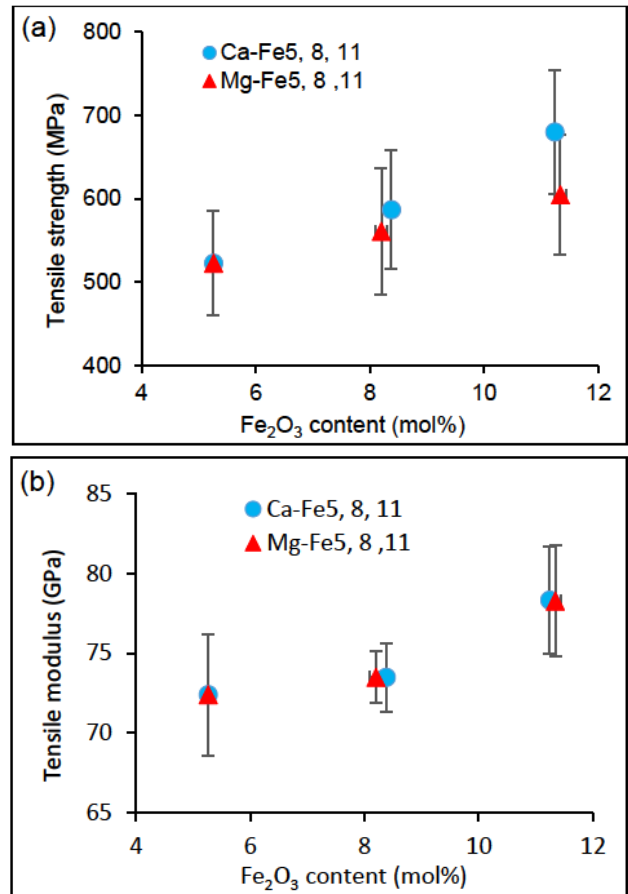
The statistical analysis was performed using the IBM SPSS (version 22). A Student's unpaired t-test was used to analyse the significance of the difference between different samples, assuming equal variance and determining two-tailed p-values. One-way analysis of variance (ANOVA) was calculated with the Bonferroni post-test to compare the significance of the change in the factors with iron content or time. The threshold value chosen for statistical significance was at the 0.05 level.

## 3 Results

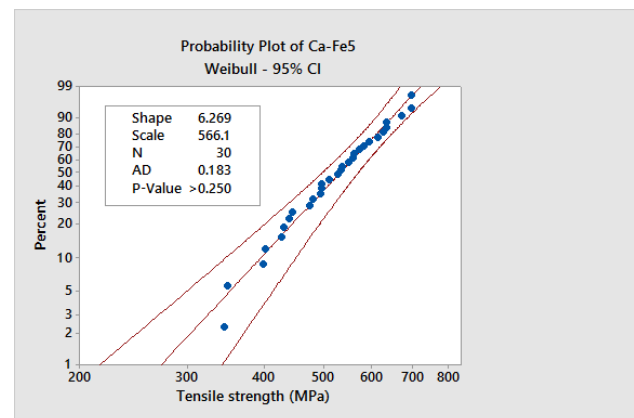
### 3.1 Mechanical properties of as-drawn fibres

As shown in Figure 1a, a significant increase ( $p < 0.05$ ) in tensile strength of the fibres was observed with increasing  $\text{Fe}_2\text{O}_3$  content at the expense of CaO/MgO. The values obtained were  $523 \pm 63$ ,  $587 \pm 72$  and  $680 \pm 75$  MPa for Ca-Fe5, 8 and 11 fibres, respectively; whilst  $523 \pm 63$ ,  $561 \pm 75$  and  $605 \pm 72$  MPa were obtained for Mg-Fe5, 8 and 11 fibres, respectively. No significant difference in tensile strength was observed between the glass series with the substitution of CaO and MgO. Table 2 represents the Weibull distribution of fibre strength. It was noted that the trend of normalising strength was consistent with that of mean tensile strength. The Weibull modulus of the fibres was seen to range from 6.0 to 8.7. Figure 2 shows an example of the Weibull plots.

Figure 1b shows a significant increase ( $p < 0.001$ ) in tensile modulus and the values were  $72 \pm 4$ ,  $74 \pm 2$  and  $78 \pm 3$  GPa for Ca-Fe5, 8 and 11 fibres, respectively. These values of modulus were numerically same as that of the Mg-Fe5, 8 and 11 fibres, respectively.



**Figure 1:** Tensile (a) strength and (b) modulus of as-drawn fibres of the composition Ca-Fe5, 8, 11 and Mg-Fe5, 8, 11 against  $\text{Fe}_2\text{O}_3$  content ( $n = 30$ ).



**Figure 2:** Weibull analysis of Ca-Fe5 was completed from the tensile strength data calculated using Minitab.

### 3.2 Mechanical properties of annealed fibres

The effect of annealing on mechanical properties was analysed. Figure 3a and b showed a decrease in tensile strength

**Table 2:** Weibull distribution of as-drawn fibres of the composition Ca-Fe5, 8, 11 and Mg-Fe5, 8, 11 ( $n = 30$ ). The tensile strength values are also included for ease of comparison.

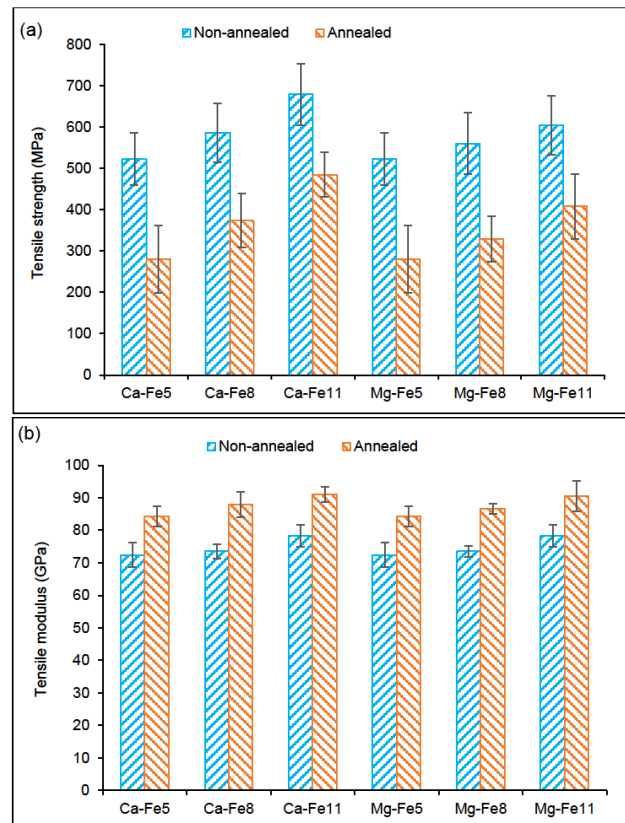
Sample codes	Diameter ( $\mu\text{m}$ )	Tensile strength (MPa)	Normalising strength (MPa)	Weibull modulus
Ca-Fe5	$24.2 \pm 1.2$	$523 \pm 63$	566	6.3
Ca-Fe8	$22.2 \pm 3.1$	$587 \pm 72$	663	6.3
Ca-Fe11	$23.2 \pm 1.9$	$680 \pm 75$	764	6.0
Mg-Fe5	$24.2 \pm 1.2$	$523 \pm 63$	566	6.3
Mg-Fe8	$21.5 \pm 1.9$	$561 \pm 75$	613	8.7
Mg-Fe11	$23.4 \pm 0.9$	$605 \pm 72$	682	6.5

and an increase in tensile modulus for the annealed fibres, respectively. The strength values of non-annealed fibres were shown in Table 2 and the values of annealed fibres were observed to be  $280 \pm 81$  MPa for Ca-Fe5,  $374 \pm 65$  MPa for Ca-Fe8,  $485 \pm 55$  MPa for Ca-Fe11,  $280 \pm 81$  MPa for Mg-Fe5,  $330 \pm 55$  MPa for Mg-Fe8 and  $408 \pm 78$  MPa for Mg-Fe11. Thus, the reductions in tensile strength after the annealing process were found to be by 46%, 36% and 29% of the as-drawn Ca-Fe5, 8 and 11 fibres ( $p < 0.001$ ), respectively; while the reductions were found to be by 46%, 41% and 32% of the as-drawn Mg-Fe5, 8 and 11 fibres ( $p < 0.001$ ), respectively.

According to Figure 3b, the tensile modulus of the annealed fibres were observed to be  $84 \pm 3$  GPa for Ca-Fe5,  $88 \pm 4$  GPa for Ca-Fe8,  $91 \pm 2$  GPa for Ca-Fe11,  $84 \pm 3$  GPa for Mg-Fe5,  $87 \pm 2$  GPa for Mg-Fe8 and  $91 \pm 5$  GPa for Mg-Fe11. Thus, the tensile modulus of the annealed fibres were observed to increase by 16.5% for Ca-Fe5, 19.6% for Ca-Fe8, 16.3% for Ca-Fe11, 16.5% for Mg-Fe5, 18.0% for Mg-Fe8 and 15.6% for Mg-Fe11 ( $p < 0.001$ ) compared to the corresponding as-drawn fibres.

### 3.3 Mechanical properties of degraded non-annealed fibres

Figure 4a shows the tensile strength of non-annealed glass fibres versus degradation time. It was observed that the tensile strength decreased rapidly during the earlier time points and then increased for day 14. After 1 day immersion, the tensile strength for Ca-Fe5, 8 and 11 fibres decreased from  $523 \pm 63$ ,  $587 \pm 72$  and  $680 \pm 75$  MPa to  $456 \pm 61$ ,  $345 \pm 81$  and  $485 \pm 98$  MPa, respectively. With further immersion up to 7 days, the strength of Ca-Fe5, 8 and 11 fibres decreased to  $323 \pm 48$ ,  $257 \pm 52$  and  $300 \pm 84$  MPa, respectively. Thus, at day 7 the reduction in strength observed was 38%, 56% and 56% of the as-drawn Ca-Fe5, 8 and 11 fibres, respectively. However, an increase in ten-

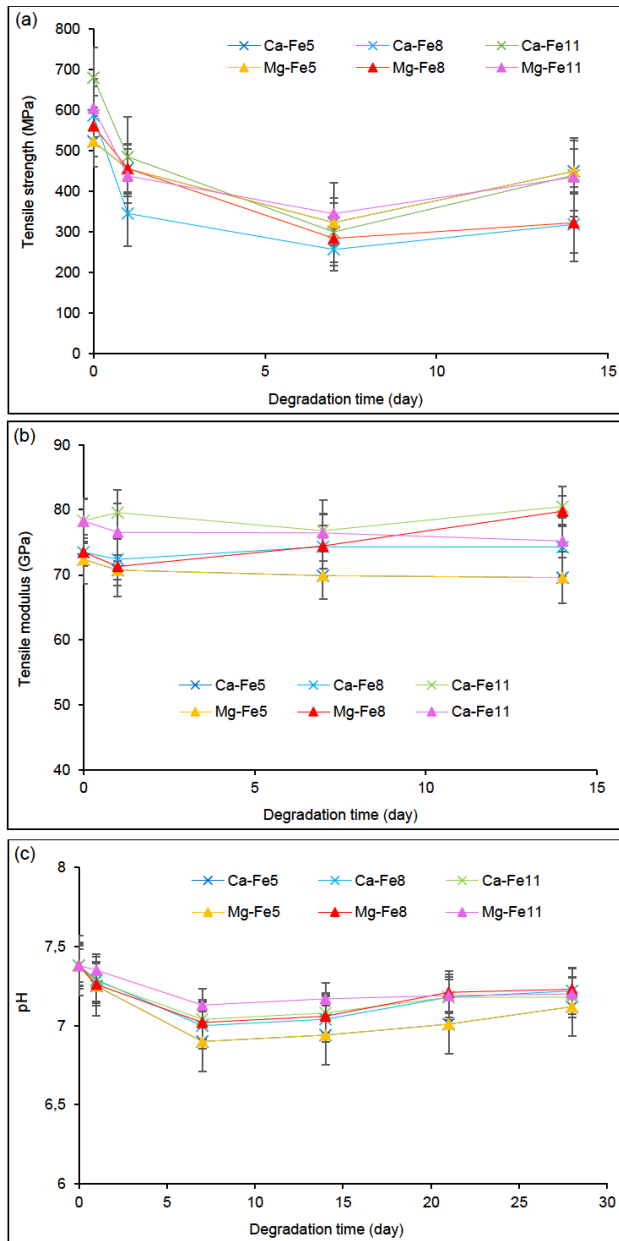


**Figure 3:** Tensile (a) strength and (b) modulus of non-annealed and annealed fibres of the composition Ca-Fe5, 8, 11 and Mg-Fe5, 8, 11 ( $n = 30$ ).

sile strength was observed for the 14 days-degraded fibres compared to the 7 days-degraded fibres, and the values of tensile strength at day 14 were  $449 \pm 54$ ,  $318 \pm 91$  and  $439 \pm 86$  MPa for Ca-Fe5, 8 and 11 fibres, respectively. Furthermore, no statistically significant difference in tensile strength was observed between the 0, 1, 7 and 14 days-degraded fibres.

The variance in tensile strength of Mg-Fe5, 8 and 11 fibres shows a similar trend with that of Ca-Fe5, 8 and 11





**Figure 4:** Tensile (a) strength and (b) modulus of non-annealed fibres of the composition Ca-Fe5, 8, 11 and Mg-Fe5, 8, 11 during degradation in PBS solution at 37°C and the pH (c) of the corresponding solutions ( $n \geq 20$ ). The lines are drawn as a guide for the eye.

fibres. After 1 day immersion, the tensile strength for Mg-Fe5, 8 and 11 fibres decreased from  $523 \pm 63$ ,  $561 \pm 75$  and  $605 \pm 72$  MPa to  $456 \pm 61$ ,  $456 \pm 57$  and  $438 \pm 67$  MPa, respectively. At day 7 the strength became  $323 \pm 48$ ,  $284 \pm 59$  and  $345 \pm 77$  MPa for Mg-Fe5, 8 and 11 fibres, respectively, which shows a reduction by 38%, 49% and 43% of the as-drawn Mg-Fe5, 8 and 11 fibres, respectively. Moreover, at day 14 the tensile strength of Mg-Fe5, 8 and 11 fi-

bres was observed to be  $449 \pm 54$ ,  $323 \pm 75$  and  $435 \pm 97$  MPa, respectively. Also, no statistically significant difference in tensile strength was observed between the 0, 1, 7 and 14 days-degraded fibres.

Figure 4b shows the tensile modulus of the corresponding degraded fibres. No statistically significant difference in fibre tensile modulus was observed throughout the degradation period.

Figure 4c shows the pH of the corresponding solutions, which endured at  $7.18 \pm 0.14$  for the whole degradation period, even though there was a slight decrease in pH during the early immersion period.

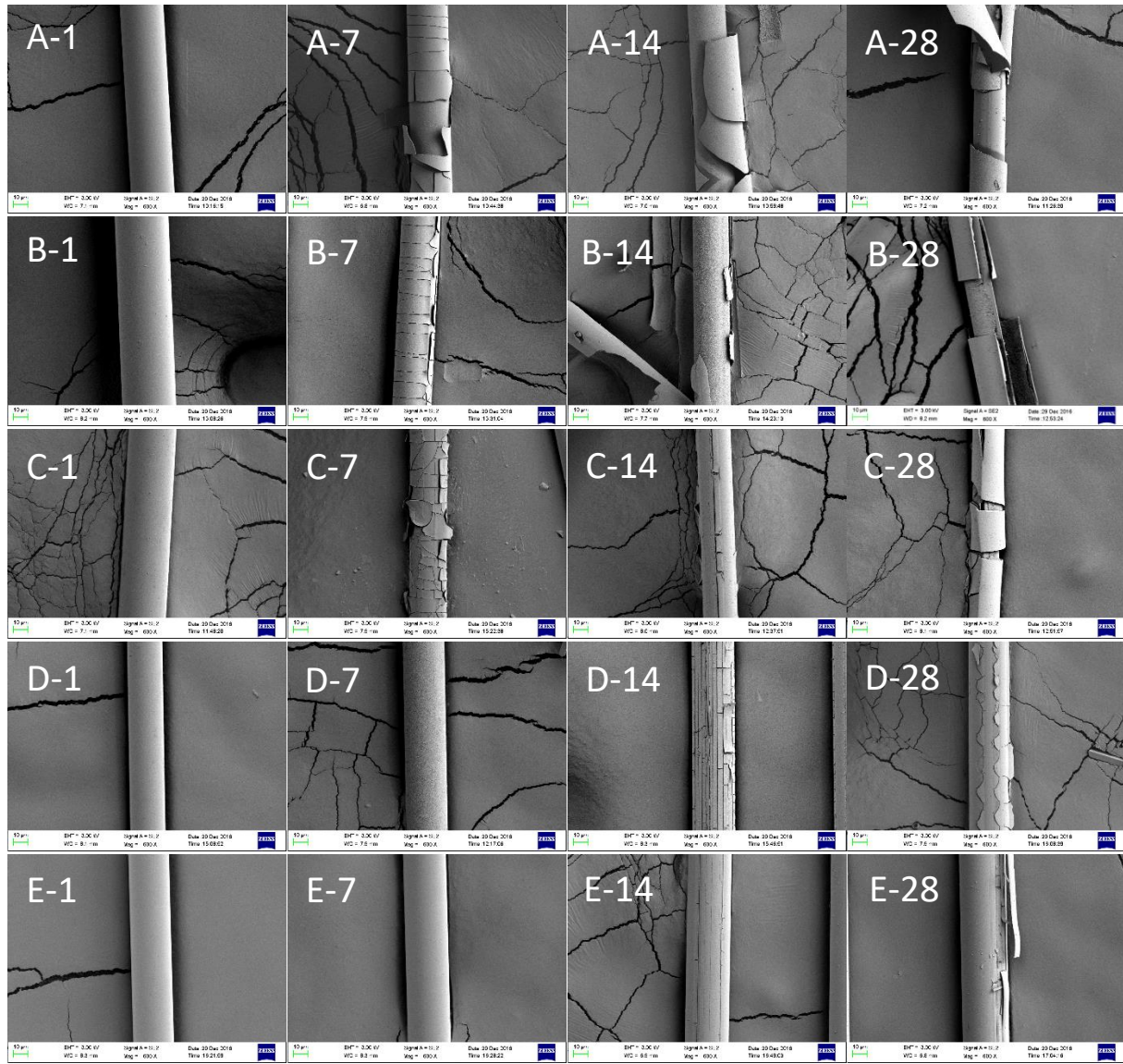
The micrographs in Figure 5 exhibit the corresponding surface morphologies of the degraded fibres. The morphologic changes in fibres revealed the effects of different immersion time and glass compositions. The cracked outer shells were observed with the degrading glass fibres.

### 3.4 Mechanical properties of degraded annealed fibres

Figure 6a shows the tensile strength of annealed glass fibres versus degradation time. Similar to the non-annealed fibres, a decrease in tensile strength was observed from day 1 to day 7, while an increase was seen from day 7 to day 14. However, no evident trend was found for the change of tensile strength on day 1 immersed samples. From day 1 to day 7, the tensile strength for Ca-Fe5, 8 and 11 fibres decreased from  $325 \pm 72$ ,  $385 \pm 76$  and  $358 \pm 63$  MPa to  $220 \pm 49$ ,  $259 \pm 77$  and  $283 \pm 64$  MPa, respectively, which shows a reduction by 21%, 31% and 42% of the corresponding day-0 fibres, respectively. At day 14, the tensile strength became  $293 \pm 62$ ,  $289 \pm 71$  and  $319 \pm 65$  MPa for Ca-Fe5, 8 and 11 fibres, respectively. No statistically significant change ( $p > 0.05$ ) in tensile strength was observed between the 0, 1, 7 and 14 days-degraded fibres.

Moreover, at day 1, the tensile strength of Mg-Fe5, 8 and 11 fibres were observed to be  $325 \pm 72$ ,  $354 \pm 73$  and  $323 \pm 86$  MPa, respectively. At day 7, the values of tensile strength became  $220 \pm 49$ ,  $280 \pm 67$  and  $260 \pm 69$  MPa for Mg-Fe5, 8 and 11 fibres, respectively, which shows a reduction by 21%, 15% and 36% of the corresponding day-0 fibres, respectively. At day 14, the strength values were observed to be  $293 \pm 62$ ,  $317 \pm 63$  and  $307 \pm 64$  MPa for Mg-Fe5, 8 and 11 fibres, respectively. Also, no statistically significant change ( $p > 0.05$ ) in tensile strength was observed between the 0, 1, 7 and 14 days-degraded fibres.

Figure 6b shows the tensile modulus of the corresponding degraded fibres. No statistically significant



**Figure 5:** Scanning electron microscopy of day 1, 7, 14 and 28 non-annealed degraded fibres of the composition Ca-Fe<sub>5</sub>/Mg-Fe<sub>5</sub> (A1-A28), Ca-Fe<sub>8</sub> (B1-B28), Mg-Fe<sub>8</sub> (C1-C28), Ca-Fe<sub>11</sub> (D1-D28) and Mg-Fe<sub>11</sub> (E1-E28). The cracked base shows the conductive tape.

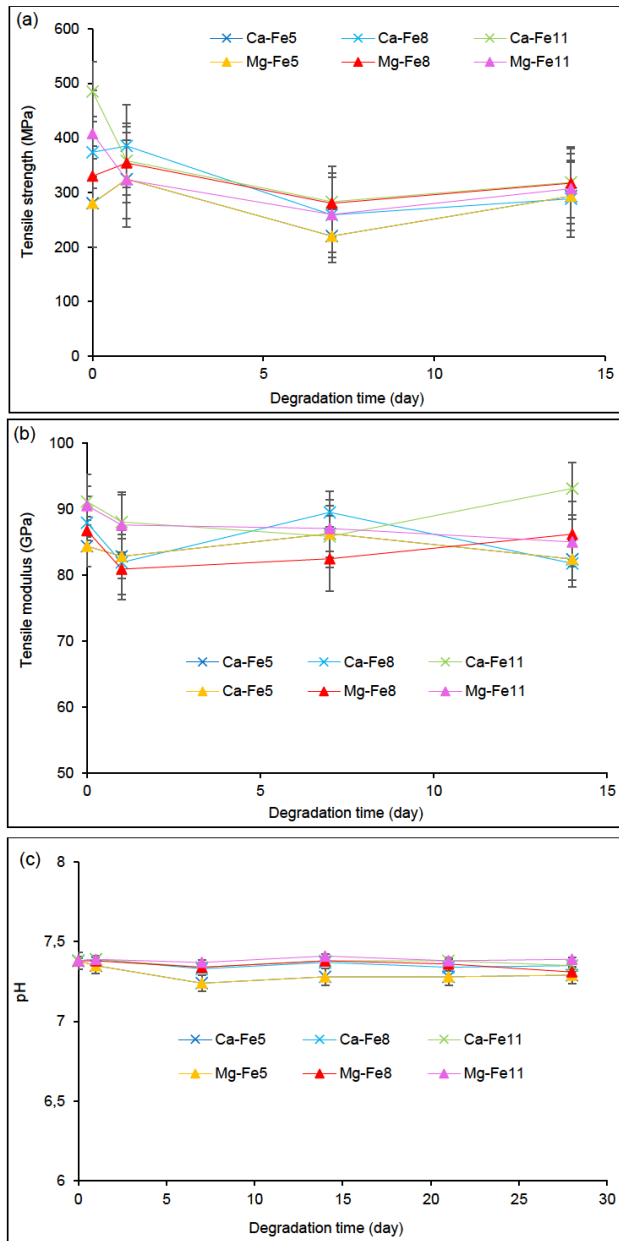
change ( $p > 0.05$ ) in the fibre tensile modulus was observed throughout the degradation study.

Figure 6c shows the pH of the corresponding solutions, which remained at  $7.36 \pm 0.04$  throughout the degradation study.

It was also observed from Figure 7 that the annealed fibres underwent a surface pitting during degradation, which was seen to be distributed heterogeneously along the fibre surface.

### 3.5 Degradation rates of fibres

Figure 8 represents degradation rates of the non-annealed and annealed fibres after 28-days degradation in PBS solution at 37°C. A 20-36% reduction in degradation rate of non-annealed fibres was observed for each 3 mol% Fe<sub>2</sub>O<sub>3</sub> increment at the cost of CaO/MgO. After 28 days, the degradation rates were observed to be  $39.6$ ,  $30.7$  and  $24.5 \times 10^{-7} \text{ kg m}^{-2} \text{ s}^{-1}$  for the non-annealed Ca-Fe<sub>5</sub>, 8 and 11 fibres, respectively, while the values were  $39.6$ ,  $29.4$  and  $18.7 \times 10^{-7} \text{ kg m}^{-2} \text{ s}^{-1}$  for the non-annealed Mg-Fe<sub>5</sub>, 8 and 11 fibres, respectively. No significant difference in degradation



**Figure 6:** Tensile (a) strength and (b) modulus of annealed fibres of the composition Ca-Fe5, 8, 11 and Mg-Fe5, 8, 11 during degradation in PBS solution at 37°C and the pH (c) of the corresponding solutions ( $n \geq 20$ ). The lines are drawn as a guide for the eye.

rate was found between the glass compositions with the replacements of CaO and MgO.

Similar trends were found for the annealed fibres and the degradation rate decreased by 38%-73% for each 3 mol%  $\text{Fe}_2\text{O}_3$  increment instead of CaO/MgO. The degradation rates were observed to be  $11.7$ ,  $6.3$  and  $3.2 \times 10^{-7} \text{ kg m}^{-2} \text{ s}^{-1}$  for the annealed Ca-Fe5, 8 and 11 fibres, respectively, and to be  $11.7$ ,  $7.3$  and  $2.0 \times 10^{-7} \text{ kg m}^{-2} \text{ s}^{-1}$  for the annealed Mg-Fe5, 8 and 11 fibres, respectively. Thus, the

annealing was observed to reduce the degradation rate by 70.4%, 79.9% and 86.9% for the Ca-Fe5, 8 and 11 fibres, respectively; while the degradation rate decreased by 70.4%, 75.3% and 89.5% for the Mg-Fe5, 8 and 11 fibres, respectively.

The decrease in degradation rate of non-annealed and annealed fibres with increasing  $\text{Fe}_2\text{O}_3$  content at the cost of CaO/MgO was shown to be statistically significant ( $p < 0.05$ ) via ANOVA data.

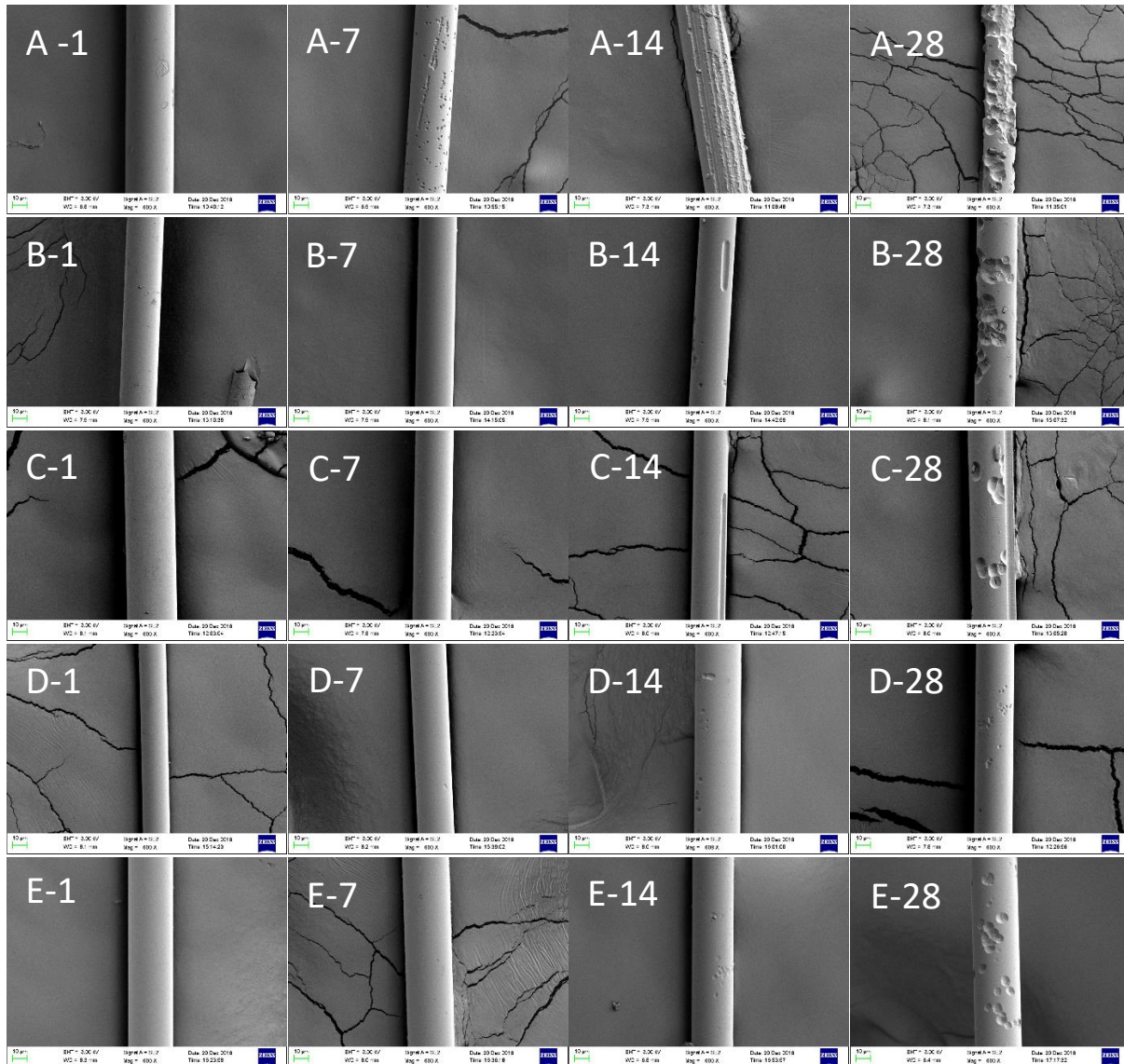
## 4 Discussion

In this study, continuous fibre was achieved for all the formulations investigated, with no breakage for at least 3 hours. As expected, the empirical evidence showed that the compositions with higher  $\text{Fe}_2\text{O}_3$  content and higher MgO/CaO ratio required higher melting temperatures and longer heating times to achieve suitable glass melts for fibre pulling.

### 4.1 Mechanical properties of as-drawn fibres

Generally, the mechanical properties of the fibres are strongly dependent on the glass formulation [34, 51]. Higher phosphate content in phosphate glasses normally results in longer phosphate chain length in the glass network [52]. The longer chain contributes to the higher mechanical strength of the fibres [53–55]. The higher field strength of the modifier ions can also increase the fibre strength by strengthening the cross-linking between phosphate chains [9, 56]. In this case, the addition of  $\text{Fe}_2\text{O}_3$  leads to the stronger P-O-Fe bonds instead of P-O-P in the chain, enhancing the mechanical properties and chemical durability, although the replacement of CaO/MgO by  $\text{Fe}_2\text{O}_3$  might increase the ionic bonds in the glass network [32, 57, 58]. Ma *et al.* [59] also stated that the dissolution behaviour of  $\text{P}_2\text{O}_5$ - $\text{Na}_2\text{O}$ - $\text{Fe}_2\text{O}_3$  glasses was much more dependent on iron content than the O/P ratio. Moreover, the higher field strength of iron ions than that of calcium and magnesium ions resulted in the stronger cross-linking within the glass structure, which could also be responsible for the increase in tensile strength and modulus of the fibres (see Figure 1a and 1b). Kurkjian [60] studied the mechanical properties of fibres drawn from a metaphosphate glass composition  $\text{P}_2\text{O}_5$ - $\text{Al}_2\text{O}_3$ -MgO- $\text{K}_2\text{O}$  and a pyrophosphate glass composition  $\text{P}_2\text{O}_5$ - $\text{Fe}_2\text{O}_3$ . They suggested that replacing monovalent cations with



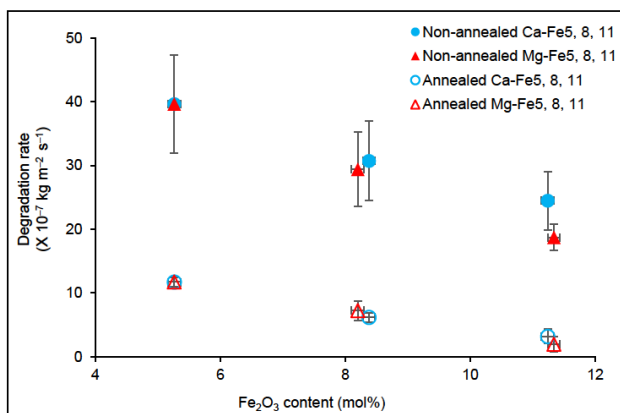


**Figure 7:** Scanning electron microscopy of day 1, 7, 14 and 28 annealed degraded fibres of the composition Ca-Fe5/Mg-Fe5 (A1-A28), Ca-Fe8 (B1-B28), Mg-Fe8 (C1-C28), Ca-Fe11 (D1-D28) and Mg-Fe11 (E1-E28). The cracked base shows the conductive tape.

trivalent/divalent iron increased the strength and modulus of the PGFs, which was suggested to be due to the higher cross-linking density between the phosphate chains caused by the higher Fe-O bond strength [23, 40].

Lin *et al.* [61] produced  $\text{P}_2\text{O}_5\text{-CaO-Fe}_2\text{O}_3$  glass fibres with diameter of  $\sim 20 \mu\text{m}$  and investigated the effect of  $\text{Fe}_2\text{O}_3$  addition on the properties of the fibres. The tensile strength of the fibres was observed to increase from  $\sim 600$  to  $\sim 1000$  MPa and the modulus increased from  $\sim 43$  to  $\sim 64$  GPa, as  $\text{Fe}_2\text{O}_3$  content increased from 5 to 22 wt%. This increase in mechanical properties was attributed to the formation of cross-linked Fe-O-P chains. Murgatroyd [62] stated that for the purpose of fibre drawing, a number of

strong bonds are required to withstand the stresses from the pulling process at high temperature. The continuity of a fibre drawing process depends on the ability of the strongest bonds to hold together as they are pulled out of the melt. In PGFs, the strongest bonds normally present the P-O bonds in the phosphate chain [63, 64]. The addition of  $\text{Fe}_2\text{O}_3$  led to a replacement of P-O-P by stronger P-O-Fe bonding [57], which increased bonding strength of the phosphate chains, resulting in an improvement in tensile strength of the fibres.



**Figure 8:** Degradation rates of non-annealed and annealed fibres of the composition Ca-Fe5, 8, 11 and Mg-Fe5, 8, 11 against Fe<sub>2</sub>O<sub>3</sub> content ( $n = 3$ ).

## 4.2 Dissolution properties of non-annealed fibres

As shown in Figure 4a, the tensile strength of the fibres was observed to decrease rapidly during the earlier immersing period and then increase for day 14. A similar observation was reported in the case of P<sub>2</sub>O<sub>5</sub>-B<sub>2</sub>O<sub>3</sub>-CaO-MgO-Na<sub>2</sub>O glass fibres [9], where the results presented a rapid loss of strength after 1 day immersion and a cumulative reduction by almost half of the initial values at day 7. However, an increase in tensile strength was observed from day 7 to day 14. This has been previously explained as the penetration of water molecules and Na<sup>+</sup>-H<sup>+</sup> exchange. The contact with aqueous media will lead to the Na<sup>+</sup>-H<sup>+</sup> exchange to form a hydrated layer and the diffusion of water will be impeded by the increasing thickness of the hydrated layer.

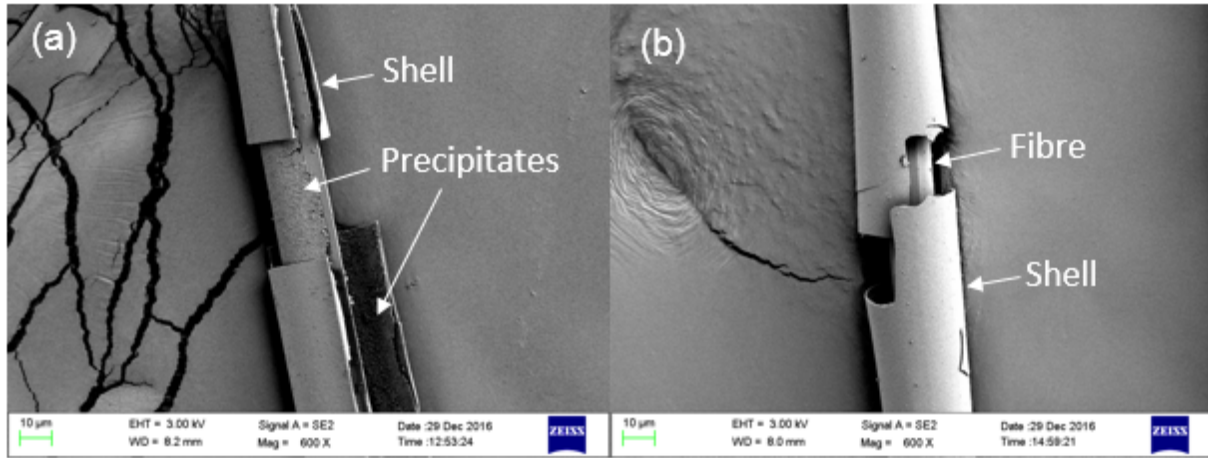
It is widely accepted that the dissolution mechanism of phosphate glasses mainly consists of two processes: dissolution by hydration of entire chains and hydrolysis of P-O-P bonds as initially proposed by Bunker *et al.* [65, 66]. Once immersed in aqueous solution, the hydrated phosphate chains are released into solution, where they undergo hydrolysis. However, an initial pH decrease (from 7.4 ± 0.1 in day-0 to 7.0 ± 0.1 in day-7) was observed in this case (see Figure 4), which was suggested to be a result of phosphate anions in solution released into the media, potentially creating acidic species [67–69]. The dissolved phosphate chains broke down to orthophosphate ultimately via hydrolysis in solution [66], leading to saturation of the media and precipitate formation in static environment. As shown in Figure 9a, the granulate layers were found at the fibre surface and the inner wall of the shell, which were most likely to be the uniformly distributed phosphate precipitates, indicating that the precipitates formed layer

by layer during degradation. Such a layer was also found by Clément *et al.* [70] when they investigated the dissolution properties of P<sub>2</sub>O<sub>5</sub>-CaO-Na<sub>2</sub>O glasses in SBF. The Raman results indicated depolymerisation of the phosphate chains due to the hydration reaction at the surface of the glass slabs. An interaction was observed between the hydrated layer at the surface of the glass and the surrounding media, resulting in a thin layer covering the glass sample. The Raman and EDX analysis indicated that this layer was composed of calcium orthophosphate groups. Abou Neel *et al.* [71] also suggested the precipitation of Ca-P at the surface of the P<sub>2</sub>O<sub>5</sub>-CaO-Na<sub>2</sub>O-Fe<sub>2</sub>O<sub>3</sub> glass fibres via ion release studies. However, it should be noted that their investigation was conducted in deionised water, which means that the cation of the precipitation came from the glass itself.

It was observed from Figure 9b that the outer shell remained at day 28, whilst the fibre core had undergone further degradation. The precipitated layer seems to be much more durable than the original glass and shows the potential to protect the inner glass from further degradation. However, this layer became very brittle after drying and disintegrated from the surface during sample preparation.

As expected, no significant change in fibre modulus was found during degradation since there was no change in the intrinsic glass structure caused by the degradation as its surface eroded [72].

Figure 8 shows that adding Fe<sub>2</sub>O<sub>3</sub> instead of CaO/MgO caused an approximate linear decrease in degradation rate. This increase in glass durability was also correlated with the micrographs shown in Figure 5. A number of studies [32, 57, 58] have given evidence for the enhancement of chemical durability of phosphate glasses caused by Fe<sub>2</sub>O<sub>3</sub> addition and suggested that it was attributed to the replacement of P-O-P bonds by more hydration resistant P-O-Fe bonds and the stronger cross-linking between phosphate chains. Yu *et al.* [32] investigated the chemical durability of the P<sub>2</sub>O<sub>5</sub>-Na<sub>2</sub>O-Fe<sub>2</sub>O<sub>3</sub> glass with varying Fe<sub>2</sub>O<sub>3</sub> content and determined the amount of Fe<sup>2+</sup> and Fe<sup>3+</sup> ions via Mössbauer spectroscopy. They found both the Fe<sup>2+</sup> and Fe<sup>3+</sup> ions in the phosphate glasses and the concentration of Fe<sup>3+</sup> increased with increasing Fe<sub>2</sub>O<sub>3</sub> content. They concluded that Fe in the form Fe<sup>3+</sup> strengthened the structure of the glasses like a network former. Thus, they supposed that the Fe<sup>3+</sup>/Fe<sup>2+</sup> ratio should be as high as possible for maximum chemical durability of the iron-containing phosphate glass. This supposition was also agreed by Mesko *et al.* [73].



**Figure 9:** Micrographs for the 28 days-degraded non-annealed (a) Mg-Fe8 and (b) Ca-Fe8 fibres. The cracked base shows the conductive tape.

### 4.3 Mechanical and dissolution properties of annealed fibres

The fibre pulling process has been shown to orientate the phosphate units along the fibre axis [62, 74, 75]. The rapid cooling will freeze the phosphate chains and the ionic bonds, which makes a highly reactive glass with inherent stresses [76]. The heat-treatment (annealing) process can allow the orientated chains to relax, removing this anisotropy in the fibres and reducing the residual stresses [74, 77, 78].

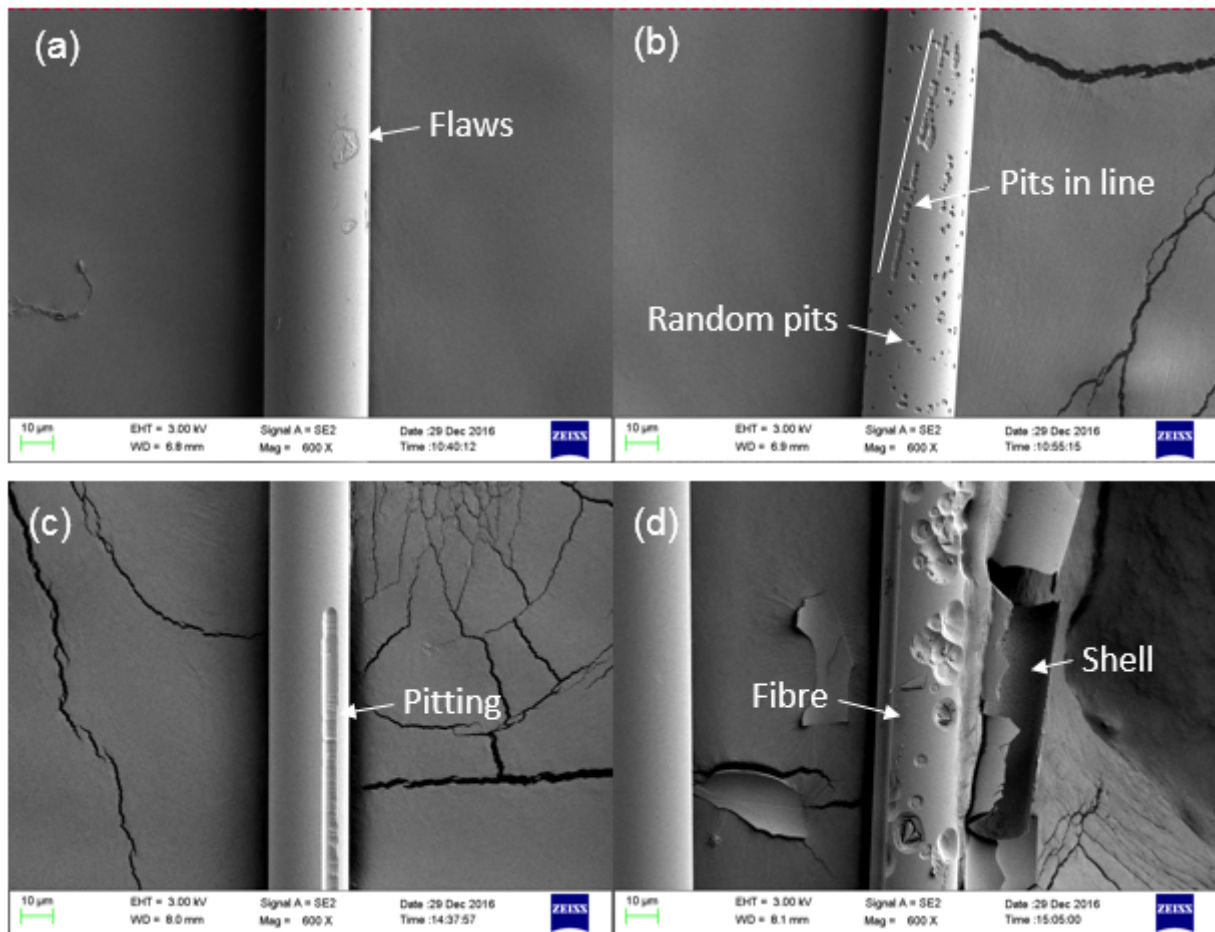
According to Figure 3a and 3b, the annealing shows a decrease in tensile strength ( $p < 0.001$ ) and an increase in tensile modulus ( $p < 0.05$ ) of the fibres. Similar observations were reported previously in the case of  $\text{P}_2\text{O}_5$ -CaO-MgO-Na<sub>2</sub>O [72] and  $\text{P}_2\text{O}_5$ -B<sub>2</sub>O<sub>3</sub>-CaO-MgO-Na<sub>2</sub>O glass fibres [9]. The increase in Young's modulus was suggested to be due to the more stable and compact glass structure. On the other hand, the enhancements of stability and cross-linking density decreased the dissolution rate of the fibres (see Figure 8) by reducing the chemical reactivity of the glass and impeding the water diffusion through the glass network, respectively.

Choueka *et al.* [76] stated that annealing of fibres close to the  $T_g$  could stabilize the bonds in the glass; however, the strength might be decreased as the properties of the fibre approached those of bulk glass. Furthermore, the loss of tensile strength could be explained with multiple factors such as manual handling. The pulling and quenching process will generate compressive stresses within the fibre. However, re-heating can cause a relaxation of compressive stresses and even a shrinkage of the fibre which may generate a tensile stress [79]. The high temperature

during annealing can also accelerate the interaction between the fibre surface and the moisture in the air [66] and the rapid loss of properties of PGFs can be ascribed to water attacking weak spots in the fibre surface that are under stress [80]. As shown in Figure 10b, the pits next to the white line were observed to arrange in line, while the rest of the pits were seen to be distributed randomly. The regularly distributed pits might indicate that pitting could be initiated from the fibre surface defect [77, 78]. Another example is shown in Figure 10c as the groove on the fibre which might occur due to a scratch on the fibre surface. The annealing operation may increase opportunities of surface damage, which may behave as stress concentration sites [81].

According to the morphological examinations, the annealed fibres in the same composition as that of non-annealed fibres exhibited heterogeneous depressions during degradation (see Figure 7). This type of pitting corrosion in the surface of annealed fibres was also observed by Choueka *et al.* [76] and Cozien-Cazuc *et al.* [20]. With a comparison between Figure 5 and Figure 7, it was noticed that non-annealed and annealed fibres underwent quite different degradation behaviours, which was suggested to be due to the rearrangement of the glass structure during annealing. The ion leach and exchange in dissolution process is believed to occur on sites of structural flaws where concentration of terminal cations interrupts the phosphate network [78]. The structural anisotropy occurs during the fibre drawing process and the terminal cations present a relative regular distribution [82]. Therefore, the diffusion-controlled leaching distributes uniformly at the surface of the fibre. However, annealing can relax the phosphate chains and disrupt them into a disordered en-





**Figure 10:** Micrographs for (a) 1 day-degraded annealed Ca-Fe5 fibre, (b) 7 days-degraded annealed Ca-Fe5 fibre (the white line is drawn as a guide for the line-arranged pits), (c) 14 days-degraded annealed Ca-Fe8 fibre and (d) 28 days-degraded annealed Ca-Fe8 fibre. The cracked base shows the conductive tape.

tangled structure, leading to an irregular distribution of structural flaws and resulting in a non-uniform water corrosion [72, 74]. In addition, the precipitate shells and the etch pits could coexist (see Figure 10d), which was also reported by Clément *et al.* [70].

In summary, an approximate linear increase in mechanical properties (tensile strength and modulus) and chemical durability were found with increasing  $\text{Fe}_2\text{O}_3$  content at the expense of CaO/MgO, which was suggested to be due to the formation of P-O-Fe bonds instead of P-O-P and the stronger cross-linking between phosphate chains. The annealing process was observed to increase the tensile modulus and chemical durability due to the loss of structural anisotropy and the relaxation of residual stresses in the glass fibres. Moreover, a reduction in tensile strength was suggested to be attributed to the relaxation of the compressive stresses, the interaction between the glass and moisture and the surface damage during annealing operation. These behaviours showed a good agreement with the

morphologic examination, which described the formation of precipitate shells and pitting corrosion during degradation.

## 5 Conclusions

The inclusion of  $\text{B}_2\text{O}_3$  made the fibre preparation continuous with no breakage for at least 3 hours for all five glass formulations. The composition with a higher  $\text{Fe}_2\text{O}_3$  content and MgO/CaO ratio was observed to require higher melting temperature and longer heating time to achieve suitable glass melts for fibre pulling. The addition of  $\text{Fe}_2\text{O}_3$  content revealed an increase in tensile strength of the fibres, whilst the annealing process decreased the strength. Chemical durability of the fibres was improved by both  $\text{Fe}_2\text{O}_3$  addition and the annealing treatment. The annealing process also impeded the precipitation reaction at the



fibre surface and revealed that pitting corrosion dominated the degradation behaviour of some PGFs.

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