

# Enantioselective Nickel-Catalyzed *anti*-Carbometallative Cyclizations of Alkynyl Electrophiles Enabled by Reversible Alkenylnickel *E/Z* Isomerization

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**S** Supporting Information

**ABSTRACT:** Nickel-catalyzed additions of arylboronic acids to alkynes, followed by enantioselective cyclizations of the alkenylnickel species onto tethered ketones or enones, are reported. These reactions are reliant upon the formal *anti*-carbometallation of the alkyne, which is postulated to occur by the reversible *E/Z* isomerization of an alkenylnickel species.

Transition-metal catalysis is utilized extensively in organic synthesis to prepare compounds required for numerous applications such as drug discovery, agrochemistry, and materials science. Despite the diversity of known transition-metal-catalyzed reactions, they all operate using combinations of only a few elementary organometallic reaction steps. One of these is the 1,2-insertion of a  $\pi$ -unsaturated component into a metal–element bond, which, owing to strict stereoelectronic requirements, is *syn*-selective. When applied to the migratory insertion of an alkyne with an organometallic species, this prevailing stereochemical principle (Figure 1A, top) provides a reliable

route to stereodefined alkenes, a feature that has been exploited in numerous synthetic methods.<sup>1</sup> Although *anti*-selective alkyne carbometalations (Figure 1A, bottom) would offer valuable new synthetic opportunities, such processes are uncommon.<sup>2</sup>

One strategy to obtain alkenylmetal species resulting from formal *anti*-carbometallation is to employ substrates with a heteroatom-containing substituent in close proximity to the alkyne. After *syn*-carbometallation, *E/Z* isomerization to the *anti*-isomer is driven by the formation of a more stable, chelated alkenylmetal species, in which the metal is coordinated to the heteroatom.<sup>1b,2a,b,d,h</sup> Although effective, the requirement for a nearby heteroatom-containing group limits broader applicability. Other reactions that provide *anti*-carbometallation products proceed via alternative mechanisms involving radical intermediates, but the substrates are restricted to terminal alkynes.<sup>2j–n,3</sup> In addition, with only a few exceptions,<sup>2h,1–n</sup> both of the aforementioned approaches require stoichiometric quantities of more reactive organometallics such as Grignard or organozinc reagents, which limit functional group compatibility and require rigorous exclusion of air and moisture. Furthermore, the integration of these processes into enantioselective reactions has not been reported. Given these issues, developing new platforms to access alkenylmetal species derived from *anti*-carbometallation of a wider range of alkynes, using organometallic reagents that exhibit high functional group tolerance, and in a manner that is amenable to asymmetric catalysis, would likely be of broad impact. Here, we describe enantioselective nickel-catalyzed domino alkyne carbometallation–cyclizations in which the products are derived from formal *anti*-carbometallation of the alkyne. The reactions use simple internal aryl-substituted alkynes and air- and moisture-stable arylboronic acids.

This study arose when we became interested in the enantioselective domino addition–cyclization of alkynyl electrophiles (Figure 1B), which are valuable complexity-generating transformations.<sup>4</sup> These reactions involve the *syn*-1,2-addition of an organometallic species across an alkyne to place the metal proximal to the electrophile (as in I), which enables cyclization to take place.<sup>5</sup> However, if *syn*-carbometallation proceeds to place the metal *distal* to the electrophile (as in II), cyclization is not possible due to geometric constraints (Figure 1C).<sup>6</sup> An *anti*-carbometallation of the alkyne would provide a solution to this problem, as now the resulting alkenylmetal species III possesses the correct geometry for cyclization (Figure 1C). However,

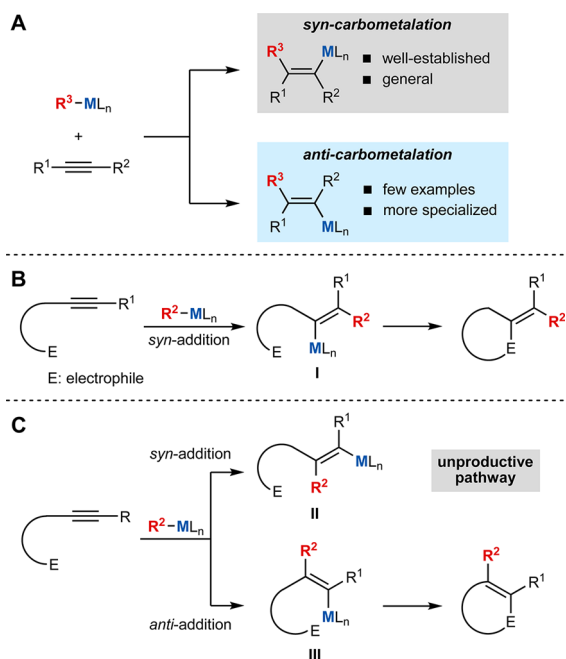


Figure 1. Carbometallation of alkynes and alkynyl electrophiles.

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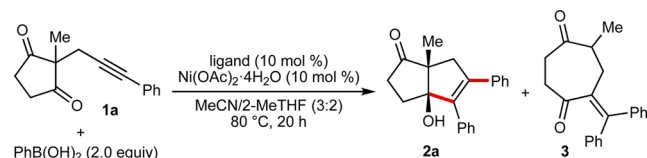
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overcoming the inherent *syn*-selectivity of alkyne migratory insertion is a considerable challenge.

A previous study has shown that the stoichiometric addition of organonickel complexes to simple alkynes that do not contain a nearby heteroatom-containing functional group can lead to alkenylnickel species resulting from formal *anti*-addition, although not always exclusively.<sup>2c</sup> A single example of the cobalt-catalyzed addition of an arylboronic acid to a symmetrical dialkyl alkyne to give a trisubstituted alkene as a 1:1 mixture of *E/Z* isomers has also been reported.<sup>2h</sup> In each case, the authors propose an initial *syn*-addition across the alkyne followed by reversible *E/Z* isomerization to give products of overall *anti*-addition.<sup>2c,h</sup> Despite the potential significance of these findings, more widespread applications to organic synthesis, such as addressing the gap in methodology outlined in Figure 1C, have yet to be reported.<sup>7</sup>

We began by reacting alkyne **1a** with  $\text{PhB(OH)}_2$  (2.0 equiv) in the presence of various cobalt and nickel salts, with or without added ligands. These experiments revealed that nickel catalysis gave promising results. For example,  $\text{Ni(OAc)}_2 \cdot 4\text{H}_2\text{O}$  (10 mol %) by itself was effective in promoting *anti*-arylmigratory cyclization at 80 °C in MeCN/2-MeTHF (3:2) to give 5,5-bicyclic **2a** in 39% NMR yield (Table 1, entry 1).<sup>8</sup> Arylyative

Table 1. Evaluation of Reaction Conditions<sup>a</sup>



Reaction conditions: ligand (10 mol %),  $\text{Ni(OAc)}_2 \cdot 4\text{H}_2\text{O}$  (10 mol %), MeCN/2-MeTHF (3:2), 80 °C, 20 h.

PhB(OH)<sub>2</sub> (2.0 equiv)

**Ligands**

- dppe (L1)
- xantphos (L2)
- pyphos (L3)
- (S)-PHOX: L4 R = *i*-Pr, L5 R = *t*-Bu, L6 R = Ph
- (S,S)-*t*-Bu-FOXAP (L7)

entry	ligand	yield of <b>2a</b> (%) <sup>b</sup>	yield of <b>3</b> (%) <sup>b</sup>	ee of <b>2a</b> (%) <sup>c</sup>
1	—	39	0	—
2	L1	64	30	—
3	L2	36	0	—
4	L3	87	7	—
5	L4	90	9	63
6	L5	56	4	93
7	L6	95	5	86
8	L7	33	3	97

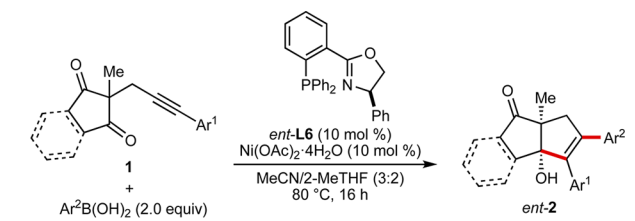
<sup>a</sup>Reactions employed 0.10 mmol of **1a**. <sup>b</sup>Determined by <sup>1</sup>H NMR analysis of the crude reactions, using 1,3,5-trimethoxybenzene as an internal standard. <sup>c</sup>Determined by chiral HPLC analysis.

cyclization via 1,4-metal migration, as we have described using iridium catalysis,<sup>9</sup> was not observed. Although the addition of dppe (L1) increased the NMR yield of **2a** to 64%, an appreciable quantity of the ring-expansion product **3** was also formed, which is derived from arylnickelation of the alkyne of **1a** with the regioselectivity opposite to that seen in the formation of **2a** (entry 2).<sup>9,10</sup> Xantphos (L2) was inferior to dppe (L1) (entry 3), but the *P,N*-ligand L3 was much more effective and gave **2a** in 87% NMR yield with only a small quantity of **3** (entry 4). Furthermore, by using commercially available chiral phosphino-oxazoline ligands<sup>11</sup> L4–L7, **2a** was obtained in 63–97% ee

(entries 5–8). Ligands L5 and L7 gave the highest enantioselectivities but the yields were modest (entries 6 and 8). (*S*)-Ph-Phox (L6) gave the best balance of yield and enantioselectivity (entry 7).

The scope of this process was then examined using (*R*)-Ph-Phox (*ent*-L6) as the ligand (Table 2). By variation of the

Table 2. Scope of Five-Membered Cyclic 1,3-Diketones<sup>a</sup>



Reaction conditions: *ent*-L6 (10 mol %),  $\text{Ni(OAc)}_2 \cdot 4\text{H}_2\text{O}$  (10 mol %), MeCN/2-MeTHF (3:2), 80 °C, 16 h.

Ar<sup>2</sup>B(OH)<sub>2</sub> (2.0 equiv)

**Arylboronic acid scope:**

- ent*-2a R = H, 89%, 86% ee
- ent*-2b R = OMe, 70%, 77% ee
- ent*-2c R = Cl, 74%, 85% ee
- ent*-2d R = Ac, 57%, 88% ee
- ent*-2e R = Me, 82%, 87% ee
- ent*-2f R = CN, 57%, 88% ee
- ent*-2g R = CO<sub>2</sub>Et, 56%, 90% ee
- ent*-2h 34%, 97% ee

**Alkynone scope:**

- ent*-2i 78%, 86% ee
- ent*-2j 75%, 66% ee
- ent*-2k R = OMe, 68%, 85% ee
- ent*-2l R = Cl, 45%, 87% ee
- ent*-2m 70%, 86% ee
- ent*-2n 70%, 42% ee

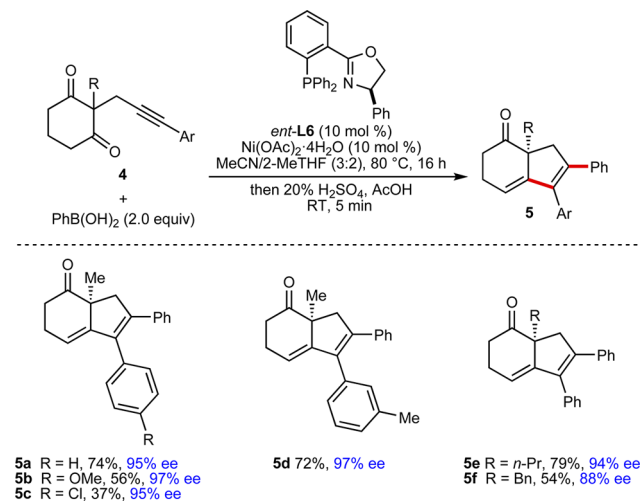
<sup>a</sup>Reactions employed 0.30 mmol of **1**. Yields are of isolated products. Enantiomeric excesses were determined by chiral HPLC analysis.

arylboronic acid and alkyne, products *ent*-2a–2n were obtained in 34–89% yield and 42–97% ee. The reaction is compatible with *para*- (*ent*-2a–2d), *meta*- (*ent*-2e–2g), and *ortho*-substituted (*ent*-2h) arylboronic acids, and the functional group tolerance is shown by examples with methoxy (*ent*-2b), halide (*ent*-2c and *ent*-2h), acetyl (*ent*-2d), nitrile (*ent*-2f), or carboethoxy (*ent*-2g) substituents. 2-Naphthylboronic acid (*ent*-2i) and a heteroarylboronic acid (*ent*-2j) are also tolerated. Variation of the alkynyl substituent to other aryl groups is possible (*ent*-2k–2m), but terminal alkynes provide complex mixtures of unidentified products. Use of an indan-1,3-dione-containing substrate gave *ent*-2n in 70% yield but in 42% ee. No reaction was observed using alkenylboronic acids or with substrates containing alkynes bearing methyl or trimethylsilyl groups in place of aryl groups.<sup>12</sup>

When the cyclic 1,3-diketone was changed to a cyclohexane-1,3-dione, significant dehydration of the initial cyclization products to give 1,3-dienes **5** was observed. The unpurified mixtures were therefore treated with 20% H<sub>2</sub>SO<sub>4</sub> in AcOH to drive dehydration to completion (Table 3). Using this procedure, substrates **4** with various substituents at the alkyne

and the 2-position of the cyclohexane-1,3-dione gave dienes **5a–5f** in 37–79% yield and 88–97% ee.

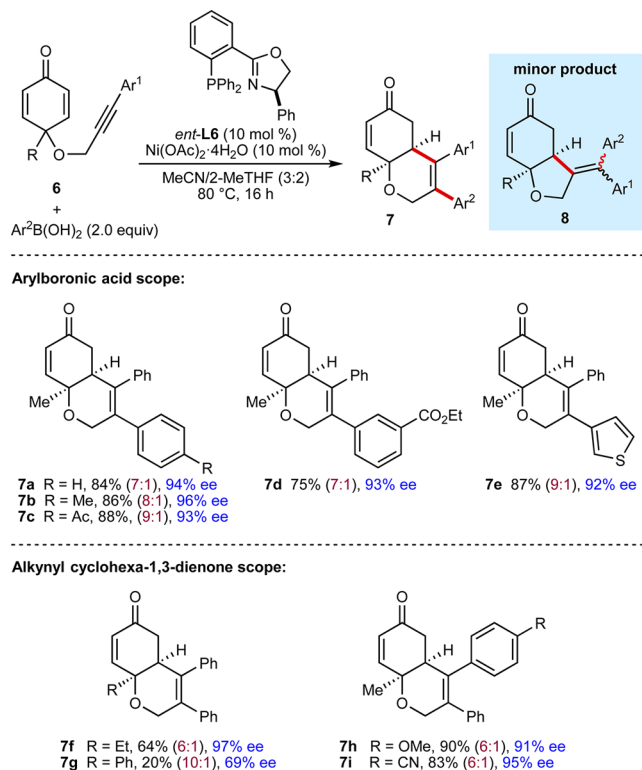
**Table 3. Scope of Six-Membered Cyclic 1,3-Diketones<sup>a</sup>**



<sup>a</sup>Reactions employed 0.30 mmol of **4**. Yields are of isolated products. Enantiomeric excesses were determined by chiral HPLC analysis.

Notably, this process is not limited to cyclic 1,3-diketones as the electrophilic trap, as substrates containing alkynes tethered to a cyclohexa-1,3-dienone are also highly effective (Table 4). For example, various arylboronic acids reacted with **6a** (R = Me, Ar<sup>1</sup> = Ph) to give 6,6-bicycles **7a–7i** in 20–90% yield and 69–97% ee.

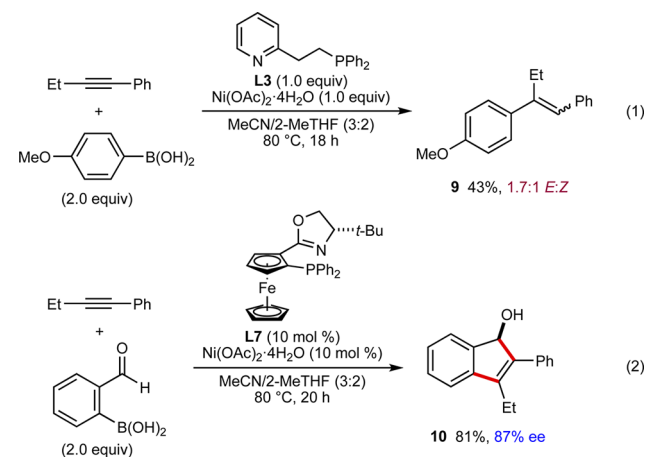
**Table 4. Scope of Alkynyl Cyclohexa-1,3-dienones<sup>a</sup>**



<sup>a</sup>Reactions employed 0.30 mmol of **6**. Yields are of isolated products. Ratios in parentheses refer to the ratio of **7**:**8** as determined by <sup>1</sup>H NMR analysis after purification. Enantiomeric excesses are of the major products **7a–7i** as determined by chiral HPLC analysis.

ee. These products were isolated together with small quantities of minor products **8**, which result from arylnickelation of the alkyne with the regioselectivity opposite to that seen in the formation of the major products **7**. Changing the substituent at the quaternary center (**7f** and **7g**) or the alkyne (**7h** and **7i**) is also possible.

To gain insight into the mechanism of these reactions, the reaction of 1-phenyl-1-butyne with 4-methoxyphenylboronic acid in the presence of 1.0 equiv each of Ni(OAc)<sub>2</sub>·4H<sub>2</sub>O and **L3** was conducted (eq 1). This experiment gave trisubstituted alkene

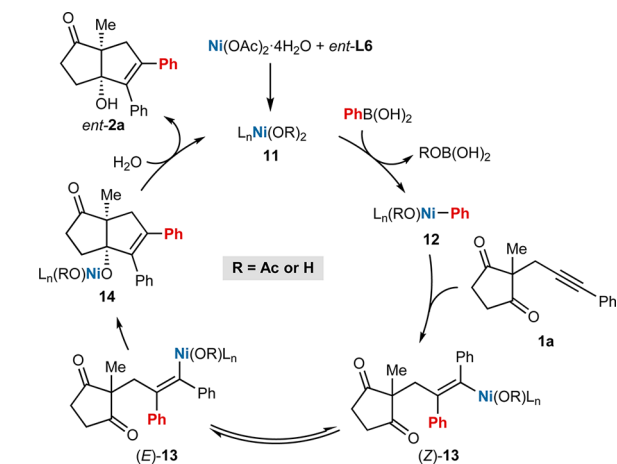


**9** as a 1.7:1 mixture of *E*- and *Z*-isomers, respectively, along with traces of isomers derived from arylnickelation with the opposite regioselectivity. As well as demonstrating the ability of nickel catalysis to provide *anti*-carbometalation products from alkynes that do not contain a tethered electrophile, this experiment suggests that the domino reactions described herein proceed through a mechanism similar to that shown in Figure 1C, rather than one involving initial “*anti*-Wacker-type” additions.<sup>6</sup> The reaction of 1-phenyl-1-butyne with 2-formylphenylboronic acid using **L7** as the chiral ligand (see Table 1, entry 8 for the use of **L7** in *anti*-carbometalative cyclization) gave indene **10** in 81% yield and 87% ee (eq 2), which is a product of *syn*-arylnickelation of the alkyne followed by cyclization of the resulting alkenylnickel species onto the aldehyde.<sup>13</sup> The ability to obtain enantio-enriched products from either *syn*- or *anti*-carbometalative cyclization illustrates the adaptive power of this nickel-based catalytic system and further suggests that reversible *E/Z* isomerization of alkenylnickel species is operative.

On the basis of these experiments, we propose a catalytic cycle for a representative example of *anti*-carbometalative cyclization, using alkynone **1a** with PhB(OH)<sub>2</sub> (Scheme 1). Transmetalation of PhB(OH)<sub>2</sub> with the nickel complex **11** formed from Ni(OAc)<sub>2</sub>·4H<sub>2</sub>O and *ent*-**L6** gives an arylnickel species **12**. *Syn*-arylnickelation of **1a** with **12** gives the alkenylnickel intermediate (*Z*)-**13**, which can undergo reversible *E/Z* isomerization,<sup>2c,14</sup> to give a new alkenylnickel species (*E*)-**13**. Cyclization of (*E*)-**13** onto one of the ketones gives nickel alkoxide **14**, which can undergo protonolysis with water to release the product **2a** and regenerate nickel complex **11**. Presumably, cyclization drives the equilibrium between (*Z*)-**13** and (*E*)-**13** toward the *E*-isomer. We assume a similar pathway is operative in the cyclizations of alkynyl cyclohexa-1,3-dienones (Table 4), with protonation of the nickel enolate formed by intramolecular 1,4-addition of the intermediate alkenylnickel species being the likely catalyst turnover step.

In conclusion, we have established that products of formal *anti*-carbometalation of alkynes can be obtained by exploiting the

## Scheme 1. Postulated Catalytic Cycle



ability of alkenylnickel intermediates to undergo reversible *E/Z* isomerization and have demonstrated the utility of this method in promoting previously inaccessible domino reactions with high enantioselectivities. We expect this underexplored mode of reactivity may have significant untapped potential, and its creative use in other contexts may lead to further new applications in organic synthesis.<sup>7</sup>

## ■ ASSOCIATED CONTENT

### Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/jacs.6b04206.

Experimental procedures and full spectroscopic data for all new compounds (PDF)

Crystallographic data (CIF)

Crystallographic data (CIF)

Crystallographic data (CIF)

Crystallographic data (CIF)

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### Notes

The authors declare no competing financial interest.

## ■ ACKNOWLEDGMENTS

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