ChemComm

COMMUNICATION

Enantioselective nickel-catalyzed arylative intramolecular 1,4allylations[†]

wReceived 00th January 20xx, Accepted 00th January 20xx

Thi Le Nhon Nguyen, ‡^{ab} Celia A. Incerti-Pradillos, ‡^{ab} William Lewis^b and Hon Wai Lam^{*ab}

DOI: 10.1039/x0xx00000x

www.rsc.org/

The enantioselective nickel-catalyzed desymmetrization of allenyl cyclohexa-2,5-dienones by reaction with arylboronic acids is described. Nickel-catalyzed arylation of the allene gives allylnickel species, which undergo cyclization by 1,4-allylation to produce hexahydroindol-5-ones and hexahydrofuran-5-ones with three contiguous stereocenters in high diastereo- and enantioselectivities.

Catalytic enantioselective 1,4-additions of organometallic reagents to electron-deficient alkenes are important reactions for the formation of new carbon–carbon bonds.¹ Despite the tremendous diversity of known examples of such reactions,¹ the catalytic enantioselective 1,4-addition of allylic nucleophiles remains considerably underdeveloped and only a handful of examples have been described.² Therefore, new types of catalytic enantioselective 1,4-allylations are highly desired to increase the range of accessible products.

In light of our discovery that chiral phosphinooxazolinenickel complexes are highly effective in promoting enantioselective anti-carbometallative cyclizations of alkynyl electrophiles with arylboronic acids,³ we questioned whether these complexes could promote a catalytic enantioselective intramolecular 1,4-allylation^{2g,i} of substrates containing an allene tethered to an electron-deficient alkene. Specifically, nickel-catalyzed addition of an arylboronic acid to an allenyl cyclohexa-2,5-dienone could provide an allylnickel species A (which could interconvert with σ -allyl isomers, Scheme 1). An enantioselective intramolecular 1,4-allylation would then likely give nickel enolate **B**, which upon protonolysis would release the Ni(II) catalyst and a cis-fused hexahydroindol-5-one or hexahydrobenzo-furan-5-one,4,5,6,7 are which important structures that appear in several natural products such as runanine,⁸ acutumine,⁹ millingtonine,¹⁰ and cryptocaryone¹¹ (Figure 1). Furthermore, the enone in these structures serves as a versatile handle for further manipulations to give octahydroindole and octahydrobenzofuran derivatives, which are present in many other natural products.



Scheme 1 Proposed arylative cyclization of allenyl cyclohexa-2,5-dienones



Figure 1 Hexahydrobenzofuran-5-one and hexahydroindol-5-one natural products

To our knowledge, however, there is only limited precedent for nickel-catalyzed 1,4-allylations, with only one reported study by Sieber and Morken.^{2b} Also relevant is work by Montgomery and co-workers, who described diastereoselective nickel-catalyzed alkylative cyclizations of allenes tethered to α , β -unsaturated carbonyl compounds that give products of formal 1,4-allylation.¹² Herein, we describe the successful implementation of the strategy shown in Scheme 1, which gives



^{a.} The GlaxoSmithKline Carbon Neutral Laboratories for Sustainable Chemistry, University of Nottingham, Jubilee Campus, Triumph Road, Nottingham, NG7 2TU, United Kingdom

^{b.} School of Chemistry, University of Nottingham, University Park, Nottingham, NG7 2RD, United Kingdom

⁺ Electronic Supplementary Information (ESI) available: Experimental procedures, full spectroscopic data for new compounds, and crystallographic data for **2a–2d**, **2g**, **2h**, and **3a**. See DOI: 10.1039/x0xx00000x

[‡] These authors contributed equally.

Communication

Table 1 Evaluation of reaction conditions^a





^{*a*} Reactions were conducted with 0.10 mmol of **1a** at 0.1 M concentration. ^{*b*} Determined by ¹H NMR analysis of the crude reactions using 1,3,5-dimethoxybenzene as the internal standard. ^{*c*} Determined by chiral HPLC analysis. ^{*d*} The major product was the enantiomer of **2a**. ^{*e*} The major product was the diastereomer of **2a**, obtained in 69% ee.

hexahydroindol-5-ones and hexahydrofuran-5-ones with three contiguous stereocenters in high diastereo- and enantioselectivities.

This study began with the reaction of allenyl cyclohexa-2,5dienone 1a with PhB(OH)₂ (2.0 equiv) and Ni(OAc)₂·4H₂O (10 mol%) in MeCN/dioxane (3:2) at 80 °C for 18 h (Table 1, entry 1). We were pleased to observe that 6,5-bicycle 2a was obtained in 35% ¹H NMR yield as a 2.0:1 ratio of diastereomers.¹³ However, this reaction also gave cyclobutane 3a in 11% ¹H NMR yield, resulting from a [2+2] cycloaddition.¹³ Heating 1a in MeCN/dioxane (3:2) at 80 °C for 24 h in the absence of PhB(OH)2 and Ni(OAc)2·4H2O also gave 3a in 44% NMR yield, which demonstrates this [2+2] cycloaddition is a thermally promoted process rather than a nickelcatalyzed reaction (see the Supplementary Information for details). The addition of 10 mol% of the P,N-ligand L1 increased the yield of 2a to 82% (entry 2). Furthermore, chiral phosphinooxazoline (PHOX) ligands L2-L5 successfully gave enantioenriched products, though with varying levels of efficiency (entries 2-6). Of these, (R)-Ph-PHOX (L2) provided 2a in the best balance between yield, diastereoselectivity, and enantioselectivity (entry 3), and this ligand was selected for further experiments.

The scope of the process with respect to the allenyl substrate was examined in reactions with PhB(OH)₂ (Table 2). Both *N*-sulfonyl-tethered and *O*-tethered substrates reacted successfully to give 6,5-bicycles **2a**–**2b** and **6a–6d**, respectively, in good yields (77–87%) and high enantioselectivities (90–94% *ee*). In all cases, small quantities of cyclobutanes (either **3a** or analogous to **3a**) were also formed, but with the exception of the reactions forming **6a**, **6b**, and **6d**, these were separable by column chromatography. The substituent at the quaternary center of the substrates can be changed from methyl (**2a** and **6a**) to ethyl (**2b** and **6b**), isopropyl (**6c**), or phenyl groups (**6d**). By increasing the reaction concentration, the

 Table 2
 Substrate evaluation^a



^{*a*} Reactions were conducted with 0.30 mmol of **1**, **4**, or **5** at 0.1 M concentration. Yields are of isolated products. Enantiomeric excesses were determined by chiral HPLC analysis. ^{*b*} Products **6a**, **6b**, and **6d** were isolated together with cyclobutane side products (see the Supplementary Information for details) as inseparable mixtures in ratios of between 11:1 and 20:1. Yields have been adjusted accordingly. ^cValues in parentheses are of a reaction conducted with 2.00 mmol of **4c** at 0.4 M concentration for 42 h, using 5 mol% each of Ni(OAc)₂·4H₂O and **L2**. ^{*d*} Conducted with 0.15 mmol of **4d**.

Table 3 Boronic acid evaluation^a



^{*a*} Reactions were conducted with 0.30 mmol of **1a** or **4a** at 0.1 M concentration. Yields are of isolated products. Enantiomeric excesses were determined by chiral HPLC analysis. ^{*b*} Isolated together with cyclobutane **3b** (see the Supplementary Information for the structure) in a 11:1 ratio. The yield of **6g** has been adjusted accordingly.

2 | J. Name., 2012, 00, 1-3

Chemical Science

catalyst loading can be reduced. For example, **6c** was obtained in 63% yield and 94% ee from a 2.00 mmol scale reaction conducted at 0.4 M concentration, using a 5 mol% catalyst loading. Allenyl substrate **5**, containing a longer tether between the allene and the cyclohexa-2,5-dienone, was largely unreactive and only a trace (<5%) of 6,6-bicycle **7** was observed, with the mass balance being mostly starting material.

The scope of the boronic acid was then examined using substrates **1a** and **4a**, which gave products **2c–2h** and **6e–6i** in 51–81% yield with generally high enantioselectivities (Table 3). The reaction is compatible with a variety of *para-* (**2c**, **6e**, and **6f**), *meta-* (**6g**), *ortho-* (**6h**), and disubstituted (**2d–2f**) arylboronic acids, and the functional group tolerance is shown by examples with vinyl (**2c**), halide (**2d**, **2e**, **6f** and **6h**), methoxy (**2f**), and acetoxy (**6e**) substituents. 2-Naphthylboronic acid (**2g**) and heteroarylboronic acids (**2h** and **6i**) are also tolerated.

 Table 4
 Reactions producing phenols 8^a



^{*a*} Reactions were conducted with 0.30 mmol of **4a**. Yields are of isolated products. Enantiomeric excesses were determined by chiral HPLC analysis.

Interestingly, the reaction of 4a with certain arylboronic acids gave 3,4-disubstituted phenols 8 in addition to the expected 6,5bicycles (Table 4). In each case, both products were formed with high enantioselectivities (89-94% ee). Most of these arylboronic acids contain strongly electron-withdrawing substituents such as 4acetyl (entry 1), 3-cyano (entry 3), or 3-carboethoxy groups (entry 4), but 4-trimethylsilylphenylboronic acid also resulted in phenol formation (entry 2). Phenols 8 might be formed by enolization of the ketone of 6 to give 9, followed by ring-opening of the furan ring, which presumably is promoted by a Brønsted acid or hydrogen bond donor (Scheme 2). Proton loss from 10 then gives 8.14 The fact that phenols 8 are observed in appreciable quantities only when certain arylboronic acids are used may indicate that the boronic acid serves as a hydrogen bond donor to promote ring-opening.¹⁵ Confirmation of the ability of acid to promote the formation of phenols 8 was provided by heating 61 with TsOH·H2O (0.5 equiv) at 80 °C, which gave 81 in 65% yield (eqn (1)).



Scheme 2 Rationale for the formation of phenols 8



To demonstrate the synthetic utility of the products, further transformations were conducted (eqn (2) and (3)). Reaction of racemic **2a** with Me₃Al in the presence of Ni(acac)₂ (10 mol%) gave **11** in 76% yield as a single observable diastereomer (>19:1 d.r.), resulting from 1,4-addition to the less hindered convex face [Eq. (2)]. Furthermore, a Luche reduction of **2e** gave allylic alcohol **12** in >99% yield, also as a single observable diastereomer (eqn (3)).



In summary, we have reported the enantioselective nickelcatalyzed desymmetrization of allenyl cyclohexa-2,5-dienones by reaction with arylboronic acids. These domino addition–cyclizations add to the currently limited body of work in catalytic enantioselective intramolecular 1,4-allylations,^{2g,i} and provide hexahydroindol-5-ones and hexahydrofuran-5-ones with three contiguous stereocenters in high diastereo- and enantioselectivities. Further enantioselective nucleophilic allylations triggered by nickelcatalyzed migratory insertions of allenes are under investigation in our group.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This work was supported by the University of Nottingham and GlaxoSmithKline.

Notes and references

 For representative reviews, see: (a) in Catalytic Asymmetric Conjugate Reactions, ed. A. Cordova, Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim, 2010, p. 447 pp. (b) S. R. Harutyunyan, T. den Hartog, K. Geurts, A. J. Minnaard and B. L. Feringa, Chem. Rev., 2008, 108, 2824-2852. (c) A. Alexakis, J. E. Bäckvall, N. Krause, O. Pàmies and M. Diéguez, Chem. Rev., 2008, 108, 2796-2823. (d) J. Christoffers, G. Koripelly, A. Rosiak and M. Rössle, *Synthesis*, 2007, 1279-1300. (e) T. Hayashi and K. Yamasaki, *Chem. Rev.*, 2003, **103**, 2829-2844. (f) N. Krause and A. Hoffmann-Röder, *Synthesis*, 2001, 171-196.

- (a) J. D. Sieber, S. Liu and J. P. Morken, J. Am. Chem. Soc., 2007, 129, 2214-2215. (b) J. D. Sieber and J. P. Morken, J. Am. Chem. Soc., 2008, 130, 4978-4983. (c) M. Shizuka and M. L. Snapper, Angew. Chem., Int. Ed., 2008, 47, 5049-5051. (d) Y. Kuang, X. Liu, L. Chang, M. Wang, L. Lin and X. Feng, Org. Lett., 2011, 13, 3814-3817. (e) Y. Yanagida, R. Yazaki, N. Kumagai and M. Shibasaki, Angew. Chem., Int. Ed., 2011, 50, 7910-7914. (f) N. R. Ball-Jones, J. J. Badillo, N. T. Tran and A. K. Franz, Angew. Chem., Int. Ed., 2014, 53, 9462-9465. (g) Z.-T. He, X.-Q. Tang, L.-B. Xie, M. Cheng, P. Tian and G.-Q. Lin, Angew. Chem., Int. Ed., 2015, 54, 14815-14818. (h) X. Li, F. Meng, S. Torker, Y. Shi and A. H. Hoveyda, Angew. Chem., Int. Ed., 2016, 55, 9997-10002. (i) Y.-X. Tan, X.-Q. Tang, P. Liu, D.-S. Kong, Y.-L. Chen, P. Tian and G.-Q. Lin, Org. Lett., 2018, 20, 248-251.
- 3 (a) C. Clarke, C. A. Incerti-Pradillos and H. W. Lam, J. Am. Chem. Soc., 2016, 138, 8068-8071. (b) C. Yap, G. M. J. Lenagh-Snow, S. N. Karad, W. Lewis, L. J. Diorazio and H. W. Lam, Angew. Chem., Int. Ed., 2017, 56, 8216-8220.
- 4 For selected examples of catalytic enantioselective domino additioncyclization reactions involving allenes, see ref. 2g and: (a) X. Yu and X. Lu, Org. Lett., 2009, 11, 4366-4369. (b) X.-X. Guo, T. Sawano, T. Nishimura and T. Hayashi, Tetrahedron: Asymmetry, 2010, 21, 1730-1736. (c) D. N. Tran and N. Cramer, Angew. Chem., Int. Ed., 2010, 49, 8181-8184. (d) D. N. Tran, N. Cramer, Angew. Chem., Int. Ed. 2013, 52, 10630-10634. (e) X. Zhang, X. Han and X. Lu, Org. Lett., 2015, 17, 3910-3913.
- 5 For reviews of asymmetric reactions of 2,5-cyclohexadienones, see: (a) G. Maertens, M.-A. Ménard and S. Canesi, *Synthesis*, 2014, 46, 1573-1582. (b) K. A. Kalstabakken and A. M. Harned, *Tetrahedron*, 2014, 70, 9571-9585.
- 6 For organocatalytic enantioselective desymmetrization of cyclohexa-2,5-dienones by intramolecular conjugate addition of an N- or O-tethered nucleophile, see: (a) Q. Liu and T. Rovis, J. Am. Chem. Soc., 2006, 128, 2552-2553. (b) R. Leon, A. Jawalekar, T. Redert and M. J. Gaunt, Chem. Sci., 2011, 2, 1487-1490. (c) R. Tello-Aburto, K. A. Kalstabakken, K. A. Volp and A. M. Harned, Org. Biomol. Chem., 2011, 9, 7849-7859. (d) M.-Q. Jia and S.-L. You, Chem. Commun., 2012, 48, 6363-6365. (e) S. Takizawa, T. M.-N. Nguyen, A. Grossmann, D. Enders and H. Sasai, Angew. Chem., Int. Ed., 2012, 51, 5423-5426. (f) S. Takizawa, T. M.-N. Nguyen, A. Grossmann, M. Suzuki, D. Enders and H. Sasai, Tetrahedron, 2013, 69, 1202-1209. (g) M. T. Corbett and J. S. Johnson, Chem. Sci., 2013, 4, 2828-2832. (h) L. Pantaine, V. Coeffard, X. Moreau and C. Greck, Org. Lett., 2015, 17, 3674-3677. (i) S. Takizawa, K. Kishi, Y. Yoshida, S. Mader, F. A. Arteaga, S. Lee, M. Hoshino, M. Rueping, M. Fujita and H. Sasai, Angew. Chem., Int. Ed., 2015, 54, 15511-15515. (j) W. Yao, X. Dou, S. Wen, J. e. Wu, J. J. Vittal and Y. Lu, Nat. Commun., 2016, 7, 13024.
- For transition-metal-catalyzed desymmetrization of cyclohexa-2,5-dienones by intramolecular conjugate addition of an N- or O-tethered nucleophile, see ref. 2g, 2i, and: (a) J. Keilitz, S. G. Newman and M. Lautens, Org. Lett., 2013, 15, 1148-1151.(b) Z.-T. He, B. Tian, Y. Fukui, X. Tong, P. Tian and G.-Q. Lin, Angew. Chem., Int. Ed., 2013, 52, 5314-5318. (c) R. Tello-Aburto, K. A. Kalstabakken and A. M. Harned, Org. Biomol. Chem., 2013, 11, 5596-5604. (d) P. Liu, Y. Fukui, P. Tian, Z.-T. He, C.-Y. Sun, N.-Y. Wu and G.-Q. Lin, J. Am. Chem. Soc., 2013, 135, 11700-11703. (e) K. Takenaka, S. C. Mohanta and H. Sasai, Angew. Chem., Int. Ed., 2014, 53, 4675-4679. (f) R. Kumar, Y. Hoshimoto, E. Tamai, M. Ohashi and S. Ogoshi, Nat. Commun., 2017, 8, 32.
- 8 M. Zhi-Da, L. Ge, X. Guang-Xi, M. Iinuma, T. Tanaka and M. Mizuno, *Phytochemistry*, 1985, 24, 3084-3085.
- 9 (a) G. Kakuji and S. Hideo, *Bull. Chem. Soc. Jpn.*, 1929, 4, 220-224.
 (b) M. Tomita, Y. Okamoto, T. Kikuchi, K. Osaki, M. Nishikawa, K. Kamiya, Y. Sasaki, K. Matoba and K. Goto, *Tetrahedron Lett.*,

1967, **8**, 2421-2424. (c) M. Tomita, Y. Okamoto, T. Kikuchi, K. Osaki, M. Nishikawa, K. Kamiya, Y. Sasaki, K. Matoba and K. Goto, *Tetrahedron Lett.*, 1967, **8**, 2425-2430.

- 10 T. Hase, K. Ohtani, R. Kasai, K. Yamasaki and C. Picheansoonthon, *Phytochemistry*, 1996, **41**, 317-321.
- (a) V. Dumontet, C. Gaspard, N. Van Hung, J. Fahy, L. Tchertanov, T. Sévenet and F. Guéritte, *Tetrahedron*, 2001, 57, 6189-6196.
- (a) M. V. Chevliakov and J. Montgomery, J. Am. Chem. Soc., 1999, 121, 11139-11143. (b) A. S. ElDouhaibi, R. M. Kassab, M. Song and J. Montgomery, Chem. Eur. J., 2011, 17, 6326-6329.
- 13 The relative and absolute configurations of **2a–2d**, **2g**, **2h**, and **3a** were determined by X-ray crystallography, and those of the remaining products were assigned by analogy. CCDC 1584893–1584899 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre.
- 14 D. L. J. Clive, S. P. Fletcher and D. Liu, J. Org. Chem., 2004, 69, 3282-3293.
- 15 For examples of arylboronic acids serving as hydrogen bond donor catalysts, see: J. Wang and Y. Zhang, ACS Catal., 2016, 6, 4871-4876.