NUMERICAL STUDIES OF SHEAR-THINNING DROPLET FORMATION IN A MICROFLUIDIC T-JUNCTION USING TWO-PHASE LEVEL-SET METHOD

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ABSTRACT

A conservative level-set method (LSM) embedded in a computational fluid dynamics 14 (CFD) simulation provides a useful approach for the studying the physics and 15 mechanism in two-phase flow. Detailed two-dimensional (2D) 16 underlvina computational microfluidics flow simulations have been carried out to examine 17 systematically the influence of different controlling parameters such as flow rates, 18 19 viscosities, surface wettability, and interfacial tensions between two immiscible fluids 20 on the non-Newtonian shear-thinning microdroplets generation process. For the twophase flow system that neglects the Marangoni effect, the breakup process of shear-21 thinning microdroplets in cross-flowing immiscible liquids in a microfluidic device with 22 a T-shaped geometry was predicted. Data for the rheological and physical properties 23 of fluids obeying Carreau-Yasuda stress model were empirically obtained to support 24 the computational work. The simulation results show that the relevant control 25 parameters mentioned above have a strong impact on the size of shear-thinning 26 droplets generated. Present computational studies on the role and relative 27 28 importance of controlling parameters can be established as a conceptual framework

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of the non-Newtonian droplet generation process and relevant phenomena for future
 studies.

Keywords: Carreau-Yasuda; microfluidics; fluid dynamics; level-set simulation;
 emulsification.

33

34 **1. Introduction**

Microfluidic applications are commonly found in coating flows, formation of 35 suspensions, emulsions and foams, heat transfer and fluid flow (Stone et al., 2004; 36 Squires and Quake, 2005; Pipe and McKinley, 2009). Since the start of the micro total 37 analysis system (µTAS), the manipulation of droplets in confined microfluidic 38 channels has been highlighted as a useful technology platform for a broad range of 39 application in many scientific fields such as biology, biomedical studies, chemical and 40 synthesis, and drug delivery. Many of these fields involve handling non-Newtonian 41 fluids characterised by a nonlinear relationship between shear stress and the 42 deformation rate. Previously, most of the studies of droplet formation in 43 microchannels have employed traditional Newtonian-Newtonian systems (Cramer et 44 al., 2004; Garstecki et al., 2006; Anna and Mayer, 2006; Utada et al., 2007, Xu et al., 45 2008), and little attention has been given to non-Newtonian fluids. However, it has 46 been identified that non-Newtonian microfluidic applications are increasingly taking 47 the centre ground of scientific and commercial interests, demanding research and 48 investigation (Stone et al., 2004). 49

Diverse forces, including surface tension and viscous forces, govern drop dynamics, 50 and the rheology of the fluid can be important when non-Newtonian fluids are involved 51 (Chhabra and Richardson, 2008). As a general rule, non-Newtonian fluids can be 52 classified into three groups, namely purely viscous fluids, time dependent fluids, and 53 viscoelastic fluids (Chung et al., 2008). Each of the above mentioned non-Newtonian 54 fluid possesses distinct characteristics of their own; however, no single constitutive 55 equation has been established to describe the rheogram of all such fluids. 56 57 Understanding the dynamical mechanisms of monodispersed micro or nanodroplets formation of non-Newtonian fluids in microfluidic channels is essential to ensure that 58 59 specific requirements such as, drop size, patterns and productivity are met.

Previously, some researchers have focused on the analysis of the dynamics and deformation of droplets with viscoelastic properties in this area (Husny and Cooper-White 2006; Steinhaus *et al.*, 2007; Arratia *et al.*, 2009; Qiu *et al.*, 2010), yet a thorough understanding of relevant physics behind their creation for other non-Newtonian fluid types is still limited.

Near monodispersed droplets can be generated by a number of methods in 65 microfluidic devices including breakup in co-flowing streams (Cramer et al., 2004, 66 Utada et al., 2007) breakup in cross-flowing stream (Garstecki et al., 2006; Xu et al., 67 2008; Husny and Cooper-White 2006; Qiu et al., 2010), breakup in elongational 68 strained flows (Anna and Mayer, 2006), and breakup in microchannel emulsification 69 (MCE) (Sugiura et al., 2001; Kobayashi et al., 2002; Kobayashi et al., 2005; Van Der 70 Zwan et al., 2009). The T-junction microchannel is one of the most frequently used 71 72 microfluidic configurations based on the physical mechanism of droplet breakup in a cross-flowing stream. According to this mechanism, viscous shear-stress from the 73 74 continuous stream of the horizontal channel overcomes surface tension at the liquidliquid interface and pulls off droplets of the dispersed phase entering from the vertical 75 channel. Cross-flow shear in a T-junction microchannel remains one of the most 76 utilised microfluidic configurations, owing to the ease of droplet formation, inherent 77 droplet size controllability, periodicity of droplet formation, and design simplicity 78 (Thorsen et al., 2001; Nisisako et al., 2002; Tice et al., 2004; Garstecki et al., 2005). 79 Those investigations of the emulsification process for non-Newtonian-Newtonian 80 systems found in the literature have been performed using T-shaped microchannels. 81 For instance, Husny and Cooper-White (2006) investigated polymeric droplet creation 82 and the breakup dynamics of low viscosity, elastic polyethylene oxide (PEO) fluids 83 and Newtonian fluids by cross-flow shear. Gu and Liow (2011) investigated the 84 formation of shear-thinning xanthan gum droplets in oil in a T-junction microchannel 85 under different flow conditions. Apart from empirical works, Sang et al. (2009) 86 87 developed both numerical and analytical models with an interface capturing scheme to explore the influence of the material behaviour of Newtonian, Bingham and power-88 law fluid on the formation of droplets. Qiu et al. (2010) also performed a numerical 89 study of the formation of Newtonian droplets in a continuous phase with shear-90 thinning properties under the cross-flow condition using the volume-of-fluid (VOF) 91 formulation. Most of the previous studies, especially the numerical ones, have 92

focused on the Newtonian droplets formed in the non-Newtonian phase; however,
there have been far fewer attempts to develop a predictive model for the relevant
physics, evolution and formation of non-Newtonian droplets in a T-junction
microfluidic cell.

Due to the existence of a high surface-to-volume ratio within a micro-scaled device, 97 interfaces between fluids play a significant role in many physical processes such as 98 droplets collision, coalescence and breakup (Christopher et al., 2009). There are two 99 main approaches to represent the interface evolution problem either explicitly or 100 101 implicitly within the Navier-Stokes (NS) equation discretized on a fixed grid: interface tracking and interface capturing. These methods are widely used in computation of 102 103 flow problems with moving boundaries and interfaces. Interface tracking of the moving boundary is based on an explicit description of the interface by the 104 105 computational mesh. Boundary integral (Hou et al., 2001) and front tracking (Tryggvason et al., 2001) methods are examples of interface tracking methods. 106 107 Interface tracking approaches provide higher numerical accuracy, but their applicability is limited to Stokes flow. In contrast to interface tracking methods, 108 109 interface motion can be simply obtained by the advection of the corresponding phase function in the interface capturing method. The interface capturing method uses a 110 separate phase function, discretised on the fixed grid, to implicitly represent the 111 interface (Bonometti and Magnaudet, 2007). In implicit approaches, there are three 112 common interface capturing methods that have been developed to solve 113 incompressible two-phase problems. These are the volume-of-fluid (VOF) (Rider and 114 Kothe, 1998), lattice-Boltzmann method (LBM) (Takada et al., 2000), and level-set 115 method (LSM) (Osher and Sethian, 1988). In each of the above reference methods, 116 the main challenge is to accurately advect the phase boundary and correctly compute 117 terms concentrated at the interface, such as interfacial tension. The functional 118 discontinuities of the VOF method present at the interface lead to low accuracy in the 119 120 computation of the mean curvature, causing errors in the surface tension force. Apart from the VOF method, the variation of interfacial tension has made difficulties in the 121 application of LBM to this problem as this parameter cannot be adjusted 122 independently. However, the above mentioned limitations can be overcome by LSM 123 as it allows any interfacial tension values to be chosen. The level-set method is a 124 simple and versatile method for tracking moving interfaces and shapes. It permits 125

numerical computations of such objects involving curves and surface to be performed
on a fixed Cartesian grid without having to parameterise them. It is a robust scheme
that is relatively easy to implement (Osher and Sethian, 1988). Although the main
drawback of classical LS description is the difficulty of ensuring volume conserving
advection of fluid phases, a conservative LSM has been devised to overcome this
(Olsson and Kreiss, 2005; Sethian, 1999).

This paper describes a numerical study carried out to develop a predictive model for 132 droplet generation of a non-Newtonian fluid at a microfluidic T junction. A conservative 133 134 LS approach coupled with the Carreau-Yasuda stress model has been adopted in the present computational model. Systematic sets of numerical simulations were carried 135 out to monitor the evolution of droplet size, which turns out to be effectively governed 136 by the physical properties of the tested fluids, flow rates, and viscosity. These key 137 138 parameters that control the formation of microdroplets of shear-thinning liquids such as sodium carboxymethylcellulose (SCMC) have not previously been examined in 139 140 detail. In the present study, the continuous phase is a Newtonian fluid with a constant viscosity, whereas the dispersed phase has properties equivalent to shear-thinning 141 142 SCMC solutions with concentration ranging from 0 wt% to 1.20 wt%. This provides a realistic means of computing the dynamics of droplet breakup process in a T-junction 143 micro-scaled device when the shear-thinning non-Newtonian effects are present. It 144 should also be noted that within the considered weight percentages, the variation in 145 density is minimal enabling the use of a constant value of 998.2 kg/m³. Furthermore, 146 the surface tension between the lowest and the highest concentrations are 22.14 and 147 18.87 mN/m. Therefore a constant value of 20.74 mN/m is used for all the cases 148 except for the parametric study on the interfacial tension. Both the concentration and 149 150 the contact angle were treated as independent parameters in this study.

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152 **2. Computational Model**

153 2.1 The Conservative Level-Set Method

The conservative level-set method (LSM) was employed to simulate the filament development and the detachment of the droplets at various operating conditions and fluid properties inside a microchannel T-junction. The conservative LSM proposed by

Olsson and Kreiss (2005) and Olsson et al. (2007) was employed in the present study. 157 The LSM relies on an implicit representation of the interface and the equation of 158 motion is numerically approximated using schemes built from hyperbolic-159 conservatives law (Osher and Sethian, 1988). This approach has been used to 160 describe the interface between two immiscible fluids which is defined by the 0.5 161 contour of the level set (phase) function (ϕ) assuming the continuous phase ($\phi = 0$) 162 flowing in the main horizontal channel and the dispersed phase ($\phi = 1$) fills a channel 163 orthogonal to the main channel. In this implicit method, the model comprises of the 164 165 following governing equations.

166 Incompressible Navier-Stokes equation

167
$$\rho \frac{\partial \mathbf{u}}{\partial t} + \rho(\mathbf{u} \cdot \nabla)\mathbf{u} = \nabla \left[-pI + \eta(\nabla \mathbf{u} + (\nabla \mathbf{u})^T) + F_{st}\right]$$
(1)

168 Continuity equation

$$169 \quad \nabla \cdot \mathbf{u} = 0 \tag{2}$$

170 Level-set equation

171
$$\frac{\partial \phi}{\partial t} + \mathbf{u} \cdot \nabla \phi = \gamma \nabla \cdot \left[\varepsilon \nabla \phi - \phi (1 - \phi) \frac{\nabla \phi}{|\nabla \phi|} \right]$$
(3)

Symbols ρ, η , and σ denote the density, dynamic viscosity, and the surface tension 172 respectively. The pressure is given by p while l denotes the identity matrix. The y173 174 denotes the re-initialization parameter approximates the maximum speed occurring in the computational domain (Sethian, 1999). The ε denotes an artificial thickness of 175 the interface, which is assumed to be the maximum mesh size in subdomains in the 176 neighbourhood of the interface. In this work, the parameters γ and ε with the value of 177 0.065 m/s and 5.8×10⁻⁶ m were selected, based on the maximum flow velocity in 178 microchannel dimensions selected for simulation and optimum mesh size, 179 respectively after a mesh convergence analysis. The surface tension force, F_{st} , acting 180 at the interface between two fluid phases can be calculated by Equation 4 below. 181

182
$$F_{st} = \sigma k(\phi) \mathbf{n}_{\Gamma} \delta_{sm}(\phi) \tag{4}$$

where *k* denotes the local interfacial curvature, \mathbf{n}_{Γ} is the unit normal vector to the interface pointing into the droplet, and the smeared out Dirac delta function concentrated at the interface between two fluids is given by $\delta_{sm}(\phi)$. These can be expressed as shown below.

187 The interfacial curvature is given by

$$188 k(\phi) = -\nabla \cdot \mathbf{n}_{\Gamma} (5)$$

189 The delta function is approximated by

190
$$\delta_{sm}(\phi) = 6 |\phi(1-\phi)| |\nabla \phi|$$
 (6)

Since the surface tension and local interfacial curvature term are easily represented in terms of the level-set function, the LSM can be used to compute the changes in the droplet topology due to the flow. The density and the viscosity of the two fluids at any point can be calculated using the two equations given below:

195
$$\rho = \rho_1 + (\rho_2 - \rho_1)\phi$$
 (7)

196
$$\eta = \eta_1 + (\eta_2 - \eta_1)\phi$$
 (8)

where ρ_1 and ρ_2 are the densities of continuous phase and dispersed phase, and η_1 and η_2 are the viscosities of continuous phase and dispersed phase.

Equations (1) – (8) were solved using the COMSOL Multiphysics (Version 4.3) two
 phase flow module.

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202 **2.2 Rheological Constitutive Model**

To implement the shear thinning non-Newtonian fluids, the Carreau-Yasuda viscosity model has been implemented within the level set formulation. The Carreau-Yasuda constitutive model is defined by the following equation (Eq. (9)) (Carreau, 1972; Yasuda and Cohen, 1981):

207
$$\eta(\dot{\gamma}) = \eta_{\infty} + (\eta_o - \eta_{\infty}) [1 + (\lambda_{CY} \dot{\gamma})^a]^{\frac{n-1}{a}}$$
 (9)

where η_o and η_{∞} denote the zero shear viscosity and the infinite shear viscosity respectively whereas λ_{CY} denotes the relaxation time and *a* denotes fitting parameter. The *n* is the shear-thinning power-law exponent and $\dot{\gamma}$ is the shear rate. We generalize the shear rate ($\dot{\gamma}$) in the Carreau-Yasuda model using the following invariant of the velocity gradient tensor in Cartesian coordinates.

213
$$\dot{\gamma} = \sqrt{\frac{1}{2} \left[4 \left(\frac{\partial u}{\partial x} \right)^2 + 2 \left(\frac{\partial u}{\partial y} + \frac{\partial v}{\partial x} \right)^2 + 4 \left(\frac{\partial v}{\partial y} \right)^2 \right]}$$
(10)

The rheological parameters of Carreau-Yasuda stress model were experimentally 214 determined using a controlled stress rheometer (MC 302, Anton Paar) for all SCMC 215 polymer concentrations. All the measurement data were seen to be fitted well to 216 Carreau-Yasuda model for all SCMC concentrations. Experimentally measured 217 normalised shear-viscosity is plotted against the normalised shear rate for SCMC 218 219 solutions at various concentrations in Figure 1a. These measurement data are also summarized in Table 1 and used in numerical calculations subsequently. For shear 220 thinning fluids, the flow modifies the viscosity and so the effective viscosity at the flow 221 conditions in a given simulation becomes an important quantity. To aid later 222 interpretation we estimate this by using the wall shear rate in the continuous phase. 223 Figure 1b shows the effective viscosity ratio ($\lambda_n = \eta_d / \eta_c$) against continuous phase 224 flow rate, Q_c, where we used the downstream wall shear rate in the continuous phase 225 as a 'typical' shear rate and computed the disperse phase viscosity via equation 9. 226

In the present work, a key output quantity is the effective droplet diameter (*d*). An integration operator was used to find the area corresponds to the dispersed phase where $\phi > 0.5$ in order to calculate the effective droplet diameter (Equation 11).

$$230 \qquad d = 2 \cdot \sqrt{\frac{1}{\pi} \int_{\Omega} (\phi > 0.5) d\Omega} \tag{11}$$

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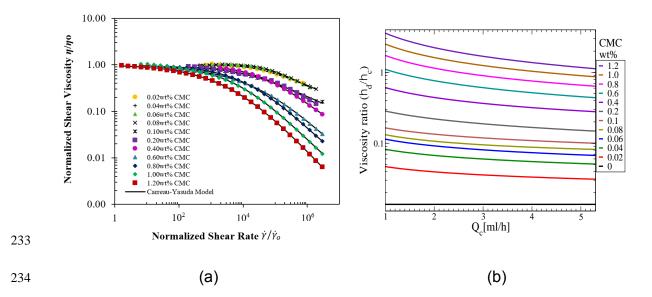


Figure 1: (a) Normalized shear viscosity plotted against normalized shear rate for a series of SCMC shear-thinning solutions with different concentrations. (b) Effective viscosity ratio against continuous phase flow rate, *Q_c*, with the shear-thinning computed from equation 9 using the wall shear rate in the continuous phase.

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Table 1: Rheological properties of SCMC polymer solution.

SCMC Solution	Carreau-Yasuda Model Constant					
wt%	η _o (Pa.s)	<i>η</i> ∞ (Pa.s)	λcy (S)	n	а	
0.02	0.0070	0.0003	0.0400	0.7121	0.9653	
0.04	0.0121	0.0000	0.0325	0.7102	1.6980	
0.06	0.0171	0.0000	0.0256	0.6775	1.3728	
0.08	0.0195	0.0028	0.0143	0.4886	1.1319	
0.10	0.0420	0.0007	0.0572	0.6242	0.4734	
0.20	0.0742	0.0006	0.0041	0.3528	0.3856	
0.40	0.1946	0.0040	0.0138	0.3157	0.5534	
0.60	0.7995	0.0022	0.0147	0.1995	0.3660	
0.80	1.6469	0.0057	0.0515	0.2444	0.4782	
1.00	4.1143	0.0031	0.1604	0.2869	0.5000	
1.20	10.2644	0.0000	0.2069	0.2297	0.4175	

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242 **2.3 Geometry and Boundary Conditions**

A T-junction geometry with 220 μ m (*w_c*) main channel width and a 90 μ m (*w_d*) side channel was created and meshed with quadrilaterals elements. These dimensions are selected to match the microchannel geometry created for validation experiments. An entrance thickness (*h*) of 73.5 μ m was prescribed in the computational domain in order to define the depth of the microchannel. The continuous phase flows along the horizontal main channel whereas the dispersed phase is introduced through the narrow side channel. The geometry of the microfluidic T-junction with structured 2D mapped mesh is illustrated in Figure 2. In microfluidics systems, all flow tends to be laminar. The domain boundary of the continuous ($\partial\Omega_1$) and dispersed phase ($\partial\Omega_2$) were set up with laminar inflow conditions by defining the volumetric flow rates of continuous phase and dispersed phase.

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$$\partial \Omega_{1} = \begin{cases} u_{x} = u \cdot \mathbf{n} = u_{o} \\ u_{y} = 0 \\ h \int_{\partial \Omega_{1}} u \cdot \mathbf{n} \, dy = Q_{c} \end{cases}$$
(12)

255

$$\partial \Omega_{2} = \begin{cases} u_{x} = 0 \\ u_{y} = u \cdot \mathbf{n} = u_{o} \\ h \int_{\partial \Omega_{2}} u \cdot \mathbf{n} \, dx = Q_{d} \end{cases}$$
(13)

where **n** is unit vector that has a direction normal to the inlet boundaries, *h* is the entrance thickness of microchannel, Q_c and Q_d denote the flow rates of the continuous and dispersed phase flow rates respectively. The domain boundary condition at the outlet ($\partial \Omega_3$) is set to a pressure with no viscous stress.

260
$$P=0 Pa$$
 (14)

No-slip boundary condition was applied to all other wall boundaries. The numerical simulations were performed at a time step $\Delta t = 2.57 \times 10^{-5}$ seconds, calculated using the Courant-Friedrichs-Lewy (CFL) number. The Courant number is defined as:

$$264 C = \frac{\Delta t \cdot U}{\Delta X} (15)$$

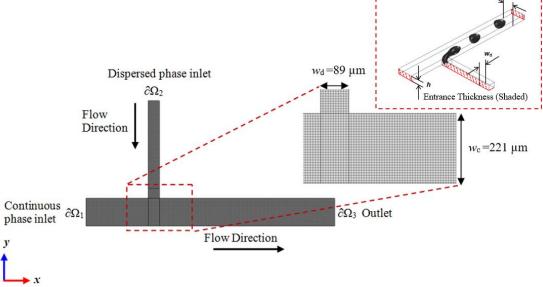
where Δt is the time-step used during simulations, *U* is the maximum fluid velocity and ΔX is the computational cell size. A Courant number of 0.25 was used in the current simulations considering as a robust value to maintain the stability of the calculations (Kashid *et al.*, 2010; Schneider *et al.*, 2011).

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Table 2: Mesh convergence analysis for the dependence of effective droplet diameter on the number of quadrilaterals elements. 273

Number of	Degree of	Effective Droplet	Relative
Elements	Freedom	Diameter (µm)	Error%
976	13590	172.46	6.960
2072	28727	162.63	1.586
4024	54152	158.81	4.684
7644	101947	150.52	0.249
12166	161405	150.90	0.388
15963	207662	150.31	



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Figure 2: The T-shaped geometry used for the simulations with two-dimensional 276 (2D) quadrilaterals mesh elements. 277

3. Mesh Convergence Analysis 279

Prior to the parametric study, mesh sensitivity analysis was performed to quantify the 280 dependency of simulation results on mesh size. A flow rate ratio, Q of 0.05 (where 281 $Q_c = 2.0$ ml/h and $Q_d = 0.1$ ml/h) was used in this analysis. The data is recorded at 282 0.0005 seconds intervals, which were sufficiently small to capture the breakup 283 phenomenon of the fluid-fluid interfaces. A 2D structured mapped mesh was 284 generated and meshes of varying degrees of resolution were set up for the domain. 285 The effect of mesh size was examined by increasing the number of mesh elements 286 from 976 (coarsest grid) to 15963 (finest grid). Table 2 summarized the mesh 287

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convergence test for the dependence of effective droplet diameter on the number of mesh elements. An acceptable relative error of 0.38% between the last two finer meshes was considered. The relative error of the measurement shows the error deviation in relation to the measured values of droplet diameter between the two different mesh resolutions. Therefore, an optimal grid resolution containing 7644 elements with cell sizes equal to 0.0082 mm was selected for the subsequent study.

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4. Numerical Model Validation

At a very preliminary stage, a validation of the 2D numerical model of formation of the water droplets was carried out against a matching laboratory experiment. The geometry and the properties of the fluids used in the numerical simulation were set to the values measured in the lab. Deionised water and olive oil (highly refined, low acidity, Sigma Aldrich) were used in the experiment to generate droplets in a Tjunction microchannel.

The comparison with the experiments is carried out to estimate the differences between the numerical and experimental approaches for a known system. Figure 3 illustrates the comparison of effective droplet diameter as a function of the applied flow rate ratio, *Q*, between numerical and experimental results.

As shown in Figure 3, both the experimental measurement and the simulated 306 prediction indicate that the effective diameter of the dispersed phase droplets 307 increases with the increasing flow rate ratio. As the flow rate ratio increases the 308 numerical computations over predicts the effective droplet diameter. The discrepancy 309 310 between the simulated and the measured droplet diameters is roughly 11% at the highest flow rate ratio tested and for lower flow rates it shows a reasonably good 311 agreement. Syringe pump induced oscillation of flow rate implied a source of 312 fluctuation in the microfluidic flow, which might affect the droplet diameter. 313 Additionally, the temperature dependence of the physical and rheological properties 314 of each working fluid in experiments may have contributed to the deviation in droplet 315 316 diameter from numerical simulation as they were assumed to remain constant in the numerical model. Besides, the difficulties of numerical dissipation in advection step 317 of fluid simulation may arise for larger flow rate ratio. In order to reduce the numerical 318

dissipation of the scheme, a reasonable range of low flow rate ratio, $0.04 \le Q <$ 320 0.0675, has been selected for the subsequent modelling studies of non-Newtonian 321 flow in order to ensure higher reliability and accuracy of the numerical model.

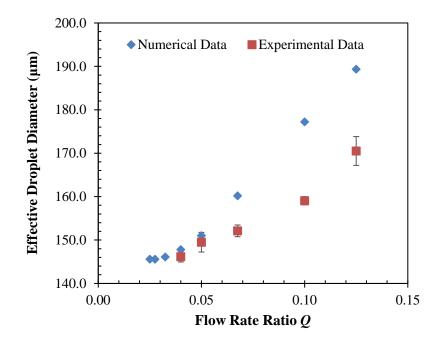


Figure 3: Comparison of effective droplet diameter between 2D numerical and experimental result in the range of Q applied. Error bars indicate the standard deviation in effective droplet size measurement of 30 droplets under fixed experimental condition.

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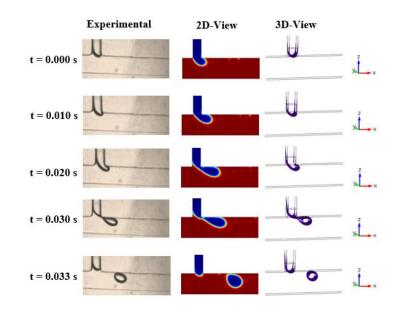
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A preliminary validation test of the 3D structure was also performed at the prescribed 328 Q of 0.05. A grid resolution containing 88257 elements with cell sizes equal to 0.0188 329 mm was selected for the study. Figure 4 illustrates a sequence of snapshot of droplet 330 formation for the 2D and 3D model representation against laboratory experiments at 331 Q of 0.05. As seen in Figure 4, a dripping phenomenon, in which droplets are being 332 sheared off before they fill the channel cross-section, were observed for the 333 experimental model. Both 2D and 3D simulations predict similar droplet breakup 334 335 process.

An effective droplet diameter was calculated based on the area of droplet in 2D model, whereas the droplet volume was used to calculate the effective diameter in 328 3D model. Note that this approach produces two different values for any given droplet

as one is on the projected area as seen from the top projection (looking down on to 339 w_c) and the other is on the volume dispensed based on the total volume fed and the 340 frequency of the droplets. An error of 1.56% between experimental value of 102.36 341 µm and numerical value of 100.77 µm were obtained in the 3D case. While the 2D 342 model presented an error percentage of 2.21% between the experimental and 343 numerical results of 149.5 µm and 152.88 µm, respectively (see Figure 3). Such 344 discrepancy between 2D and 3D implementation are may attributed to the boundary 345 effects in the third coordinate's direction. 346

The velocity profiles are not significantly affected in the simplest 2D model when 347 compared to that of 3D channel. A comparison on both, the main and the side 348 channels show (see Figure 5) the maximum velocity remains same while the 349 distribution changed due to wall effects in the 3D case. Nevertheless, based on this 350 351 comparison, we have concluded that the flow system can be modelled using 2D geometry with acceptable results compared to 3D and experiments. As it is 352 computationally too costly for extensive tests of 3D model to resolve the typical nano-353 sized interface and motion (Liu and Zhang, 2009) of microdroplets, the present 354 parametric analyses were implemented based on 2D model. 355



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Figure 4: Snapshots of three-dimensional (3D) and two-dimensional (2D) simulations of the water-in-oil droplet breakup process in microchannel and corresponding to experiments (for system: Q_c: 2.0 ml/hr and Q_d: 0.1ml/hr, Q=0.05). The colour bar specifies the dispersed phase as the area where colour distinction is from 0 to 0.5 and continuous phase the area from 0.5 to 1.0. The iso-surface of 0.5 denotes the interface where the droplet is identified as solid blue rounded shape.

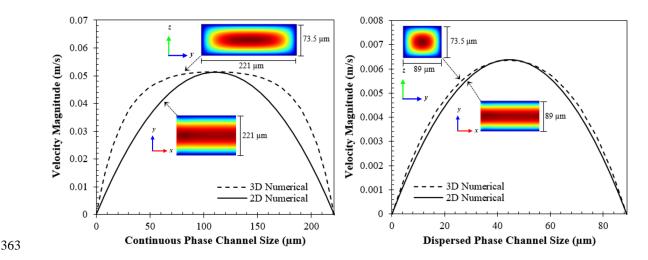


Figure 5: Comparison of velocity profiles for 3D and 2D continuous and dispersed phase microchannel.

367 5. Results and Discussion

5.1 Droplet formation and breakup mechanism

The simulations closely follow the experimental observations of the formation and the 369 detachment of a droplet in the microfluidics T-junction. First, the tip of the dispersed 370 phase filament enters the main channel and partially blocks it, which results in the 371 continuous phase having to travel through a smaller cross-sectional area at the T-372 373 junction. This causes higher drag at the interface between the two phases and, over time, the dispersed phase entering the T-junction is slowly convected downstream. 374 Due to the drag applied by the continuous phase, the dispersed phase filament thins 375 out forming a neck. The pressure gradient across the forming liquid filament and the 376 drag at the interface further distort the droplet in the downstream direction until they 377 overcome the surface tension and a droplet is formed (Garstecki et al., 2006; 378 Garstecki et al., 2005; Glawdel et al., 2012). The fully detached droplet flows 379 downstream in the main channel, while the dispersed phase retracts into side 380 channel. This break-up process is continuously repeated, and the size of the droplets 381 produced can be varied by altering the fluid flow rates, the channel dimensions or by 382 383 changing the physical properties of the two phases. The results in Figure 6 and 7 were computed using the numerical scheme outlined in Section 2 with both the 384 dispersed and continuous phases being Newtonian fluids with viscosities of 9.3×10⁻⁴ 385 Pa.s and 6.8×10⁻² Pa.s, respectively. 386

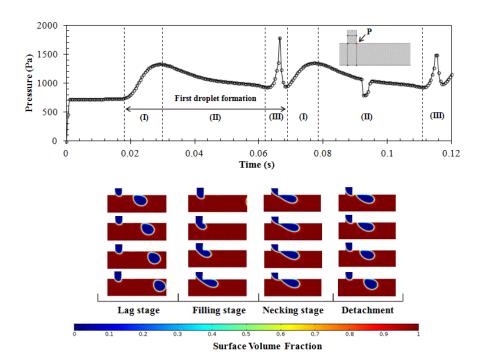
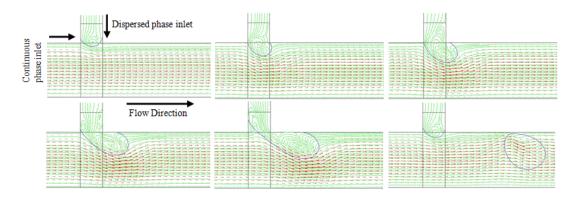


Figure 6: Evolution of the pressure at point P during droplet breakup. P denotes the point at the junction and reflects the evolution of breakup mechanism. The droplet formation process can be divided into three stages: (I) lag stage, (II) filling stage, and (III) necking stage.



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Figure 7: Streamline and velocity field representation of the process of droplet breakup in a model of the T-shaped.

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Variations in pressure at the leading edge of the T junction indicated by point P during the growth of the dispersed phase filament and the detachment of the drop is given in Figure 6. It shows that the pressure increases gradually until the drop starts to yield under the shear and the pressure build up and then decreases until the breakup. The breakup of the droplet is marked by the sharp peak. Then the filament starts to grow again and the pressure at point P starts to increase. The sudden depression in the pressure during the formation of the second drop indicates first drop exiting the 403 computational domain boundary. It should be noted that the pressure inside the 404 droplet is higher than that of the continuous phase in the neighbourhood. Since the 405 boundary condition applied assumes the droplets to be entering the constant 406 pressure as they leave the domain, the pressure field shows a sudden reduction 407 spread over the time it takes a droplet to cross the outlet boundary completely. This 408 phenomenon can be observed in all subsequent parametric studies.

The streamlines are overlaid on the velocity field in Figure 7 to investigate the flow development within the dispersed phase filament and the growing droplet. In Figure 7, at the start of the formation process, a recirculation flow is generated close to the interface due to the shear applied by the continuous phase flow. This region expands gradually when the dispersed phase enters the main stream and elongates downstream. When detached, the droplet is carried by the continuous phase and the initial flow is disturbed and the recirculation flow no longer exists.

The droplet detachment mechanism has been modelled by Husny and Cooper-White (2006) for Newtonian fluids, and we use this model here to illustrate the influence of physical properties of the fluids on the droplet size. We consider forces and velocities along the main channel. The droplet detaches when it becomes sufficiently large that the cross-flow drag force from the continuous phase, which pulls the droplet downstream, matches the interfacial tension force joining the droplet to the side channel. The interfacial tension force is given by

$$F_{\sigma} = \frac{\pi \sigma w_d^2}{d},\tag{16}$$

where *d* is the droplet diameter. The cross-flow drag force is given by

425
$$F_D = 3\pi\eta_c (v^* - v_d) df(\lambda_\eta)$$
 (17)

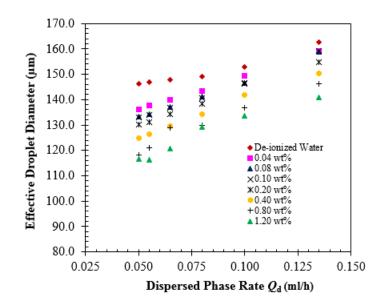
where v^* is the continuous phase velocity at the height of the droplet centre, v_d is the droplet velocity, $\lambda_{\eta=} \eta_d / \eta_c$ is the viscosity ratio and $f(\lambda_{\eta}) = (\frac{2}{3} + \lambda_{\eta})/(1 + \lambda_{\eta})$ captures the effect of the disperse phase viscosity on the droplet drag. When $\lambda_{\eta} \gg 1$ then the disperse phase viscosity is sufficiently high that the droplet is solid-like and equation 17 reduces to the drag from Stokes flow around a solid sphere. Conversely, for smaller values of λ_{η} , internal flow within the droplet becomes possible and this acts to reduce the drag on the droplet from the continuous phase. We note that $2/3 < f(\lambda_{\eta}) < 1$, so this model predicts a fairly modest effect of the disperse phase viscosity.

435

436 **5.2 Effect of Dispersed Phase Rate**

437 In the first series of numerical simulations, the effect of the dispersed phase flow rate, Q_d , on the droplet diameter was systematically examined for a fluid with physical 438 properties equivalent to SCMC solutions with concentrations ranging from 0.00 wt% 439 440 to 1.20 wt%. Similarly, the continuous phase properties were selected to reflect those of olive oil ($\eta_c = 0.068$ Pa.s). A contact angle of 180°, representing complete 441 repulsions of the working liquid by the channel surface was applied. This is to reflect 442 the wall being wetted with the oil, forming a lubricating layer preventing the formation 443 of three phase contact line at the channel surface. 444

For a constant continuous phase flow rate, Q_c , of 2.00 ml/hr, the droplet size, d, was 445 446 measured with increasing Q_d between 0.05 ml/hr and 0.135 ml/hr. Results are shown in Figure 8. Here d increases with increasing Q_d and with increasing SCMC 447 448 concentration and, as Q_d increases then the results for different SCMC concentration 449 becomes more tightly bunched. The droplet size increases with Q_d as, for larger Q_d , larger volumes of the dispersed phase fluid enter the main channel before the filament 450 is pinched off. The increase in drop size does not replicate the threefold increase of 451 the dispersed phase flow rate, because the formation time shortens. However, it 452 453 should be noted that the frequency of droplet shedding increases in line with 454 conservation of mass. The effect of SCMC concentration can be explained as follows. 455 Figure 1b shows that, for all Q_c in this study, there is a considerable increase in the effective η_d with SCMC concentration even when shear thinning is accounted for. This 456 increase in η_d leads to smaller droplets because, as equation 17 shows, larger η_d 457 leads to an increase in the cross-flow drag force, due to internal flow inside the droplet 458 459 being suppressed. This effect, combined with the shear thinning, also explains why the results at different SCMC concentrations become more closely packed with 460 461 increasing Q_d . As Q_d increases flow within the droplets becomes faster and hence shear thinning lowers the effective viscosity. As figure 1 shows, shear thinning is 462 463 stronger in the more concentrated SCMC solutions, thus with increasing flow rates the range of effective viscosity ratios narrows and so does the range of droplet sizes. 464





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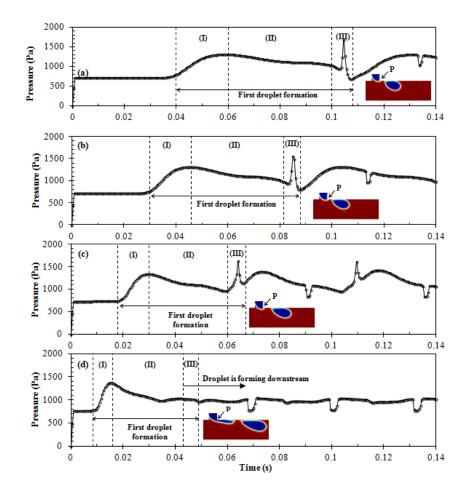
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Figure 8: Effect of Q_d on the effective droplet diameter of SCMC solutions for various SCMC concentrations. The continuous phase flow rate Q_c is kept constant at 2.00 ml/hr for all cases reported.

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The time dependence of the pressure at the corner of T-junction during SCMC droplet 470 breakup at different Q_d were systematically investigated and are plotted in Figure 9. 471 At the necking point, there is a pressure spike at the time of detachment. Moreover, 472 the magnitude of pressure depends on the specific location of the droplet detachment. 473 When the Q_d is small, the droplets are pinched off at the T-junction. When $Q_c > Q_d$, 474 droplet does not have time to fill the main channel fully producing elongated slugs. 475 The droplet breakup occurs immediately after a short filament is formed filling the 476 main channel partially and the droplet detachment occurs at the junction⁴¹. These 477 observations are also supported by Tan et al. (2008) who reported that the shear force 478 from the continuous phase is sufficiently large to induce the dispersed phase to form 479 droplets at the T-junction of the microchannel at the low dispersed phase rates. When 480 Q_d increases, the filament grows at faster rates, partially blocking the main channel. 481 This leads to detachment of the droplets at increasing frequency. This is clearly shown 482 in the Figure 9. The distinct pressure peak at point P at the corner decreases as the 483 filament no longer retracts into the side channel. As the Q_d increases further, the 484 filament begins to extend in the main channel and the droplet detachment takes place 485 downstream of the T-junction. As the filament grows beyond the corner of the T-486 junction, the pressure at corner of the junction becomes less important. Downstream 487

detachment of the droplets occurs more readily for fluids with high η_d because they are more able to resist fragmentation of the filament of disperse phase as it forms.



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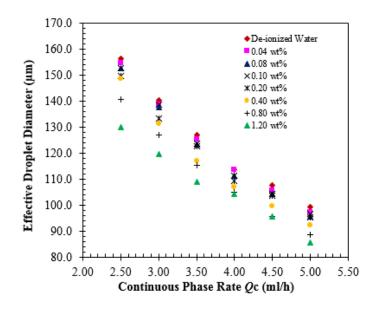
Figure 9: Pressure profile of 0.20 wt% SCMC for the effects of Q_d on SCMC droplet breakup process at each Q_d : (a) 0.050 ml/hr, (b) 0.065 ml/hr, (c) 0.100 ml/hr, (d) 0.200 ml/hr (for system: Q_c = 2.00 ml/hr). P denotes the pressure at the corner of Tjunction. The sharp peak with given data values reflects the droplet breakup at the corner of the side channel as shown in the extracted images. At high Q_d , the pressure peak becomes insignificant as the detachment point is located downstream of the junction.

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499 **5.3 Effect of Continuous Phase Rate**

As discussed in section 5.1, droplet formation at the T junction is dominated by the balance of the interfacial force with the viscous stresses imposed by the continuous phase. The size of the droplets can be tuned by changing Q_c , which alters the viscous stresses on the droplets (DeMenech *et al.*, 2008). Since the channel width of the continuous phase ($w_c = 221 \ \mu m$) is larger than dispersed phase channel width ($w_d =$ 90 μm), the dispersed phase stream does not occupy the entire width of the channel downstream of T-junction. Thus the breakup is controlled by the local shear-stress of the continuous phase acting on the dispersed phase droplet, which can be estimated via equation 17. For a constant Q_d at 0.20 ml/h, the droplet diameter was measured with the variation in Q_c (2.50 ml/h to 5.00 ml/h). Results are illustrated graphically in Figure 10.





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Figure 10: Effect of Q_c on the effective droplet diameter of SCMC solutions for various SCMC concentrations. The dispersed phase flow rate Q_d is kept constant at 0.20 ml/h for all cases reported.

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Figure 10 shows that SCMC droplet size can be controlled in the T-junction 518 519 microchannel by varying the flow rates of the continuous phase when the disperse phase is kept constant. As Q_c is increased, the simulation results show a reduction in 520 521 droplet size (see Figure 10). This is because at higher Q_c , the shearing effect imparted by the continuous phase on the dispersed phase is larger, and as a result, smaller 522 droplets will be pinched off. According to Husny et al. (2003), this is because, a higher 523 Q_c imparts a higher cross-flow drag force on the droplet. Consequently, this leads to 524 a decrease in the droplet growth time, resulting in a reduction in final droplet volume. 525 This result also agrees with Thorsten et al. (2001) and Tan et al. (2008) who reported 526 527 that droplet size decreased with increased continuous phase flow rate. Figure 10 also shows that decreased SCMC concentration leads to larger droplets. This is for the 528

same reason as in Figure 8, namely that a lower disperse phases viscosity enables 529 stronger internal flow in the droplet, reducing the cross-flow drag force. Also similarly 530 to Figure 8, increasing the flow rate Q_c causes the variation of droplet size with SCMC 531 concentration to lower. As above, this is because the more concentrated SCMC 532 solutions have a larger zero shear viscosities but also stronger shear thinning. Hence 533 534 increasing Q_c narrows the separation of η_d values across the concentration range (see figure 1b). With increased SCMC concentration we also see increased formation 535 of extended filaments into the downstream region. The increased η_d opposes the 536 breakup of the disperse phase filament and results in longer threads (Zhang and 537 Basaran, 1995; Nunes et al., 2013). This prediction is corroborated by reported 538 experimental observations at a T-junction microfluidics (Gu and Liow, 2011). 539

Figure 11 illustrates the evolution of pressure of the entire droplet formation process 540 as observed at the trailing edge of the side channel (marked by point P in the sub-541 figure) for 0.20 wt% SCMC at each continuous phase flow rate. The average pressure 542 increases from Figure 11 (a) to (d) in accordance with the increase of the flow rate. 543 However, the pronounced sharp pressure peaks, as previously mentioned in section 544 5.1, are not observed at lower Q_c. As Q_c decreases, the position of the necking and 545 546 detachment point is shifted downstream, leading to the jetting phenomenon where a long neck of dispersed thread is developed before the droplets are sheared off in the 547 548 main channel. It is worth noting that the pressure at which the droplets pinch off initially varies slightly, as it takes several cycles of droplet production to reach steady 549 550 state. As Q_c increases, the gap between the emerging droplet and main channel wall is increased. Here the emerging dispersed phase filament does not have sufficient 551 552 time to obstruct the main channel before detachment.

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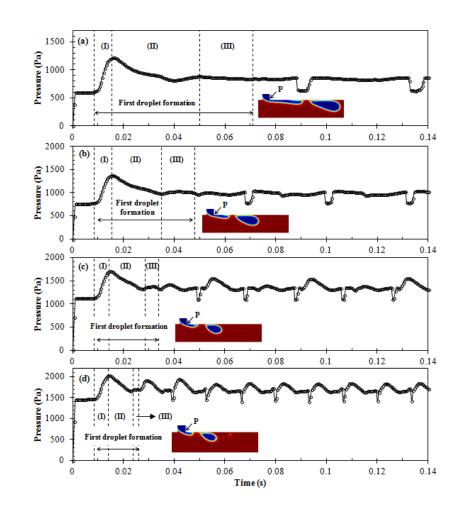


Figure 11: Pressure profile of 0.20 wt% SCMC for the effects of Q_c on SCMC droplet breakup process at each Q_c : (a) 1.50 ml/hr, (b) 2.00 ml/hr, (c) 3.00 ml/hr, (d) 4.00 ml/hr (for dispersed phase flow rate Q_d =0.20 ml/h). P denotes the pressure at the corner of T-junction. After the first droplet is pinched off, the pressure fluctuates abruptly and repeatedly. The lack of a sharp pressure peak occurs when the detachment point is located downstream of the junction.

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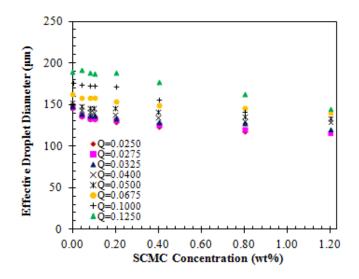


Figure 12: Effect of SCMC concentration on effective droplet diameter for various flow rate ratios Q (with fixed Q_c = 2.00 ml/h and 0.05 ml/h < Q_d <0.25 ml/h).

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574 5.4 Effect of Sodium Carboxymethylcellulose (SCMC) Concentration

575 The present parametric analysis aims to assess the role of the dispersed phase SCMC concentration on the formation of droplets in a Newtonian continuum. The 576 viscosity and degree of shear thinning both increase with the increasing concentration 577 and molecular weight. Figure 12 plots our results to emphasize the influence of SCMC 578 concentration. Increasing the concentration of SCMC reduces d, and increasing the 579 flow rate reduces the influence of concentration. These two effects are explained by 580 the viscosity and degree of shear thinning, respectively, and their increase with 581 concentration (see Figure 1b and equation 17). This is consistent with experiments 582 by Gu and Liow (2011) who saw a decrease of droplet size with increasing polymer 583 concentration, for fluids in which the polymers increase the viscosity and the strength 584 of shear thinning. There is a notable contrast with experiments of Husny and Cooper-585 White (2006) who reported no observable change in droplet diameter with increasing 586 polymer molecular weight but a strong change in the filament characteristics. 587 However, their polymer solutions were Boger fluids, meaning that the polymer 588 influenced the elasticity but not the shear viscosity. This illustrates the potential to 589 tailor the droplet size and formation mechanism by controlling the shear-thinning and 590 elasticity through choice of polymer concentration and molecular weight. 591

Figure 13 demonstrates the pressure distribution during droplet breakup at the T-junction corner for the concentration of 0.00 wt%, 0.20 wt% and 0.80 wt% SCMC solution. For the fixed Q of 0.05, the droplets pinch off near to the corner of junction, resulting sharp pressure peaks when the SCMC concentration is lower. However, when higher concentrations are encountered, a progressive necking and breakup of the dispersed phase filament occurs downstream, resulting in less pronounced pressure peaks. This is due to the higher viscosity of the dispersed phase stabilising the filament and not due to the influence of elasticity, which is absent in our calculations.

5.5 Effect of Continuous Phase Viscosity

Several types of oil with different viscosities, namely mineral oil, olive oil and peanut oil, were modelled as the continuous phase (From data in Loizou et al. (2014)). Table 3 shows the experimental data of viscosities and interfacial tensions of different continuous phase systems and we note that as η_c increases the interfacial tension decreases, both of which are associated with smaller droplets. Numerical simulations were performed to examine the effect of the continuous phase on the SCMC droplet breakup process at a constant Q value of 0.05 (where $Q_d = 0.10$ ml/hr; $Q_c = 2.00$ ml/hr).

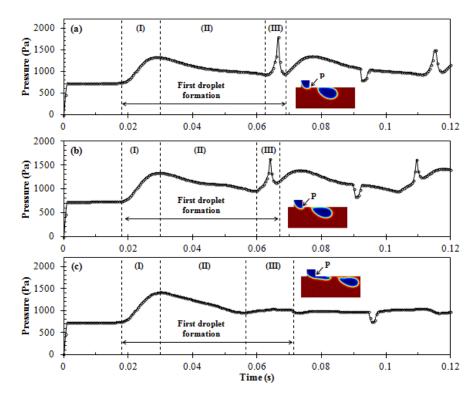




Figure 13: Pressure profile of (a) 0.00 wt%, (b) 0.20 wt%, (c) 0.80 wt% SCMC droplet breakup process (for system: $Q_d = 0.10$ ml/h; $Q_c= 2.00$ ml/h). P denotes the pressure at the corner of T-junction. The sharp pressure peak reflects the droplet breakup point as shown in the extracted images. The absence of a sharp pressure peak means that the detachment point is located downstream of the junction.

These results are presented in Figure 14, which illustrates that the droplet diameter 627 decreases as η_c increases, as anticipated. Larger η_c produces a larger drag force 628 from the continuous phase onto the droplet, meaning that it detaches at a smaller 629 size (see equation 17). Data from Husny and Cooper-White (2006) show that 630 increasing the viscosity of continuous phase decreases the droplets size. The findings 631 of this study are also consistent with those earlier experimental studies (Garstecki et 632 633 al., 2006; Gu and Liow, 2011; Yeom and Lee, 2011). For the oils we modelled, the interfacial tension also reduces as the viscosity increases and this is expected to 634 further reduce the droplet size, as the interfacial tension resists detachment of the 635 growing droplet (see equation 16). Indeed, Figure 14 shows that the properties of the 636 continuous phase strongly influence the droplet size, whereas the disperse phase 637 properties, which are varied via the SCMC concentration, have a smaller secondary 638 effect. 639

Figure 15 illustrates the variation in SCMC droplet diameter for different viscosities of 640 the continuous phase. As above, increased viscosity ratio reduces the droplet size, 641 but this effect is small compared to the influence of the continuous phase viscosity. 642 We note that the strongest effect of the SCMC occurs at lower concentrations where 643 the difference in viscosity ratio between neighbouring concentrations is most 644 pronounced (see Figure 1b). The pressure profile of 0.20 wt% SCMC droplets under 645 different η_c at fixed flow rate ratio Q are illustrated in Figure 16. As observed from 646 Figure 16, the build-up of the pressure at the corner of the junction increases as η_c 647 decreases. The pressure peaks are less marked in the case of larger η_c . This appears 648 to correlate with droplet size and also downstream distance of the detachment point. 649 Consequently, sharp peaks were clearly observed only for smaller η_{c} . 650

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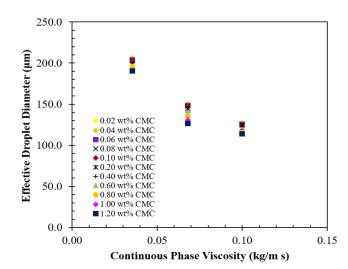
Table 3: Physical properties of the continuous phase systems.

Continuous Phase System	Dynamic Viscosity (mPa.s)	Interfacial tension between oil and water (0wt% SCMC) (mN/m)
Mineral oil	35.40 ± 0.00	41.37 ± 2.90
Olive oil	68.10 ± 0.67	20.74 ± 0.47
Peanut oil	98.80 ± 0.39	18.80 ± 1.31

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Figure 14: Effect of η_c on effective droplet diameter of SCMC droplets (for system: $Q_d = 0.10 \text{ ml/hr}; Q_c = 2.00 \text{ ml/hr}$).

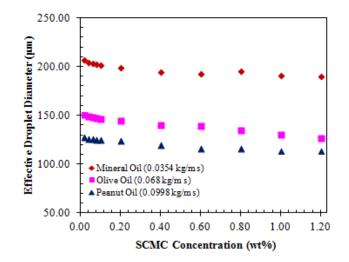
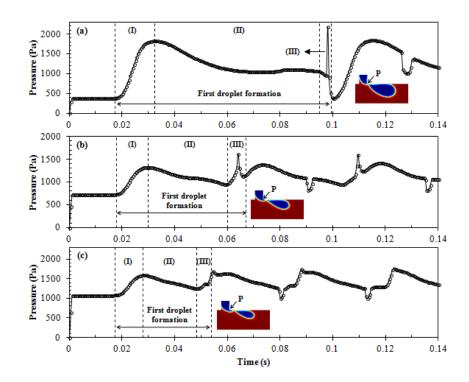


Figure 15: Effect of η_c on effective droplet diameter of SCMC droplets (for system: $Q_d = 0.10 \text{ ml/hr}; Q_c = 2.00 \text{ ml/hr}.$



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Figure 16: Pressure profile of 0.20wt% SCMC droplets breakup for the system with different continuous phase viscosities: (a) Mineral oil ($\eta_c = 0.0354$ Pa.s), (b) Olive oil ($\eta_c = 0.068$ Pa.s), (c) Peanut oil ($\eta_c = 0.0998$ Pa.s) (for system: $Q_d = 0.10$ ml/hr; Q_c = 2.00 ml/hr). P denotes the pressure at the corner of T-junction. The sharp peak with given data values reflects the droplet breakup point as shown in the extracted images. The pressure peak becomes insignificant when the detachment point is located downstream of the junction.

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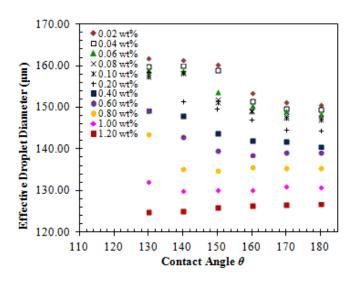
5.6 Effect of Contact Angle

The influence of the contact angle (θ) on drop shape size and production has been 673 the focus of several investigations of Newtonian fluids (Bashir et al., 2011; 674 Kumacheva and Garstecki, 2011). However the role of the contact angle on non-675 Newtonian droplets with shear-thinning behaviour has not been widely studied. Due 676 to the resulting high surface area to volume ratios of micro-sized devices, the 677 interaction between the solid surface of the microchannel and the fluid has been a 678 major focal point in affecting the dynamics of the droplet formation process. In a 679 microfluidic geometry with a T-junction configuration, the continuous and dispersed 680 phases were initially dispensed separately at the desired flow rates through the 681 microchannel with hydrophobic walls. The continuous phase needs to wet the 682 channel walls preferentially in order to separate the dispersed phase, which tends to 683 684 be non-wetting, from the walls and eventually it breaks into droplets. The propensity of the two flowing immiscible liquids to wet the channel walls is of utmost importance 685 686 for the droplet formation process to take place and be stable (Rosengarten et al., 2006). The magnitudes of surface wettability can be characterized by static contact 687 angles, whereas the dynamic effects are not explicitly taken into account. 688

A focus of the present study is to highlight the effect of the wetting characteristic of 689 the channel walls on the non-Newtonian SCMC droplet formation process. The 690 contact angle of the SCMC droplets at the channel walls was varied from 130° to 180° 691 and implemented in the simulations. Considering a three-phase coexisting system, 692 this range of contact angles covers partial wetting behaviour to complete non-wetting 693 behaviour. Olive oil was selected as the carrier fluid along the main channel, while a 694 SCMC polymer solution was used as the dispersed phase fluid supplied from the 695 orthogonal channel. Figure 17 illustrates the effect of different wetting conditions on 696 697 the droplet diameter of each SCMC concentrations. As elucidated in Figure 17, the droplet diameter decreases as θ increases for a fixed flow rate ratio Q of 0.05, for the 698 lower CMC concentrations. Here, θ has a strong impact on the reduction in droplet 699 size for concentrations from 0.02 wt% to 0.80 wt%, causing a step change in droplet 700 diameter at about 150°. Interestingly, the effect of θ on the droplet diameter was found 701 to be opposite at higher concentration at CMC solution (> 0.80 wt%), showing a weak 702 increase with increasing θ for almost all rates. 703

In addition, the numerical simulation predicts that the droplet diameter was observed 704 to be largely independent of θ when it is larger than 150°, which is in agreement with 705 numerical results from Sang et al. (2009) and Bashir et al. (2011). Sang et al. (2009) 706 reported that larger contact angles ($\theta \ge 165^\circ$), corresponding to poor wettability of the 707 dispersed phase to the channels wall, had a negligible effect on the resulting diameter 708 of 1,6-hexanediol diacrylate droplet in polyvinyl aqueous solution, in their VOF 709 simulations. Bashir et al. (2011) also reported that the size of the resulting water 710 droplets in n-Dodecane oil solution remains constant in the super-hydrophobic (θ 711 712 \geq 165°) regime using LSM. For smaller θ , Liu and Zhang (2009) reported that the droplet interfaces are prone to be normal to the channel walls. 713

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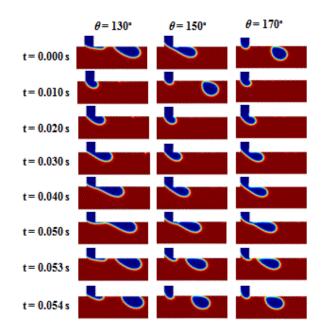
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Figure 17: Effect of contact angle on the effective droplet diameter of SCMC solutions (for system: $Q_d = 0.10$ ml/hr; $Q_c= 2.00$ ml/hr).

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Our simulations show that decreasing the contact angle promotes the formation of 719 downstream filaments. For example, in Figure 18 decreasing θ causes the formation 720 mechanism to change from corner detachment to downstream detachment via a 721 filament, because the dispersed thread has strong tendency to attach to the wall 722 surface at smaller θ . Thus a larger contact angle reduces the adhesion force between 723 the dispersed fluids with the channels wall, meaning that the cross-flow drag force 724 will detach the droplet earlier, leading to smaller droplets (Bashir et al., 2011; 725 Davidson and Cooper-White, 2006). When the concentration increases above 0.80 726

wt%, the droplet diameter increases with increasing θ , but the dependence was very 727 slight (see Figure 17). Correspondingly, Figure 19 shows higher SCMC 728 concentrations and for these high-viscosity disperse phases, the filament is present 729 even at high θ . Thus increasing θ merely shortens the filament and does not remove 730 it. As there is no change in the mechanism, the effect of θ is weaker at these higher 731 SCMC concentrations. Indeed for the highest SCMC concentrations the filament is 732 already so long that the corner has virtually no effect. The solutions, in these cases, 733 appear to be self-similar, with θ determining the downstream position of the 734 735 detachment but with little effect on the detachment dynamics. Hence, once the initial filament is formed, θ has very little effect on the droplet size. 736



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Figure 18: Snapshots of two-dimensional (2D) simulations of the effect of contact angle on 0.20 wt% SCMC-in-olive oil droplet breakup process at different wetting conditions: (a) θ =130° (b) 150° (c) θ =170° (for system: Qd = 0.10 ml/hr; Qc= 2.00 ml/hr).

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Our above findings are confirmed in Figure 20, which shows the variation in droplet diameter for various concentrations at fixed θ . Here, the droplet diameter decreases with increased SCMC concentration, for the reasons discussed in section 5.2, and θ has less influence over the droplet diameter at higher viscosity because the droplets form via a downstream filament for all θ values.

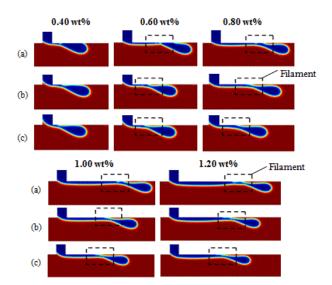
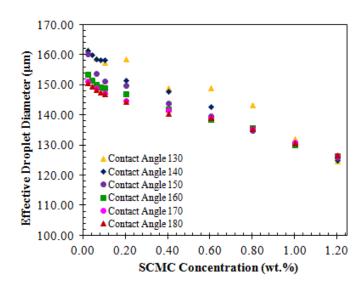


Figure 19: Filament thinning profile prior to pinch-off for SCMC concentrations above 0.40wt% at contact angle: (a) 130° , (b) 150° , (c) 170° (for system: $Q_d/Q_c=0.05$). The dashed box denotes the region where the formation of thin filament.

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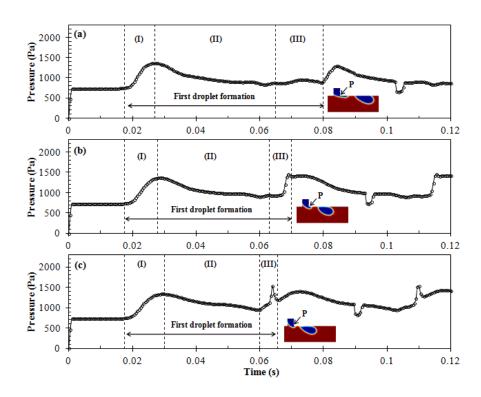
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Figure 20: Variation in effective droplet diameter of SCMC droplets for various SCMC concentrations at each wetting condition (for system: $Q_d = 0.10$ ml/hr; $Q_c = 2.00$ ml/hr).

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Figure 21 illustrates the pressure profile of 0.20 wt% SCMC droplet breakup at different wetting conditions. On the basis of the results shown in Figure 21, smaller pressure peaks for the breakup process were found for the smaller θ . This is because the wettability drives the fluid toward the surface, thus the contact area between the droplets and solid surface increases which contributes to an elongation of the filament prior to droplet pinch off.





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Figure 21: Pressure profile of 0.20 wt% CMC droplets breakup for different wetting conditions: (a) 130°, (b) 150°, (c) 170° (for system: $Q_d = 0.10$ ml/hr; $Q_c= 2.00$ ml/hr). P denotes the pressure at the corner of T-junction. The sharp peak in the pressure reflects the droplet breakup point as shown in the extracted images. The pressure peak becomes insignificant when the detachment point is located downstream of the junction.

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780 5.7 Effect of Interfacial Tension

By considering the resisting force in the process of emulsification, it is generally accepted that interfacial tension is a significant element that markedly influences the evolution of the interface when droplets are forming⁵⁶⁻⁵⁷. In our simulations, six groups

of cases with increasing values of interfacial tension from 0.010 N/m to 0.035 N/m 784 were simulated. To obtain preferentially complete wetting of the continuous phase, a 785 contact angle θ of 180° was employed. Figure 22 illustrates the interfacial tension 786 effect on the droplet diameter. The simulation predicts that the SCMC droplet 787 diameter increases when the interfacial tension is increased, which is in agreement 788 with the numerical results from Bashir et al. (2011) and Peng et al. (2011). The larger 789 interfacial tension gives rise to a stronger driving force that attempts to minimize the 790 surface area, ultimately inhibiting the break-up process. A larger interfacial force 791 792 tends to delay the droplet formation process and resists the detachment of the droplets at higher surface tension (see equation 16). Consequently, a larger droplet 793 diameter is required to balance the cross-flow drag force and the interfacial tension. 794 This is illustrated in Figure 23, which shows a sequence of snapshots of droplet 795 formation with the interfacial tension of 0.010 N/m, 0.020 N/m, and 0.035 N/m for 0.20 796 wt% concentration of SCMC. 797

798 According to Laplace's law, the interfacial tension forces necessitate a pressure drop 799 across the interface to maintain the interface at equilibrium against collapse. As 800 described by the Young-Laplace law (Adamson and Gast, 1997), the pressure difference across a fluid interface is directly proportional to the interfacial tension and 801 inversely proportional to the radius of curvature. Therefore, the higher the surface 802 tension, the larger the required pressure difference across the interface. Figure 24 803 illustrates the plot of pressure profile for the 0.04 wt% SCMC solution at the corner of 804 the T-junction at different interfacial tension. Here, higher interfacial tension leads to 805 a greater pressure peak when the droplets break-up. A study made by Bashir et al. 806 807 (2011) has reported that the tension across the interface of two phases may not be 808 sufficient to prevent filament formation at lower interfacial tensions.

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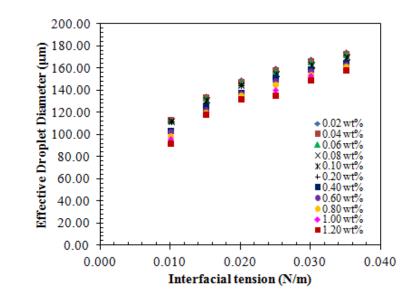


Figure 22: Effect of interfacial tension on the effective droplet diameter of SCMC solutions (for system: $Q_d = 0.10 \text{ ml/hr}$; $Q_c= 2.00 \text{ ml/hr}$).

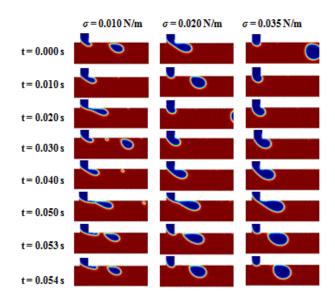


Figure 23: Snapshots of two-dimensional (2D) simulations of the effect of interfacial tension on 0.20 wt% SCMC-in-olive oil droplet breakup: (a) σ = 0.010 N/m, (b) σ = 0.020 N/m, (c) σ = 0.035 N/m (for system: Q_d = 0.10 ml/hr; Q_c= 2.00 ml/hr).

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> 1500 (II) (III) **(I**) (a) Pressure (Pa) 1000 ₹ P First droplet formation 500 0 0.02 0.04 0.06 0.08 0.1 0.12 0 2000 **(I**) (II) (III) (b) **Lressure (Pa)** 1000 200 First droplet formation 0 0.02 0.04 0.06 0.08 0.1 0.12 0 2000 (c) Pressure (Pa) 1500 1000 First droplet formation 500 **(I**) (II) (III) 0 0 0.02 0.04 0.06 Time (s) 0.08 0.1 0.12

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Figure 24: Pressure profile of during 0.20wt% SCMC droplet breakup for different interfacial tensions: (a) $\sigma = 0.010$ N/m, (b) $\sigma = 0.020$ N/m, (c) $\sigma = 0.035$ N/m (for system: $Q_d = 0.10$ ml/hr; $Q_c = 2.00$ ml/hr). P denotes the pressure at the corner of Tjunction. The sharp peak with given data values reflects the droplet breakup point as shown in the extracted images. The pressure peak becomes insignificant when the detachment point is located downstream of the junction.

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837 6. Conclusions

Emulsifying non-Newtonian solution in microfluidics in a controllable manner is one 838 of the most difficult and least understood emulsification problems. The droplet 839 formation of SCMC droplets in a Newtonian continuum was investigated for different 840 flow conditions such as the volumetric flow rates of the continuous and dispersed 841 phases, SCMC polymer concentration, continuous phase viscosity, interfacial 842 tension, and contact angle. With thorough parametric studies, the evolution of SCMC 843 droplet diameter and pressure distribution as function of these controlling parameters 844 was elaborated. 845

We varied the SCMC concentration, which modified the rheology of the disperse 846 phase, increasing the viscosity and degree of shear thinning with increased SCMC 847 concentration. We investigated the effect on droplet size. The strongest effects on the 848 droplet size come from the continuous phase properties and flow rate, the interfacial 849 tension and the contact angle. However, the disperse phase rate and rheology has a 850 smaller, but still important effect. The effect of the disperse phase rheology is manifest 851 in the viscosity ratio, which influences the mechanism by controlling the appearance 852 and length of the downstream filament and has a direct effect on the droplet diameter. 853 854 Diameters generally get smaller with increased SCMC concentration. Differences between the SCMC fluids become smaller with increased flow rates because stronger 855 shear thinning in the more concentrated fluids narrows the difference in effective 856 viscosity at higher flow rates. 857

858 All of these effects can be explained qualitatively by the idea that droplets detach when the drag on the droplet from the continuous phase balances the interfacial force 859 that joins the droplet to the bulk region of disperse phase. Increasing the continuous 860 phase viscosity or flow rate increase the drag force. Conversely, increasing the 861 862 interfacial tension increases the force resisting detachment. The effect of the disperse phase viscosity is less direct. As the disperse phase viscosity decreases, internal 863 flows within the droplet become possible and these lower the drag on the droplet. 864 This explains the direction of the changes in droplet size with η_d . However, based on 865 the modelling of Husny and Cooper-White¹¹, the effect is expected to be fairly modest 866 and may not fully explain the changes in droplet size seen in our simulations. For 867 example figure 8 shows a fairly marked drop in the diameter between de-ionised 868 water and 0.04wt% SCMC solution. This may be because, even for Newtonian fluids, 869 870 the full effect of is not correctly captured, perhaps because the walls interrupt the flow past the droplet in ways that are not included in the model of Husny and Cooper-871 White. Alternatively there may be qualitatively new effects that are introduced by 872 873 shear thinning. Candidates for these non-Newtonian flow effects include the following. We assumed that the effective viscosity inside the droplet is determined by the wall 874 875 shear rate in the continuous phase. However the internal flow in the droplet may be lower than this, meaning that the true n_d is lower than our estimate, which would lead 876 to smaller droplets. Secondly, shear thinning may drastically change the nature of the 877 878 internal flow inside the droplet, leading to quantitatively different effects on the drag.

For example, it is likely that a shear thinning fluid can localise the shear inside the droplet to the region close to the droplet interface, in an effect similar to plug-flow in a channel. This would suppress most of the internal flow and give greater drag and hence smaller droplets.

883 Finally, we contrasted our finding with the experiments of Husny and Cooper-White (2006), who studied Boger fluids. By varying the elasticity at fixed η_d , Husny and 884 Cooper-White (2006) changed the formation mechanism with negligible effect on the 885 886 droplet diameter. In contrast, by varying the viscosity and shear thinning of the 887 disperse phase, without elasticity, we showed a direct influence on the droplet size. The combination of these two studies shows the potential to tailor the droplet size 888 889 and formation mechanism by controlling the shear-thinning and elasticity through choice of polymer concentration and molecular weight. 890

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- 898 Nomenclature
- 899 a Fitting parameter in Carreau-Yasuda model
- 900 *d* Effective droplet diameter (µm)
- 901 F_{st} Surface tension force acting on the interface (N/m³)
- 902 F_D Cross flow drag force (N/m³)
- 903 *h* Depth of the channel (μm)
- 904 I Identity matrix
- 905 *k* Curvature of fluid-fluid interface

- *n* Power-law exponent for Carreau-Yasuda Model
- \mathbf{n}_{Γ} Unit normal vector at the interface
- *p* Pressure (N/m²)
- **Q** Flow rate ratio
- Q_c Flow rate of the continuous phase (ml/hr)
- Q_d Flow rate of the dispersed phase (ml/hr)
- *t* Time-step (s)
- *u* Velocity component in x-direction (m/s)
- *U* Maximum fluid velocity (m/s)
- $u \cdot \mathbf{n}$ Velocity component normal to the domain boundary (m/s)
- u_0 Initial velocity (m/s)
- *v* Velocity component in y-direction (y-direction) (m/s)
- *v** Continuous phase velocity (m/s)
- **u** Velocity field
- w_c Width of the continuous phase channel (μ m)
- w_d Width of the dispersed phase channel (μ m)
- 922 x Longitudinal coordinates axes
- 923 X Computational cell size (m)
- *y* Lateral coordinates axes

926 Greek Symbols

- η Dynamic viscosity of fluid (Pa.s)
- η_d Dynamic viscosity of the dispersed phase (Pa.s)

- η_c Dynamic viscosity of the continuous phase (Pa.s)
- η_o Zero shear viscosity (Pa.s)
- η_{∞} Infinite shear viscosity (Pa.s)
- λ_{η} Viscosity ratio (η_d/η_c)
- λ_{CY} Relaxation Time in Carreau-Yasuda Model(s)
- ρ Fluid density (kg/m³)
- 935 7 Shear stress (Pa)
- $\dot{\gamma}$ Shear rate (1/s)
- γ Reinitialization parameter (m/s)
- 938 ε Thickness of the interface (m)
- σ Interfacial tension (mN/m)
- ϕ Level set function
- δ_{sm} Dirac delta function concentrated at interface
- 942 Ω Computational domain
- $\partial \Omega$ Domain boundary

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