1 Parametric analysis of a cross-flow membrane-based parallel-plate

2 liquid desiccant dehumidification system: numerical and experimental

- 3 data
- 4 Hongyu Bai, Jie Zhu^{*}, Ziwei Chen, Junze Chu
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Department of Architecture and Built Environment, the University of Nottingham, University Park, Nottingham, NG7 2RD, UK

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10 Abstract

11 Operating parameters of a membrane-based parallel-plate liquid desiccant dehumidification 12 system are investigated in this paper. The liquid desiccant and air are in a cross-flow 13 arrangement, and separated by semi-permeable membranes to avoid carry-over problem. A 14 numerical model is developed to simulate the system performance, and validated by 15 experimental and analytical results. Impacts of main operating parameters on the system 16 performance (i.e. sensible, latent and total effectiveness) are evaluated, which include 17 dimensionless parameters (i.e. solution to air mass flow rate ratio m^* and number of heat 18 transfer units NTU), solution properties (i.e. concentration C_{sol} and temperature T_{sol}) and inlet 19 air conditions (i.e. temperature $T_{air,in}$ and relative humidity $RH_{air,in}$). It is found that m^* and 20 NTU are two of the most important parameters influencing the system effectiveness. Even 21 though the system performance can be improved by m^* and NTU, its increasing gradient is 22 limited when m^* and NTU exceed 1 and 4 respectively. Decreasing solution temperature does 23 not make a great improvement to the system performance, however, increasing solution 24 concentration is a good approach to enhance the latent effectiveness without influencing the 25 sensible effectiveness. The system shows the broad adaptability in various weather conditions, 26 and has the ability to provide relative stable state supply air.

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28 *Keywords:* liquid desiccant, membrane-based, dehumidification, numerical modelling

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35 * Corresponding author. Tel: +44 1158466141. E-mail address: jie.zhu@nottingham.ac.uk.

37 Nomenclature

Α	membrane surface area (m^2)
c_p	specific heat capacity (J/kgK)
С	concentration (%)
C_r^*	thermal capacity ratio
d	width of the rectangular channel (m)
D	diffusivity (m ² /s)
h	convective heat transfer coefficient (W/m ² K)
h_{fg}	condensation heat of water (J/kg)
h^*	operating factor
Н	height of the dehumidifier unit (m)
k	thermal conductivity (W/m K)
L	length of the dehumidifier unit (m)
m^*	solution to air mass flow rate ratio
'n	mass flow rate (kg/s)
NTU	number of heat transfer units
NTU_m	number of mass transfer units
Р	atmospheric pressure (pa)
ре	Peclet number
P_{v}	equilibrium vapour pressure of desiccant solution (pa)
Re	Reynolds number
RH	relative humidity (%)
Т	temperature (°C)
U	overall heat transfer coefficient (W/m^2K)
U_m	overall mass transfer coefficient (kg/m ² s)
\dot{V}	volumetric flow rate (l/min)
W	humidity ratio (kg/kg dry air)
Χ	solution mass fraction

Greeks

ε	effectiveness
δ	thickness of membrane (m)
ρ	density (kg/m ³)

Superscripts	
*	dimensionless
Subscripts	
air	air flow
crit	critical value
desi	desiccant
exp	experimental
in	inlet
lat	latent
т	mass transfer
тет	membrane
пит	numerical
out	outlet
sen	sensible
sol	solution flow

total

38

39 **1. Introduction**

tol

40 The global energy consumption has increased significantly in past decades as a result of 41 growing population and rapidly developing economy. Buildings contribute to a significant part 42 of the global energy consumption. In particular, heating, ventilation and air-conditioning 43 (HVAC) systems are responsible for around 50% of the energy consumed in buildings [1]. In 44 hot and humid regions, occupants would feel uncomfortable and mildew would grow on 45 building interior walls without proper air dehumidification [2]. Thus the energy efficient air 46 conditioning system is of vital importance with considerations of improving occupant thermal 47 comfort and productivity. It has been shown that the building energy consumption could be 48 decreased by 20-64% with efficient dehumidification technologies [3].

49 The traditional cooling coil system has several advantages for its ability to remove sensible heat 50 load within a conditioned space effectively, good stability in performance, long life and a 51 reasonable electrical coefficient of performance (COP) of between 2 and 4. However it is 52 inefficient in dealing with latent heat load [4, 5]. In the conventional cooling coil system, air 53 dehumidification is achieved simply by cooling the air below its dew point for condensation in 54 order to reduce its moisture content. This results in wet cooling coil surfaces that may cause 55 growths of mould and bacteria, and consequently leads to undesirable healthy issues and poor 56 indoor air quality [4, 6]. Furthermore, the air leaving the cooling coil is normally overcooled 57 and needs to be re-heated to an appropriate supply temperature. Therefore, this combined 58 process consumes a considerable amount of energy to cool (typically using a vapour 59 compression system) and heat (using hot water or electricity) the supply air [7]. Drawbacks of 60 the traditional cooling coil system can be avoided by using a liquid desiccant dehumidification 61 system. In this system dehumidification is achieved by using liquid desiccant to absorb water 62 vapour from moist air directly. The liquid desiccant system has some merits, for example, it is 63 more energy efficient, healthy and environmentally friendly than the conventional system [8-64 10]. Furthermore, it has a better ability in handling latent heat load and removing pollutants with low temperature heat sources, such as solar thermal energy and waste heat [4, 11]. The 65 66 traditional liquid desiccant system uses packed beds in the dehumidifier and regenerator, where 67 air and desiccant solution are in direct contact. In such a system small desiccant droplets are 68 carried over by the supply air to the conditioned environment, which badly affects occupant 69 health, building structure and furniture [2]. Furthermore, this system has potential drawback of 70 large pressure drop when the air flows through packed beds, which increases the operating cost. 71 Selectively permeable membrane has been used to replace packed beds as heat and mass 72 transfer medium to overcome the desiccant droplet carryover problem. In the membrane-based 73 liquid desiccant dehumidification system air and desiccant are separated by the membranes. 74 Furthermore, other harmful gases are also prevented from permeating to supply air side through 75 membranes.

76 Many researches on the membrane-based liquid desiccant dehumidification have been 77 conducted. Mahmud et al. [3] tested a novel run-around membrane energy exchanger (RAMEE), 78 which consists of two counter-cross-flow membrane energy exchangers. According to their 79 results, during summer test conditions, the total effectiveness increases with desiccant flow rate, 80 but decreases with air flow rate. By contrast, the total effectiveness changes little with air and 81 desiccant flow rates under winter test conditions. Moghaddam et al. [12] experimentally and 82 numerically evaluated the performance of liquid desiccant system. They focused on the effect 83 of thermal capacity ratio (Cr^*) on the performance of a counter-flow liquid-to-air membrane 84 energy exchanger (LAMEE), and found that numerical model agrees with experimental data 85 for the latent and total effectiveness of the LAMEE, and all effectiveness increase with Cr^* under all test conditions. They [13] also investigated the influences of various heat and mass 86 87 transfer direction and desiccant solution on the steady state performance of a small-scale 88 counter-flow LAMEE, and their research results indicate that changing the solution 89 concentration is one way to control the supply air outlet humidity ratio. Moghaddam [14] 90 experimentally and numerically studied solution-side effectiveness for a membrane energy 91 exchanger as dehumidifier and regenerator, and found that the difference between the air-side 92 and solution-side latent effectiveness is negligible. Moghaddam et al. [15] further tested the 93 performance of a small-scale single-panel LAMEE under different conditions (e.g. heating and

94 humidifying, cooling and humidifying and cooling and dehumidifying), and discovered that the 95 system effectiveness always increases with number of heat transfer units (NTU) under all test 96 conditions. Abdel-Salam et al. [16] numerically investigated the performance of a counter-flow 97 membrane liquid desiccant air-conditioning system. They focused on the effects of different 98 operating parameters on the system overall energy performance and revealed that the system 99 COP at the design condition is 0.68, while the sensible heat ratio (the ratio of the sensible to 100 total energy removed from the supply air) is in the range of 0.3 to 0.5 under different climatic, 101 operating and design conditions. Vali et al. [17] developed a numerical model to evaluate the 102 performance of a counter-cross LAMEE by considering effect of the system design parameters, 103 such as aspect ratio and entrance ratio. They found that the effectiveness is in the range between 104 the effectiveness of pure counter-flow and pure cross-flow LAMEEs with the same membrane 105 area, and the counter-cross-flow LAMEE would have the same performance as a counter-flow 106 LAMEE when the membrane area of practical design is increased by 10%. Huang et al. [18] 107 numerically and experimentally assessed the performance of a quasi-counter flow parallel-plate 108 membrane contractor, and observed that the cooling and dehumidification effectiveness are 109 deteriorated significantly compared to a cross-flow one's. Huang et al. [19, 20] also investigated 110 novel internally-cooled parallel-plate membrane contractors with cross-flow and quasi-counter 111 flow configurations, and found that their effectiveness can be significantly improved compared 112 to adiabatic one. Hollow fiber membrane contractor is another type and has attracted more 113 attention recently. Performances of hollow fiber contractors with parallel and cross flow 114 configurations have been studied in many papers [21-24]. Huang et al. [25, 26] conducted 115 several researches into an elliptical hollow fiber membrane tube bank, which is a more normal 116 case in reality, the local and average Nusselt and Sherwood numbers under the conjugate heat 117 and mass transfer boundary condition are obtained. Applications of membrane-based liquid 118 desiccant humidification in real industry have also been reported [27-29].

119 As discussed previously, many experimental studies have been carried out to investigate the 120 performance of the membrane-based system [3, 12-15, 30]. A few studies evaluate the 121 performance of liquid desiccant system numerically [12-17] by examining the effects of several 122 operating and design parameters on the effectiveness of LAMEE or the overall energy 123 performance. Several numerical investigations [18-20, 31, 32] used models that solve 124 momentum equation and continuity equation to obtain velocity field, energy and mass equation 125 to obtain concentration and temperature distributions. These studies focus on the basic heat and 126 mass transfer of a parallel plate membrane-based contractor. However, limited researches have 127 been carried out to evaluate the performance of a cross-flow membrane-based parallel plate 128 liquid desiccant system numerically by considering comprehensive operating parameters. In 129 this study, the conjugate heat and mass transfer in a cross-flow membrane-based parallel-plate 130 liquid desiccant (using lithium chloride as desiccant) dehumidification system is investigated

- by numerical modelling, the influences of three groups of main parameters (i.e. dimensionless
- parameters: NTU and solution to air mass flow rate ratio m^* , solution properties: temperature
- 133 T_{sol} and concentration C_{sol} , and inlet air conditions: temperature $T_{air,in}$ and relative humidity

134 RH_{air,in}) are assessed. Moreover, the interactions of different parameters on the system

135 performance are also investigated. This work provides a comprehensive parametric study on

- the membrane-based liquid desiccant dehumidifier performance, which supplies valuable data
- 137 for the optimum design in the dehumidification and air-conditioning systems in practice.
- 138

139 2. Numerical model

140 2.1. Governing equations

The structure of the membrane-based parallel-plate dehumidifier is depicted in Fig. 1(a), and its air and solution flows are in a cross-flow arrangement. The air and solution channels are seperated by semi-permeable membranes, thus heat and vapour can be transferred through membranes while the desiccant solution is prevented from going through them. The coordinate system used in numerical modelling is given in Fig. 1(b). As can be seen, the air and solution flow in x and y directions respectively, while heat and mass are exchanged in z direction. One air channel and one neighbouring solution channel are selected as the calculating domain.



Fig. 1. Structure of membrane-based parallel-plate dehumidifier (a), and coordinate system
used for numerical modelling (b).

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Heat and mass transfer phenomena of the dehumidifier are complicated, many factors have the influences on heat and mass transfer characteristics, such as the fluid thermal properties, vapour condensation, fluid flow rate, etc. In order to develop the numerical model, the following assumptions for simplification in this study are made:

- The dehumidifier is well-insulated, which means heat transfer between the
 dehumidifier and environment is not considered.
- 159 2) Heat and mass transfer through membrane is normal to the membrane in z direction.

- Axial conductions in both air and solution channels are neglected since Peclet numbers
 (*pe*) in the air and solution channels are much larger than 20 [33, 34].
- 162 4) Condensation heat is released to solution side only since the solution side mass transfer163 coefficient is much higher than that in the air side.
- 164 5) Both the air and solution flows are treated as laminar flows since their Reynolds 165 numbers (*Re*) are much lower than 2300 in most cases [12].
- 6) Both the air and solution flows are fully developed, while their temperature andconcentration vary along channel length only.
- 168
- 169 2.1.1. Solution side governing equations
- 170 Heat and mass balance equations in the solution side are given as:

171
$$\left(\frac{m_{sol}}{L} \cdot \frac{\partial T_{sol}}{\partial y} \cdot C_{p,sol}\right) \cdot dxdy = \left[U(T_{air} - T_{sol}) + h_{fg} \cdot U_m (W_{air} - W_{sol,mem})\right] dxdy$$
(1)

172
$$\frac{\dot{m}_{desi}}{L} \cdot \frac{\partial X_{sol}}{\partial y} \cdot dx dy = U_m \cdot (W_{air} - W_{sol,men}) dx dy$$
(2)

173 Where in equations, \dot{m}_{sol} is solution mass flow rate (kg/s); \dot{m}_{desi} is desiccant mass flow rate 174 (kg/s); *L* is the length of dehumidifier (m) as illustrated in Fig. 1(a); T_{sol} is solution 175 temperature (°C); T_{air} is air temperature (°C); W_{air} is air humidity ratio $(kg/kg \, dry \, air)$; 176 $W_{sol,men}$ is humidity ratio of membrane surface on solution side $(kg/kg \, dry \, air)$; X_{sol} is 177 solution mass fraction, which is calculated as:

178
$$X_{sol} = \frac{m_{water}}{m_{desi}} = \frac{1 - C_{sol}}{C_{sol}}$$
(3)

179 Where *C*_{sol} is solution mass concentration:

$$180 C_{sol} = \frac{m_{desi}}{m_{sol}} (4)$$

181 h_{fg} is water condensation heat (J/kg); $C_{p,sol}$ is solution specific heat capacity (J/kgK); 182 $U(W/m^2K)$ and $U_m(kg/m^2s)$ are heat transfer and mass transfer coefficients respectively, 183 which are given by:

184
$$U = \left(\frac{1}{h_{air}} + \frac{\delta}{k_{mem}} + \frac{1}{h_{sol}}\right)^{-1}$$
(5)

185
$$U_m = \left(\frac{1}{h_{m,air}} + \frac{\delta}{k_{m,mem}}\right)^{-1}$$
(6)

186 Where h_{air} and h_{sol} are convective heat transfer coefficients in air and solution sides 187 respectively (W/m^2K) ; $h_{m,air}$ is air side mass transfer coefficient (kg/m^2s) ; δ is membrane 188 thickness (m); k_{mem} (W/mK) and $k_{m,mem}(kg/ms)$ are membrane thermal conductivity and 189 mass transfer conductivity respectively.

- 190 2.1.2. Air side governing equations
- 191 Similarly, heat and mass balance equations in the air side are given below:

192
$$\frac{\dot{m}_{air}}{H} \cdot C_{p,air} \cdot \frac{\partial T_{air}}{\partial x} + U(T_{air} - T_{sol}) = 0$$
(7)

193
$$\frac{\dot{m}_{air}}{H} \cdot \frac{\partial W_{air}}{\partial x} dx + U_m (W_{air} - W_{sol,mem}) dx dy = 0$$
(8)

194 Where *H* is height of the dehumidifier (*m*), as shown in Fig. 1(a); $C_{p,air}$ is air specific heat 195 capacity (J/kgK).

196

197 2.2. Normalization of governing equations

198 Governing equations (1)(2)(7)(8) are normalized as:

199
$$\frac{\partial T_{sol}^{*}}{\partial y^{*}} - NTU_{m}h^{*}\frac{1}{cr^{*}}\left(W_{air}^{*} - W_{sol,mem}^{*}\right) - NTU\frac{1}{cr^{*}}\left(T_{air}^{*} - T_{sol}^{*}\right) = 0$$
(9)

200
$$\frac{\partial X_{sol}}{\partial y^*} - NTU_m \frac{1}{m^*} W_0 (1 + X_{sol}) (W_{air}^* - W_{sol,mem}^*) = 0$$
(10)

$$201 \qquad \frac{\partial T_{air}^*}{\partial x^*} + NTU(T_{air}^* - T_{sol}^*) = 0 \tag{11}$$

$$202 \quad \frac{\partial W_{air}^*}{\partial x^*} + NTU_m \left(W_{air}^* - W_{sol,mem}^* \right) = 0 \tag{12}$$

Where in the above equations, the following dimensionless properties have been defined.Dimensionless length is defined by:

$$205 \qquad x^* = \frac{x}{L} \tag{13}$$

206 Dimensionless height is defined by:

$$207 \qquad y^* = \frac{y}{H} \tag{14}$$

208 Dimensionless temperature is defined by:

209
$$T^* = \frac{T - T_{air,in}}{T_0}$$
 (15)

- 210 Where T_0 is equal to $(T_{sol,in} T_{air,in})$.
- 211 Dimensionless humidity ratio is defined by:

212
$$W^* = \frac{W - W_{air,in}}{W_0}$$
 (16)

- 213 Where W_0 is equal to $(W_{sol,in} W_{air,in})$.
- 214 m^* is mass flow rate ratio, which is defined by:

$$215 \qquad m^* = \frac{\dot{m}_{sol}}{\dot{m}_{air}} \tag{17}$$

216 Cr^* is thermal capacity ratio, which is defined by:

217
$$Cr^* = \frac{(\dot{m}c_p)_{sol}}{(\dot{m}c_p)_{air}}$$
(18)

218 h^* is operating factor, which is a dimensionless number defined by:

219
$$h^* = \frac{W_0}{T_0} \frac{h_{fg}}{c_{p,air}}$$
(19)

220 NTU and NTU_m are numbers of heat and mass transfer respectively, which are defined by:

$$221 NTU = \frac{UA}{(\dot{m}c_p)_{air}} (20)$$

$$222 NTU_m = \frac{U_m A}{m_{air}} (21)$$

- 223 Where A is total membrane area (m^2) .
- 224

225 2.3. Boundary conditions

226 Boundary conditions for the solution side are:

227
$$T_{sol}^* = 1$$
, at $y^* = 0$ (22)

228
$$X_{sol} = X_{sol,in}$$
, at $y^*=0$ (23)

229 While the air side boundary conditions are:

230
$$T_{air}^* = 0$$
, at $x^* = 0$ (24)

231
$$W_{air}^* = 0$$
, at $x^* = 0$ (25)

232

233 2.3.1. Heat transfer boundary conditions on membrane surfaces

To solve the governing equations, heat and mass transfer boundary equations on membrane surfaces need to be established. Heat transfer boundary conditions are based on thermal energy balance through the membrane:

237
$$h_{sol}(T_{sol,mem} - T_{sol}) = U(T_{air} - T_{sol,mem}) + h_{fg}U_m(W_{air} - W_{sol,mem})$$
(26)

Eq. (26) can be normalized as:

239
$$NTU_{sol}(T_{sol,mem}^* - T_{sol}^*) = NTU(T_{air}^* - T_{sol,mem}^*) + NTU_m h^*(W_{air}^* - W_{sol,mem}^*)$$
 (27)

240 Where NTU_{sol} is number of heat transfer unit in solution side and defined by:

241
$$NTU_{sol} = \frac{h_{sol}A}{(mc_p)_{air}}$$
(28)

242

243 2.3.2. Mass transfer boundary conditions on membrane surfaces

244 Similarly, mass transfer boundary conditions are based on mass balance through the membrane:

245
$$U_m(W_{air} - W_{sol,mem}) = h_{m,sol}(C_{sol} - C_{sol,mem})$$
(29)

Eq. (29) can be normalized as:

247
$$NTU_m W_0 \left(W_{air}^* - W_{sol,mem}^* \right) = NTU_{m.sol} \left(C_{sol} - C_{sol,mem} \right)$$
(30)

248 Where $C_{sol,mem}$ is solution concentration in the interface between the solution and membrane

surface; $NTU_{m.sol}$ is number of mass transfer unit in the solution side, which is defined by:

$$250 \qquad NTU_{m.sol} = \frac{h_{m,sol}A}{\dot{m}_{air}} \tag{31}$$

251 Where $h_{m,sol}$ is solution side mass transfer coefficient (kg/m^2s) .

252

253 2.4. Air and solution property equations

- In numerical modelling, the air specific humidity or humidity ratio $(kg/kg \ dry \ air)$ is derived
- from its relative humidity by applying a correlation introduced in [35].

The solution equilibrium specific humidity (W_{sol}) is used to calculate both the sensible and latent effectiveness, the relationship between the specific humidity and vapour pressure is given by [36]:

259
$$W_{sol} = 0.62198 \frac{P_v}{P - P_v}$$
 (32)

260 Where *P* is atmospheric pressure (*Pa*) and P_v is vapour pressure of desiccant solution (*Pa*).

261 The equilibrium vapour pressure of desiccant solution is a function of T_{sol} and C_{sol} ($P_v = f(T_{sol}, C_{sol})$), the correlation is given by [37]:

263
$$Log P_{v} = KI \left[A - \frac{B}{T - E_{s}} \right] + \left[C - \frac{D}{T - E_{s}} \right]$$
(33)

264 Where P_v is solution equilibrium vapour pressure (kPa), K is an electrolyte parameter relating

to solute; A, B, C, D and E_s are parameters regarding to solvent. A psychrometric chart of LiCl

solution is plotted and shown in Fig. 2.

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268

269 270

Fig. 2. Psychrometric chart of LiCl.

271 2.5. *Performance evaluation*

272 Effectiveness is the most important parameter used to evaluate the performance of a heat and 273 mass exchanger [38]. Three types of effectiveness have been defined in this study: sensible 274 effectiveness (ε_{sen}), latent effectiveness (ε_{lat}) and total effectiveness (ε_{tot}). ε_{sen} is the ratio 275 between the actual and maximum possible rates of sensible heat transfer in a heat exchanger. 276 ε_{lat} is the ratio between the actual and maximum possible moisture transfer rates in a mass 277 exchanger. ε_{tot} is the ratio between the actual and maximum possible energy (enthalpy) 278 transfer rates in a heat and mass exchanger. The capacity rate of desiccant solution is higher 279 than that of the air, which means $Cr^* \ge 1$, then the sensible, latent and total effectiveness are 280 defined by Eqs. (34) - (36) [39].

$$281 \qquad \varepsilon_{sen} = \frac{T_{air,in} - T_{air,out}}{T_{air,in} - T_{sol,in}} \tag{34}$$

$$282 \qquad \varepsilon_{lat} = \frac{W_{air,in} - W_{air,out}}{W_{air,in} - W_{sol,in}} \tag{35}$$

283
$$\varepsilon_{tol} = \frac{\varepsilon_{sen} + H^* \varepsilon_{lat}}{1 + h^*}$$
(36)

Where $T_{air,out}$ is air temperature at the outlet (°C) and $W_{air,out}$ is air humidity ratio at the outlet ($kg/kg \, dry \, air$).

286

287 **3. Simulation procedure**

- 288 3.1. Discretization of governing equations
- Governing equations in section 2.1 are solved by finite difference method, and discretized by a
 forward difference scheme. Discretization equations are given below:

291
$$T_{sol(m+1,n)}^{*} - T_{sol(m,n)}^{*} - dy^{*}NTU_{m}h^{*}Cr[W_{air(m+1,n)}^{*} - W_{sol,mem(m+1,n)}^{*}] -$$

292 $dy^{*}NTUCr[T_{air(m+1,n)}^{*} - T_{sol(m+1,n)}^{*}] = 0$ (37)

293
$$X_{sol(m+1,n)} - X_{sol(m,n)} - dy^* m^* W_0 NTU_m [1 + X_{sol(m+1,n)}] [W_{air(m+1,n)}^* - W_0 NTU_m [1 + X_{sol(m+1,n)}]]$$

294
$$W_{sol,mem(m+1,n)}^* = 0$$

295
$$T_{air(m,n+1)}^* - T_{air(m,n)}^* + dx^* NTU[T_{air(m,n+1)}^* - W_{sol(m,n+1)}^* = 0$$
 (39)

(38)

296
$$W_{air(m,n+1)}^* - W_{air(m,n)}^* + dx^* NTU_m [W_{air(m,n+1)}^* - W_{sol,mem(m,n+1)}^*] = 0$$
(40)

297 Where m is number of girds in x direction, and n is number of girds in y direction.

Since the air and desiccant solution are closely interacted with each other, the governing equations are solved in Matlab iteratively until converged. In order to guarantee the accuracy of numerical results, numerical tests have been conducted to determine the grid size. It has been found that 30×60 grids are adequate in this study, the result difference is less than 1.0% compared with 50×100 grids. The numerical uncertainty is 1.0%.

303

304 *3.2. Numerical solving scheme*

- 305 The numerical solution scheme used to solve interacted governing equations are given below:
- Set the initial temperature and concentration fields for air and solution as boundary
 conditions.
- 308 2) Assume the initial humidity ratio on the membrane surface as the solution inlet309 equilibrium specific humidity.
- 310 3) Solve the energy equation (9) to get the solution temperature field (T_{sol}^*) .
- 311 4) Figure out the mass equation (10) to obtain the solution concentration field (X_{sol}) .
- 312 5) Deal with the energy equation (11) to acquire the air temperature field (T_{air}^*) .
- 313 6) Solve the mass equation (12) to get the air humidity field (W_{air}^*) .

- 314 7) Based on the temperature and humidity fields for air and solution flows, solve heat and 315 mass transfer boundary conditions on the membrane surface (26) and (27) to obtain the 316 membrane surface temperature and concentration fields in the solution side $(T_{sol.mem})^*$ 317 and $C_{sol.mem}$).
- 8) Calculate the membrane surface humidity field in the solution side $(W_{sol,mem}^*)$ based 318 319 on $T_{sol,mem}^*$ and $C_{sol,mem}$.
- 9) Adopt new $W_{sol,mem}^*$ as a default value and return to step 7 until $W_{sol,mem}^*$ is 320 321 converged.
- 10) Return to step 3 with the new $W_{sol,mem}^*$ until T_{sol}^* , X_{sol} , T_{air}^* and W_{air}^* are 322 323 converged.
- 324

4. Experimental work 325

326 In order to assess the performance of a membrane-based parallel-plate liquid desiccant 327 dehumidification system, a test facility is designed and built in the laboratory, which is depicted 328 in Fig. 3.



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330

331

Fig. 3. Schematic diagram of the laboratory test rig.

332 The test rig mainly consists of a dehumidifier, a regenerator, two solution tanks and three heat 333 exchange units. The outdoor air with high temperature and relative humidity is generated in the 334 environmental chamber, and it flows into the dehumidifier where both its moisture content and

- temperature are reduced by cold desiccant solution, then it leaves the dehumidifier unit at a dry
- and cool state. Its flow rate is controlled by adjusting an AC axial fan rotation speed (ebm-papst
- 337 Mulfingen GmbH & Co. KG). The dehumidifier has a dimension of 410mm (L) x 230mm (W)
- 338 x 210mm (H) with 11 air channels and 11 solution channels. The regenerator has the same
- 339 structure as the dehumidifier. Three gauze layers are paved on the top surface of the
- 340 dehumidifier unit to ensure even solution distribution. The dehumidifier specifications and
- 341 membrane physical properties are given in Table 1.

342 **Table 1**

343 Dehumidifier specifications and membrane physical properties.

Symbol	Unit	Value
L	т	0.41
W	т	0.23
Н	m	0.21
d_{air}	т	0.0077
d_{sol}	m	0.0043
δ_{mem}	т	0.5×10 ⁻³
k _{mem}	W/mK	0.3
km, mem	kg/ms	3.87×10 ⁻⁶

344

Lithium chloride (LiCl) is used as the desiccant in the system. The desiccant solution is circulated in the system by two identical pumps (15W centrifugal magnetically driven type with flow rate range of 0-10L/min) and their flow rates are measured by two liquid flow indicators (Parker UCC PET 1-15 L/min). The desiccant solution and air transport properties are listed in Table 2.

Table 2

351 Air and desiccant solution transport properties.

Symbol	Unit	Value
k _{air}	W/mK	0.03
k_{sol}	W/mK	0.53
D_{air}	m^2/s	2.46×10 ⁻⁵
D_{sol}	m^2/s	0.892×10 ⁻²
$C_{p,air}$	J/kgK	1020
$C_{p,sol}$	J/kgK	3200
$ ho_{air}$	kg/m^3	1.29
$ ho_{sol}$	kg/m ³	1247

352

Air velocities through the dehumidifier and regenerator are measured at the air duct outlets by a thermo-anemometer (Testo 405) with a measuring range up to 10m/s. All fans at the inlets of the dehumidifier and regenerator are equipped with infinitely variable speed controllers to adjust air flow rates. All air inlets and outlets are instrumented with humidity and temperature sensors (Sensirion Evaluation KIT EK-H4). The desiccant solution and water temperatures are measured with K-type thermocouples, and all sensors are connected to a DT500 data logger. The dehumidifier, regenerator, heat exchangers, storage tanks and pipes are well insulated to

- 360 reduce the environment influence. All measurement devices and their accuracies are listed in
- 361 Table 3. Uncertainty analysis has been conducted for all experimental data by applying a
- 362 method of propagation introduced by Taylor [40] to estimate uncertainties for experimental
- 363 data.
- **Table 3**
- 365 Measurement devices and uncertainties.

Device	Measurement	Range	Uncertainty
Testo thermos-anemometer	Air velocity	0-10 m/s	±5%
Sensiron Evaluation KIT EK-H4	Temperature	-40-125 °C	$\pm 0.4\%$
	Relative humidity	0-100 %	±3%
K-type thermocouple probe	Temperature	0-1100 °C	$\pm 0.75\%$
DT500 Datalogger	Data acquisition	-	$\pm 0.15\%$
Parker UCC PET liquid flow indicator	Solution flow rate	1-15 L/min	$\pm 5\%$
Parker liquid flow indicator	Water flow rate	2-22 L/min	±2%

³⁶⁶

367 **5. Results and discussion**

368 5.1. Model validation

Analytical solutions and experimental data are used to validate the numerical results. 12 groups of experimental data under different operating conditions are used to validate numerical results. Under each operating condition, the numerical calculation and experimental results of sensible effectiveness (ε_{sen}) and latent effectiveness (ε_{lat}) are compared, as shown in Table 4. It can be seen that generally numerical modelling results of both ε_{sen} and ε_{lat} agree well with experimental data. The maximum discrepancy between numerical results and experimental data for ε_{sen} is 8.756%, while the maximum discrepancy for ε_{lat} is 9.822%.

377 Table 4

378 Comparisons between numerical results and experimental data.

Operating conditions			Comparisons						
NTU	m*	m _{air} (kg/s)	m _{sol} (kg/s)	E _{sen,num}	E _{sen,exp}	Error (%)	Elat,num	E _{lat,exp}	Error (%)
2	1	0.1216	0.1216	0.5699	0.520	8.756	0.5101	0.460	9.822
2	2	0.1216	0.2431	0.7113	0.671	5.665	0.5311	0.500	5.856
2	3	0.1216	0.3647	0.7608	0.708	6.940	0.5380	0.508	5.576
2	4	0.1216	0.4862	0.7861	0.735	6.500	0.5413	0.511	5.597
4	1	0.0608	0.0608	0.6276	0.576	8.221	0.7226	0.653	9.632
4	2	0.0608	0.1216	0.8116	0.753	7.220	0.7632	0.742	2.778
4	3	0.0608	0.1823	0.8704	0.810	6.940	0.7756	0.756	2.527
4	4	0.0608	0.2432	0.8990	0.849	5.562	0.7816	0.750	4.043
8	1	0.0304	0.0304	0.7376	0.686	6.996	0.8778	0.818	6.812
8	2	0.0304	0.0608	0.9043	0.857	5.231	0.9270	0.916	1.187
8	3	0.0304	0.0912	0.9440	0.894	5.297	0.9396	0.918	2.299
8	4	0.0304	0.1216	0.9608	0.918	4.455	0.9452	0.928	1.820



381

Fig. 4. Comparisons among numerical modelling, experimental data and analytical solutions
 for sensible and latent effectiveness under different NTU values: NTU=2(a); NTU=4(b);
 NTU=8(c).

The numerical modelling results and experimental data are presented in Fig. 4, the numerical modelling results follow the same trends of experimental data for both the sensible and latent effectiveness under different *NTUs*. Based on the uncertainty analysis, which is shown as error

- bars for experimental data, it can be seen that numerical modelling results are within the tolerance range of experimental data. So it can be concluded that the agreements between the numerical modelling and experimental data for both the sensible and latent effectiveness are satisfied. It should be emphasized that under each NTU, the discrepancy reduces with the solution mass flow rate. This is because that the lower of solution mass flow rate, the greater
- influence of solution mal-distribution on the effectiveness.
- Analytical solutions to the model of an enthalpy exchanger with membrane core have been presented in several literatures. According to these literatures, the sensible effectiveness is a function of two dimensionless parameters (*NTU* and Cr^*) for an unmixed cross flow, and given as [41,42]:

399
$$\varepsilon_s = 1 - exp\left[\frac{\exp\left(-NTU^{0.78}C_r^{*-1}\right) - 1}{NTU^{-0.22}C_r^{*-1}}\right]$$
(41)

400 Similar to the sensible effectiveness, the latent effectiveness is calculated as [43]:

401
$$\varepsilon_l = 1 - exp\left\{\frac{NTU_m^{0.22}}{m^{*-1}}\left[exp\left(-m^{*-1}NTU_m^{0.78}\right) - 1\right]\right\}$$
 (42)

402 Results of the analytical solution are also given in Fig. 4. The numerical modelling and 403 analytical solution results have similar variation trends, an acceptable agreement between them 404 is achieved. Based on the comparisons, it is valid to predict the performance of the membrane-405 based cross-flow dehumidifier by the numerical model.

406

407 5.2. Temperature and humidity fields

408 Temperature and humidity fields of the air and solution channels, and the membrane surface 409 are obtained based on the numerical model. The distributions of temperature and humidity ratio in the air channel and membrane surface under NTU = 8 and $m^* = 2$ are plotted in Fig. 5. The 410 411 inlet temperatures of the air and solution are 30°C and 20°C respectively, while the inlet air 412 relative humidity and solution concentration are set as 70% and 39% respectively. It is observed 413 that the air has the lowest temperature (20.00°C) and humidity ratio (0.0042 kg/kg dry air) 414 at the right top corner of the air outlet. This is because the air at the top side of the dehumidifier 415 interacts with stronger and cooler desiccant solution. Furthermore, the air becomes cooler and drier along the air channel length. It is noticed that the air temperature at left bottom corner of 416 417 the dehumidifier is slightly higher than its inlet temperature. This is owing to the increase of 418 the solution temperature during the dehumidification process.



420 Fig. 5. Air temperature field (a); air humidity ratio field (b); temperature field on membrane
421 surface (c); humidity ratio field on membrane surface (d).

- Furthermore, Fig. 5 (c) and (d) reflect the temperature and humidity boundary conditions on the membrane surface, it is clear that the boundary condition is neither uniform temperature nor uniform humidity ratio. They are both non-uniform and two-dimensional profiles, the temperature and humidity ratio decrease along the diagonal line of the membrane surface.
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428 5.3. Effects of dimensionless parameters

- 430

431 **Fig. 6.** Variations of effectiveness: (a) sensible effectiveness; (b) latent effectiveness; (c) total

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433 434 The variations of sensible, latent and total effectiveness with m^* and NTU are given in Fig. 6. 435 The maximum values of sensible, latent and total effectiveness are 0.9849, 0.9845 and 0.9846 436 respectively when $m^* = 4$ and NTU = 12. The minimum values of sensible, latent and total

effectiveness with m^* and NTU.

- 437 effectiveness are 0.3232, 0.4682 and 0.4353 when $m^* = 0.5$ and NTU = 2. Separate effects of
- m^* and *NTU* are discussed then.
- m^* is the relative mass flow rate of two fluids in the dehumidifier, and changed by adjusting
- solution mass flow rate while keeping air mass flow rate under each NTU. The variations of
- sensible, latent and total effectiveness with m^* at different NTUs are depicted in Figs. 7-9.



Fig. 7. Sensible effectiveness variations with m^* under different NTUs.



Fig. 8. Latent effectiveness variations with m^* under different NTUs.



450

Fig. 9. Total effectiveness variations with m^* under different NTUs.

451 It is evident that m^* has significant influences on sensible, latent and total effectiveness, which 452 all increase with m^* . The gradient of change for the sensible effectiveness is more considerable 453 than that for latent and total effectiveness. For instance, at NTU = 8, the sensible effectiveness rises from 0.2917 to 0.9608 as m^* increases from 0.5 to 4. In the meanwhile, the latent and total 454 455 effectiveness vary from 0.7476 to 0.9452 and from 0.6441 to 0.9487 respectively. However, 456 the gradients of their changes become moderate gradually and only a slight variation is observed 457 once m^* exceeds 1. Take the latent effectiveness as an example, under NTU = 4, the 458 effectiveness increases by 43.15% as m^* rises from 0.25 to 1.0, while the effectiveness only 459 increases by 8.16% when m^* rises from 1.0 to 4.0. Therefore a critical value of m^* can be defined as m_{crit}^* , and the effectiveness are more sensitive to m^* when m^* is lower than m_{crit}^* . 460 461 Once m^* exceeds m^*_{crit} , there is not much significant change any more. Similar trends can be found in literatures [43, 44], in which their results show that both sensible and latent 462 463 effectiveness increase with m^* when $m^* < 1$, and they are nearly constant when $m^* \ge 1$. Similar to m_{crit}^* , another important indicator Cr_{crit}^* has been introduced in literatures [45, 46], 464 as all effectiveness increase with Cr^* and are more sensitive before Cr^* reaching a critical 465 value. This is easily explainable since Cr^* is proportional to m^* . Compared to Cr^* , m^* is a 466 467 more straightforward parameter for the system. As a result, it is desirable to maintain the dehumidification system operating at a condition where m^* is close to m^*_{crit} . It is also worth 468 469 mentioning that the gradient of change with m^* gets smoothly at low NTU. This is more 470 obvious for the latent and total effectiveness. For example, the latent effectiveness increases by 471 36.21% (i.e. from 0.3974 to 0.5413) when m^* varies from 0.25 to 4 under NTU = 2, while 472 under NTU = 12, an growth of 73.63% (i.e. from 0.5670 to 0.9845) is observed for the same

473 m^* range. This means there is hardly benefit by increasing m^* at low *NTU* to improve the 474 system performance.

475 On the other hand, NTU is another important dimensionless parameter affecting the 476 effectiveness. In literature [46], NTU is treated as the most important parameter with the most significant impact on the dehumidification system. Compared to the flow rate, the non-477 478 dimensional group NTU is a comprehensive indicating parameter because it eliminates the 479 impact of channel geometric properties. In numerical modelling, NTU is changed by adjusting 480 air mass flow rate, while the solution mass flow rate is changed proportionally to maintain a 481 constant m^* accordingly. Variations of effectiveness with NTU under $m^* = 4$ are shown in Fig. 482 10.

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484

485

Fig. 10. Variations of effectiveness with NTU under $m^* = 4$.

487 All effectiveness are seriously affected by NTU as indicated in Fig. 10. Under the same m^* , the sensible effectiveness is the highest while the latent effectiveness is the lowest among the three 488 489 effectiveness. Similar to the influences of m^* , the three effectiveness vary more remarkably 490 when NTU is in the range of 1 to 4. Once NTU exceeds 4, variations of these effectiveness tend 491 to level off. For instance, under $m^* = 4$, the sensible effectiveness increases by 53.02% (i.e. 492 from 0.5875 to 0.899) when NTU changes from 1 to 4, conversely it only increases by 9.56% 493 (i.e. from 0.899 to 0.9849) when NTU varies from 4 to 14. Cases are similar for the latent and 494 total effectiveness. Thus similar to m_{crit}^* or Cr_{crit}^* , a critical value of NTU exists and is defined as NTU_{crit}. Increasing NTU beyond NTU_{crit} would not enhance the system efficiency. 495 496 Furthermore, it is observed from Figs. 7-9 that when m^* is low, there is hardly benefit to 497 improve these effectiveness by increasing NTU. This is more obvious for the sensible effectiveness. For instance, under $m^* = 0.5$, increasing NTU from 2 to 12 will only slightly 498

- 499 enhance the sensible effectiveness from 0.3232 to 0.3311. However, under $m^* = 4$, the sensible
- 500 effectiveness improves from 0.7861 to 0.9849 when NTU increases from 2 to 12. Therefore 501 there is hardly significance to increase NTU at low m^* for performance improvement.

To sum up, these effectiveness increase with m^* and NTU, but their increase gradients are limited as $m^* > m_{crit}^*$ (i.e. 1 in this study) and $NTU > NTU_{crit}$ (i.e. 4 in this case). Effects of m^* and NTU are interacted on each other. Under a relatively low NTU there is hardly benefit to increase m^* , especially for the latent effectiveness. By contrast, with low m^* , no obvious performance improvement could be achieved by increasing NTU, especially for the sensible effectiveness.

- 508
- 509 5.4. Effects of solution properties

510 Solution temperature (T_{sol}) and concentration (C_{sol}) affect the system performance, variations

511 of sensible, latent and total effectiveness with T_{sol} and C_{sol} are presented in Fig. 11, while *NTU*

512 and m^* are set as 8 and 2 respectively.



513

514 **Fig. 11.** Variations of effectiveness: (a) sensible effectiveness; (b) latent effectiveness; (c) 515 total effectiveness with T_{sol} and C_{sol} .

516

517 The sensible effectiveness reaches the maximum value of 0.9458 when $T_{sol} = 12^{\circ}$ C and $C_{sol} =$ 518 21%, while its minimum value is 0.8848 when $T_{sol} = 22^{\circ}$ C and $C_{sol} = 39\%$. The maximum latent effectiveness is 0.9366 when $T_{sol} = 12^{\circ}$ C and $C_{sol} = 39\%$, and its minimum value is 519 0.8449 when $T_{sol} = 22$ °C and $C_{sol} = 21$ %. For the total effectiveness, it reaches the maximum 520 value of 0.9380 when $T_{sol} = 12^{\circ}$ C and $C_{sol} = 39\%$, while the minimum value is 0.8757 when 521 $T_{sol} = 22^{\circ}$ C and $C_{sol} = 21\%$. Effects of T_{sol} and C_{sol} on the system performance are 522 523 discussed separately then. 524 Solution temperature has a great influence on the system performance since it is closely related

525 Solution temperature has a great influence on the system performance since it is closely related 525 to the solution surface vapour pressure. The variations of the effectiveness with T_{sol} under 526 different C_{sol} are shown in Figs. 12-14.

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531 532

Fig. 12. Sensible effectiveness variations with T_{sol} under different C_{sol} .





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Fig. 13. Latent effectiveness variations with T_{sol} under different C_{sol} .

1.00 0.95 Total effectiveness 0.90 0.85 0.80 0.75 21% Δ 33% Exp. Data 39% Exp. Data 0.70 20 12 14 16 18 22 Temperature (°C)

Fig. 14. Total effectiveness variations with T_{sol} under different C_{sol} .

538 It is clearly reflected in Figs. 12-14 that all effectiveness decrease accordingly with the solution 539 temperature. The decrease of sensible effectiveness is easily understood since the increase of 540 solution temperature narrows the temperature difference between the solution and air. The 541 decrease of latent effectiveness is because vapour pressure in the solution side increases with 542 the solution temperature, which weakens the dehumidification potential. Although decreasing 543 the solution temperature can improve the sensible effectiveness, the improvement is not 544 significant. The sensible effectiveness is weakened by increasing dehumidification effect since 545 more latent heat released to the solution side would increase the solution temperature, which 546 would affect the sensible effectiveness negatively. It also should be mentioned that at a low 547 solution concentration, the sensible effectiveness variation is almost negligible. For instance, 548 the sensible effectiveness increases by only 1.58 % (i.e. from 0.9311 to 0.9458) when the solution temperature decreases by 10 °C under $C_{sol} = 21\%$. Thus when the solution 549 concentration is low, decreasing solution temperature will not improve the sensible 550 551 effectiveness significantly. The case is opposite for the latent effectiveness as it is more 552 sensitive to T_{sol} when C_{sol} is low. For example, under $C_{sol} = 21\%$, the latent effectiveness 553 increases by 5.3% (i.e. from 0.8449 to 0.8896) when the solution temperature decreases from 554 22°C to 12°C. By contrast, it only increases by 1.33% (i.e. from 0.9243 to 0.9366) under $C_{sol} =$ 555 39%. This indicates that the latent effectiveness is more sensitive to solution temperature at a 556 low solution concentration. Nevertheless, the effect of T_{sol} on the latent effectiveness is insignificant compared to C_{sol} as discussed subsequently. 557

Solution concentration affects the system performance as it is directly related to surface vapour pressure. However, increasing C_{sol} has different impacts on the sensible, latent and total effectiveness, which can be observed from Fig. 15.



Fig. 15. Variations of effectiveness with C_{sol} under $T_{sol} = 18^{\circ}$ C.

- 563 As presented in Fig. 15, increasing solution concentration from 21% to 39% would decrease 564 the sensible effectiveness from 0.9366 to 0.9176, and increase the latent and total effectiveness 565 from 0.8656 to 0.9296 and from 0.8916 to 0.9266 respectively. Similar to the effects of T_{sol} , 566 increasing solution concentration will decrease the solution surface vapour pressure, thus the 567 solution absorption ability will be enhanced and the latent effectiveness will be increased. The increased dehumidification ability will negatively affect the sensible effectiveness since more 568 569 latent heat will be released to the solution side. The total effectiveness is mainly dominated by 570 the latent effectiveness since they have the same variation trends. It can be observed in Fig. 12 571 that the sensible effectiveness is neither sensitive to T_{sol} nor C_{sol} . For the latent effectiveness, 572 the effect of solution concentration is far more obvious than that on the sensible effectiveness. 573 For instance, at $T_{sol} = 12^{\circ}$ C, the sensible effectiveness decreases by 0.51% (i.e. from 0.9458) 574 to 0.941) when C_{sol} changes from 21% to 39%, while the latent effectiveness increases by 5.28% 575 (i.e. from 0.8896 to 0.9366). The latent effectiveness improvement is more significant at a 576 higher solution temperature. For example, at $T_{sol} = 22^{\circ}$ C, the latent effectiveness increases by 9.40% (i.e. from 0.8449 to 0.9243) when C_{sol} varies from 21% to 39%. 577
- To sum up, the sensible effectiveness is insensitive to both T_{sol} and C_{sol} , while the latent effectiveness is comparatively more sensitive. Thus in practical applications, a low solution temperature is not necessary since it is energy consuming and no significant performance improvement could be achieved. In the meanwhile, at a relatively high solution temperature, increasing C_{sol} will improve dehumidification ability more considerably. It implies that increasing solution concentration is a better approach to improve the latent effectiveness without decreasing the sensible effectiveness.
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586 5.5. Effects of inlet air condition

587 Inlet air condition (i.e. temperature $T_{air,in}$ and relative humidity $RH_{air,in}$) is of vital 588 importance to evaluate the adaptivity of the system. Variations of the sensible, latent and total 589 effectiveness with $T_{air,in}$ and $RH_{air,in}$ are presented in Fig. 16, while *NTU* and m^* are set as 8 590 and 2 respectively.





The maximum sensible effectiveness is 0.924 when $T_{air,in} = 36$ °C and $RH_{air,in} = 60\%$, while the minimum value is 0.8818 when $T_{air,in} = 27$ °C and $RH_{air,in} = 75\%$. Comparatively, the latent effectiveness reaches the maximum value of 0.9289 when $T_{air,in} = 27$ °C and $RH_{air,in} =$ 75%, while the minimum value is 0.9222 when $T_{air,in} = 39$ °C and $RH_{air,in} = 60\%$. For the total effectiveness, the maximum value is 0.9249 when $T_{air,in} = 33$ °C and $RH_{air,in} = 60\%$, while its minimum value is 0.9174 when $T_{air,in} = 39$ °C and $RH_{air,in} = 75\%$. Then effects of $T_{air,in}$ and $RH_{air,in}$ on the system performance are discussed separately.

601





Fig. 17. Sensible effectiveness variations with $T_{air,in}$ under different $RH_{air,in}$.

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Fig. 18. Latent effectiveness variations with $T_{air,in}$ under different $RH_{air,in}$.





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Fig. 19. Total effectiveness variations with $T_{air,in}$ under different $RH_{air,in}$.

611 The effectiveness variations with $T_{air.in}$ under different $RH_{air.in}$ are shown in Figs. 17-19, it 612 can be seen that the sensible effectiveness is more sensitive to $T_{air.in}$ and $RH_{air.in}$ compared 613 with the latent and total effectiveness. When $T_{air,in}$ increases from 27°C to 36°C, the sensible 614 effectiveness gradually rises and reaches the peak value at 36°C. This is because that increasing 615 $T_{air,in}$ at a constant $RH_{air,in}$ will enhance the heat transfer potential [16]. However the sensible effectiveness starts to decrease slightly when $T_{air,in}$ is higher than 36°C. For instance, under 616 $RH_{air,in} = 75\%$, the sensible effectiveness rises by 2.42% (i.e. from 0.8818 to 0.9031) when 617 $T_{air,in}$ increases from 27°C to 36°C. From 36°C to 39°C, the sensible effectiveness decreases 618 619 slightly (i.e. from 0.9031 to 0.9018). For the latent effectiveness, the effect of $T_{air.in}$ is negligible, for example, under $RH_{air,in} = 75\%$, the latent effectiveness decreases by 0.74% 620 (i.e. from 0.9289 to 0.9220) when $T_{air,in}$ rises from 27°C to 39°C. This is attributed to the 621 increased membrane moisture resistance [46]. The case is similar for the total effectiveness, an 622 623 obvious relationship between the total effectiveness and $T_{air,in}$ can hardly be found. Effect of 624 $RH_{air,in}$ on the sensible effectiveness is clear, the lower $RH_{air,in}$, the higher sensible effectiveness. Furthermore, the improvements of the sensible effectiveness by decreasing 625 626 $RH_{air.in}$ at different inlet air temperatures are basically similar. For example, by decreasing 627 RH_{air,in} from 75% to 60%, the sensible effectiveness improvements are 3.22%, 2.63%, 2.39%, 628 2.31% and 2.31% respectively at Tair.in of 27°C, 30°C, 33°C, 36°C and 39°C. For the latent and 629 total effectiveness, similar to the effects of $T_{air,in}$, they can barely be affected by $RH_{air,in}$ as 630 well. In other words, the latent and total effectiveness are insensitive to both $T_{air,in}$ and 631 RH_{air.in}.

- 632 In fact, the inlet air state depends on the local weather condition, thus $T_{air,in}$ and $RH_{air,in}$ are 633 input parameters for the system rather than controllable ones. Fig. 20 illustrates inlet and outlet 634 air conditions on the psychrometric chart.
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Fig. 20. Inlet and outlet air conditions on the psychrometric chart for numerical modelling.

640 In traditional cooling coil air-conditioning system, the inlet air is dehumidified by reducing its 641 temperature below its dew point temperature, afterwards the air relative humidity is normally high, which means its temperature is too low for the supply air requirement, and subsequently 642 643 re-heating is required and more energy is consumed. As seen in Fig. 20, both inlet air 644 temperature and humidity content are reduced after passing through the dehumidifier, and the 645 system outlet air has relatively low temperature and relative humidity. Furthermore, the outlet 646 air is at the very similar condition in spite of different inlet conditions with the same NTU, and 647 this is more obvious at high NTU. This means this system has broad adaptability in different 648 weather conditions, and can produce relative stable state supply air.

To sum up, the influences of $T_{air,in}$ and $RH_{air,in}$ on the latent and total effectiveness are negligible, while their effects on the sensible effectiveness are weak as well but evident. The cross-flow membrane-based parallel-plate liquid desiccant dehumidification system has better air-conditioning ability to provide the supply air with relatively stable condition despite diverse outdoor conditions.

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657 **6. Conclusions**

A numerical model is developed to investigate the performance of a cross-flow membrane-658 659 based parallel-plate liquid desiccant dehumidification system, the interaction between the 660 solution and air through membranes is studied by solving heat and mass governing equations 661 in Matlab. The influences of main parameters on dehumidification effectiveness (sensible, latent and total effectiveness) are assessed respectively, which include: number of heat transfer 662 units (NTU), solution to air mass flow rate ratio (m^*), solution temperature (T_{sol}), solution 663 concentration (C_{sol}), inlet air temperature ($T_{air.in}$) and relative humidity ($RH_{air.in}$). The 664 conclusions can be drawn as follows: 665

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668

- m^* and NTU are two of the most important parameters influencing the system effectiveness. Although all effectiveness increase with m^* and NTU, their increasing gradients hardly change when m^* and NTU exceed m^*_{crit} and NTU_{crit} respectively.
- It is desirable to operate the system at the critical condition where m_{crit}^* and NTU_{crit} 670 are 1 and 4 respectively in this study.
- Effects of m^* and NTU on the system performance are interacted with each other. There is hardly benefit to the system performance improvement by increasing m^* at low NTU, especially for the latent effectiveness. By contrast, no obvious performance enhancement is achieved by increasing NTU at low m^* , especially for the sensible effectiveness.
- There is no obvious improvement in the sensible effectiveness with low temperature and high concentration solution, which is insensitive to both T_{sol} and C_{sol} .
- The latent effectiveness increases significantly with C_{sol} at a high solution temperature,
 while the system sensitive effectiveness does not decrease observably.
- The system has broad adaptability in different weather conditions by providing relative
 stable state supply air, in particular at high *NTU*.
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