Fabrication of Novel Transparent Co₃O₄- TiO₂ nanowires p-n heterojunction diodes for Multiband Photodetection Applications

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Abstract:

Axial Co₃O₄ - TiO₂ heterojunction nanowires (NWs) were synthesized by glancing angle deposition (GLAD) technique. The p-n heterojunction showed excellent rectification ratio of 2.26×10^2 at ± 3.4 V. The forward turn on voltage of 1.5 V in the dark was reduced to 1 V under white light excitation on the device. The diode showed a maximum half-wave rectification efficiency of 7.77% at 200 Hz frequency operated with maximum ± 10 V. The device showed maximum peak responsivity of 4.01 A/W and internal gain of 13.1 at 380 nm wavelength. The detectivity was calculated to be 2.82×10^{11} and 1.69×10^{11} Jones and the noise equivalent power was estimated to be 14.9 and 24.8 pW at 380 and 620 nm wavelength, respectively. The device spatial response showed sharp transition with rise and fall time of ~ 0.17s and 0.21s, respectively.

Keywords:

Semiconductors, optical materials, oxide materials, thin films, electron-phonon interactions, photoconductivity and photovoltaics

1. Introduction

Metal oxide semiconductors have become dominating materials in the field of nanoscience research due to higher efficiency, diverse properties (ferromagnetic, ferroelectric, supercapacitive, semiconducting and superconducting) and relatively easy preparation methods [1-3]. The electrical, optical, physical, chemical and other properties of the oxide semiconductors can be altered very easily as per requirement by simply tuning the bandgap structure through doping impurities, morphology, Schottky or Ohmic contact formation etc. [4-6]. The majority of these oxide semiconductors are inherently n-type such as TiO₂, ZnO, SnO₂, Fe₂O₃, In₂O₃ etc. [7] due to the interstitial defects and related oxygen vacancies. Considerable efforts have been paid to synthesize transparent homojunctions by synthesis of p-type counterpart of the mentioned material. However, due to the instability of the p-type oxides, fabrication of the transparent homojunction is still a challenge to fabricate. A p-n heterojunction is a junction between two materials with different bandgaps. These structures have the superiority to prevent the generated electron-hole pairs to recombine due to the presence of bandgap discontinuity at the p-n interface. It is important to mention that there are some inherently p-type oxide semiconductors such as Cr₂O₃, CuO, NiO, Co₃O₄, Mn₃O₄ etc [8]. Several heterojunction combinations of these materials have been studied vastly for gas sensing purposes. Among these materials Co₃O₄ has a direct bandgap varying between 2 to 2.8 eV which falls in the visible region [9-11], and is able to absorb the visible portion of solar radiation spectrum. Co₃O₄ and its heterojunctions with TiO₂ may be able to reveal a wide range of solar spectrum from UV to near infrared region. Research on nanowires nowadays are expanding at a very rapid rate due to some of their outstanding properties such as guidance of light owing to high refractive index, ability to form interpenetrating co-axial

and longitudinal heterojunction interface and mostly for strong light trapping [12] on the front panel of the optoelectronic devices. Previously, synthesis of Co_3O_4 was reported by solvothermal process [13], hydrothermal process [14], sol-gel method [15], and oxidation of cobalt [16]. The Glancing Angle Deposition (GLAD) method is a cost effective technique to grow different nanostructure morphology by simply altering the substrate azimuthal rotation speed, angle of deposition and rate of deposition [17]. Hence, the investigation of optoelectronic properties of GLAD synthesized Co_3O_4 - TiO₂ longitudinal nanowire heterojunctions is carried out.

In this article, we report the synthesis of Co_3O_4 - TiO_2 longitudinal heterojunction nanowires grown by GLAD technique. The absorption and transmission spectra were measured and analyzed. Room and low temperature measurements of the current densityvoltage characteristics have been made and half wave rectification efficiency of the p-n heterojunctions and ideality factors were calculated, respectively. In addition, the spectral responsivity and spatial response measurements of the detector were performed.

2. Experimental

2.1. Synthesis of Co₃O₄-TiO₂ nanowire heterojuntion (NWH)

Co₃O₄ pellets were prepared from Co₃O₄ nanopowder (Sigma Aldrich-637025) without using any binder by a KBr hydraulic press under a pressure of 60 Kg/cm² for 10 minutes duration. The formed pellets were then sintered inside a quartz tube furnace (GSL-1700X, MTI, USA) in open air condition at 600° C for 30 minutes using even heating and cooling ramp of 4°C/min. These pellets were used to deposit Co₃O₄ nanowires (NWs) inside an electron beam evaporator. Prior to the deposition, n-Si <100> (30 Ω -cm) substrates were cleaned using trichloroethylene, methanol, acetone, and a solution of hydrofluoric acid and de-ionized water at 1:50 volume ratio. A 50 nm thick TiO₂ film was grown on the substrate using an electron beam evaporator (BC-100, Hind High vacuum Co. (p) Ltd.) at a base pressure of 10^{-6} mbar. Next, 250 nm long perpendicular TiO₂ NWs were grown inside the chamber of the same ebeam evaporator using GLAD technique. The substrate holder was kept at a distance of 24 cm from the evaporated material source at an orientation of 85° with respect to the perpendicular line between the source and the substrate, having an azimuthal rotation of 10 rpm. This was followed by 250 nm Co₃O₄ NWs deposited from Co₃O₄ pellet source using the same experimental setup mentioned above. Finally, 30 nm thick gold (Au) was evaporated to form the top electrode using an aluminium mask with hole area of 1.77×10^{-6} m². The growth rate was kept constant at 1.2 Å s⁻¹ for deposition of TiO₂, Co₃O₄ and Au.

2.2 Characterization

Scanning electron microscopy (SEM) (Sigma, Zeiss) and energy dispersive X-ray (EDX) (ZEISS EVO-MA 10) analysis were performed on the samples to investigate the morphological orientation. The transmission electron microscopy (TEM) (JEM-2100, 200 kV, Jeol) investigation with selected area electron diffraction analysis were carried out on the samples. A UV-Visible near-infrared optical absorption spectrophotometer (Lambda 950, Perkin Elmer) was used to measure the optical absorption and transmission spectra of the samples. The p-n junction half wave rectification was observed by connecting the junction diodes in series with a 1 M Ω resistance and using a digital storage oscilloscope (TBS1062, Tektronix). The current density (J) vs. voltage (V) and photocurrent measurements were performed using a Keithley 2401 source-measure unit, a 300 W xenon arc lamp and a monochromator (Sciencetch Inc. Canada) in open beam configuration. The transient time response of the device was measured under white light on-off irradiation. The temperature dependent J-V characteristics from 50K-300K were measured using a closed cycle helium cryostat (Advanced Research System Inc.).

3. Results and discussion

Fig. 1 shows the top view SEM micrograph of Co_3O_4 -TiO₂ nanowire heterojuntion (NWH) synthesized on indium tin oxide (ITO) coated glass (Fig. 1a) and n-Si substrates (Fig. 1b). The images show that the NWs, grown on ITO coated glass and n-Si substrates, have typically diameter ranges of 40-50 nm and 40-60 nm, respectively. Fig. 1c reveals the EDX spectrum for the Co_3O_4 -TiO₂ NWH. The EDX spectrum verifies the presence of titanium (Ti), cobalt (Co) and oxygen (O₂) in the prepared samples.



Fig. 1: Top view SEM images of TiO₂-Co₃O₄ NW arrays prepared on (a) ITO coated glass and (b) n-Si substrate; (c) EDX spectrum

The morphology of the NWH has been investigated separately using TEM characterization technique to verify the formation of perpendicular NWs. The color contrast in the cross-

section micrograph confirms the formation of the heterojunction between TiO₂ and Co₃O₄. The observation reveals that TiO₂ NWs have a diameter of ~20 nm at bottom which gradually increases to 50 nm at the interface of TiO₂ and Co₃O₄. The top diameter of Co₃O₄ was measured to be 40 nm. The height of TiO₂ and Co₃O₄ were 170 nm and 90 nm, respectively. The selected area electron diffraction (SAED) analyses of Co₃O₄ show its amorphous nature (Fig. 2 c, d) with diffuse concentric rings, while TiO₂ shows polycrystalline nature with bright concentric spotty rings having typical d spacing of 0.21 nm attributed to (210) plane (Fig. 2e, f) of rutile TiO₂ [18]



Fig. 2. TEM images of the Co_3O_4 -TiO₂ NWH (a) NWs arrays, (b) individual NW, (c) magnified view of upper portion (Co_3O_4) of the heterojunction, (d) SAED of Co_3O_4 section, (e) magnified view of lower portion (TiO₂) of the heterojunction, (f) SAED of TiO₂ section

3.2. Optical Characterization

Fig. 3a shows the room temperature UV-Vis optical absorption spectra for $TiO_2-Co_3O_4$ heterostructure deposited on ITO coated glass substrate. The peak at 380 nm may be due to the main band transition in TiO_2 [19] whereas the small shoulder at 425-445 nm can be assigned to the absorption via intrinsic surface states [20]. The broad peak at 620 nm may be

ascribed to the main band transition in Co₃O₄ [10]. The $(\alpha hv)^2$ vs *hv* characteristics for TiO₂-Co₃O₄ heterostructure on ITO coated glass and n-Si substrates are represented in Fig. 3b where α is the absorption coefficient and *hv* is the photon energy. Extrapolation of linear part of the spectra yielded band gap values of 3.29 eV, 2.75 eV, 2.25 eV and 1.6 eV for ITO coated glass substrate based samples and 3.28 ev and 2.23 eV for n-Si substrate based samples. Among them, 3.29 and 3.28 eV are due to the main band transition in TiO₂ [19] whereas the value at 2.75 eV (450.9 nm) is very close to the reported value of TiO₂ (450 nm) [21]. The 2.25 eV and 2.23 eV energy may be attributed to the band to band transitions due to oxygen to Co (II) and Co (III) metal to ligand charge transfer [22], and the 1.6 eV band is due to indirect bandgap transition in Co₃O₄ [23]. Fig. 3c shows the room temperature optical transmittance spectra for the TiO₂-Co₃O₄ heterostructure. In a range between 375 nm - 800 nm, the heterostructure allowed 73% photon to pass through it.





Fig. 3: UV-Vis optical spectra of TiO₂- Co₃O₄ NWs (a) absorption, (b) $(\alpha hv)^2$ versus energy curve for sample prepared on ITO coated glass and n-Si substrate (inset) (c) transmission *3.3.* Co₃O₄ - TiO₂ heterostructure based p-n diode characteristics

The axial Co_3O_4 - TiO₂ heterojunction NWs based diode (HND) is fabricated on TiO₂ TF coated n-Si substrate as shown schematically in Fig. 4a. A transparent top Ohmic contact was made of Au on Co₃O₄ NWs and Al was used as back contact on the Si substrate. James E. Moore et al reported a conduction band offset of 0.4 eV at the CdS/CIGS interface [24] and 0.15 eV for a-Si:H/c-Si heterojunction solar cell [25]. We have calculated the conduction band offset of 0.4 eV and valence band offset of 0.64 eV for the Co₃O₄ - TiO₂ heterointerface (Fig.4b), where the electron affinity of Co₃O₄ was considered as 1.99 eV using Vegard's law from the electron affinity of CoO_n [26].This band offset is expected to drive the photogenerated electron-hole pair to the oppositely charged electrodes [27].



Fig. 4: (a) Schematic diagram of the prepared HND, (b) device band diagram at equilibrium

Fig.5a shows the current versus voltage (I-V) characteristic of the Au/Co₃O₄/TiO₂ HND at room temperature under dark and white light exposure. The forward bias is considered here as a positive DC voltage applied on the Au top electrode. The maximum rectifying ratio measured between -10 V to +10 V has a value of 2.26×10^2 at ± 3.4 V (Fig.5.a inset) which is 3.8 times superior to the reported value of 59 for ZnO/Si nano-heterojunction array [28]. A turn-on voltage of 1.5 V is recorded for dark, which reduced to 1 V under white light excitation. Large current was recorded for white light excitation, which may be due to barrier height lowering and multiple carriers generation. To understand the AC operation of the HND, we have measured the rectification efficiency under applied AC voltage with maximum \pm 10 V at various frequencies from 200 Hz – 2 kHz. The output voltage vs time characteristics at 200 and 1000 Hz are displayed in Fig. 5b. A maximum efficiency of 7.77% was observed at 200 Hz frequency which gradually decreased with increase in signal frequency as shown in Fig. 5c. It is well known that the maximum efficiency for ideal diode halfwave rectifier is 40.6%. In our case, the low rectification efficiency is due to the presence of non-zero forward voltage drop [29]. Furthermore, it was observed that (Fig. 5b) with increase in frequency, the clipping abilities of the diode started deteriorating which may be due to the prominent reverse recovery transient related to the high frequency operation of



Co₃O₄/TiO₂ HND junction diode [30]. Hence our HND is suitable for low frequency operation.

Fig. 5: (a) Room temperature current density-voltage characteristic under dark and white light illumination for Co_3O_4 -TiO₂ NW device, inset shows the rectification ratio as a function of applied potential; (b) output and input characteristics comparison of half-wave rectifier at 200 and 1000 Hz; (c) Rectification efficiency-frequency spectrum for the heterostructure device.

The current density versus voltage (J-V) characteristics of the Au/Co₃O₄/TiO₂ HND at 50 K - 300 K are displayed in Fig. 6a. It can be clearly seen that with an increase in temperature there is an enhancement in the current through the diode. Averagely 6.87×10^2 times enhancement in current was achieved at 300 K compared with 50 K.

The ideality factor of the HND was calculated using the following formula at different temperatures [31]

$$n = \frac{q}{kT} \left(\frac{dV}{d(\ln(I))} \right)^{*}$$
(1)

where n=ideality factor; q=charge of electron; k=Boltzmann's constant; T=temperature; V= voltage; I= current through the HND

Fig. 6b shows the variation of the ideality factor with change in temperature from 50 K - 300 K. The ideality factor was observed to decrease with an increase in temperature. In the Co₃O₄ – TiO₂ HND the electrons gain more energy to cross the barrier height the temperature is increased. Hence, the effective barrier height is increased [32] resulting in an influence over the current transport along with thermionic emission [33]. Due to an increase in temperature, the energy difference between conduction band and Fermi level decreases resulting in an increase in flat band barrier [34,35]. Hence, a very high ideality factor (196) was observed at 50 K which gradually decreased with increase in temperature (15 at 300 K).



Fig. 6: (a) Current density-voltage characteristic for Co_3O_4 -TiO₂ NW device at different temperatures from 50-300 K; (b) Ideality factor-temperature characteristic for Co_3O_4 -TiO₂ NWs device

Fig. 7 shows the room temperature responsivity vs wavelength and internal gain vs wavelength spectra (inset) for Co_3O_4 -TiO₂ HND at + 5 V applied bias. The internal gain was calculated using Eq. (2) [36].

$$G = \frac{h \times c \times R_{\lambda}}{\lambda \times \eta \times e} \tag{2}$$

where h = Planck's constant, c = speed of light, R_{λ} = responsivity, λ = wavelength of light, η = quantum efficiency ~1 and e = electron charge.

The maximum peak responsivity of 4.01 A/W and internal gain of 13.1 at 380 nm wavelength were due to the main band gap related transition in TiO_2 [16]. A small peak responsivity of 2.4 A/W and internal gain of 4.81 both at 620 nm wavelength were both due to main band gap related transition in Co_3O_4 [10].



Fig. 7: Room temperature responsivity spectra at +5 V; inset shows the internal gainwavelength spectrum at +5 V)

The detectivity and noise equivalent power (NEP) are two important parameters for a photodetector to evaluate its performance. Detectivity is given by

$$D^* = \frac{R_{\lambda}}{\sqrt{2eJ_{dark}}} \tag{3}$$

where J_{dark} is the dark current density.

The NEP is expressed as

$$NEP = \frac{\sqrt{A}\sqrt{B}}{D^*} \tag{4}$$

Where *A* is the area of the device $(1.77 \times 10^{-2} \text{ cm}^2)$ and *B* is the bandwidth (1 kHz).

With these relations, detectivity and NEP were plotted as functions of operating voltage in a range of 3-7 V at 380 and 620 nm. These characteristics are displayed in Fig. 8. At 5 V, the detectivity was calculated to be 2.82×10^{11} and 1.69×10^{11} Jones under 380 and 620 nm wavelength photon illumination, respectively. The NEP at 5 V was estimated to be 14.9 and 24.8 pW under 380 and 620 nm photon illumination.



Fig. 8: Detectivity and noise equivalent power versus applied bias voltage characteristics under (a) 380 nm, (b) 620 nm wavelength illumination

Fig. 9a depicts the current density as a function of time under white light on/off switching irradiation at +5 V for the Co₃O₄ - TiO₂ HND. The rise and fall time were calculated using the following formulas [37]

$$I_{photo} = I_{dark} + A_1 \left[exp(-\frac{t}{\tau_{1_{fall}}}) \right] + B_1 \left[exp(-\frac{t}{\tau_{2_{fall}}}) \right]$$
(5)

$$I_{photo} = I_{dark} + A_2 [1 - exp(-\frac{t}{\tau_{1_rise}})] + B_2 [1 - exp(-\frac{t}{\tau_{2_rise}})]$$
(6)

where I_{photo} is photocurrent, I_{dark} is dark current, A₁, B₁, A₂ and B₂ are scaling constants, and τ_1 and τ_2 are the time constants. By fitting the photoresponse (Fig. 9b) using the mentioned equations, the time constants for decay were estimated to be $\tau_{1_{fall}} = 4 \times 10^{-3}$ s and $\tau_{2_{fall}} = 0.25$ s. For rise, the same parameters were estimated to be $\tau_{1_{rise}} = 3 \times 10^{-3}$ s and $\tau_{2_{rise}} = 0.2$ s. The time constant $\tau_{1_{fall}}$ and $\tau_{1_{rise}}$ may be associated with carrier generation and recombination in nanowires, and $\tau_{2_{fall}}$ and $\tau_{2_{rise}}$ may be related to the trapping and release process of carriers. [37]



Fig. 9: (a) White light on-off switching irradiation of heterostructure device at +5 V; (b) Equation fitting of the rising and decaying photocurrent under white light illumination

4. Conclusion

In conclusion, we have prepared Co_3O_4 - TiO₂ HND using the GLAD technique. The TiO₂ exhibited a main band and sub-band related transitions at 3.29 and 2.75 eV, respectively. For Co_3O_4 a band to band and indirect band gap related transitions were observed at 2.25 and 1.6 eV. The HND showed a rectification ratio of 2.26×10^2 at ± 3.4 V. The turn on voltage of 1.5 V at dark was reduced to 1 V with light excitation. The rectification efficiency of the half-wave rectifier using the diode showed a decrease with an increase in operating frequency

having a maximum value of 7.77% at 200 Hz. At 380 nm wavelength, the maximum peak responsivity of 4.01 A/W and internal gain of 13.1 were recorded due to main band gap related transition in TiO₂. For Co₃O₄, at 620 nm wavelength excitation a small peak in responsivity of 2.4 A/W and internal gain of 4.81 were observed due to main band gap related transition. At 5 V, the detectivity was calculated to be 2.82×10^{11} and 1.69×10^{11} Jones under 380 and 620 nm wavelength photon illumination, respectively. The NEP at 5 V was estimated to be 14.9 and 24.8 pW under 380 and 620 nm wavelength photon illumination, respectively. From white light on-off switching, the rise and fall time were calculated to be 0.2 and 0.25 s. Therefore, the diode can be used as multiband detector for the detection of UV and visible light simultaneously.

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