

Perspective:

All surfaces are not equal in contact transmission of SARS-CoV-2

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Summary

The world faces a severe and acute public health emergency due to the ongoing Coronavirus Disease 2019 (COVID-19) global pandemic caused by the Severe Acute Respiratory Syndrome Coronavirus 2 (SARS-CoV-2). Healthcare workers are in the front line of the COVID-19 outbreak response and are exposed to the risk of SARS-CoV-2 infection daily. Personal protective equipment (PPE) is their main defence against viral contamination; gloves, visors, face masks and gown materials are designed to eliminate viral transfer from infected patients. Here we review research investigating the stability of SARS-CoV-2 and similar viruses on surfaces, and highlight opportunities for materials that can actively reduce SARS-CoV-2 surface contamination and associated transmission and improve PPE.

Introduction

Respiratory diseases caused by viruses have become a serious global public health concern, in particular, as new viruses emerge. The most well known pandemic in recent history is the Spanish flu pandemic, caused by a strain of the influenza A virus known as H1N1, which led to at least 50 million killed worldwide in 1918.¹ Swine flu, caused by H1N1, was responsible for the pandemic in 2009 to 2010, which spread rapidly from country to country.² Additionally, recent coronavirus outbreaks that caused the respiratory diseases called Severe Acute Respiratory Syndrome (SARS, the virus known as SARS-CoV-1) originated in China in 2002 and Middle East Respiratory Syndrome (MERS, the virus known as MERS-CoV) first identified in the Middle East in 2012, both caused major disruption and threatened people's life and livelihood.^{3,4}

The ongoing COVID-19 pandemic, the latest example, has spread rapidly since December 2019. Although the rate of new infections in some countries has reduced significantly recently, other countries are still facing exponential growth. Like other human respiratory viruses, SARS-CoV-2 is reported to spread primarily in small droplets released when an infected person sneezes or coughs, although conversation and breathing have recently been highlighted as potential routes for virus transmission.⁵ While larger droplets are expected to land on surrounding surfaces within a relatively short timeframe, some smaller ones (<5 µm in diameter) may remain airborne and potentially travel larger distances, for example, up to 7-8 m if turbulent cloud is created.^{6,7} Consequently, the virus is envisaged to spread through direct inhalation of virus laden droplets or, more rarely, aerosols, or by hand contact with contaminated surfaces and subsequent transfer to mucus membranes. The infectious dose of SARS-CoV-2, namely the average number of the viral particles required to establish an infection for COVID-19, is unknown, but data from other respiratory viruses such as influenza, indicate that the initial viral dose is directly correlated with the severity of disease symptoms.⁸

The best approach to prevent viral infections is vaccination, and antiviral drugs are the only treatment option once infected. However, at the moment in the absence of an effective vaccine or drug, and

the majority of the population unexposed or without immune protection after recovering from COVID-19,⁹ there is considerable risk of largescale future outbreaks across the world. Healthcare workers are in the front line of the COVID-19 outbreak response and are exposed to the risk of SARS-CoV-2 virus infection daily. PPE is their main defence against viral contamination. In addition to the effort made on vaccines and drugs, new and effective prevention strategies, for example the development of effective antiviral protective materials for PPE or environmental infection control, potentially reducing the contact transmission of virus pandemics in the future, need to be developed urgently.

SARS-CoV-2 and surfaces

Viruses cannot reproduce themselves outside the host, relying instead, on the host's cellular machinery to produce RNA and to build proteins for their own use. A completely assembled virus, ready to infect a host, is known as a virion. The basic structure of a virion usually composes a nucleic acid (RNA or DNA) core and a protein capsid to protect their nucleic acid. Some viruses, such as coronaviruses, have an additional lipid envelope. The first step of host infection/cell entry is virus attachment to host tissue by recognizing and binding to cell surface receptors with their externally displayed proteins. It is anticipated that the viral surface proteins also play a role when viruses interact with manmade substrates outside a host.

Viral attachment to manmade surfaces is expected to be a function of the physicochemical properties of the material. The binding of viruses on synthetic surfaces has typically been described using non-specific electrostatic¹⁰⁻¹³ and hydrophobic¹¹⁻¹⁴ interactions. Early studies have shown that the viral survival and their ability to infect mammalian cells increases with the amount of adsorption to the contact transfer surfaces.¹⁵ However, more recent studies have demonstrated that strong irreversible adsorption on hydrophobic and polycationic surfaces damaged and inactivated viruses.¹¹⁻¹³ To date, there is still limited information regarding the correlation of virus attachment to materials and their stability outside a host. Virus stability and their infectivity in ambient conditions are affected by a combination of the biology: type of virus; the environment: temperature, light, relative humidity (RH); and physicochemical characteristics of the surface along with local environment factors including carrier liquid identity (saliva/mucus), toxic agents, pH and salts.¹⁶ It is therefore difficult to compare among published studies as experimental conditions, methods and types of viruses vary in the conditions chosen to model this complex situation, in the absence of information on the conditions dominant in transmission. The classification of surfaces as either 'soft' or 'hard' has gained attention. 'Soft surfaces' e.g. cardboard, paper, fabric, have been grouped since they have been observed to support infectious virus for shorter periods whilst 'hard surfaces', e.g. plastic and steel have been shown to support active virus for longer periods.^{17,18} Here we look in more detail at the importance of polymer surfaces, and outline what research is required to choose optimal materials to reduce contact transmission. We also consider what material aspects influence adsorption and virus survival and in the context of PPE suggest what improvements can be achieved.

It has been reported that the transmission of SARS-CoV-1 (the most closely related human coronavirus to SARS-CoV-2) starts only after symptoms develop and is associated with detectable viral loads in the respiratory tract that reach a peak after about 10 days.¹⁹ In contrast, SARS-CoV-2 virus has been widely detected in the upper respiratory tract in the absence of symptoms,²⁰ highlighting the possibility of viral spread through microdroplets (bioaerosol) generated during breathing or conversation.⁵ These asymptomatic cases have been shown to have a similar viral load to symptomatic patients.²¹ Like other coronaviruses, SARS-CoV-2 particles are spherical and have proteins protruding from their surfaces with the spike-protein being the most characteristic. Binding of spike protein to the angiotensin-converting enzyme 2 (ACE2) receptors on human cell surfaces, drives structural changes to the protein, which facilitates membrane fusion and entry of the nucleocapsid into the cell cytoplasm.^{22,23} It has been found that the SARS-CoV-2 spike protein has between a 10- and 20-fold increased affinity to the ACE2 receptor, compared to SARS-CoV-1, which might facilitate efficient replication in upper respiratory tract and enable more efficient human transmission.²⁴ Whilst the major routes of transmission are not fully understood it is likely that transfer of virus from contaminated surfaces plays a major role. Importantly, virus-laden aerosols may also be generated by the doffing (removing) of PPE, through cleaning, or via the movement of staff, so there could be a benefit to a surface which encourages strong binding.^{25,26}

Critically, it is unclear how long the SARS-CoV-2 virus can remain viable outside the host. Viral degradation and inactivation on surfaces is influenced by environmental factors such as humidity, temperature and light, but critically also the identity of the surface.¹⁵ A review article²⁷ published early this year summarized research on the persistence of coronavirus on a range of potential contact transfer surfaces: SARS-CoV-1 and MERS-CoV were found to remain infectious on metal (steel),^{28,29} glass²⁸ and plastic (type not specified)²⁸⁻³⁰ for up to 9 days, under ambient conditions.³¹ Higher temperature and humidity appear to hasten viral degradation on surfaces, for example, SARS-CoV-1 was reported to remain infectious on plastic (polystyrene) at room temperature and RH of 40 – 50% with only 1 log₁₀ loss of titer after 5 days, while a 0.25 – 2 log₁₀ loss of titer was observed at 38 °C and 80 – 90% RH within 24 h. Although the surface state of the 24-well plastic plate was unknown, for tissue culture applications oxygen modification of the polystyrene surface is often used.³² More recently, human coronavirus 229E (HCoV-229E) was inoculated onto a variety of surfaces, including steel, polytetrafluoroethylene (Teflon, PTFE), polyvinyl chloride (PVC), ceramic tiles, glass, and silicone rubber. The virus remained infectious for at least 5 days on all surfaces, as shown by a plaque assay which measures the infection of host cells in vitro³³, but reduced to 3 days for silicon rubber.³⁴ Additionally, influenza A viruses, the respiratory viruses which cause avian flu, were found to remain detectable up to 2 weeks on stainless steel, and up to 1 week on cotton and microfibre under ambient condition.³⁵ Instead of using quantitative Polymerase Chain Reaction (qPCR) to measure the amount of viral nucleic acid, which could arise from inactivated viruses, in most of these studies, the fifty-percent-tissue-culture-infective-dose (TCID₅₀) end-point titration and viral plaque assays were used. This measures the viral infectivity in tissue culture cells and are therefore represents a more relevant functional quantification of infectious virus on surfaces.

A more recent study³⁶ compared the stability of SARS-CoV-2 and SARS-CoV-1 in aerosols and on various surfaces at 21 to 23 °C and 40% relative humidity (65% relative humidity for aerosols). A virus dose of 50 µl of 10⁵ TCID₅₀/mL, which has been shown by qPCR to be equivalent to viral loads present in the upper respiratory tract of infected individuals,²¹ was placed on a variety of surfaces. Due to the nature of liquid absorption on cardboard, the inoculum was recovered from this by first swabbing the surface, and then extracting with media from the swab at predetermined time-points, while those on the other materials were recovered directly from the surfaces. All samples were then quantified by TCID₅₀ end-point titration on Vero E6 cells. In particular, the researchers estimated the decay rates of viable viruses on surfaces using a Bayesian regression model, which allowed them to account for differences in the sources of experimental noise including the initial inoculum level. Under the experimental conditions, SARS-CoV-2 remained detectable up to 4 hours on reddish copper (99.9%, Metal Remnants), up to 24 hours on cardboard, and up to 2 to 3 days on plastic (polypropylene) and AISI 304 alloy stainless steel. SARS-CoV-1 showed similar stability on these tested surfaces to SARS-CoV-2, with shorter viability times reported for cardboard. More comprehensive follow-up studies, for example experiments performed at varying levels of temperature and relative humidity, are being planned by the authors.

Polymers and SARS-CoV-2

The majority of PPE (gloves, gowns, visors and face masks), and indeed many work surfaces where contact transfer may occur, are polymeric. Disposable polymer gloves are ubiquitous in clinical settings to help prevent cross-contamination between patients and healthcare workers and to protect against hazardous chemicals and bacteria/viruses. For instance, healthcare workers wear medical gloves as a barrier to contact with blood, other body fluids, wounds or mucus membranes and the skin of patients, and higher risk surfaces to reduce the chance of bacteria/virus transfer. Currently, there are three main types of commercially available medical gloves, including latex, nitrile and vinyl gloves. Specifically, latex gloves, made of natural rubbers with polyisoprene as their primary chemical constituent, offer a great deal of flexibility, comfort and fit for healthcare givers who perform sensitive work. However, as the number of people suffering from allergies to latex is increasing,³⁷ latex-free gloves are now chosen by many professional users. Vinyl gloves, made of poly(vinyl chloride) (PVC), are the most cost-efficient latex-free gloves and are often preferred for low risk and shorter tasks that still require some level of protection. Polyethylene gloves are not recommended for medical use as they are loose fitting with limited tensile strength, but are often used in food preparation and serving. Nitrile gloves, made of a synthetic copolymer of acrylonitrile and butadiene

monomers, are preferred in clinical areas as they are stretchable, highly durable and puncture resistant. Importantly, the protection levels offered by each type of material also vary. While latex gloves were reported to provide the best protection against bacteria and viruses after puncture,³⁸ synthetic gloves offer a higher degree of chemical protection. Medical visors, gowns and face masks vary in terms of the polymers from which they are made and for what purpose they are worn. Face visors are often made from polyethylene terephthalate (PET), such as, for example a recent COVID-19 specific CE approved design.³⁹ Disposable aprons, usually made from polyethylene to protect against fluids, designed with over-the-head neck and long sleeves, are recommended for use in high risk areas or aerosol generating procedures by the UK government's '*Guidance on Infection Prevention and Control for COVID-19*'.⁴⁰ Face masks to reduce the risk of exhaling and/or inhaling viruses, usually made of an outer hydrophobic nonwoven layer (e.g. polypropylene, polyester and polyaramid), a middle melt-blown/filter layer (e.g. synthetic fibres - nonwoven polypropylene), and an inner soft absorbent nonwoven layer (e.g. terry cloth towel, quilting cotton, and flannel), vary by their quality and levels of protection.⁴¹⁻⁴⁴

An early study compared the maintenance of influenza A virus infectivity on PPE surfaces including rubber glove (type not specified), N95 particulate respirator, surgical mask (non-woven fabric), gown made of Dupont Tyvek (polyethylene fibre), coated (coating material not specified) wooden desk, and stainless steel.⁴⁵ The influenza A viruses were found to remain infectious ($\geq 10^{2.8}$ TCID₅₀/mL) for at least 8 hours on all the surface materials at 25.2 °C and 55% RH, but increased to 24 hours for rubber glove surface. A recent preprint article studied the SARS-CoV-2 stability on some of the currently used PPE by healthcare workers.⁴⁶ A virus dose of 10 µl of $10^{7.88}$ TCID₅₀/mL in organic components to mimic the typical virus-containing fluids was loaded on each sample surface, and dried before starting the study. SARS-CoV-2 were found to be detectable, determined by endpoint titration in Vero E6 cells, up to 7 days on nitrile gloves (copolymer of acrylonitrile and butadiene), 4 days on chemical resistant gloves (type not specified, but usually made of nitrile rubber), 21 days on plastic face shield (type not specified, but usually made of PET) and N95/N100 particulate respirators, 14 days on Tyvek (polyethylene fibre) and stainless steel under ambient conditions (20 °C and 35-40% RH). The infectivity of SARS-CoV-2 on cotton was reduced within 4 hours of drying, and not detectable by 24 hours in the same study. To date, we have been unable to find a systematic study of the stability of SARS-CoV-2, nor of its similar viruses, on different polymers.

In order to render polymers antimicrobial, the addition of toxic substances to kill surface located cells has been widely employed, and silver has been proposed as a strategy to inactivate SARS-CoV-1.⁴⁷ PPE made of PVC with silver coating are commercially available, but these are normally much more expensive than the non-coated ones. Antiviral textiles, made of polymer fibres (e.g. nylon) embedded with nano-copper, are potentially functional and cost-effective materials for PPE manufacture.⁴⁸ The incorporation of metals has been proposed as a fast and efficient way to improve the function of the existing PPE or coating materials. However, when the metals are employed as nanoparticles for example silver, this approach does have the disadvantage that it is likely to release these into the environment with can potentially contaminate the food chain and be toxic to humans at the cellular and organismic levels.⁴⁹ A recent review has summarised the advances in antiviral materials and their mechanisms of activity, and highlighted that nanomaterials, including metal-based, carbon-based, silicon-based, organic-based and intrinsic antiviral materials, should be studied to improve the antiviral capability of PPEs (e.g. filtering materials for face masks).⁵⁰ In addition, an emerging strategy to treat microbial infection is to create cell-membrane mimics as decoys to trap and to detain the pathogens,⁵¹ which gives the chance to inactivate the viruses/bacteria while retaining immunogenicity. However, the active-loaded and protein-coated products currently available have limitations for long-term use as they deplete or degrade over time.

An alternative approach taking advantage of the polymer surface chemistry to modulate cellular attachment has been employed to induce significant effects ranging from bacterial and mammalian cell death to modified viability and phenotype.^{52,53} There is now theoretical and experimental evidence to suggest that polymer surface chemistry may also be used to control virion adsorption strength and therefore reduce the viable lifetimes for viruses attached to surfaces.^{16,54,55} For example, cationic polymers have shown high-affinity for virus binding,¹⁰ and hydrophobic polycationic surfaces have been reported to inactivate influenza viruses owing to irreversible attachment on the surface,

and related viral structure damage and inactivation, which in turn altered release of their genomic materials.^{12,13} This proposed mechanism was supported by the loss of infectivity and viral proteins of the exposed solution, indicating the virion attachment on surface, and the significant amount of viral RNA quantified in the disinfected solution due to the damage of viral structure.¹³ Similarly, a recent study reported that surfaces of polypropylene and other PPE materials engineered with both positive charge and hydrophobic features achieved strong binding with the surface spike-protein of SARS-CoV-2 viruses.¹¹ The strong binding between the hydrophobic spike protein and surfaces through electrostatic and hydrophobic interactions was proposed to cause protein conformational change and consequently inactivate the viruses. Non-charged oleophilic surfaces have also been demonstrated to destroy the viral lipid envelope upon contact.¹⁴ In another study the anionic copolymers with both hydrophobic and hydrophilic blocks have been shown to be inherently self-sterilizing and have broad-spectrum antimicrobial efficacy to successfully inactivate a broad range of bacteria and viruses including both enveloped vesicular stomatitis virus (VSV) and influenza A virus, and non-enveloped human adenovirus-5 (HAd-5) virus, after only 5 mins of contact.⁵⁶ The authors proposed that the high antimicrobial efficacy of these polymers is due to the ability of hydrated sulfonic acid groups to dramatically reduce the pH of the media, leading to the enzyme damage, protein denaturation and microbe death.

Progress

Looking forward, a systematic study on the lifetime of infectious viruses on a range of existing polymers under ambient conditions would be useful for those choosing which PPE to use; e.g. are viral particles lifetimes similar on vinyl and latex gloves? There also appears to be an opportunity to develop new polymers for use, i.e. those which bind virus particles strongly and speed the inactivation of adsorbed virus in conditions modelling ambient transmission. (Figure 1B) The theoretical basis for describing viral particle interactions at synthetic surfaces is not well developed. One means of identifying materials that does not require such a theoretical framework and could help to build it, is to use high throughput polymer micro array screening. Polymer micro array screening has mainly been explored in identifying simple polymers to control bacterial biofilm formation,⁵⁷ stem cell pluripotency,⁵⁸ and phenotype control.⁵⁹ Recent work in screening polymer libraries using micro arrays for Rubella and Lassa Fever virus like particles illustrates its potential in identifying materials for differential virus binding (Figure 1C),⁶⁰ but this needs to be progressed to include SARS-CoV-2 viral particle inactivation as a selection criteria in order to allow new polymers to be identified with desirable attachment and inactivation profiles.

Figure 1. A) An image showing fluorescent powder, representing viruses, transferred from door handle to hand after contacting; B) The to-be-developed new polymer surface binding virus particles strongly and speeding the inactivation of adsorbed virus under ambient conditions (blue: infectious virus particle; orange: inactivated virus particle); C) Schematic depiction of the process, showing the printing of the monomers, the in situ UV polymerisation of the monomers and finally their incubation with differently fluorescently tagged virus like particles.⁶⁰

The many viral outbreaks in human history, and especially the ongoing COVID-19 pandemic, are urging us to look for effective strategies that might be used to deal with emergent viral diseases in the future. The best long-term population-wide approach to counter viral infections is prevention of infection by good public health measures, but if outbreaks do occur, vaccination is required. However, vaccines are not yet available against Covid-19, and despite the unprecedented speed in developing new candidates, which are showing promise in pre-clinical models,^{61,62} there is still a lag time between developing a promising lead formulation in the lab to bringing a safe and effective vaccine to the market. In the absence of effective therapeutics and vaccines, there is an urgent quest for broad-spectrum antiviral protective materials that can prevent the contact transmission of existing and emerging viruses. Generally, there are a range of considerations on the design of a material for antiviral surface transmissions, for example, the interaction between virions (e.g. enveloped or non-enveloped) and surface materials (e.g. incorporation of antiviral agents or cell-membrane mimics, charge and hydrophobicity), the durability of the antiviral function of the surface, the cost-

effectiveness, the toxicity of the materials to environment and mammalian cells, and the environmental conditions (e.g. temperature, RH and light). However, most antiviral materials proposed so far have limitations for wide use as PPE or as surface coatings, including environmental issues, mammalian cell contact toxicity, loss of effectiveness over time, virus specificity and cost. A route to identify polymers using high throughput polymer micro array screening and predictive quantitative structure-activity relationship (QSAR) has been exemplified for controlling bacterial-surface interactions.⁶³ This approach may be adopted to the development of novel antiviral polymers with the potential to reduce contact transfer of a wide range of infectious viruses including the emerging ones in the future.

Potential

Since December 2019 the COVID-19 pandemic has affected more than 200 countries and impacted people's lives and livelihoods. With no vaccine available at this time and uncertainty over the duration of immunity following infection, the outlook is that we will be dealing with this virus for some time. To date, SARS-CoV-2 appears to spread easily in droplets in the air and via surfaces by infected individuals. Once outside the host cells, the virus cannot replicate, either going on to infect another host or degrade in the environment. Early work indicates that virus survives longer on some surfaces compared to others. However, it is currently unclear what role the surface chemistry plays in viral survival, infectivity and denaturation and the role of the local environment (e.g. media) is unclear. It remains to be seen if any of the existing materials commonly used to fabricate PPE may offer improvements in viral binding and inactivation rate at their surface, for example comparative viral lifetime data cannot be found on the three types of commonly used polymer gloves used clinically. These are experiments that can readily be undertaken in virology laboratories with access to virus in biohazard containment conditions. In the medium term, there are a number of lines of enquiry for the development of anti-SARS-CoV-2 PPE materials that could reduce contact transfer of infectious virus in clinical and public settings.

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Author Contributions

M.R.A. and X.X. conceived the paper. X.X. wrote the original draft, revised and edited the manuscript. J.K.B., C.A. and M.R.A. commented and revised the paper.

Declaration of Interests

The authors declare no competing interest.

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