

A molecular dynamics analysis of the influence of iron corrosion products on the healing process of bitumen

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ABSTRACT

Corrosion of iron materials in the asphalt concrete pavement occurs commonly when the bitumen film peels off, and the generation of corrosion products would affect the healing performance of bitumen. To identify the affection, this research focuses on the influence of iron corrosion products on the healing process of bitumen by molecular dynamics simulation. Firstly, bitumen model and iron corrosion products model were built. Then the healing systems of sandwich structure were constructed, and the simulated temperature were applied to reach equilibrium in the healing process with NVT ensemble (constant number of atoms, volume, and temperature). Dynamic movements of bitumen were characterized by appearance qualitatively. Healing rate of crack and healing rate of bitumen aggregation were held to evaluate the healing effect. Diffusion behaviors, internal force of motivation and interaction effect were also analyzed. The results indicate the duplicity of iron corrosion products in the healing process including the ease for bitumen climbing and the obstruction of bitumen movement. The comprehensive healing index demonstrated that iron corrosion products would reduce the healing degree, which was mainly caused by the obstruction effect and large internal stress generated by severe aggregation of bitumen in the limited space. From the perspective of crack closure and bitumen aggregation degree in the corrosion area, FeO healing systems were healed best, followed by Fe₃O₄, Fe₂O₃ and FeOOH. Furthermore, diffusion period of bitumen molecules on the surface of iron corrosion products during the healing process should be regarded as the important period affecting healing.

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34 Key words: Bitumen, Healing, Corrosion products, Molecular dynamic

35

36 1 Introduction

37 Highways are always considered as the most important systems in the transportation
38 infrastructures, where asphalt concretes are widely applied on the highway due to its marked
39 service characteristics including comfort driving, low noise, outstanding skid resistance,
40 simplicity of maintenance [1–3]. More than 90% roads of Europe are constructed by asphalt
41 concrete and its mileage in China is in excess of 1.2 million kilometers [4–6]. However, it should
42 be admitted that asphalt concrete pavement has been constantly exposed to the damage of
43 heavy ultraviolet light, moist climate and continuous loading since it was built [7–9]. The
44 damages will oxidize and harden the bitumen and then result in various deteriorations [10,11].
45 According to the official report by the Ministry of Transport of the People’s Republic of China,
46 the accumulated maintenance road length has exceeded 535.03 kilometers by 2022, accounting
47 for 99.9 % of total road length [12]. Once the damages are not maintained and repaired in time,
48 it will cause immeasurable damage to pavement safety and transportation. Therefore, the
49 development of advanced and effective maintenance technology has become the emergency
50 task.

51 The current advanced maintenance technology include encapsulation technology,
52 microwave heating technology and induction heating technology, which can accelerate the
53 healing of cracks by reducing the viscosity and enhancing the fluidity of bitumen [13–15].
54 Among these technologies, induction heating technology is considered the renewable and
55 sustainable method with the advantages of high efficiency and cleaner production, and would
56 be potential to become to a common maintenance method soon. The method was developed
57 on excellent magnetic conductivity of steel products like steel wool fibers, steel shaves and steel
58 grits [16]. The asphalt concrete with the steel products can be induced-heated firstly and then
59 flow to heal the microcrack generated by aging, hardening and loading under variable
60 magnetic field [17–19]. However, while recognizing the advanced and suitable characteristics
61 of induction heating technology, it can’t be ignored that the activity of iron materials in steel
62 products lead to their easy reaction and action with the environment, especially the potential

63 situation of direct exposure of metal materials [20].

64 Steel products consist of a large amount of iron and a small amount of carbon, where the
65 reactivity of iron would make it susceptible to severe corrosion while it is constantly exposed
66 to environment and loading. Suda et al and Duffo et al have investigated the corrosion of steel
67 bar in concrete, and the results show that even if the iron material is wrapped, it may still be
68 subjected to erosion, and the erosion products of which consist of the crystal substances like
69 magnetite, goethite, lepidocrocite, meanwhile the corrosion products are in a state of changing
70 composition and related to the environment [21,22]. In generally, the steel products in the
71 asphalt concrete would be considered being wrapped in the bitumen during the service life.
72 Though asphalt concrete was exposed to extreme environment and intense loading, the steel
73 products were also hard to get damage, because the effect of environment can hardly penetrate
74 the wrapped bitumen film and the loading can't strip bitumen from its surface [23]. However,
75 once cracks occur in asphalt concrete, the bitumen wrapped around steel products will peel off
76 and the cross section of the steel products would be exposed to the environment directly.
77 Furthermore, it is hard for the roads agency to judge the time to fix the crack, which would
78 worsen the erosion. Then, the deposits of generated corrosion products on the interface of
79 asphalt concrete would obstruct diffusion between bitumen and bitumen, and finally affect the
80 healing performance of induction heating technology [24]. In addition, the corrosion process
81 will make the oxide layer on the steel products grow, thus generating internal stress. If this
82 corrosion process is allowed to continue, it will lead to the situation that the old cracks have
83 not been closed and new cracks have been produced [25]. Therefore, it is necessary to clarify
84 the influence of corrosion products of steel products on healing effect of bitumen for the better
85 maintenance effect.

86 Molecular Dynamics (MD) simulation is a suitable method for investigating the
87 movements and dislocations of nanoparticles [26]. It describes the motions and positions of
88 atoms based on Newton's second law. Empirical force-field equations are used to describe
89 interatomic forces, electron interactions, and other energies during the simulation. The MD
90 method can describe various scales and indicators, such as the change in energy of the proton
91 system, mean-square displacement of the atomic system, and adsorption state between atoms
92 [27,28]. These indicators are physical quantities that have been demonstrated to be describable

93 by MD, both theoretically and experimentally [29–31]. Molecular dynamic (MD) simulation
94 would be advantageous compared with the phenomenological method: it is not limited by
95 experimental methods or specimen preparation conditions and can effectively address the
96 shortcomings of macroscopic scales. Several studies by MD simulation have been carried out
97 to investigate the healing properties of bitumen. Bhasin et al used MD simulation to investigate
98 the mechanism of self-healing of bitumen and demonstrated the correlation of chain length and
99 chain branching to self-diffusivity of bitumen molecules [32]. Sun et al have conducted
100 molecular dynamics simulation to evaluate the hypothesis of healing mechanism and evaluate
101 the self-healing capability of virgin and SBS modified asphalt binder by introducing the micro-
102 crack model [33,34]. Qu et al and Zhang et al focused on the behavior characteristics of fractions
103 of bitumen during the healing process [35,36]. The influence of the addition on the healing
104 performance of bitumen have been also investigated, which includes waste vegetable oil,
105 soybean-oil, graphene, iron oxide, nanomaterials and rubbers [37–42]. Furthermore, the
106 investigations related to bitumen and iron oxide have also been conducted, but it mainly
107 focuses on the interaction effect and the adhesion property between bitumen and iron oxide
108 [43,44]. Therefore, it can be found that the current research mainly focused on the healing
109 process and healing mechanism of bitumen, the materials that would be beneficial to the
110 healing property, and the interaction effect and the adhesion property between bitumen and
111 iron oxide, while few work has been conducted to investigate the effect of diffusion-blocking
112 substances like iron corrosion products on self-healing properties of bitumen at the molecular
113 scale. In general, it is necessity for understanding the influence of iron corrosion products on
114 the healing of bitumen at atomic scale, which would be investigated in the study.

115 The main purpose of this research is to investigate the influence of iron corrosion products
116 on the healing process of bitumen by molecular dynamics simulations. The bitumen model and
117 iron corrosion products model including Fe_2O_3 , Fe_3O_4 , FeO and FeOOH were built. Then the
118 healing systems were constructed, and the simulated temperature were applied to reach
119 equilibrium in the healing process. The dynamic movements of bitumen were characterized,
120 and healing rate indexes of crack closure, healing rate of bitumen aggregation and their
121 comprehensive indexes were held to evaluate the healing effect. Diffusion behavior of the
122 bitumen molecules and the internal force motivating bitumen molecules in the healing systems

123 were evaluated. Eventually, interaction effects between bitumen and iron corrosion products
124 were obtained and analyzed.

125

126 2 Molecular dynamic simulation models and simulation details

127 2.1 Bitumen

128 The bitumen molecular model was built based on the AAA-1 bitumen model proposed by
129 Li and Greenfield. As shown in the Figure 1, it has 12 components and its characteristics were
130 proved closer to experimental data [45]. Materials studio software was used for the
131 establishment and thermodynamic properties calculation for the models. 12-components
132 molecular models were built in 3D Atomistic. Condensed-phase optimized molecular
133 potentials for atomistic simulation studies (COMPASS) was selected as the force field, which
134 can predict and calculate the structure and thermophysical properties of common inorganic
135 and organic system over a large temperature and pressure range [46]. The model was
136 constructed with the following step: Firstly, the model was constructed by Amorphous Cell
137 tools with an initial density of 0.1 g/cm³ under the three-dimensional cycle condition. The
138 geometric optimization with 5000 iterations was followed to eliminate unreasonable
139 configurations in the model, leveling off the energy of the molecule to reach minimum energy.
140 Then, Forcite tools was used to reach dynamic equilibrium for the stable structure and density,
141 where a canonical ensemble (NVT, constant molecule number, model volume, and temperature)
142 with 298 K, 1 fs time step for 100 ps and an isothermal-isobaric ensemble (NPT, constant atomic
143 number, pressure, and temperature) with 298 K and 1.0 atm were conducted successively. The
144 temperature and pressure of the block were controlled by Andersen barostat and Nose-
145 Hoover-Langevin thermostat. Moreover, the Ewald with the accuracy of 0.001 kcal/mol and
146 Atom-based with the cutoff distance of 15.5 Å are assigned as the Electrostatic and van der
147 Waals summation method. Finally, the models have been established for further performance
148 prediction and analysis in terms of thermodynamics parameters, structural characteristics, and
149 dynamic behaviors. The rationality and reality of this model have been proved in our previous
150 studies [47].

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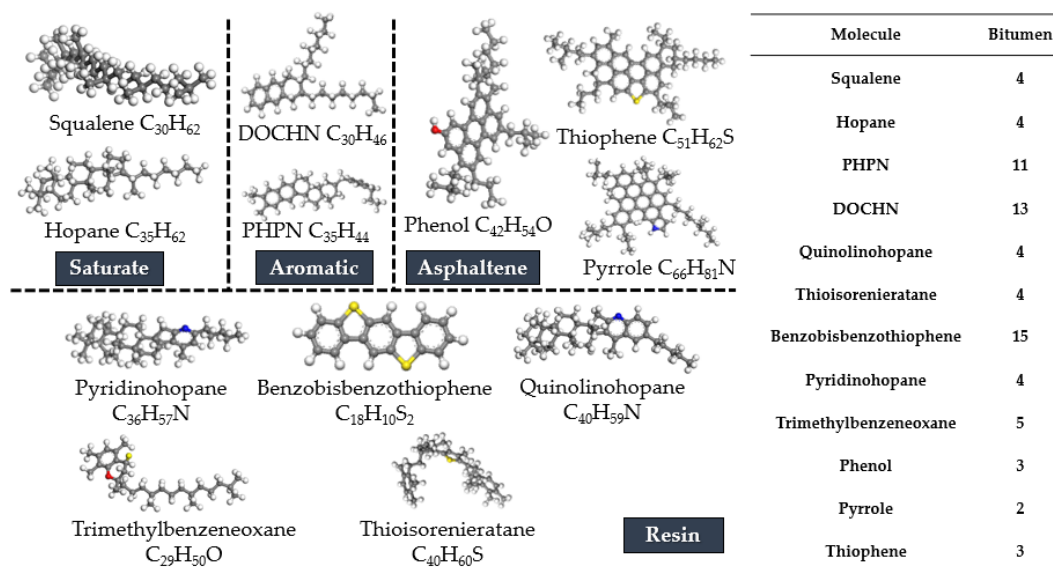


Figure 1. 12-components molecules models and molecular compositions of bitumen

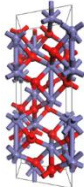
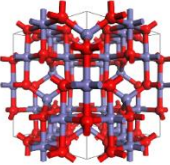
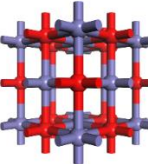
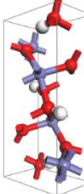
2.2 Iron corrosion products

The iron products will be corroded in the environment with air and water, and the corrosion process is a dynamic process, whose products will be constantly transformed. Commonly, the corrosion products are mainly composed of Fe_2O_3 , Fe_3O_4 , FeO and $FeOOH$ [20,48], therefore α - Fe_2O_3 (Hematite), Fe_3O_4 (Magnetite), FeO and α - $FeOOH$ (Goethite) were selected as the iron corrosion products in this research. The unit cell constants and unit cell structure models of Fe_2O_3 , Fe_3O_4 , FeO and $FeOOH$ are shown in Table 1 [49–52]. The models of iron corrosion products were also established by Materials studio software and the steps were as follows: Firstly, the lattice was converted into an orthorhombic lattice. The plane's crystal face was sliced using the Cleave Surface tool with specific Miller index where (1 0 0) was for Fe_3O_4 , FeO , (1 1 0) was for Fe_2O_3 and (0 1 0) was for $FeOOH$. The reason for the adoption of crystallographic surface plane was that these crystallographic surfaces showed the least lattice mismatch shown in the previous research that adsorption and diffusion of organic molecules happened on iron oxide surfaces, meanwhile the oxygen ions were full bulk-coordination on the surfaces [53–55]. Then, before performing energy reduction on the unit cell structure, it was necessary to ensure that the COMPASS force field of each atom in the iron corrosion products unit cell was accurately assigned and that chemical linkages between atoms were deleted, which can describe the different valence states of iron atoms. Finally, the optimized unit cell model was enlarged to a supercell model with similar size and volume to eliminate the size

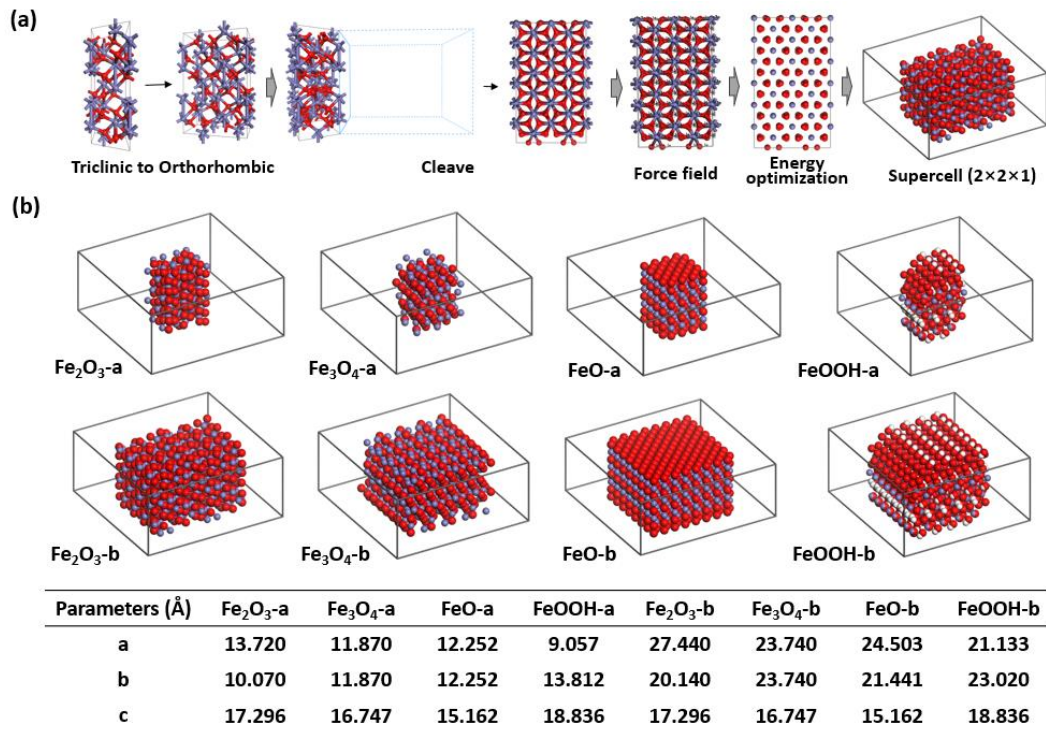
174 effect and a certain lattice same to bitumen model was added to the unit cell model. The
 175 establishment of the supercell model of Fe₂O₃ molecule was as an example shown in Figure 2.
 176 The supercell models of iron corrosion products and their parameters are also shown, where
 177 there are two sizes of each iron corrosion products were established to investigate the influence
 178 of iron corrosion products volume on the healing performance.

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Table 1. Unit cell of iron corrosion products and lattice parameters

Corrosion Products	Unit cell structure	Unit cell constants	
Fe ₂ O ₃		a = 5.035 Å, b = 5.035 Å, c = 13.720 Å	α= 90 °, β = 90 °, γ=120 °
Fe ₃ O ₄		a = 8.394 Å, b = 8.394 Å, c = 8.394 Å	α=90 °, β =90 °, γ=90 °
FeO		a = 4.332 Å, b = 4.332 Å, c = 4.332 Å	α=90 °, β =90 °, γ=90 °
FeOOH		a = 4.604 Å, b = 9.951 Å, c = 3.019 Å	α=90 °, β =90 °, γ=90 °

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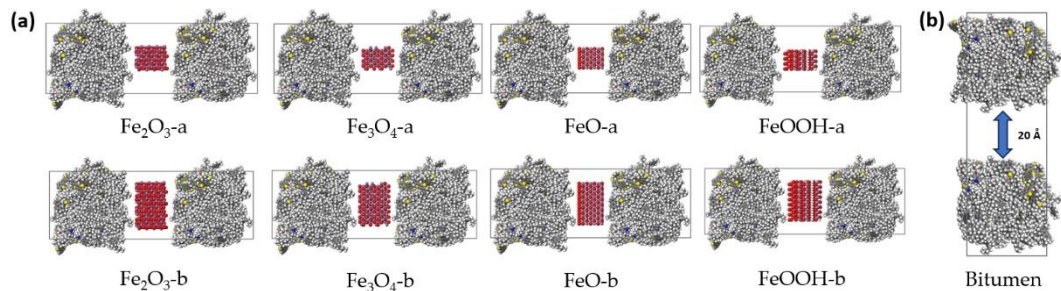


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Figure 2. The supercell models of iron corrosion: (a) Establishment process; (b) The models and the size parameters

186 2.3 Healing systems

187 The healing systems were developed by the Materials studio software's Build Layers tools
188 with the obtained bitumen models and iron corrosion products models. Iron corrosion
189 products healing systems have triple layers: a bitumen layer as the foundation, an iron
190 corrosion products layer as the middle layer, and another bitumen layer on top of the iron
191 corrosion products layer. Bitumen healing systems were constructed by bitumen layer, vacuum
192 layer and another bitumen layer. The healing systems are shown in Figure 3: Bitumen healing
193 system and iron corrosion products healing systems including group-a (Fe₂O₃-a, Fe₃O₄-a, FeO-
194 a and FeOOH-a) and group-b (Fe₂O₃-b, Fe₃O₄-b, FeO-b and FeOOH-b).
195



196
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Figure 3 The model of healing systems: (a) Iron corrosion products

200 2.4 Simulation details

201 In this study, a classical molecular dynamics code: the large-scale atomic/molecular
 202 massively parallel simulator (LAMMPS) was used to perform the simulation. The polymer
 203 consistent force field (PCFF) was chosen for bitumen simulation, which has been validated to
 204 describe the organic, inorganic, and organic–inorganic interface systems. The force field is an
 205 empirical expression of the potential energy surface, and the total energy of the molecules is
 206 the sum of kinetic energy and potential energy. Moreover, the total potential energy is
 207 composed of bond angle bending potential energy, bond stretching potential energy, dihedral
 208 angle twisting potential energy, off-plane vibration potential energy, Waals potential energy
 209 and Coulomb electrostatic potential energy, shown in Equations (1).

$$\begin{aligned}
 E_{\text{potential}} = & \sum_{\text{cross}} E(b, \theta, \varphi) + \sum_{\text{bond}} E_b(b) + \sum_{\text{torsion}} E_\varphi(\varphi) \\
 & + \sum_{\text{angle}} E_\theta(\theta) + \sum_{\text{inversion}} E_x(x) + E_{ele} + E_{vdw}
 \end{aligned} \quad (1)$$

210 where $E_{\text{potential}}$ is the total energy; $\sum_{\text{cross}} E(b, \theta, \varphi)$ represents the cross term potential
 211 energy; $\sum_{\text{bond}} E_b(b)$ is the bond stretching potential energy; $\sum_{\text{torsion}} E_\varphi(\varphi)$ is the dihedral
 212 angle twisting potential energy; $\sum_{\text{angle}} E_\theta(\theta)$ is the bond angle potential energy;
 213 $\sum_{\text{inversion}} E_x(x)$ is the off-plane vibration potential energy; E_{ele} is the Coulomb electrostatic
 214 potential energy and E_{vdw} is the Waals potential energy. The interaction between bitumen and
 215 corrosion products can be described by the 6/9 Lennard–Jones potential (Sun 1998), as shown
 216 in Equations (2)-(3). The LJ 9-6 and Coulombic interactions are truncated to 10 Å and 8 Å.

$$E_{ele} = \sum_{i>j} \frac{q_i q_j}{r_{ij}} \quad (2)$$

$$E_{vdw} = \sum \epsilon_{ij} \left[2 \left(\frac{r_{ij}^0}{r_{ij}} \right)^9 - 3 \left(\frac{r_{ij}^0}{r_{ij}} \right)^6 \right] \quad (3)$$

217 where q_i and q_j are the charges of atomic i and j ; r_{ij} is the distance of atomic i and j
 218 and ϵ_{ij} is the well depth of atomic i and j , respectively.

219 Each simulation process consists primarily of the following steps: (1) Energy minimization
 220 was used to remove any potential energy excess that existed in the initial configuration. (2) The
 221 healing system was then relaxed using NVT (constant number of atoms, volume, and

222 temperature) ensemble at 373 K for 1000 ps, which was the common healing temperature used
 223 in the induction heating [24]. Simultaneously, the iron corrosion products layer in each model
 224 was fixed and each valence state of iron atom was arranged to the corrosion products to
 225 precisely calculate the indicators of healing system.

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227 3 Evaluation indexes

228 3.1 Healing rate indexes

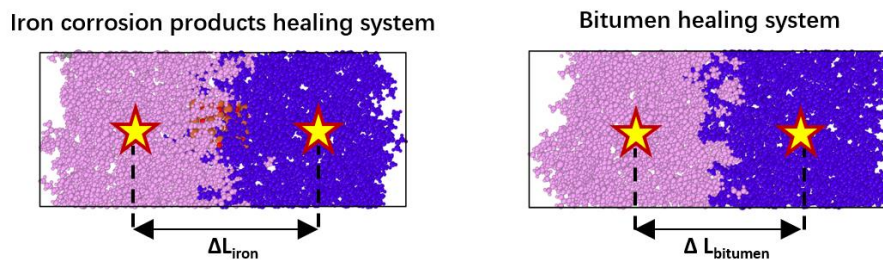
229 3.1.1 Healing rate of crack

230 The closure of cracks is considered as a sign of healing of bitumen, but it is hard to calculate
 231 the accurate volume of crack during the simulation. So the difference value of centroid position
 232 between bitumen molecules are used to propose the healing rate of crack as shown in Figure 4
 233 and its calculation is accorded to Equation (4) and (5):

$$\Delta L = |Z_{bitumen1} - Z_{bitumen2}| \quad (4)$$

$$H_1 = 1 - \frac{\Delta L_{iron} - \Delta L_{bitumen}}{\Delta L_{bitumen}} \times 100\% \quad (5)$$

234 Where ΔL represents the absolute value of the difference value of centroid position between
 235 bitumen layers in the healing systems; $Z_{bitumen1}$ represents Z coordinate of centroid position
 236 of one bitumen layer in the healing systems; $Z_{bitumen2}$ represents Z coordinate of centroid
 237 position of the other bitumen layer in the healing systems; ΔL_{iron} represents the average value
 238 of difference value of centroid position between bitumen layers in the iron corrosion products
 239 healing systems from 900 ps to 1000 ps; $L_{bitumen}$ represents the average value of difference
 240 value of centroid position between bitumen layers in the bitumen healing systems from 900 ps
 241 to 1000 ps; H_1 represents the healing rate of crack, and greater value indicates the closer of the
 242 bitumen combination in this system compared to that of bitumen healing systems, as well as
 243 that the better the cracks are closed.



244

245 Figure 4 Difference value of centroid position between bitumen in the healing systems

246

247 3.1.2 Healing rate of bitumen aggregation

248 Healing rate of bitumen aggregation is held to evaluate the diffusion of bitumen molecules
249 after the cracks are closed. Firstly, the relative concentration of bitumen molecules should be
250 obtained. It was computed as the ratio of the atom number in the unit volume perpendicular
251 to the axis to the atom number in the unit volume of the amorphous unit. The three-
252 dimensional periodic system should be partitioned into evenly spaced plates in the interfacial
253 system. The distribution of atomic density in each plate was then computed to give the overall
254 structure's relative concentration distribution. As shown in Figure 5, iron corrosion products
255 healing systems are divided into three parts: free area 1, free area 2 and corrosion area; the free
256 area is also separated in bitumen healing systems. The standard of demarcation is the position
257 of iron corrosion products and the peak of relative concentration of bitumen. The average value
258 of relative concentration of bitumen in free area of bitumen healing systems should calculated
259 firstly as the reference. Then the average value of relative concentration of bitumen in the
260 selected area of iron corrosion products healing systems should be calculated respectively.
261 Healing rate of bitumen aggregation can be calculated as Equation (5)-(9):

$$M_0 = \frac{R_0}{V_0 - V_{iron}} \quad (5)$$

$$M_f = \frac{\left(\frac{R_1}{V_1} + \frac{R_2}{V_2}\right)}{2} \quad (6)$$

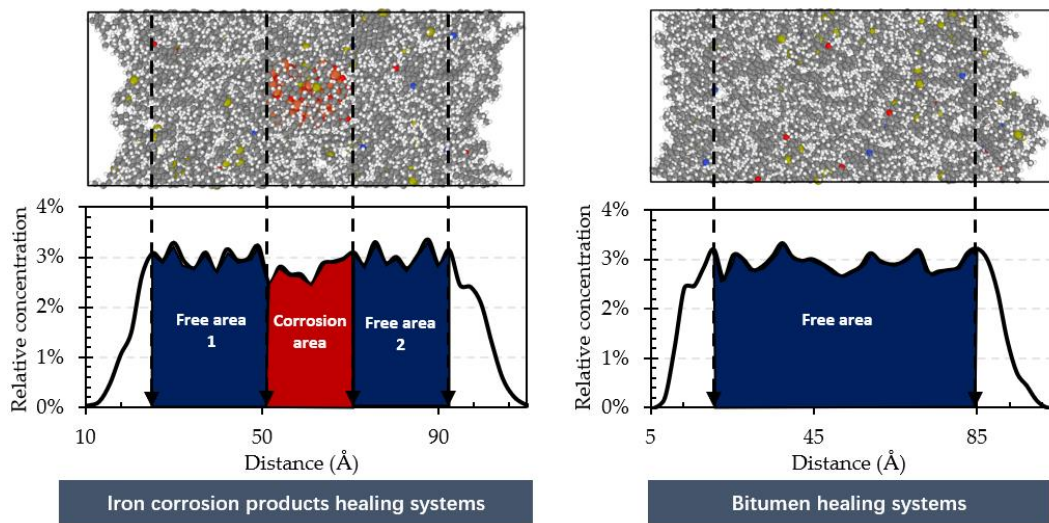
$$M_b = \frac{R_3}{V_3} \quad (7)$$

$$H_{2a} = \frac{M_0}{M_b} \times 100\% \quad (8)$$

$$H_{2b} = \frac{M_f}{M_b} \times 100\% \quad (9)$$

262 Where M_0 represents the relative concentration per unit volume of bitumen in corrosion area
263 of iron corrosion products healing systems; R_0 represents the sum of relative concentration of
264 bitumen in corrosion area of iron corrosion products healing systems; V_0 represents the
265 volume of corrosion area of iron corrosion products healing systems; V_{iron} represents the
266 volume of iron corrosion products; M_f represents the relative concentration per unit volume
267 of bitumen in free area of iron corrosion products healing systems; R_1 and R_2 represent the
268 sum of relative concentration of bitumen in free area of iron corrosion products healing systems

269 respectively; V_1 and V_2 represent the volume of free area 1 and free area 2 of iron corrosion
 270 products healing systems respectively; M_b represents the relative concentration per unit
 271 volume of bitumen in free area of bitumen healing systems; R_3 represents the sum of relative
 272 concentration of bitumen in free area of bitumen healing systems; V_3 represent the volume of
 273 free area of bitumen healing systems; H_{2a} and H_{2b} represent the healing rate of bitumen
 274 aggregation in corrosion area and free area of iron corrosion products healing systems, and
 275 greater value indicates serve bitumen aggregation in this area of iron corrosion products
 276 healing systems compared to that of bitumen healing system.



277
 278 Figure 5 Schematic diagram of relative concentration of bitumen in the healing systems
 279

280 3.2 Diffusion and interaction indexes

281 3.2.1 Mean square displacement and diffusion coefficient

282 The core regulation of diffusion phenomena was the movements of atoms in three-
 283 dimension space, which was vital to analyze the interaction between bitumen and corrosion
 284 products. However, due to the enormous number of atoms in the interface system, detecting
 285 each atom's motion trajectory is difficult. As a result, mathematical statistics method was held
 286 to describe the regularity of particle movement. The most commonly used indicator was mean
 287 square displacement (MSD), which would be represented and calculated by Equation (10):

$$288 \text{MSD}(t) = \langle |r_i(t) - r_i(0)|^2 \rangle \quad (10)$$

289 Where, $\text{MSD}(t)$ indicated as the mean value of all atoms' movement positions in the molecular
 290 system, $r_i(0)$ indicated the original position of particle i , and $r_i(t)$ indicated the position of
 particle i at the time of t .

291 Diffusion coefficient, for the measurement of the molecule's capacity for diffusion, rate at
 292 which a quantity diffuses per unit area while the concentration gradient is the same unit.
 293 MSD had a linear relationship with time and was correlated with the diffusion coefficient after
 294 diffusion relaxation process. After this period, the linear slope of the MSD curve might be used
 295 to compute the diffusion coefficient of the contact system, as indicated by Equation (11):

$$D = \frac{1}{6N} \lim_{t \rightarrow \infty} \frac{d}{dt} \sum_{i=1}^N (|r_i(t) - r_i(0)|^2) \approx \frac{1}{6} K_{MSD} \quad (11)$$

296 Where, the diffusion coefficient was recorded as D in the interface system, N indicated the
 297 whole number of molecules in the interface system, and the differential term was equal to the
 298 linear slope of the interface system's MSD curve, K_{MSD} was equal to the linear slope of the
 299 interface system's MSD curve.

300

301 3.2.2 Interaction energy and debonding energy

302 Interaction energy could be used to evaluate the stability of the selected molecules. The
 303 greater its absolute value, the more interaction there was between the selected molecules. When
 304 the value of interaction was zero or positive, adsorption was minor or non-existent. Its
 305 calculation formula was shown in Equation (12):

$$E_{inter} = (E_{pe1} + E_{pe2} + \dots + E_{pei}) - E_{pe-total} \quad (12)$$

306 Where E_{inter} represented the interaction energy between the selected molecules, $E_{pe-total}$
 307 represented that the total potential energy of the selected molecules in a steady state, E_{pei}
 308 represented the total potential energy of every component of the selected molecules
 309 individually.

310 Debonding energy is defined as the energy for the separation of the bitumen in the healing
 311 systems, which can be calculated by Equation (13):

$$E_{debonding} = E_{inter2} - E_{inter1} \quad (13)$$

312 Where $E_{debonding}$ is the debonding energy; E_{inter1} is the interaction energy between bitumen
 313 and bitumen in the iron corrosion products healing systems; E_{inter2} is the interaction energy
 314 between bitumen and iron corrosion products in the iron corrosion products healing systems.

315 Another parameter used in this study to further define the debonding property was the
 316 Energy Ratio, which was computed as the ratio of the interaction energy between bitumen of

317 bitumen healing systems to the debonding energy of iron corrosion products healing systems,
318 as calculated as Equation (14) [56]:

$$ER = \frac{E_{inter3}}{E_{debonding}} \times 100\% \quad (14)$$

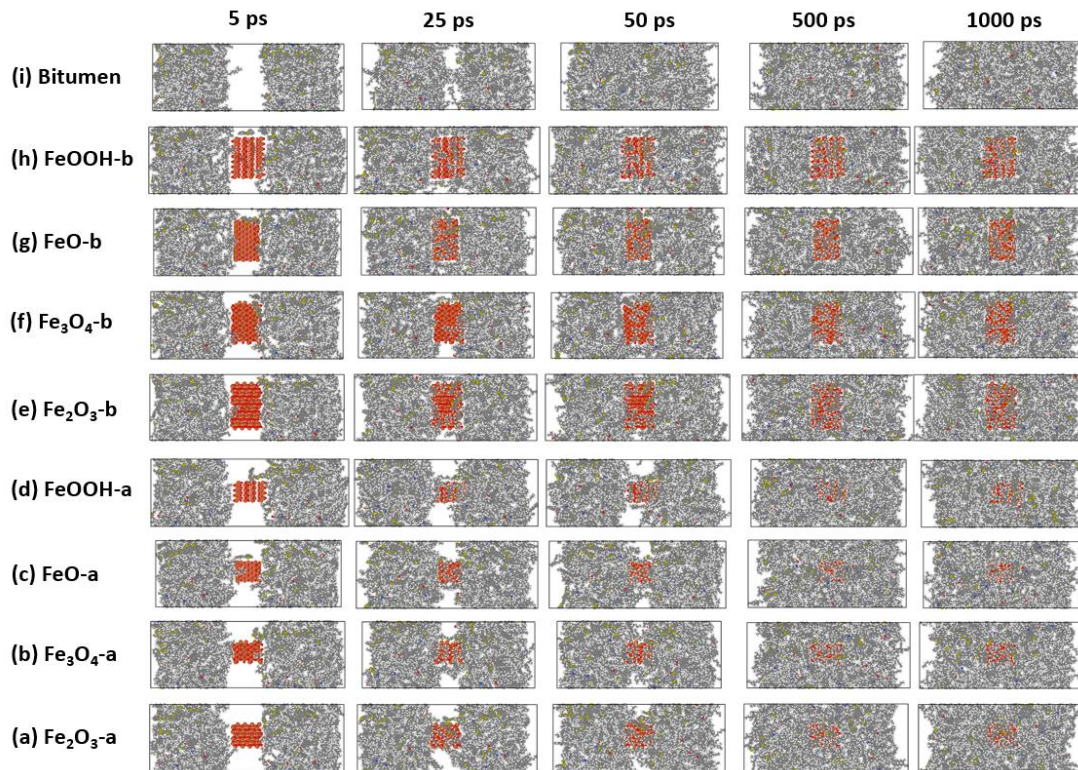
319 Where ER is the Energy Ratio; E_{inter3} is the interaction energy between bitumen and bitumen
320 in the bitumen healing systems; $E_{debonding}$ is the debonding energy.

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322 4 Results and discussion

323 4.1 Dynamic movement and healing rate of crack

324 Figure 6 shows the healing process of bitumen in the different healing systems where the
325 snapshots of the healing systems at 5 ps, 25 ps, 50 ps, 500 ps and 1000 ps are selected. It can be
326 found in the bitumen healing systems that the bitumen molecules would get close firstly, and
327 then the molecules would seize each other to reduce the vacuum area. Finally, the crack was
328 filled with bitumen completely and the bitumen molecules were keeping diffusion. Similarly,
329 the bitumen molecules in the iron corrosion products healing systems would also get touch
330 firstly, capture the iron corrosion products, and finally swallow the iron corrosion products.
331 In this process, it can be observed that the bitumen molecules would climb on the surface of
332 iron corrosion products firstly and then cover the iron corrosion products completely, rather
333 than filling the vacuum area directly like that in bitumen healing systems. It was caused by the
334 absorption force generated by iron corrosion products that was obvious larger than that of
335 vacuum, therefore the iron corrosion products could drag the bitumen molecules forwards to
336 its position. In addition, during the absorption process, the aggregation degree of bitumen on
337 the iron corrosion products were quite different. It is clearly that the aggregation of bitumen
338 on Fe_2O_3 , Fe_3O_4 and FeO were more severe than that of $FeOOH$ at 50 ps. The volume of iron
339 corrosion products would also affect the diffusion of bitumen molecules: at 1000 ps, the iron
340 corrosion products of Fe_2O_3 -b, Fe_3O_4 -b, FeO -b and $FeOOH$ -b systems were more obvious to
341 observe than Fe_2O_3 -a, Fe_3O_4 -a, FeO -a and $FeOOH$ -a systems. Larger iron corrosion products
342 volume means that there are few space for the bitumen molecules to move, which results in
343 that only small molecules are accessible to diffusion while macromolecules were blocked out
344 the space.



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Figure 6 Trajectories of bitumen in the healing systems at 373 K

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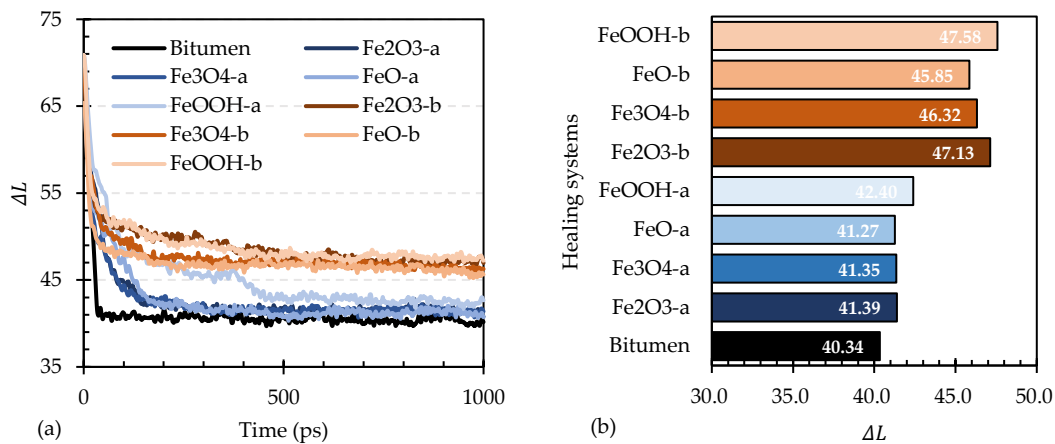
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Figure 7 shows the changing of ΔL of bitumen molecules in different healing systems. It can be found from Figure 7 (a) that the ΔL curve of bitumen healing systems can be divided into two parts (rapidly decrement and equilibrium): first, it would decrease to the equilibrium value straightly, and then it would keep the equilibrium value dynamically. The addition of iron corrosion products into the healing systems made the crack closure process dissimilarly. The ΔL curves of iron corrosion products healing systems can be divided into three parts: rapidly decrement, slowly decrement and equilibrium. It is clearly that the rapidly decrement part were affected significantly and rapidly crack closure was retarded, which was caused by the obstruction of iron corrosion products. The slowly decrement part of group-a and group-b of iron corrosion products healing systems were affected by the volume of iron corrosion products obviously. The bitumen molecules could differ through the side surface of iron corrosion products in group-a while the bitumen molecules only diffused on the top and bottom surface of iron corrosion products in group-b and could hardly catch the molecules in the other side. Therefore, the healing process of group-a was controlled by movements of bitumen and attraction of iron corrosion products collectively while that of group-b was

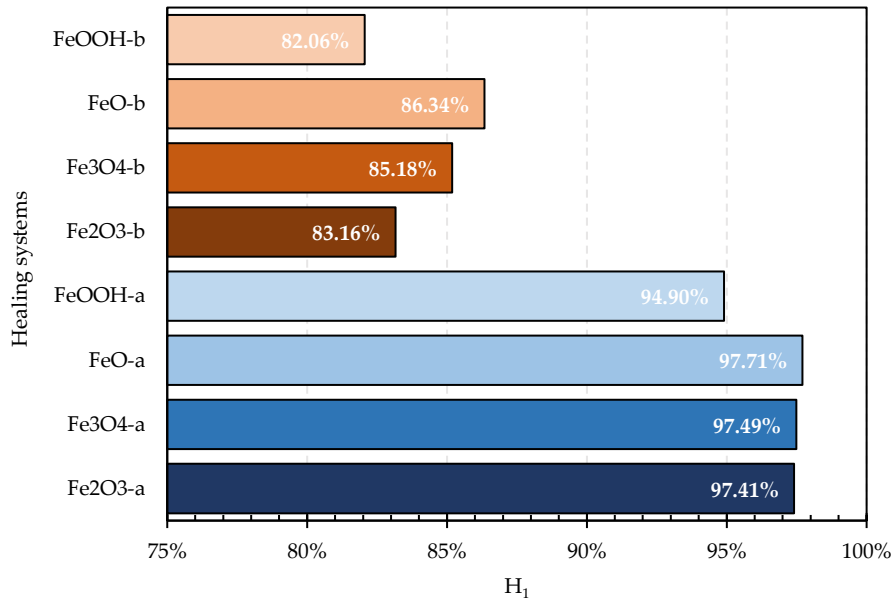
364 controlled by attraction of iron corrosion products mainly. Figure 7 (b) shows ΔL of different
 365 healing systems at equilibrium. It can be found that ΔL of bitumen system was the least, which
 366 was the results of free diffusion without the obstruction and attraction of iron corrosion
 367 products. ΔL of group-a was obviously less than that of group-b, also indicating that the
 368 obstruction and attraction were affected by the volume changing of iron corrosion products.
 369 With healing system of bitumen regarded as none iron corrosion products, the influence of the
 370 volume of iron corrosion products could be analyzed by linear fitting and the slope can reflect
 371 the sensitivity. The results of slope indicate that healing system of FeOOH was the most
 372 sensitive to iron corrosion products volume, followed by Fe₂O₃, Fe₃O₄, and FeO.
 373



374 Figure 7 Dynamic movements of bitumen molecules: (a) ΔL of different healing systems with
 375 the simulated time; (b) ΔL of different healing systems at equilibrium
 376

377 Figure 8 shows the healing rate of crack in the healing systems, which was calculated
 378 according to Equation (4) in section 3.1.1 with ΔL of bitumen healing system as the reference. It
 379 means that the crack closure in the bitumen healing systems was considered completely, It
 380 can be found that H_1 of group-a of iron corrosion products healing systems were obviously
 381 larger than H_1 of group-b of iron corrosion products healing systems. It indicates that a larger
 382 volume of iron corrosion products would weaken the crack closing of bitumen molecules. The
 383 reason is that the larger obstruction area made bitumen molecules unable to diffuse freely.
 384 Inside the group-a and group-b of iron corrosion products healing systems, H_1 both show the
 385 regularity that FeO > Fe₃O₄ > Fe₂O₃ > FeOOH, which was consistent with the results of Figure 7
 386 (b). In addition, the difference values of H_1 between different healing systems were more

387 obvious in group-b of iron corrosion products healing systems. It demonstrates that the
388 increment of iron corrosion products volume will enhance the difference of crack closing.
389



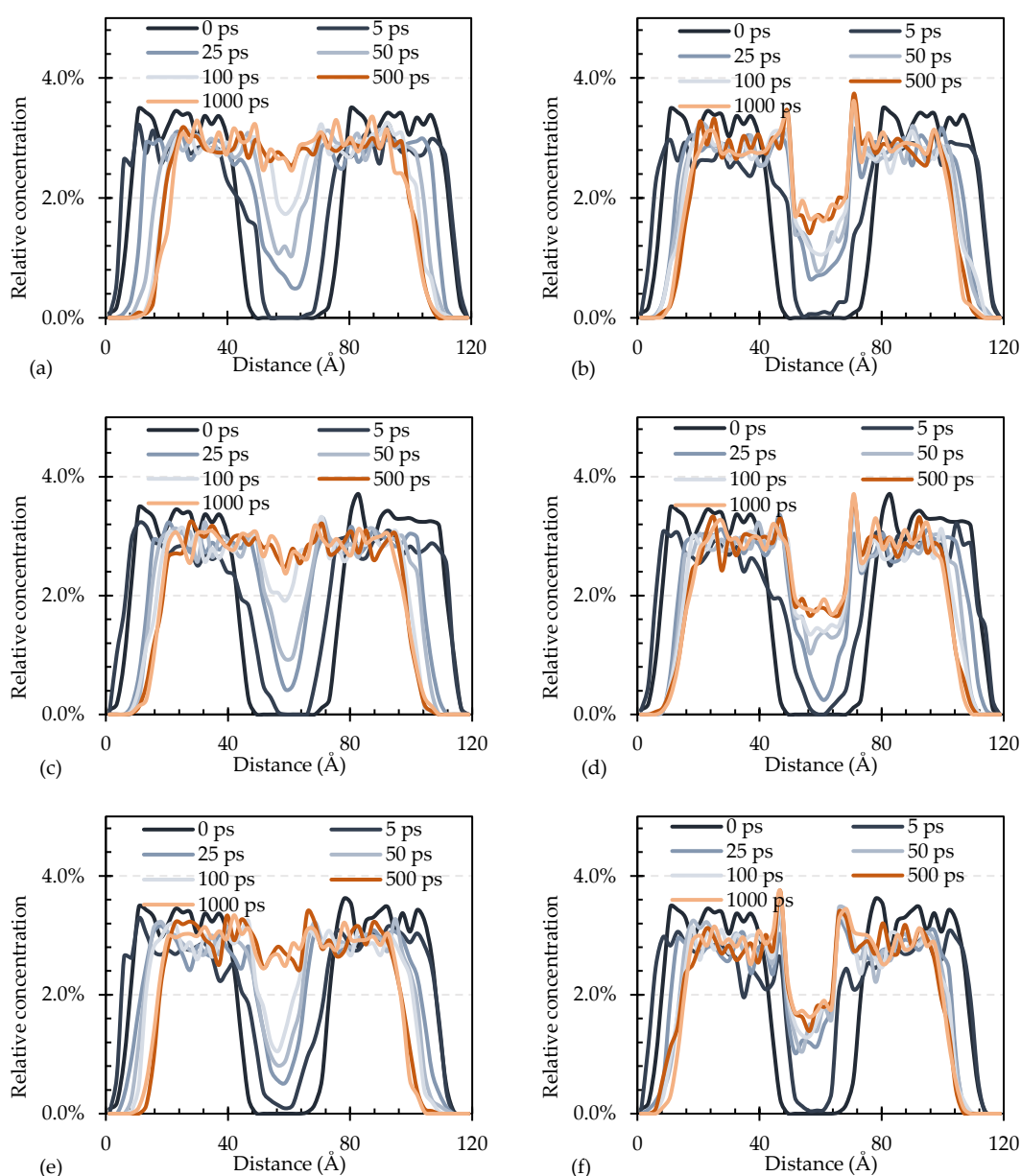
390
391 Figure 8 Healing rate of crack in the healing systems
392

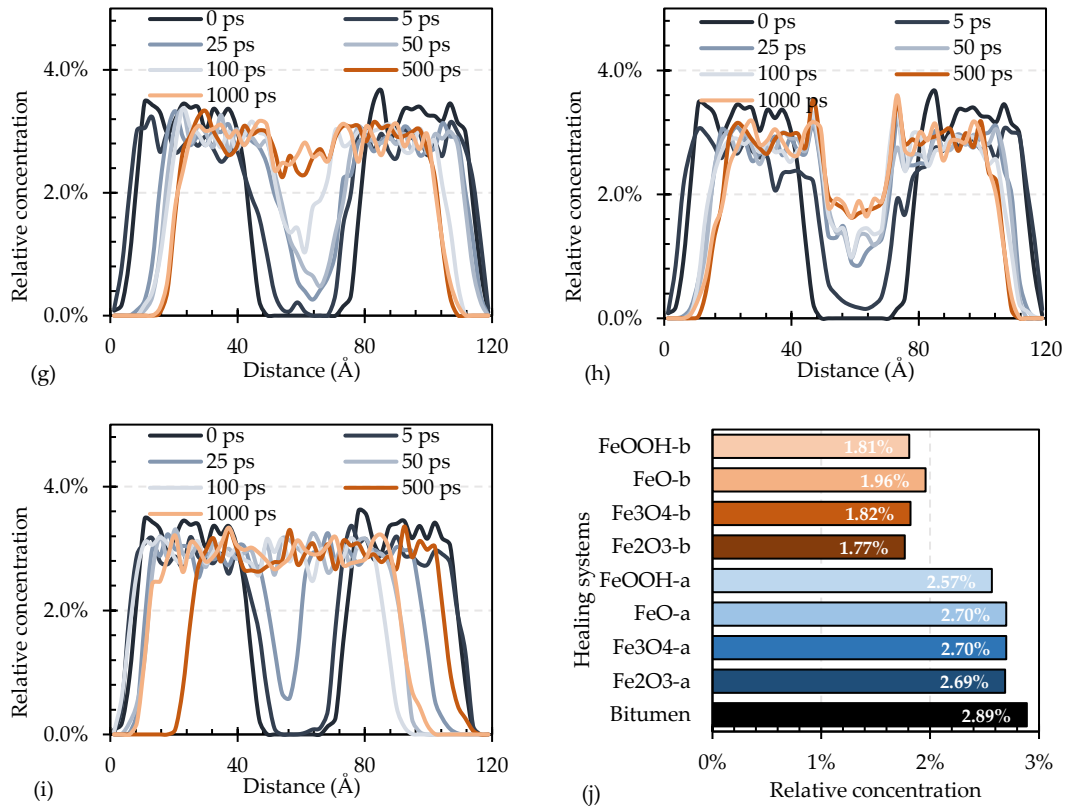
393 4.2 Healing rate of bitumen aggregation

394 The closure of cracks cannot represent healing of bitumen completely. In the healing
395 process, bitumen molecules would continue diffusion to recover strength after the cracks are
396 closed. Therefore, another index of bitumen aggregation in the healing systems were held for
397 the evaluation. Relative concentration of bitumen in the healing systems should be
398 demonstrated firstly, which is shown in Figure 9, representing the aggregation degree of
399 bitumen in the healing systems. It can be found that the relative concentration of the bitumen
400 in the healing systems shows bimodal shape initially: there were few bitumen molecules in the
401 middle area. Within 100 ps of simulated time, the bottom of recess changed from plate shape
402 to sharp shape gradually, which indicates that the bitumen molecules would move to the crack
403 area gradually and the center of middle area was the hardest area of arrive. Meanwhile it is
404 observed that the relative concentration of bitumen on both sides of the healing systems
405 decreased. In group-a, the relative concentration of bitumen in the middle would slowly
406 increase approaching to that of both sides, while the relative concentration of bitumen in the
407 middle of group-b was still less than that of both sides obviously. At 1000 ps of simulated time,

408 the relative concentration of bitumen in group-b still shows bimodal shape while that in group-
 409 a have changed to the plate shape. Moreover, it can be also found that the bitumen molecules
 410 contracted to the middle, and the overall volume decreased. Figure 9 (j) shows the average
 411 value of relative concentration of bitumen molecules in the middle area. The relative
 412 concentration of bitumen in bitumen healing systems was the largest, and the order of relative
 413 concentration of bitumen that $\text{FeO} > \text{Fe}_3\text{O}_4 > \text{Fe}_2\text{O}_3 > \text{FeOOH}$ were found in group-a and $\text{FeO} >$
 414 $\text{Fe}_3\text{O}_4 > \text{FeOOH} > \text{Fe}_2\text{O}_3$ were found in group-b.

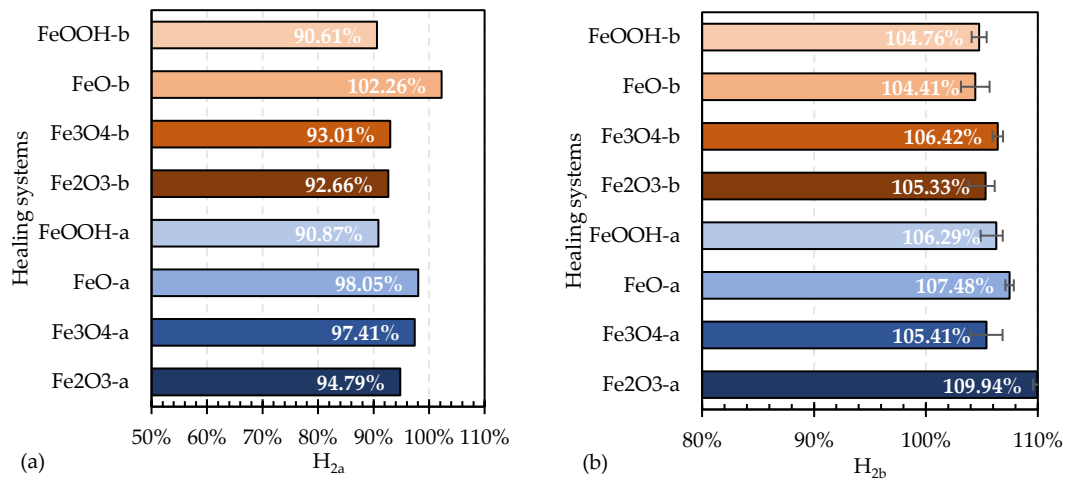
415





416 Figure 9 Relative concentration of bitumen in the healing systems at 373 K: (a) Fe₂O₃-a; (b)
 417 Fe₂O₃-b; (c) Fe₃O₄-a; (d) Fe₃O₄-b; (e) FeO-a; (f) FeO-b; (g) FeOOH-a; (h) FeOOH-b; (i) Bitumen; (j)
 418 Average value in the middle area
 419

420 Based on the results of Figure 9, the healing rate of bitumen aggregation in the healing
 421 systems was calculated according to Equation (5)-(9) with the value of bitumen healing systems
 422 as the reference. It means that bitumen aggregation degree in bitumen healing systems was
 423 considered as 100%. It can be found in Figure 10 (a) that relative concentration of bitumen in
 424 the healing systems was ordered as FeO > Fe₃O₄ > Fe₂O₃ > FeOOH, which indicates that the
 425 bitumen in the middle area of FeO healing systems possessed the most serve aggregation
 426 degree. The H_{2a} value in FeO-b healing systems has exceeded 100% demonstrating that the
 427 addition of sufficient volume of FeO resulted in closer combination of bitumen molecules in
 428 the corrosion area. The result also indicated that the few movements space in the corrosion area
 429 of group-b wouldn't reduce the bitumen aggregation degree. Conversely, an influx of more
 430 small molecules might intensify the aggregation degree. Figure 10 (b) shows the healing rate
 431 bitumen aggregation of free area in the healing systems. It can be observed that the values all
 432 have exceeded 100%, which was caused by the insufficient aggregation in corrosion area and
 433 the occupation of iron corrosion products.



435 Figure 10 Healing rate of bitumen aggregation in the healing systems: (a) Corrosion area; (b)
 436 Free area
 437

438 To combine the healing rate of crack and healing rate of bitumen aggregation, a
 439 comprehensive healing rate (H_3) of bitumen in the healing systems were held, which was
 440 defined as the product of healing rate of crack (H_1) and healing rate of bitumen aggregation
 441 (H_{2a}). The index combines the evaluation of crack closure and intrinsic diffusion of the bitumen
 442 in the healing systems. As shown in Figure 11, H_3 of the healing systems demonstrated the
 443 regularity that FeO > Fe₃O₄ > Fe₂O₃ > FeOOH in both group-a and group-b. The results indicate
 444 that the bitumen in the FeO healing systems could be healed best while that of FeOOH healing
 445 systems was the worst. The existence of FeO can make the bitumen molecules closer to each
 446 other and attract more bitumen molecules around the FeO crystals to realize better healing
 447 effect. It can be also found that the healing rate of group-a was obviously larger than that of
 448 group-b, which was mainly caused by the difference of iron corrosion products volume.

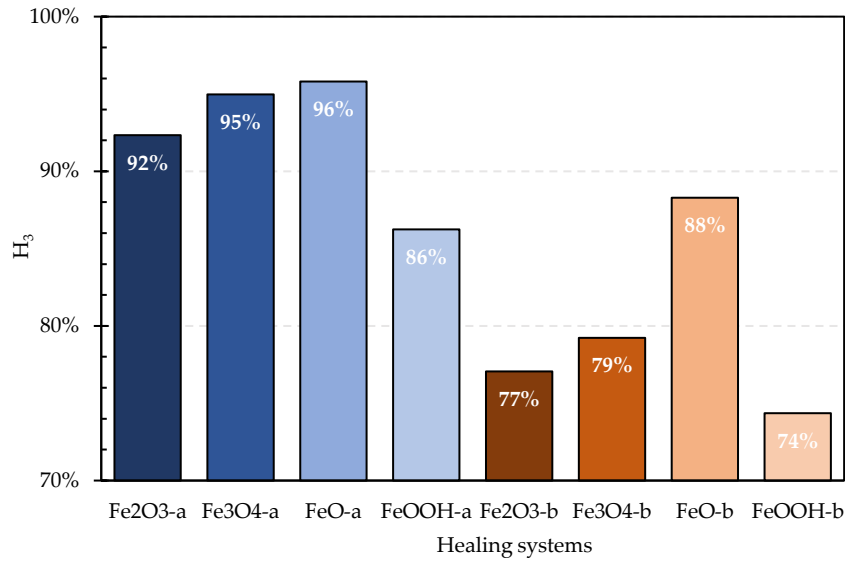


Figure 11 Comprehensive healing rate of bitumen in the healing systems

4.3 Diffusion behavior of bitumen in the healing systems

Figure 12 shows the MSD curves of bitumen in the different healing systems. It is clearly that there was a rapidly increasing period of MSD curves around 0-40 ps. In this period, the bitumen in the bitumen healing systems increased the most rapidly; then, the movements in the healing systems were gradually reduced around 40-150 ps when the reduction of bitumen healing systems was obviously less than that of other systems; another reduction period around 150-400 ps of the movements were followed with the less slope; finally the MSD would keep increment dynamically in the period of 400-1000 ps. Before 400 ps of simulated time, the MSD value followed the order that Fe₃O₄-a > Fe₂O₃-a > Fe₃O₄-a > bitumen > FeOOH-a > FeO-a > FeOOH-b > Fe₃O₄-b > Fe₂O₃-b > FeO-b. The above phenomena indicates that group-a possessed larger diffusion range than group-b and diffusion range of bitumen in the FeO healing systems was the narrowest in each group. It might be controlled by the combination of the attraction and appearance of iron corrosion products. After 400 ps of simulated time, diffusion range of bitumen in the FeOOH-a healing systems had a sudden increment and become the largest range.

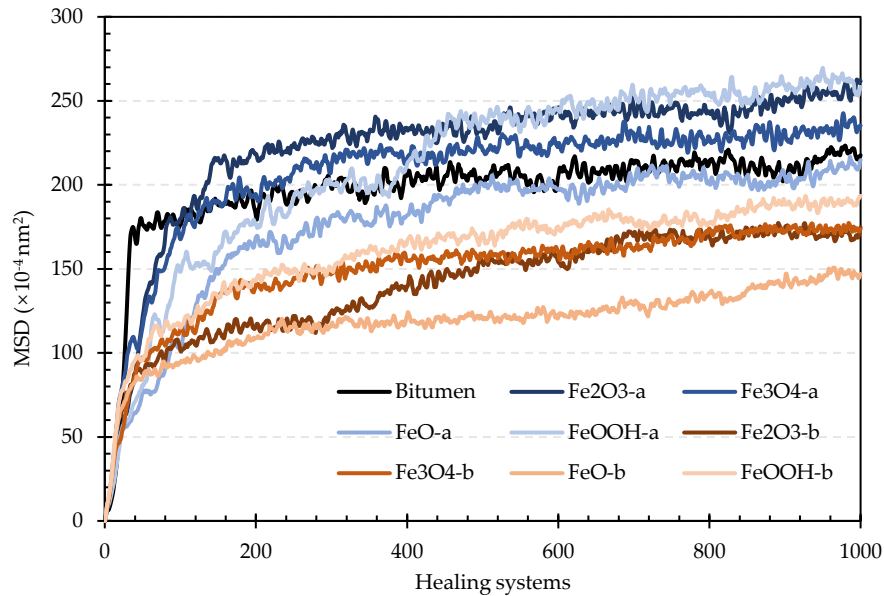


Figure 12 MSD curves of bitumen in the healing systems

467

468

469

470 According to the slope changing of MSD curves, the curves can be divided into four parts:

471 0-40 ps, 40-150 ps, 150-400 ps and 400-1000 ps. Commonly the diffusion coefficient could be

472 only used for the equilibrium statement to obtain the diffusion characteristic of the material. In

473 this section, diffusion coefficient was also used in the increment period to obtain the diffusion

474 rate approximately. Table 2 shows the diffusion coefficients of bitumen in the healing systems

475 at different time periods. With the extension of time, the diffusion coefficients of bitumen in

476 the healing systems would decrease rapidly at first and then slowly. It can be found that the

477 movement rate in bitumen healing systems was the fast in the 0-40 ps. There were no obvious

478 differences between the diffusion coefficients of group-a and group-b except 40-150 ps. In this

479 period, the diffusion coefficients of group-a were obvious larger, indicating faster movement

480 rate of bitumen molecules. Meanwhile, the movement of bitumen mainly focused on the

481 diffusion on the surface of iron corrosion products in this period. This period should be

482 considered the important period that could affect the healing degree.

483 The diffusion process was controlled by the interaction force inside the bitumen of the

484 healing systems, which was changing with the acceleration linearly in the molecular dynamics

485 based on Newton's second law. It can be defined as the second derivative of $MSD^{0.5}$ to the

486 simulated time (T) accorded with dimension. It was also found that $\lg \frac{d\sqrt{MSD}}{dT}$ and $\lg T$ shows

487 strong liner relationship, therefore the slope value of the relationship can represent the internal

488 force inside the bitumen in the healing systems. As shown in the Table 3, the value of R² have
 489 proved the linear relationship. It can be observed that the internal force that motivated bitumen
 490 molecules in the bitumen healing systems was the largest, and the internal force of group-a
 491 were larger than that of group-b except FeOOH healing systems. The results also indicate that
 492 Fe₃O₄ healing systems possess the largest internal force inside the bitumen in the iron corrosion
 493 products systems.

494

495 Table 2 Diffusion coefficients ($\times 10^{-4}$ nm²/ps) of bitumen in the healing systems

Healing systems	Time period			
	0-40 ps	40-150 ps	150-400 ps	400-1000 ps
Bitumen	0.863	0.022	0.009	0.003
Fe ₂ O ₃ -a	0.372	0.154	0.015	0.006
Fe ₃ O ₄ -a	0.517	0.133	0.021	0.004
FeO-a	0.268	0.133	0.019	0.005
FeOOH-a	0.315	0.131	0.025	0.010
Fe ₂ O ₃ -b	0.368	0.036	0.017	0.009
Fe ₃ O ₄ -b	0.358	0.055	0.014	0.006
FeO-b	0.362	0.024	0.010	0.008
FeOOH-b	0.419	0.053	0.018	0.007

496

497 Table 3 Evaluation of internal force motivating bitumen molecules in the healing systems

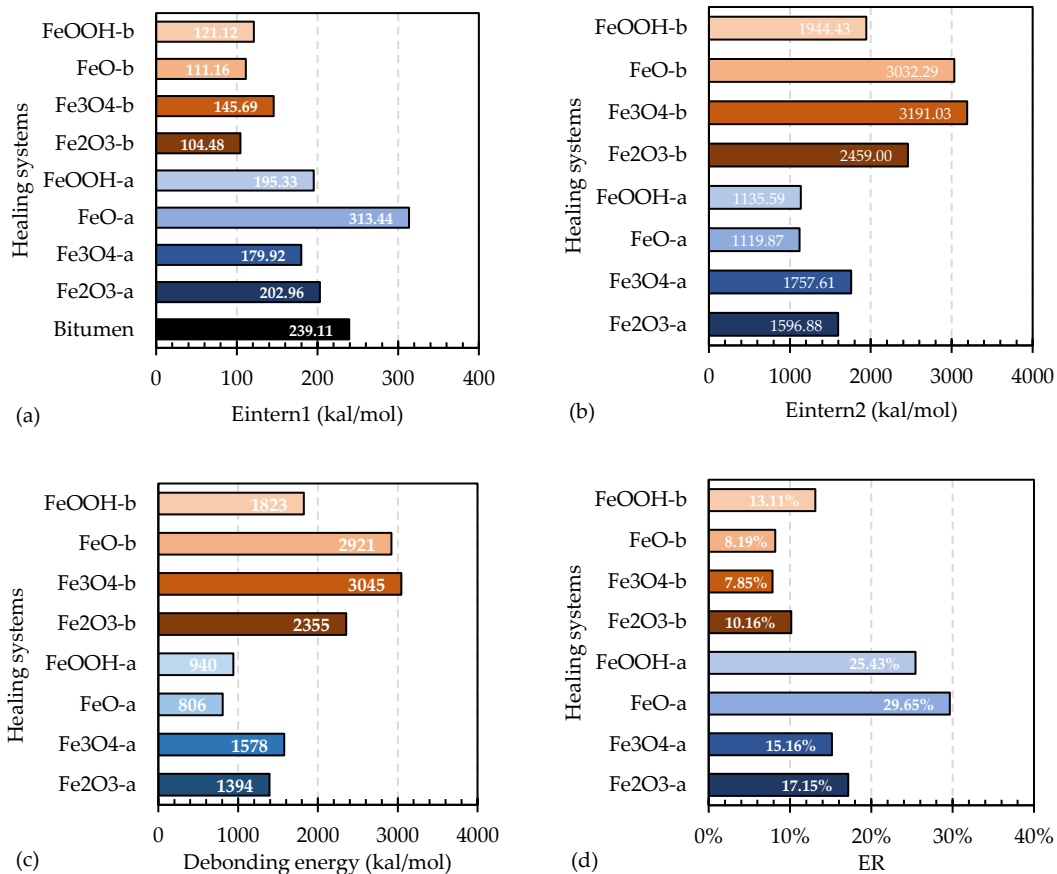
Healing systems	Relationships	R ²	Slope value
	($\lg \frac{d\sqrt{MSD}}{dT} - \lg T$)		
Bitumen	y = -2.0259x + 2.5933	R ² = 0.9099	2.03
Fe ₂ O ₃ -a	y = -1.6323x + 1.9762	R ² = 0.9512	1.63
Fe ₃ O ₄ -a	y = -1.8288x + 2.4216	R ² = 0.9897	1.83
FeO-a	y = -1.5388x + 1.7244	R ² = 0.9568	1.54
FeOOH-a	y = -1.3347x + 1.3952	R ² = 0.975	1.34
Fe ₂ O ₃ -b	y = -1.3447x + 1.2406	R ² = 0.9198	1.35
Fe ₃ O ₄ -b	y = -1.53x + 1.6271	R ² = 0.9817	1.53
FeO-b	y = -1.3985x + 1.2431	R ² = 0.8522	1.40
FeOOH-b	y = -1.5024x + 1.6233	R ² = 0.9759	1.50

498

499 *4.4 Interaction effect in the healing systems*

500 Figure 13 demonstrates interaction evaluation in the healing systems. The interaction
501 energy of bitumen and bitumen, and Interaction energy of bitumen and iron corrosion products
502 were shown in Figure 13 (a) and (b) respectively. The interaction energy of bitumen and
503 bitumen in group-a was larger than that in group-b due to the distance difference. There was
504 insignificant regulation of the interaction energy of bitumen and bitumen between different
505 iron corrosion products. It was caused by that with the expansion of iron corrosion products,
506 the bitumen molecules that crossed the corrosion area were different, resulting in the
507 insignificant regularity. It can be observed that the interaction energy of bitumen and iron
508 corrosion products in group-b was larger than that in group-a caused by the larger volume and
509 contact surface. The interaction in the Fe_3O_4 healing systems was the most severe. The results
510 also indicate that the force by iron corrosion products should be regarded as the main power
511 that motivate the movement of bitumen molecules. It can be also found that the interaction
512 between bitumen and iron corrosion products was obviously larger than the interaction energy
513 of bitumen and bitumen. This difference would cause stress concentration easily in the
514 aggregation area on the iron corrosion product surface and generate weak areas. It indicates
515 that the main failure form should be cohesive failure once the cracks occurred again.
516 Meanwhile, the tearing on the sides of iron corrosion products between bitumen molecules was
517 accompanied. However, the accumulation of bitumen in the free area led to excessive internal
518 stress, which would make failure easier and weaken the healing effect. This can also be proved
519 by Figure 13 (c) and (d) where the debonding energy in the healing systems was quite larger
520 than the interaction energy between bitumen and bitumen.

521



522 Figure 13 Interaction evaluation in the healing systems: (a) Interaction energy of bitumen and
 523 bitumen; (b) Interaction energy of bitumen and iron corrosion products; (c) Debonding
 524 energy; (d) ER
 525

526 5 Conclusions

527 The investigation has been carried out to identify the influence of iron corrosion products
 528 on the healing process of bitumen by molecular dynamics simulation approach. Based on the
 529 results, the primary conclusions are as follows:

- 530 (1) The iron corrosion products have a dual effect on bitumen molecules: the attraction of iron
 531 corrosion products would make bitumen molecules climb to its surface to diffuse, while
 532 the products with larger volume would also reduce space of bitumen molecules movement
 533 and then obstruct the diffusion. The spatial structure of actual corrosion products is more
 534 complicated, and this dual effect may be affected.
- 535 (2) The iron corrosion products would result in the reduction of healing degree. Compared to
 536 other iron corrosion products, FeO healing systems had the best healing degree from in
 537 terms of crack closure and bitumen aggregation degree in the corrosion area

538 comprehensively, followed by Fe₃O₄, Fe₂O₃ and FeOOH.

539 (3) Diffusion coefficients of the period that bitumen molecules diffused on surface of iron
540 corrosion products were significantly different, which would make it vital in the healing
541 process. It was also related to the internal force and there is the largest internal force to
542 motivate bitumen molecules moving in the Fe₃O₄ healing systems.

543 (4) The interaction effect between bitumen and iron corrosion products is obviously larger
544 than the interaction effect between bitumen and bitumen. The difference would make more
545 bitumen molecules absorbed on the surface of the iron corrosion products where the
546 bitumen film would be thicker. It would result in cohesive failure becoming the main
547 failure form once the cracks re-occur.

548

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550 Original draft, Writing - Review & Editing. Yingxue Zou: Methodology, Investigation, Writing
551 - Review & Editing. Gordon Airey: Investigation, Writing - Review & Editing. Haopeng Wang:
552 Methodology, Investigation, Data curation. Hanyu Zhang: Methodology, Investigation, Data
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554 Conceptualization, Methodology, Project administration.

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