

## Supporting Information

# **A novel "holey-LFP / graphene / holey-LFP" sandwich nanostructure with significantly improved rate capability for lithium storage**

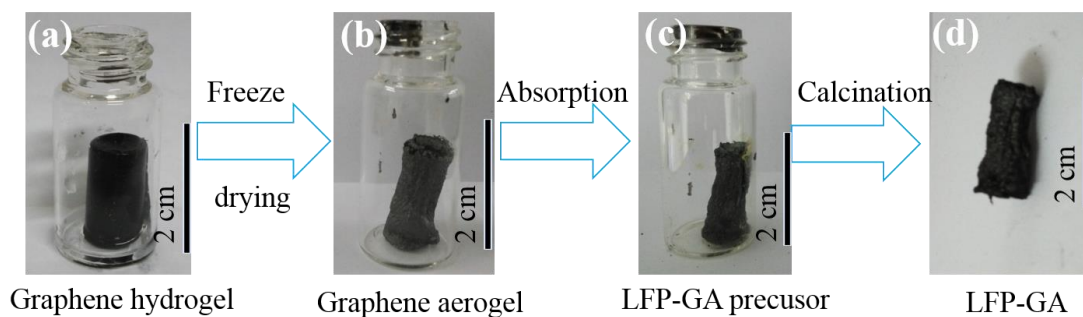
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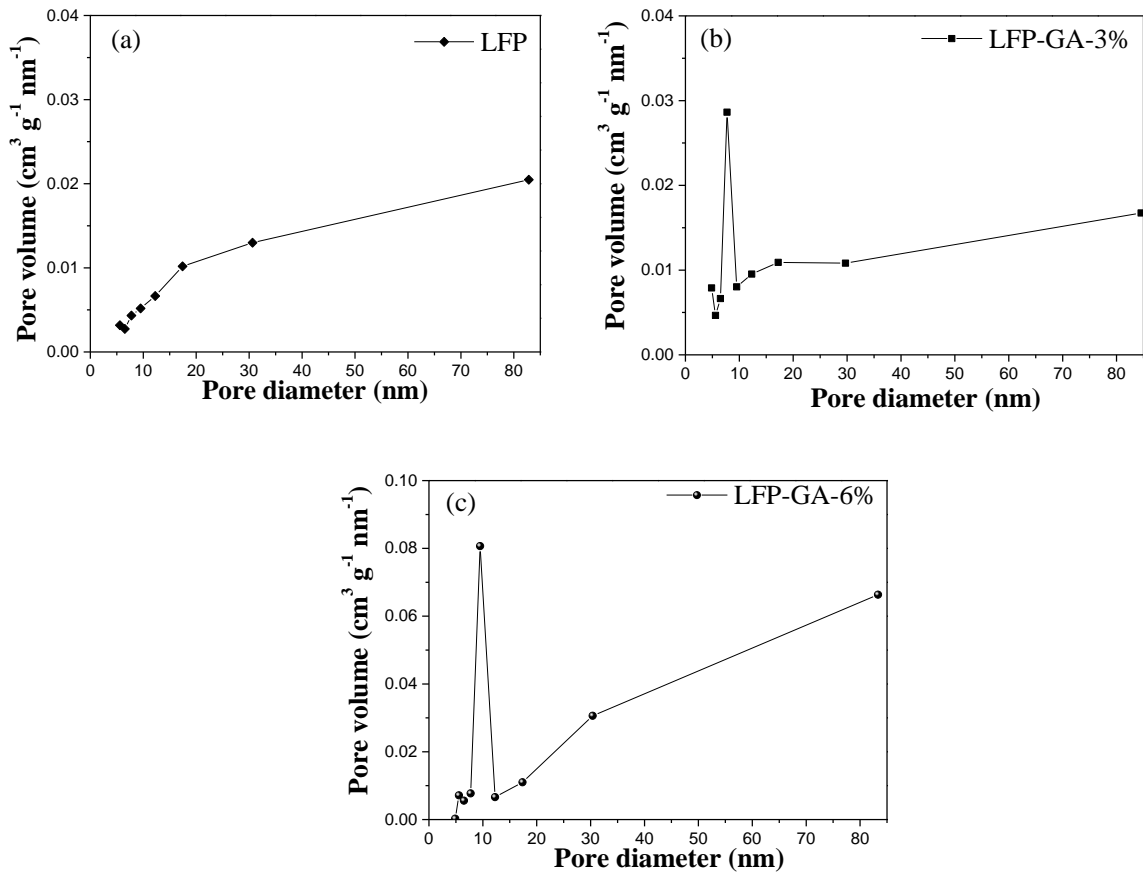
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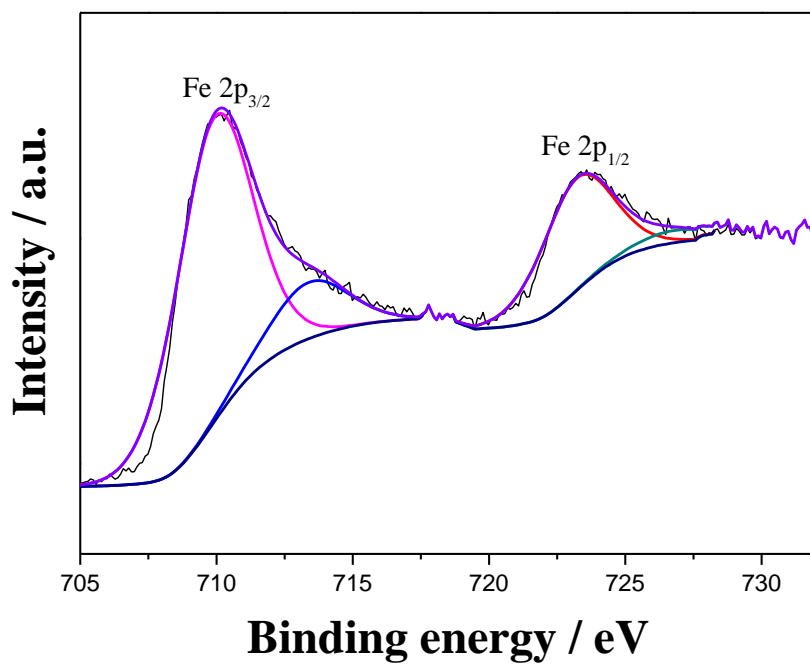
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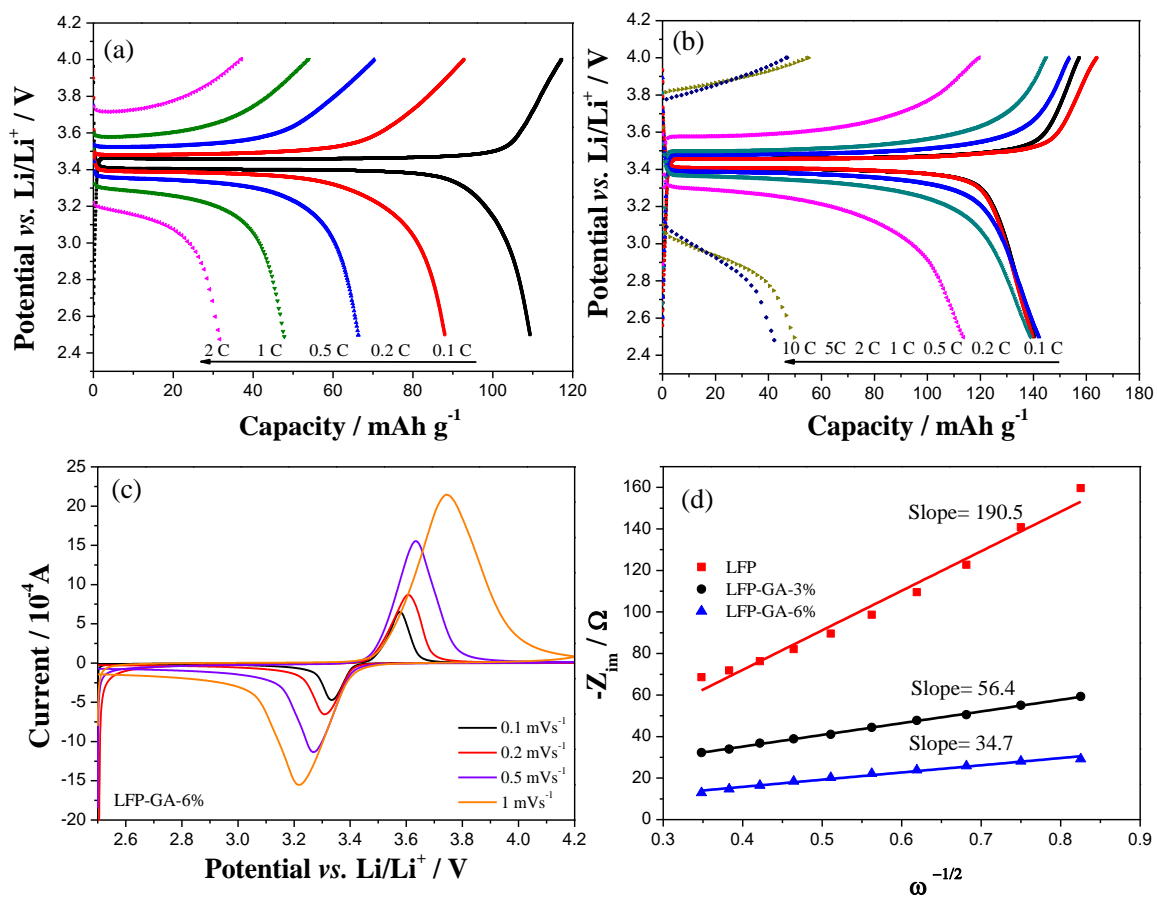
**Figure S1.** The digital photograph illustration of the synthesis of LFP-GA: (a) Graphene hydrogel synthesized by a hydrothermal process; (b) graphene aerogel obtained after the freeze drying; (c) LFP colloidal was adsorbed on graphene aerogel and dried to form the LFP-GA precursor; (d) The LFP-GA composite materials were obtained by calcination.



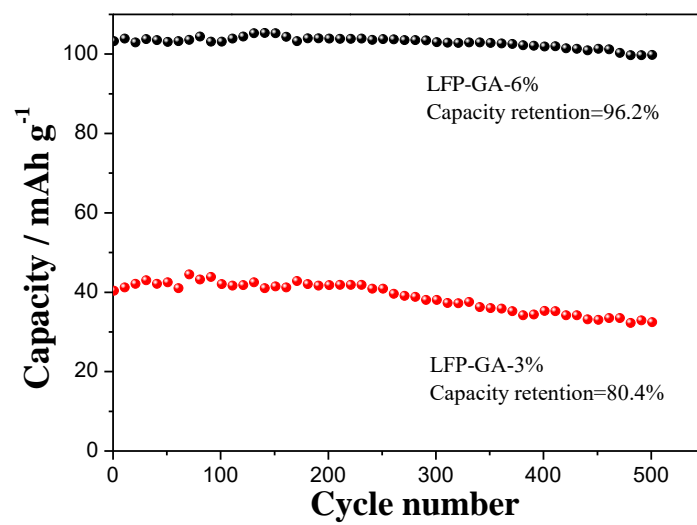
**Figure S2.** Pore-size distributions of (a) LFP, (b) LFP-GA-3% and (c) LFP-GA-6%.



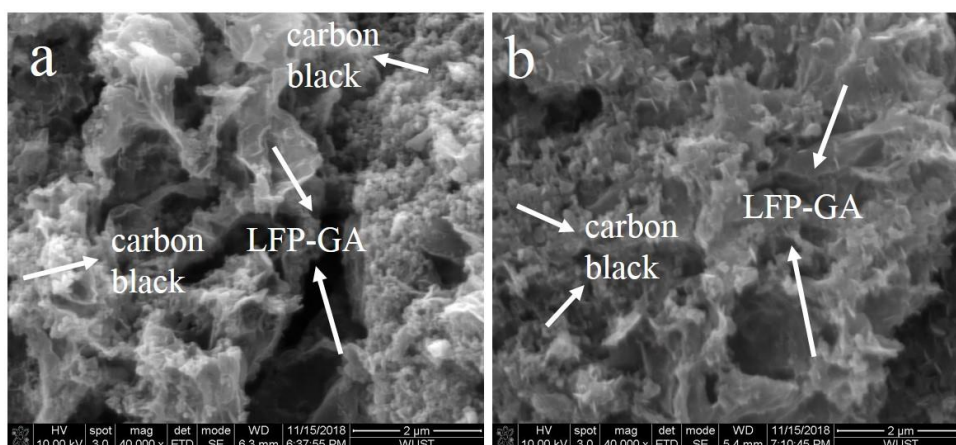
**Figure S3.** XPS core level spectrum of Fe 2p of LFP.



**Figure S4.** The discharge curves of (a) LFP and (b) LFP-GA-3% at different discharge rates, (c) CV curves of LFP-GA-6% at different scan rates, and (d) the fitting curves of  $-Z_{\text{im}}$  and the reciprocal square root of the angular frequency at low frequency region of LFP, LFP-GA-3% and LFP-GA-6%.



**Figure S5.** The cyclic performance of LFP-GA-3% and LFP-GA-6% during 500 cycles at 10 C.



**Figure S6.** The SEM images of LFP-GA-6% electrode (a) before and (b) after cycling.

**Table S1.** Summary of the EIS fitted parameters, Li<sup>+</sup> diffusion coefficient and Warburg factor of the samples.

Samples	Fitted parameters			
	R <sub>s</sub> / (Ω)	R <sub>ct</sub> / (Ω)	σ / (Ω s <sup>-1/2</sup> )	D / (cm <sup>2</sup> s <sup>-1</sup> )
LFP	11	292	190.5	6.76×10 <sup>-15</sup>
LFP-GA-3%	8.6	245.1	56.4	6.55×10 <sup>-14</sup>
LFP-GA-6%	7.5	87.5	34.7	2.04×10 <sup>-13</sup>