

Cyclization

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Enantioselective Synthesis of Chiral Cyclopent-2-enones by Nickel-Catalyzed Desymmetrization of Malonate Esters

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Abstract: The enantioselective synthesis of highly functionalized chiral cyclopent-2-enones by the reaction of alkynyl malonate esters with arylboronic acids is described. These desymmetrizing arylytic cyclizations are catalyzed by a chiral phosphinooxazoline/nickel complex, and cyclization is enabled by the reversible *E/Z* isomerization of alkenylnickel species. The general methodology is also applicable to the synthesis of 1,6-dihydropyridin-3(2*H*)-ones.

Chiral cyclopent-2-enones are versatile building blocks for synthesis^[1] and are present in many biologically active natural products^[1] such as (+)-achalensolide,^[2] phorbol,^[3] and (–)-kjellmanianone^[4] (Figure 1). In view of their broad signifi-

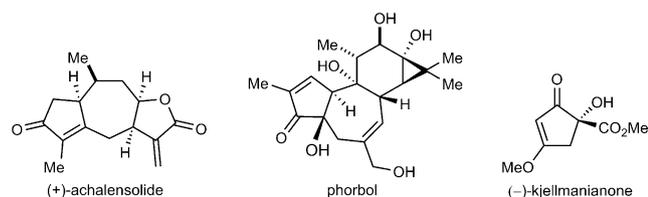
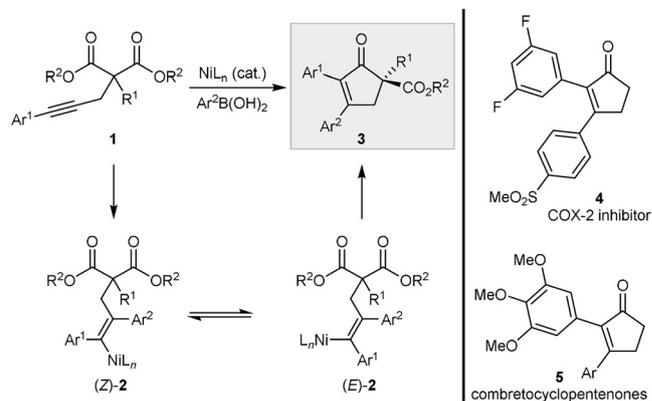


Figure 1. Natural products containing cyclopent-2-enones.

cance, various methods have been developed for the de novo construction of enantiomerically enriched chiral cyclopent-2-enones,^[1,5–7] such as Pauson–Khand reactions,^[5] Nazarov cyclizations,^[6] and several other approaches.^[7] However, given the wide structural diversity of chiral cyclopent-2-enones in target compounds, the development of new strategies to these structures continues to be highly valuable.

We envisaged that chiral cyclopent-2-enones might be prepared by the enantioselective nickel-catalyzed reaction of



Scheme 1. Proposed synthesis of chiral cyclopent-2-enones.

alkynyl malonate esters (**1**) with arylboronic acids (Scheme 1). Specifically, nickel-catalyzed *syn* addition of an arylboronic acid to the alkyne of **1** would give the alkenylnickel species (**Z-2**), which possesses the incorrect stereochemistry for cyclization onto one of the esters. However, reversible *E/Z* isomerization^[8,9] of (**Z-2**) would give the alkenylnickel species (**E-2**), which could now attack an ester in an enantioselective desymmetrization^[10] to give 2,3-diaryl cyclopent-2-enones (**3**).^[11] The 2,3-diaryl cyclopent-2-enone scaffold is present in the highly potent COX-2 inhibitor **4**,^[12a] as well as in the combretocyclopentenones **5**^[12b] and related compounds,^[12c] which exhibit antitumor activity. Moreover, there are few asymmetric methods for the de novo construction of cyclopent-2-enones with a quaternary stereocenter at the 5-position (as in **3**).^[5k,6g] Although our previous work on enantioselective nickel-catalyzed arylytic cyclizations of alkynyl electrophiles showed that ketones^[8a] and activated alkenes^[8a,b] are competent reaction partners for alkenylnickel species, the ability of less electrophilic esters to undergo analogous cyclizations was less certain. Herein, we report the successful implementation of this strategy. Not only can this methodology produce highly functionalized, enantiomerically enriched chiral cyclopent-2-enones, but 1,6-dihydropyridin-3(2*H*)-ones are also accessible.

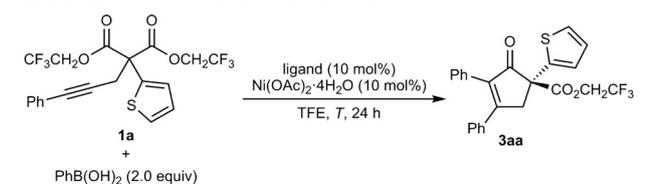
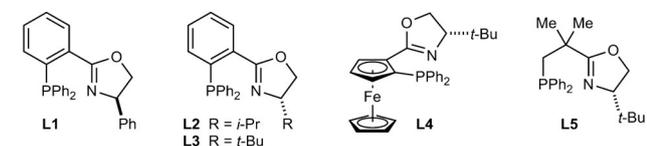
Our initial experiments revealed that the substrates **1**, containing ethyl esters, are insufficiently reactive under a range of reaction conditions that are effective in our nickel-catalyzed *anti*-carbometallic cyclizations described previously.^[8] However, the more electrophilic bis(2,2,2-trifluoroethyl) malonate **1a** reacted successfully with PhB(OH)₂ (2.0 equiv) in the presence of 10 mol % each of Ni(OAc)₂·4H₂O and various chiral P,N-ligands (**L1–L5**) in 2,2,2-trifluoroethanol (TFE) to give the cyclopent-2-enone

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Table 1: Evaluation of reaction conditions.^[a]**Ligands**

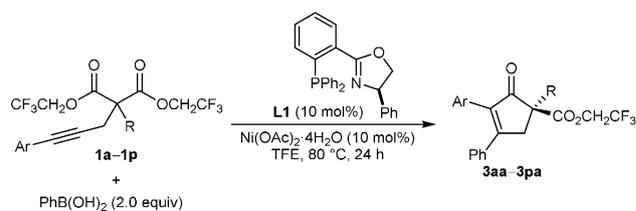
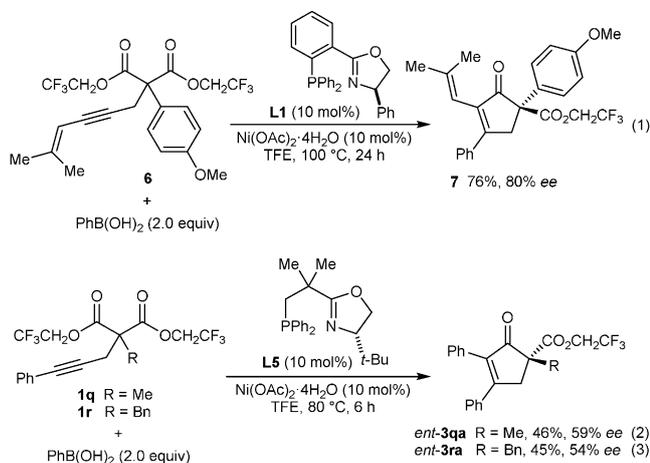
Entry	Ligand	T [°C]	Yield [%] ^[b]	ee [%] ^[c]
1	L1	100	98	91
2	L1	80	99	94
3	L2	80	94	−81 ^[d]
4	L3	80	84	−94 ^[d]
5	L4	80	49	88
6	L5	80	61	−78 ^[d]

[a] Reactions were conducted with 0.05 mmol of **1a** in TFE (0.5 mL).

[b] Determined by ¹H NMR analysis using 1,4-dimethoxybenzene as an internal standard. [c] Determined by HPLC analysis on a chiral stationary phase. [d] These reactions gave *ent*-**3aa** as the major enantiomer.

3aa (Table 1). At 100 °C, (*R*)-Ph-PhOX (**L1**) gave **3aa** in 98 % yield (by ¹H NMR analysis) and 91 % *ee* (entry 1).^[13] Reducing the temperature to 80 °C improved the enantioselectivity to 94 % *ee* with no loss of yield (entry 2). Other phosphino-oxazolines, **L2**–**L5**, are effective at 80 °C (entries 3–5), but with the exception of **L3** (entry 4), the yields and enantioselectivities are appreciably lower than with **L1**.

The scope of this process with respect to the alkynyl bis(2,2,2-trifluoroethyl) malonate was then explored using **L1** as the chiral ligand in reactions with PhB(OH)₂, which gave cyclopent-2-enones (**3aa**–**3pa**) in 46–98 % yield and 77–94 % *ee* (Scheme 2). As well as a 2-thienyl group (**3aa**, **3ia**, and **3ja**), the substituent at the 2-position of **1** can be changed to a phenyl group (**3ba**), mono- and disubstituted benzenes with electron-donating or electron-withdrawing substituents (**3ca**–**3ga**, **3ka**, and **3la**), and a 2-naphthyl group (**3ha**). Ethoxy (**3ma**), benzyloxy (**3na**), 3-thienylmethoxy (**3oa**), and anilino groups (**3pa**) at this position are also tolerated. The reaction is



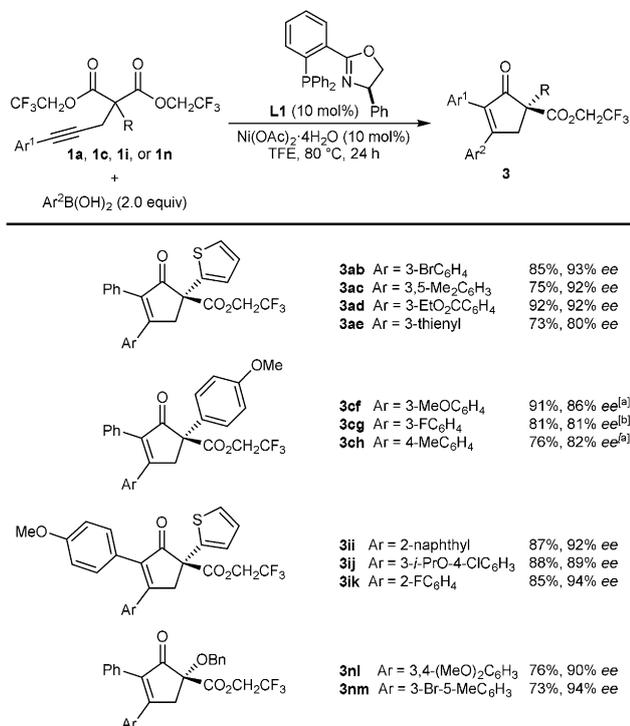
	3aa Ar = 2-thienyl	82%, 94% <i>ee</i>
	3ba Ar = Ph	97%, 82% <i>ee</i>
	3ca Ar = 4-MeOC ₆ H ₄	98%, 88% <i>ee</i> ^[a]
	3da Ar = 3-MeC ₆ H ₄	96%, 87% <i>ee</i>
	3ea Ar = 3,4-Me ₂ C ₆ H ₃	75%, 92% <i>ee</i>
	3fa Ar = 3,5-Me ₂ C ₆ H ₃	83%, 92% <i>ee</i>
	3ga Ar = 3-O ₂ NC ₆ H ₄	96%, 77% <i>ee</i>
	3ha Ar = 2-naphthyl	87%, 94% <i>ee</i>
	3ia Ar = 4-MeOC ₆ H ₄	96%, 94% <i>ee</i>
	3ja Ar = 3-MeC ₆ H ₄	58%, 91% <i>ee</i>
	3ka Ar = 4-ClC ₆ H ₄	87%, 87% <i>ee</i> ^[a]
	3la Ar = 2-thienyl	84%, 86% <i>ee</i> ^[a]
	3ma R = Et	46%, 92% <i>ee</i> ^[b]
	3na R = Bn	78%, 93% <i>ee</i>
	3oa	78%, 93% <i>ee</i> ^[b]
	3pa	98%, 90% <i>ee</i>

Scheme 2. Scope with respect to the alkynyl bis(2,2,2-trifluoroethyl) malonate. Reactions were conducted with 0.30 mmol of **1a–p** in TFE (3 mL). Yields are those of the isolated products. Enantiomeric excesses were determined by HPLC analysis on a chiral stationary phase. [a] Conducted at 100 °C. [b] Conducted with 20 mol% each of Ni(OAc)₂·4H₂O and **L1**.

compatible with various other (hetero)aryl groups at the alkyne, such as 4-methoxyphenyl (**3ia**), 3-methylphenyl (**3ja**), 4-chlorophenyl (**3ka**), and 2-thienyl (**3la**). In a few cases, reaction at 100 °C (**3ca**, **3ka**, and **3la**) or use of a 20 mol% catalyst loading (**3ma** and **3oa**) were required for complete consumption of the starting material.

The process is not limited to aryl groups at the alkyne, as shown by the reaction of the 1,3-enyne **6** to give the cyclopent-2-enone **7** in 76 % yield and 80 % *ee* [Eq. (1)]. (*R*)-Ph-PhOX (**L1**) is less effective for substrates with alkyl groups at the 2-position. For example, the cyclization of **1q** and **1r** (see [Eqs. (2) and (3) for the structures] gave cyclopent-2-enones in 29 and 0 % *ee*, respectively, with **L1** as the ligand. However, somewhat improved results were obtained with (*S*)-*t*-Bu-NeOPHOX (**L5**),^[14] which gave *ent*-**3qa** and *ent*-**3ra** in 59 and 54 % *ee*, respectively [Eqs. (2) and (3)].

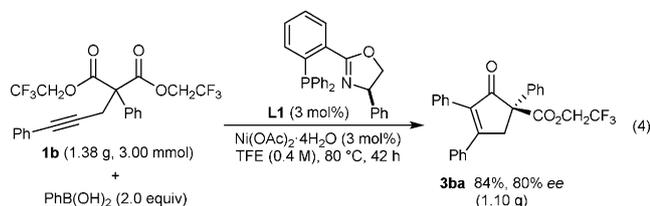
The reactions of a range of (hetero)arylboronic acids with representative substrates, **1a**, **1c**, **1i**, and **1n**, are presented in Scheme 3. Pleasingly, these reactions gave cyclopent-2-enones in generally good yields (73–92%) and enantioselectivities (80–94% *ee*). The process is compatible with arylboronic acids containing halide (**3ab**, **3cg**, **3ij**, **3ik**, and **3nm**), methyl (**3ac**, **3ch** and **3nn**), carboethoxy (**3ad**), or alkoxy



Scheme 3. Scope with respect to the boronic acid. Reactions were conducted with 0.30 mmol of **1a**, **1c**, **1i**, or **1n** in TFE (3 mL). Yields are those of the isolated products. Enantiomeric excesses were determined by HPLC analysis on a chiral stationary phase. [a] Conducted at 100 °C. [b] Conducted at 120 °C.

(**3cf**, **3ij**, and **3nl**) substituents. 2-Naphthylboronic acid (**3ii**) and 3-thienylboronic acid (**3ae**) are also effective.

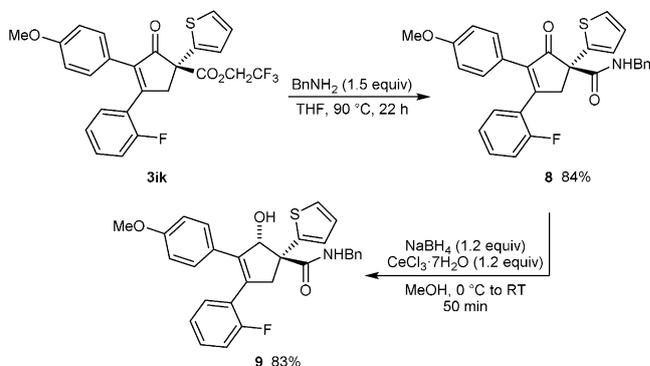
The process also works well for gram-scale reactions. For example, the reaction of **1b** (1.38 g, 3.00 mmol) with PhB(OH)₂ gave 1.10 grams of **3ba** (84% yield) in 80% *ee* [Eq. (4)]. Importantly, by conducting this reaction a higher



concentration of 0.4M, rather than at 0.1M used in the experiments shown in Scheme 2 and Scheme 3, the catalyst loading was lowered to 3 mol%.

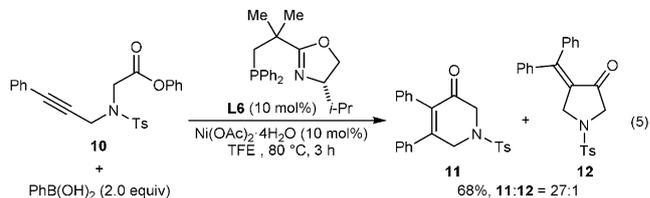
To demonstrate the synthetic utility of the products, further transformations of a representative cyclopent-2-enone

were conducted. Trifluoroethyl esters are moderately active acylating agents^[15] and could therefore serve as useful functional handles. Indeed, heating **3ik** with benzylamine (1.5 equiv) in THF at 90 °C smoothly gave the amide **8** in 84% yield without affecting the enone (Scheme 4). A Luche reduction of **8** then gave allylic alcohol **9** as a single observable diastereomer in 83% yield.



Scheme 4. Further transformations of the cyclopent-2-enone **3ik**.

Finally, although chiral cyclopent-2-enones were the primary targets of this study, the general methodology can be applied to the synthesis of other products. For example, reaction of the alkyne phenyl ester **10** with PhB(OH)₂ using (*S*)-*i*-Pr-NeoPHOX (**L6**)^[14] as the ligand gave a 27:1 mixture of the 1,6-dihydropyridin-3(2*H*)-one **11** together with a minor product (**12**) in 68% yield [Eq. (5)].^[16] Other P,N-ligands resulted in lower yields and less favorable ratios of **11**:**12**.



In conclusion, we have reported the enantioselective synthesis of chiral cyclopent-2-enones by the nickel-catalyzed desymmetrizing arylation/cyclization of alkyne bis(2,2,2-trifluoroethyl) malonates with arylboronic acids. The reactions proceed in good yields and generally high enantioselectivities to give cyclopent-2-enones containing a fully substituted alkene and a quaternary stereocenter at the 5-position. This work further demonstrates the utility of reversible *E/Z* isomerization of alkyne nickel species in promoting new domino addition/cyclizations of alkyne electrophiles, reactions that would otherwise be impossible.^[8,9d,e] Investigation of this reactivity in other contexts is ongoing and will be reported in due course.

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Conflict of interest

The authors declare no conflict of interest.

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- [17] The research data associated with this publication can be found at <https://doi.org/10.17639/nott.356>.

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